

Supporting Information for

Green synthesis of BiOBr modified $\text{Bi}_2\text{O}_2\text{CO}_3$ nanocomposites with
enhanced visible-responsive photocatalytic properties

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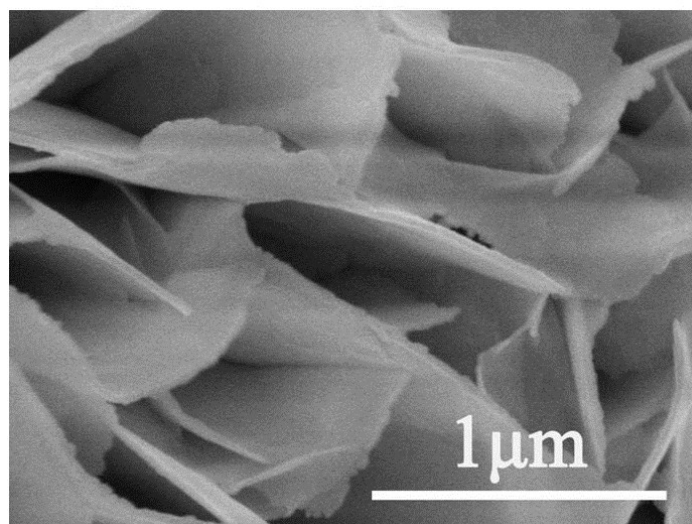


Fig. S1. SEM images of 60% $\text{Bi}_2\text{O}_2\text{CO}_3/\text{BiOBr}$ sample.

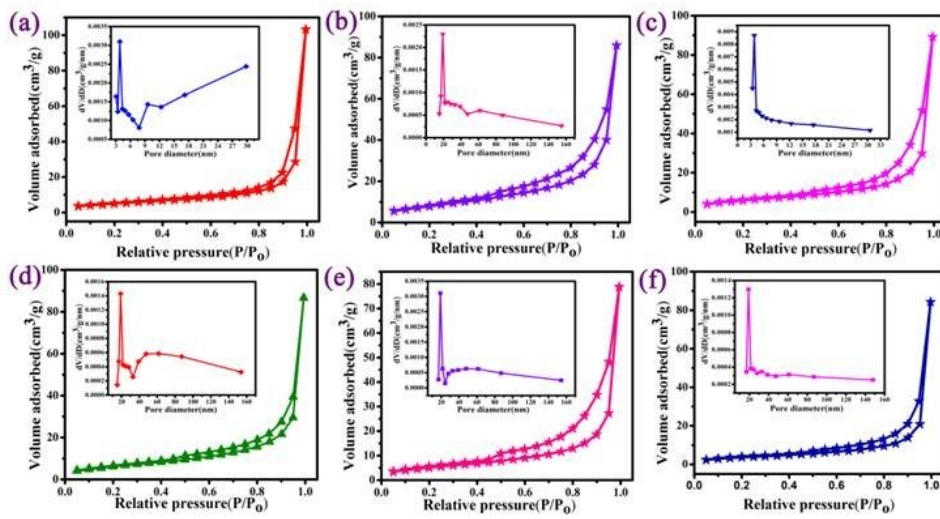


Fig. S2. Nitrogen adsorption-desorption isotherms and pore size distribution for: (a) $\text{Bi}_2\text{O}_2\text{CO}_3$, (b) 20% $\text{Bi}_2\text{O}_2\text{CO}_3/\text{BiOBr}$, (c) 40% $\text{Bi}_2\text{O}_2\text{CO}_3/\text{BiOBr}$, (d) 60% $\text{Bi}_2\text{O}_2\text{CO}_3/\text{BiOBr}$, (e) 80% $\text{Bi}_2\text{O}_2\text{CO}_3/\text{BiOBr}$ and (f) BiOBr .

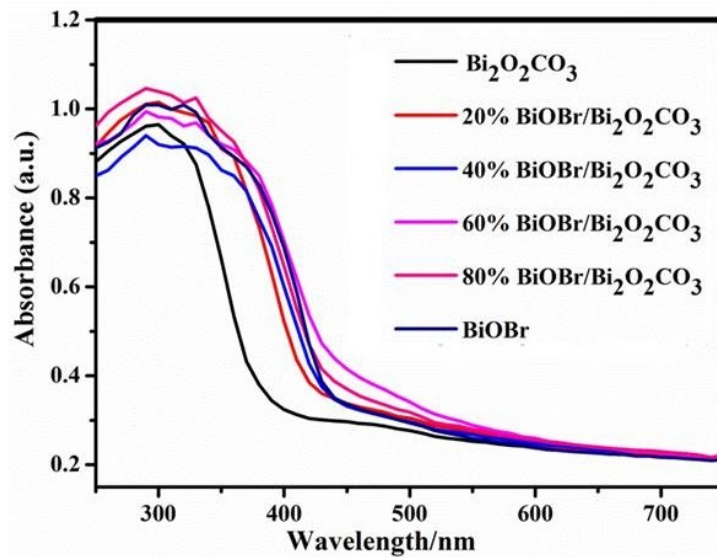


Fig. S3. UV-visible diffuse reflectance spectra of the as-prepared samples.

