

## Supporting Information

### Simple synthesis of MoO<sub>2</sub>/Carbon aerogels anodes for high performance lithium ion batteries from seaweed biomass

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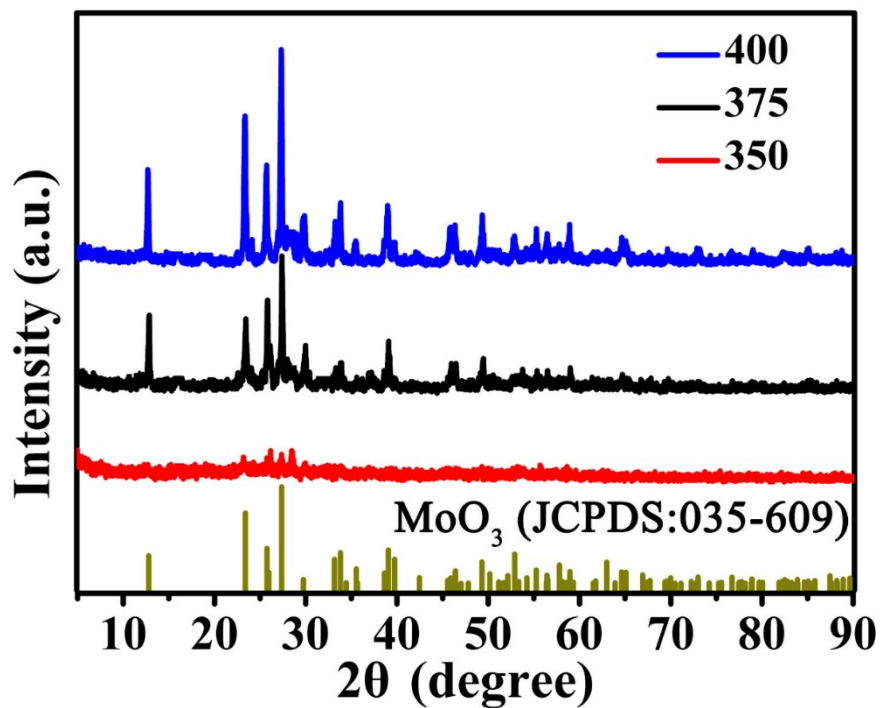


Fig. S1 XRD patterns of AMM aerogels stabilized at 350, 375 and 400 °C in air for 1 h.

