

## Supporting information

### Polymer electronic composites that heal by solvent vapour

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**Table S 1.** Molecular weight of the components used in the conducting polymer composite.

Component	$M_w$ (g/mol)	$M_n$ (g/mol)	PDI
PDMS-urea	27900	14700	1.9
P3HT	131000	26300	5.0



**Figure S 1.** Photograph of 30 wt% P3HT composite in THF (total polymer concentration of 3 wt%).

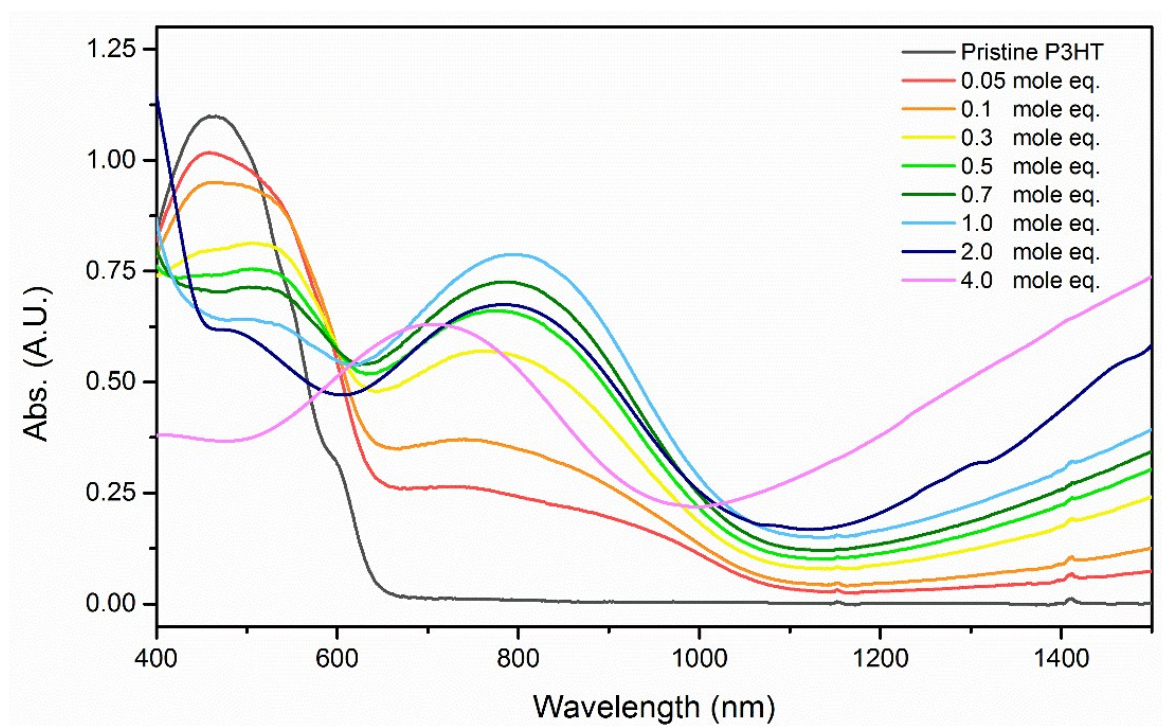


Figure S 2. Solution doping of dilute P3HT solution in chloroform with various mole equivalence of FeCl<sub>3</sub>.