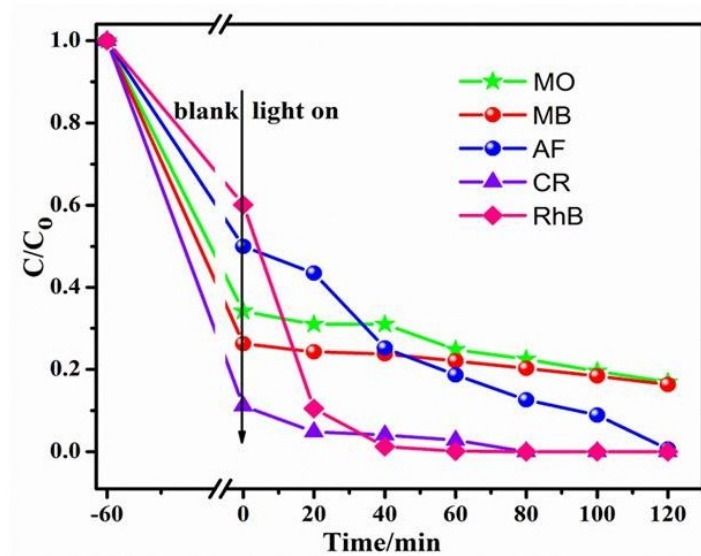


Fig. S4. Photocatalytic degradation curves of 60% Bi₂O₂CO₃/ BiOBr sample for methyl orange (MO, 10 mg/L), methyl blue (MB, 20 mg/L), acid fuchsin (AF, 20 mg/L), congo red (CR, 20 mg/L) and rhodamine B (RhB, 20 mg/L) under visible light irradiation.

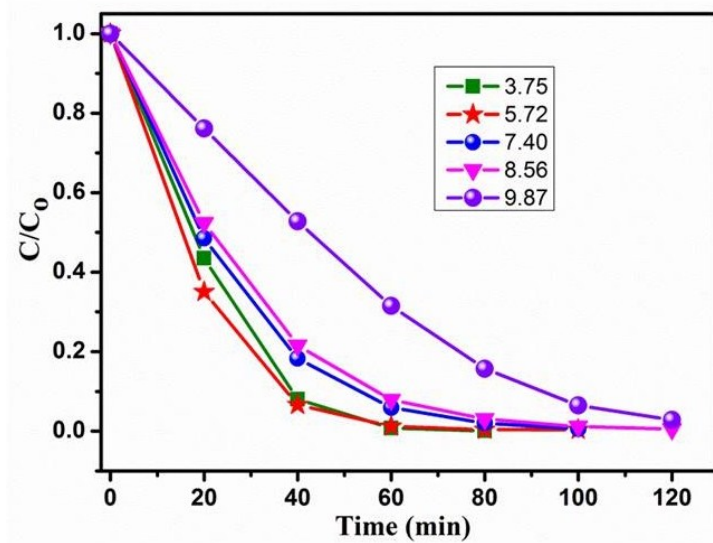


Fig. S5. Photocatalytic degradation curves of RhB solutions with different initial pH over 60% BiOBr/Bi₂O₂CO₃ sample.

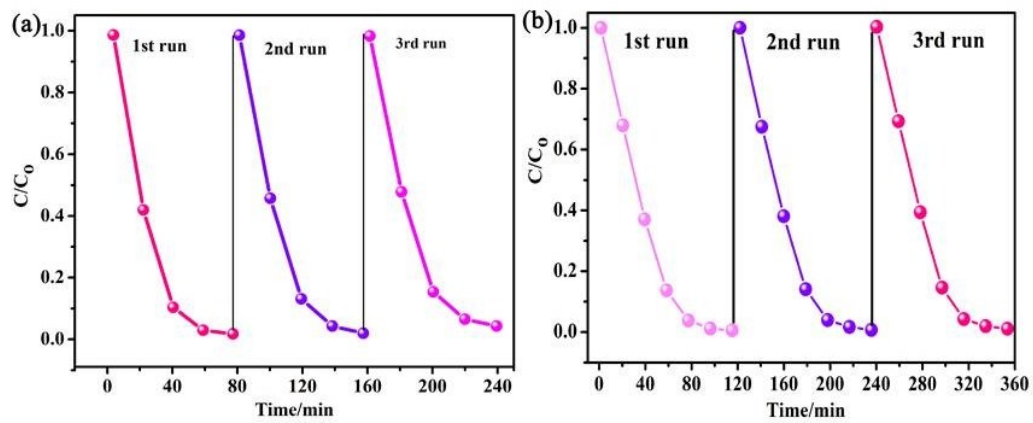


Fig. S6. Cycling runs of (a) 60% BiOBr/Bi₂O₂CO₃ and (b) single BiOBr samples under visible-light irradiation.