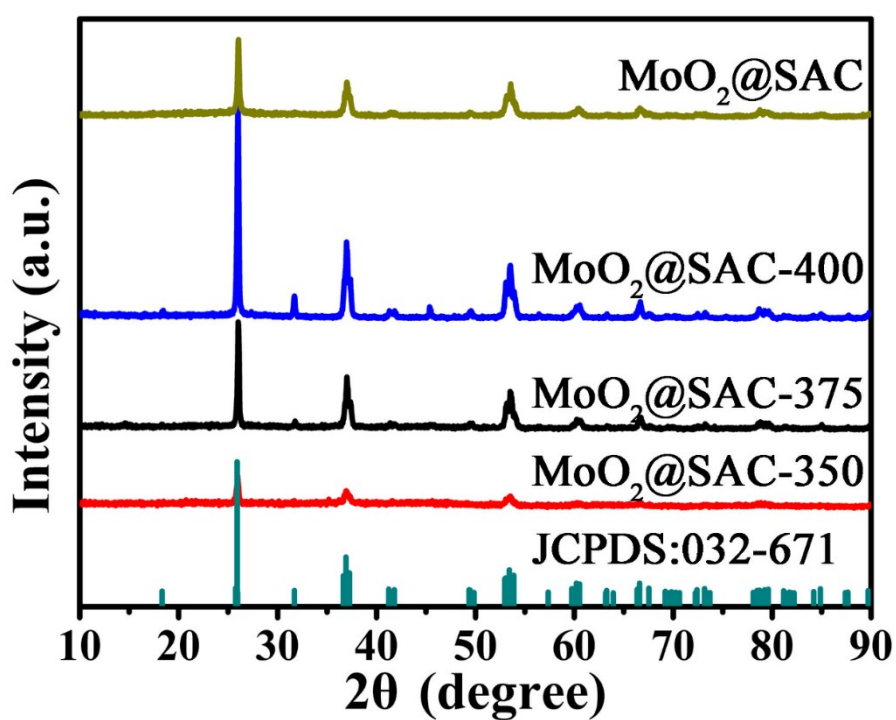
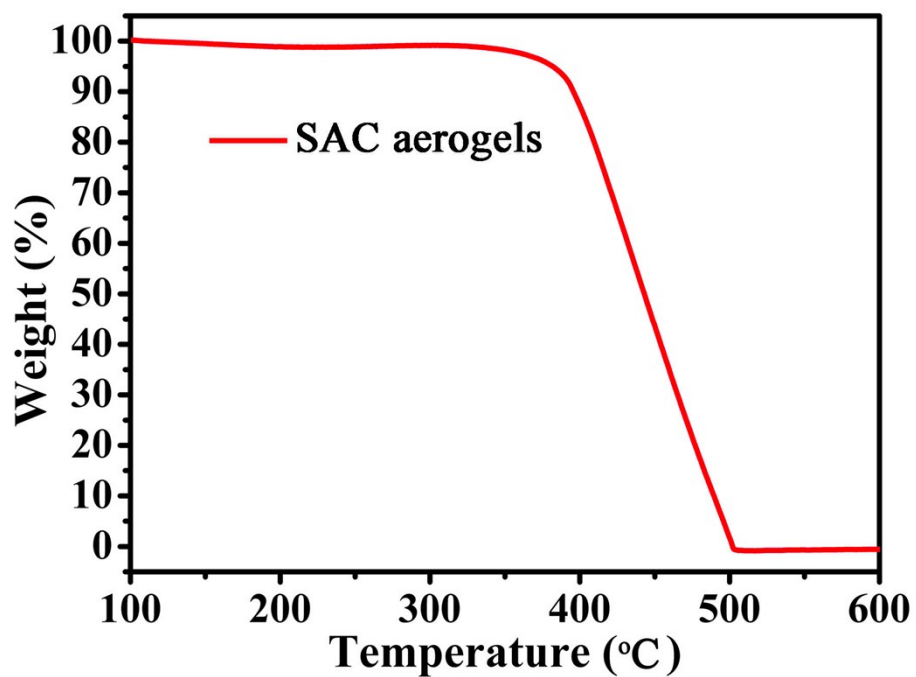
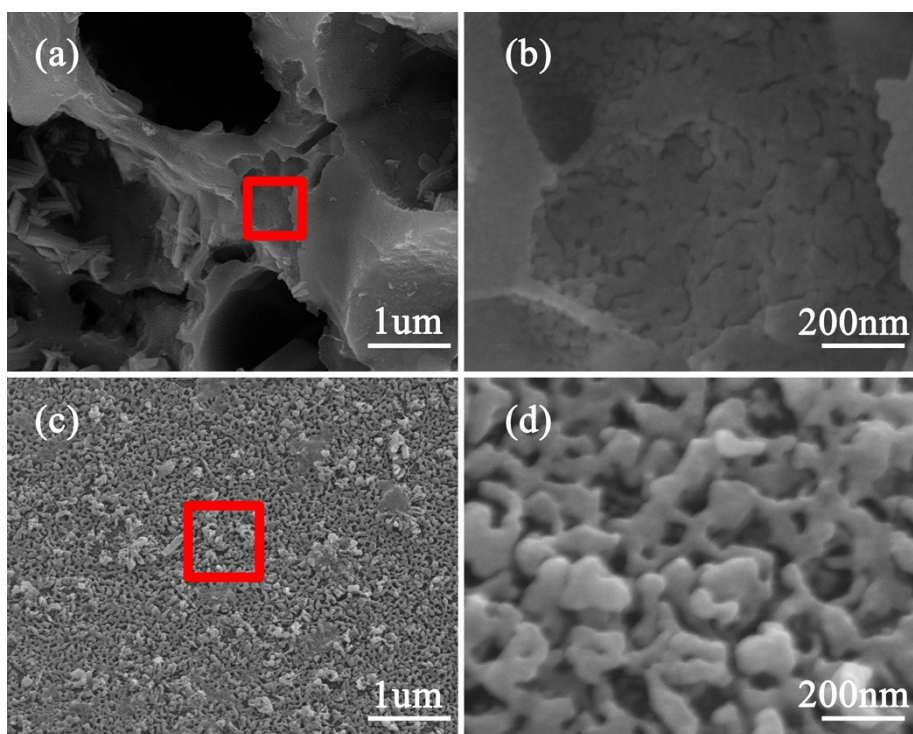