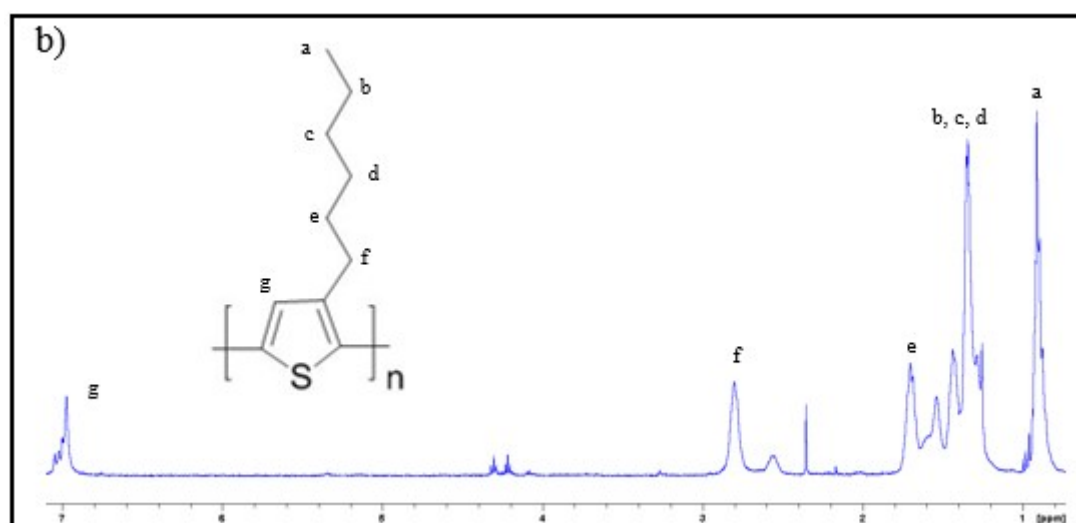
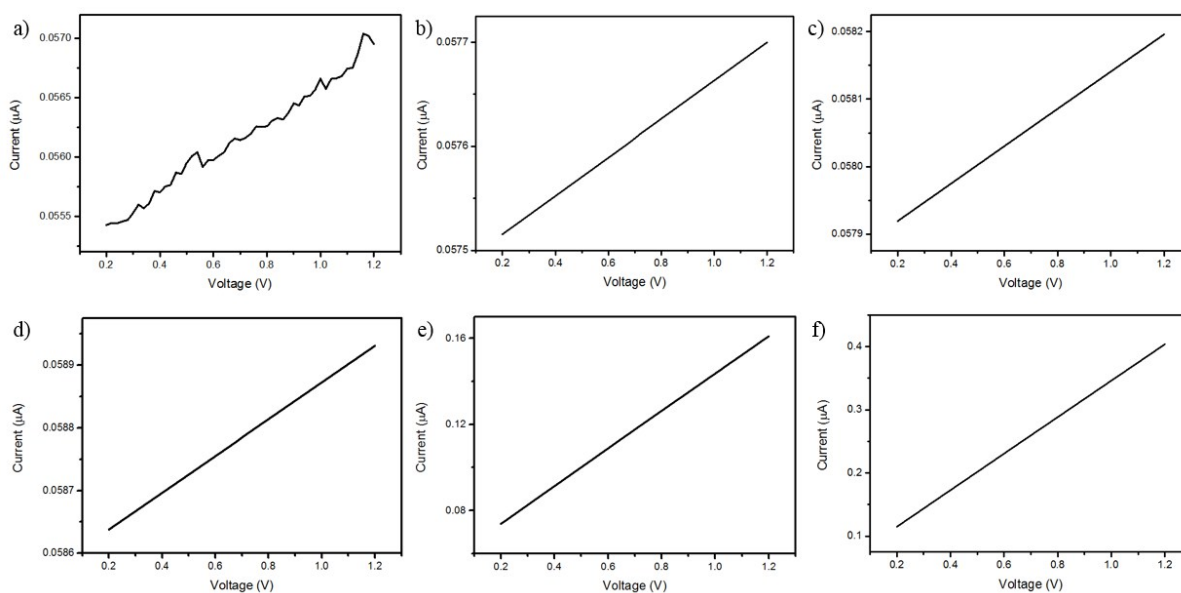
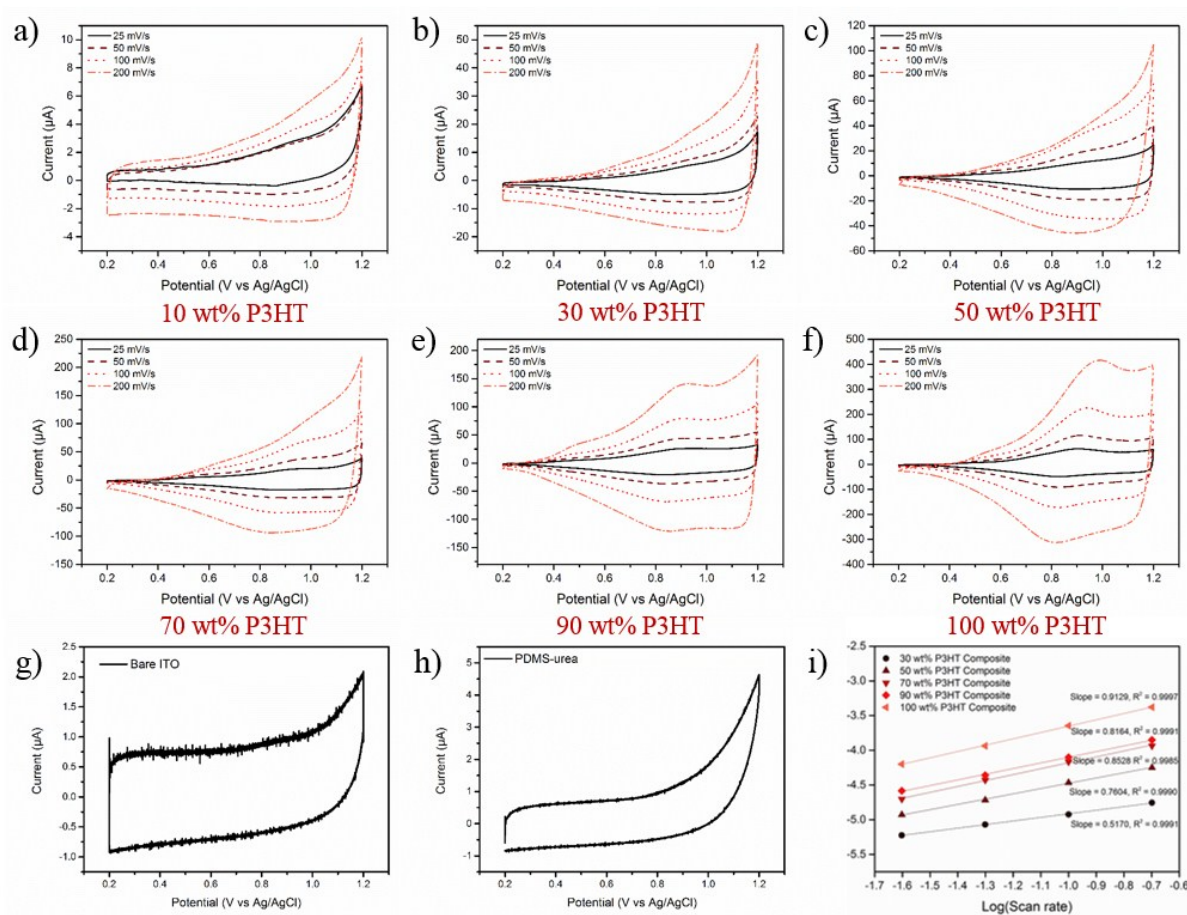


