

Supplementary Information for

The Preparation of High-Quality Water-soluble Silicon Quantum Dots and Their Application in Detection of Formaldehyde

Xiaoling Xu, Shiyao Ma, Yan Hu, Xincai Xiao and Dan Zhao*

School of Pharmaceutical Sciences South-Central University for

Nationalities, Wuhan 430074, P. R. China

****Corresponding Author:***

E-mail address: wqzhdpai@163.com (D. Zhao)

QY Calculations

The QYs of SiQDs were calculated by comparing the integrated PL intensities and absorbance values of the samples (excited at 320 nm), using quinine sulfate dissolved in 0.1 mol/L H₂SO₄ aqueous solution (refractive index (η) of 1.33) as the standard (QY = 58%). All samples dissolved in water ($\eta = 1.33$) had absorbance less than 0.1 at 320 nm. The relative QY can be calculated using the below equation:

$$\Phi_X = \Phi_{ST} (\text{Grad}_X / \text{Grad}_{ST}) (\eta_X^2 / \eta_{ST}^2)$$

Where Φ is the QY, Grad is the gradient from the plot of integrated fluorescence intensity versus absorbance, and η is the refractive index of the solvent; ST denotes the standard and X denotes the sample.

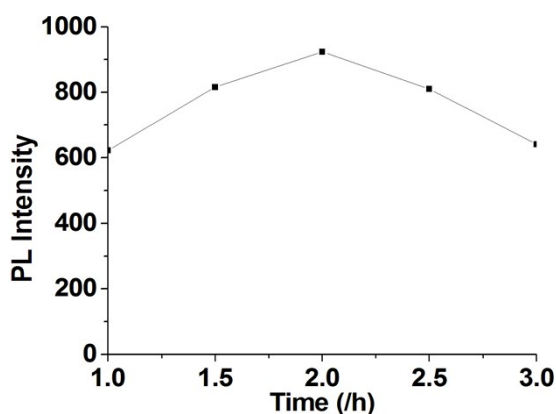


Fig.S1 PL spectra of Si QDs grown at different heating temperature.

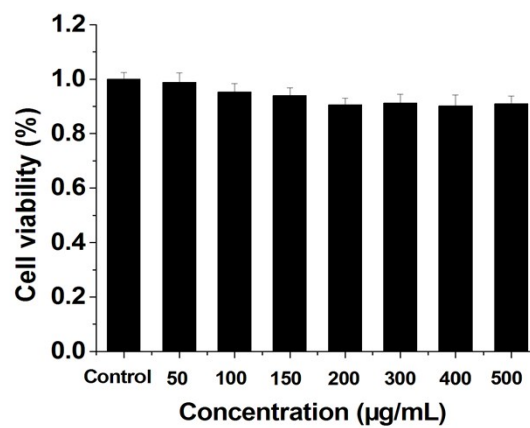


Fig.S2 Cell viability of L-929 cells treated with SiQDs of serial concentrations.

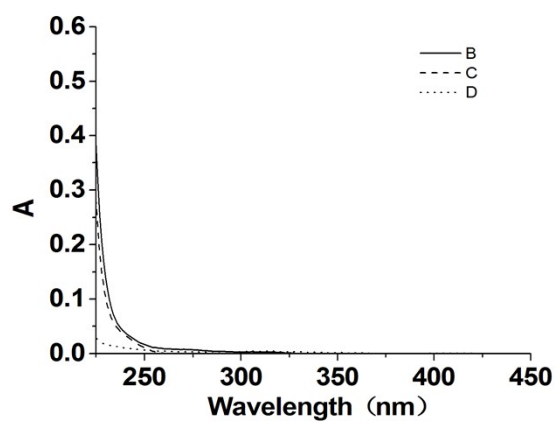


Fig.S3 UV-visible absorption spectra of APTMS, APTES and UPTES.

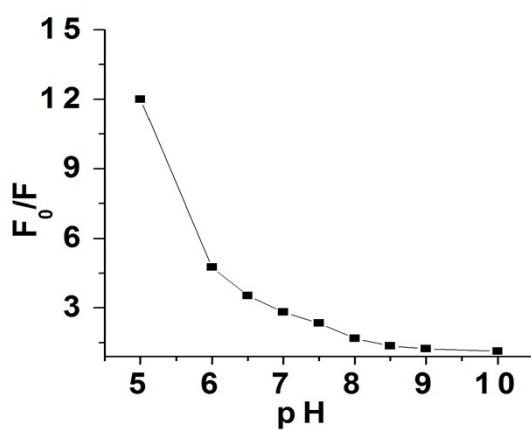


Fig.S4 The quenching degree of Si QDs at different pH values.