Fig. S2 XRD patterns of  $\text{MoO}_2$ @SAC-X aerogels and  $\text{MoO}_2$ @SAC aerogels which were synthesized by annealing the AMM aerogels at 600 °C under  $\text{N}_2$  without a stabilization process.



**Fig. S3** TG curve of SAC aerogels synthesized by pyrolysis of alginic acid aerogels at 600 °C under N<sub>2</sub> without a stabilization process.



**Fig. S4** SEM images of (a, b) MoO<sub>2</sub>@SAC-350 and (c, d) MoO<sub>2</sub>@SAC-400. (b) and (d) Images showing a magnified view of the area enclosed by the red box in (a) and (c).

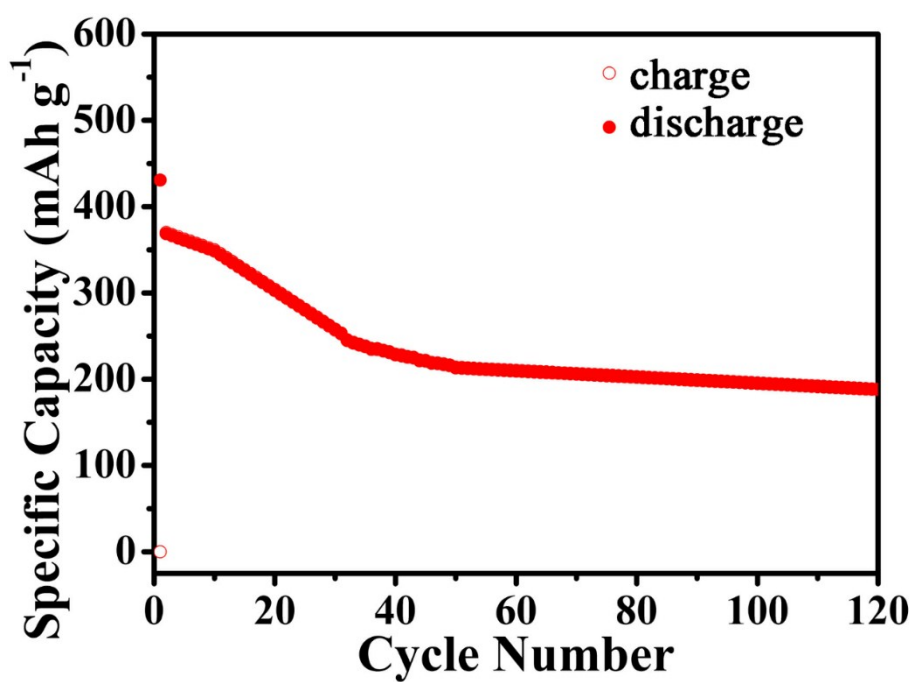
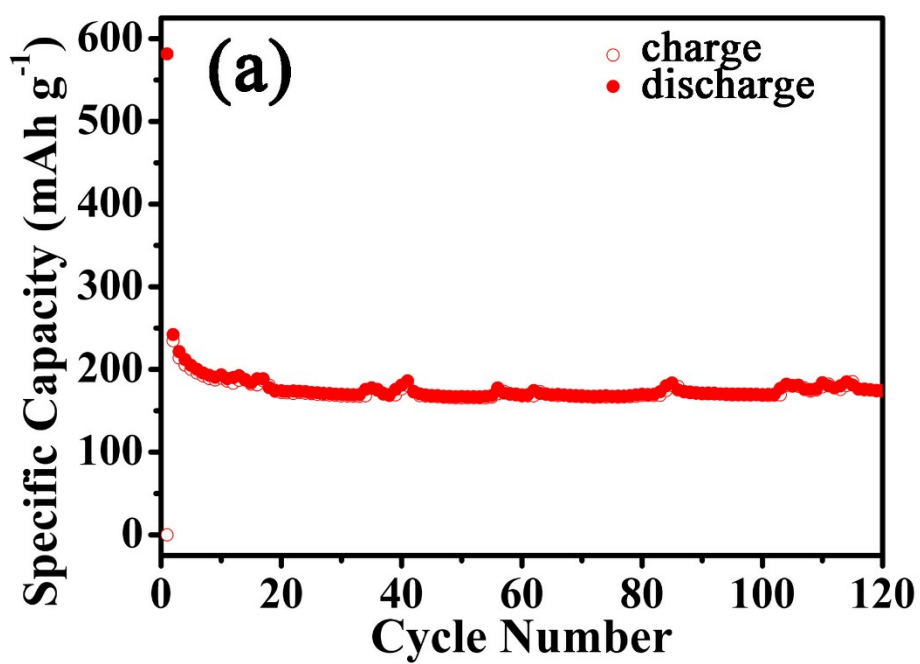
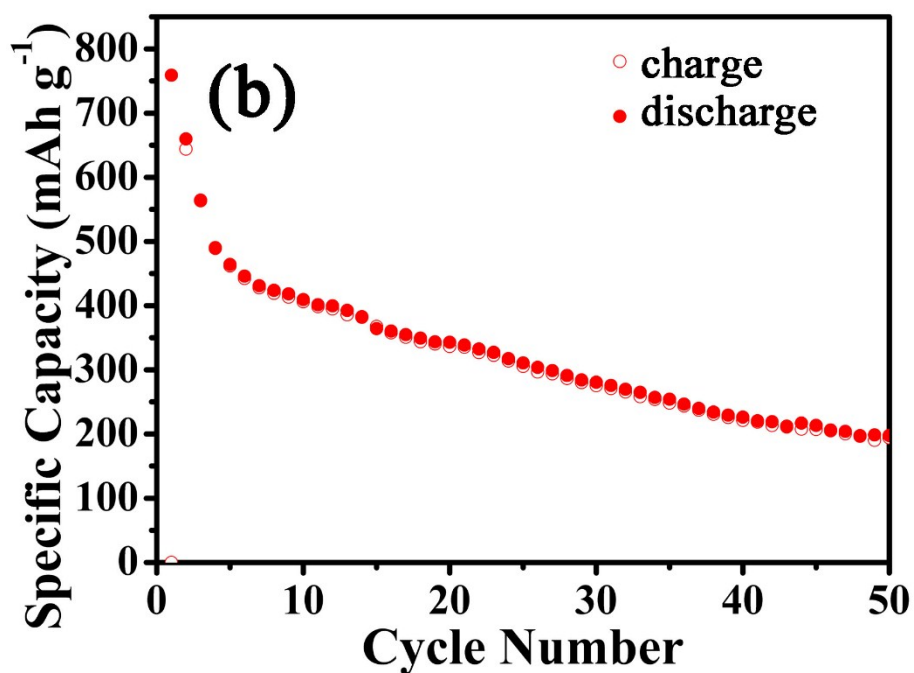
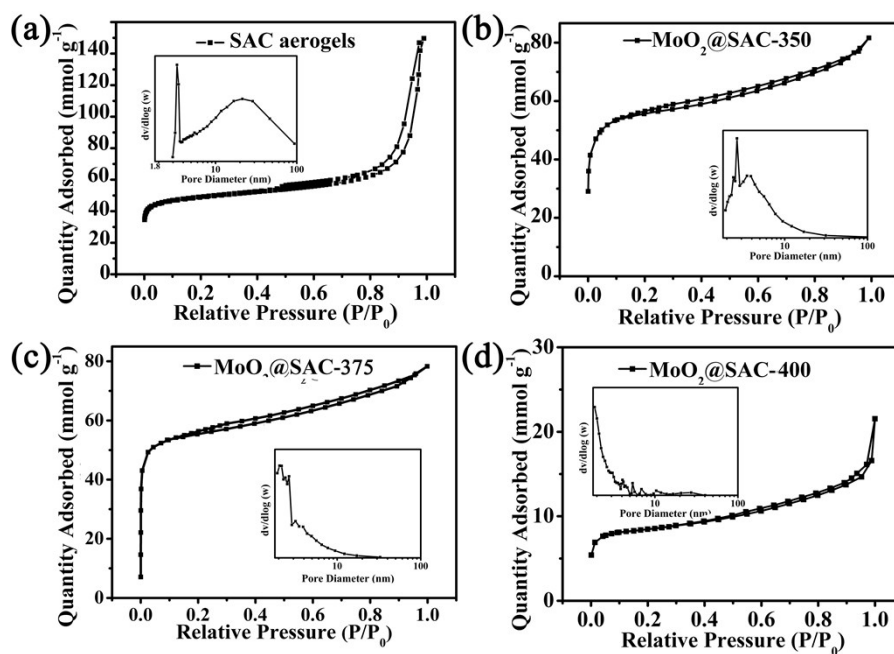