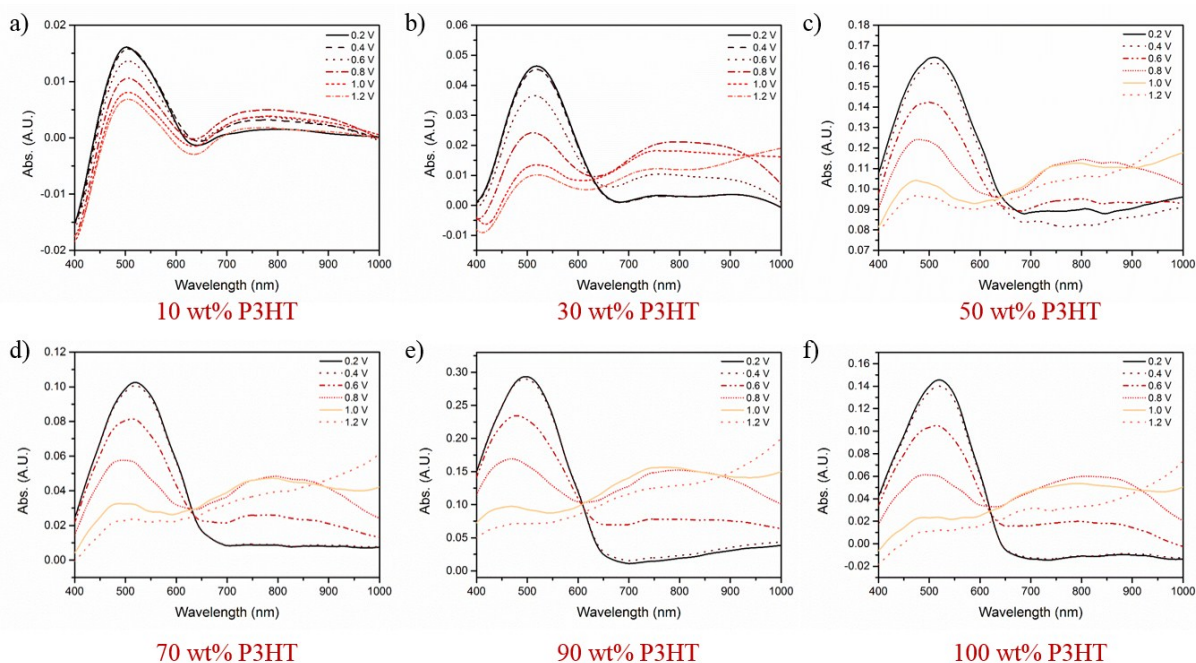
Figure S 3. <sup>1</sup>H NMR of P3HT in CDCl<sub>3</sub>.



**Figure S 4.** Current-voltage curves for a) 10, b) 30, c) 50, d) 70, e) 90, f) 100 wt% P3HT composite thin films prepared on ITO coated glass. Total polymer concentration of 3 wt% in THF was used to spin coat thin films.



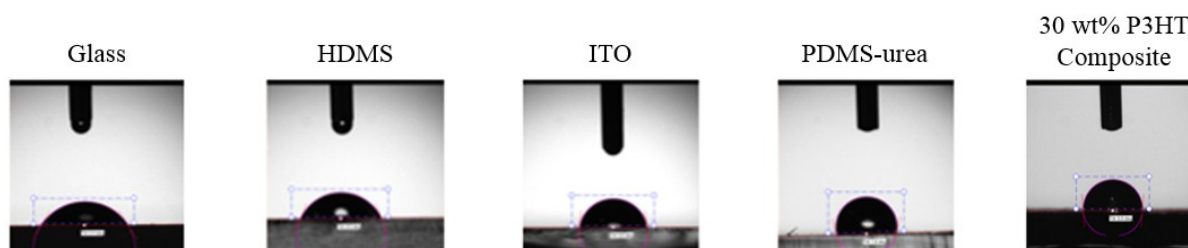
**Figure S 5.** Cyclic voltammograms of a) 10, b) 30, c) 50, d) 70, e) 90, f) 100 wt% P3HT composite thin films prepared on ITO coated glass, g) bare ITO electrode, h) pristine PDMS-urea. Total polymer concentration of 3 wt% in THF was used to spin coat thin films. Applied potential between +0.2 and +1.2 V in 0.1 M TBAHFP in MeCN. i) Linear dependence of log of scan rate over log of current of composite thin films with various weight fractions (30 to 100 wt%) of P3HT.



**Figure S 6.** Spectroelectrochemistry of a) 10, b) 30, c) 50, d) 70, e) 90, f) 100 wt% P3HT composite thin films prepared on ITO coated glass. Total polymer concentration of 3 wt% in THF was used to spin coat thin films. The applied potential was between +0.2 and +1.2 V in 0.1 M TBAHFP in MeCN.

### Surface chemistry

In order to find the optimal surface for spin coating of the composite films, various substrates of surface chemistry were analysed by water contact angle measurements (**Figure S6**). The composite thin films showed good adhesion on hydrophobic surfaces, such as of bis(trimethylsilyl)amine silanised glass and indium tin oxide (ITO)-coated glass with an average water contact angle of  $75.9 \pm 1.38^\circ$  and  $82.3 \pm 3.55^\circ$ , respectively (**Table S2**). Thus for conductivity measurements, composite films were deposited between two parallel gold contacts on HMDS-silanised glass substrates, whereas ITO-coated glass was used for electrochemical studies.



**Figure S 7.** Water contact angle of various substrates, PDMS-urea and the 30 wt% P3HT composite.

**Table S 2.** Contact angle measurements on various substrates.

	<b>Glass</b>	<b>HMDS</b>	<b>ITO</b>	<b>PDMS-urea</b>	<b>30 wt% P3HT composite</b>
<b>Contact angle</b>	55.3 ± 2.39	75.9 ± 1.38	82.3 ± 3.55	106 ± 1.01	97.5 ± 2.53