Fig. S5 Cycling performance of  $\text{MoO}_2@\text{SAC}$  electrode at current density of  $200 \text{ mA g}^{-1}$ .





**Fig. S6** Cycling performance of (a) SAC aerogels and (b) commercial  $\text{MoO}_2$  at current density of  $200 \text{ mA g}^{-1}$ .



**Fig. S7** Nitrogen adsorption-desorption isotherms and the pore diameter distribution of (a) SAC aerogels, (b)  $\text{MoO}_2$ @SAC-350, (c)  $\text{MoO}_2$ @SAC-375 and (d)  $\text{MoO}_2$ @SAC-400.

**Table. S1** BET, pore volume and average pore width of SAC aerogels, MoO<sub>2</sub>@SAC-350, MoO<sub>2</sub>@SAC-375 and MoO<sub>2</sub>@SAC-400.

Samples	BET Surface Area /m <sup>2</sup> g <sup>-1</sup>	Adsorption volume of pores /cm <sup>3</sup> g <sup>-1</sup>	Adsorption average pore width /nm
SAC aerogels	360.03	0.42	18.90
MoO <sub>2</sub> @SAC-350	210.14	0.12	5.21
MoO <sub>2</sub> @SAC-375	196.73	0.10	4.06
MoO <sub>2</sub> @SAC-400	25.22	0.09	3.59

**Table. S2** Comparison of electrochemical performance of MoO<sub>2</sub>/carbon composites electrodes prepared by different methods.

Ref	Materials	Synthesis method	Cycle numbers	Initial coulombic efficiency(%)	Reversible capacities (mA h g <sup>-1</sup> )	Rates (mA g <sup>-1</sup> )
	<b>MoO<sub>2</sub>/Carbon aerogels</b>	<b>Present study</b>	<b>120</b>	<b>72</b>	<b>490</b>	<b>200</b>
9	MoO <sub>2</sub> /carbon nanocomposites	Hydrothermal and annealing	50	52.2	629	200
11	MoO <sub>2</sub> ordered mesoporous carbon hybrids	Solvothermal	50	63.9	1049.1	100
30	Carbon-coated MoO <sub>2</sub> nanofibers	Electrospinning and annealing	50	--	430.6	200
32	MoO <sub>2</sub> /ordered mesoporous carbon nanocomposites	Thermal reduction	50	61.4	689	50
S1	MoO <sub>2</sub> /graphene composites	Solution and annealing	50	75	640	200
S2	MoO <sub>2</sub> /graphene thin film	Layer-by-layer self-assembly	100	71.5	675.9	47.8
S3	MoO <sub>2</sub> /graphene hierarchical nanoarchitectures	In-situ reduction	50	75.4	997.1	167.6

S4	Carbon coated MoO <sub>2</sub> nanobelts	Hydrothermal and annealing	30	60	617.2	100
S5	Ultrafine MoO <sub>2</sub> nanoparticles/carbon composites	Impregnation and annealing	50	62.5	409	800
S6	MWCNTs@MoO <sub>2</sub> -C nanocable composites	Electrospinning and annealing	30	--	832.2 425	50 200

## Notes and references

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