Supporting information

Multistimuli-responsive organogels based on hydrazide and

azobenzene derivatives

Table 1 Gelation properties of E5, E6 and E10 in various solvents

Solvent	E5	E6	E10
DMF	S	G (8)	G (11.5)
THF	S	S	S
DMSO	S	S	S
CHCl ₃	Ι	Ι	Ι

I: insoluble; G: stable gel formed at room temperature; S: soluble; Numbers in parentheses present the critical gel concentration (CGC, mg/mL).

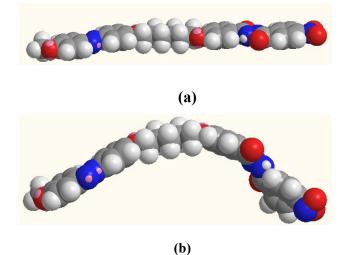


Fig. S1 The molecular shapes of: (a) **E6**, an even; (b) **E5**, an odd numbered central spacer in the all-*trans* conformation (calculated by MM2).

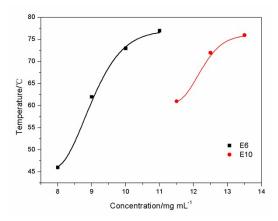


Fig.S2 Plots of T_{gel} versus the concentration of E6 () and E10 () in DMF.

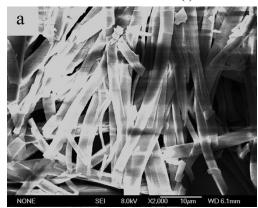


Fig. S3 SEM images of E5 solution in DMF(25 mg/ml).

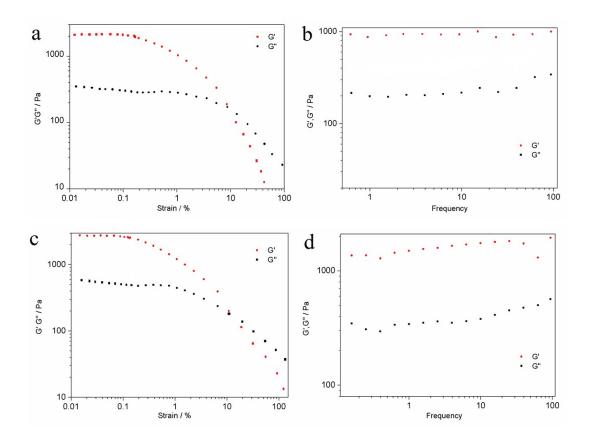


Fig. S4 (a) Amplitude dependencies and (b) Frequency dependency of storage modulus (G') and lose moulus (G") of **E6** gel in DMF (12mg/ml). (c) Amplitude dependencies and (d) Frequency dependency of storage modulus (G') and lose moulus (G") of **E10** gel in DMF (The frequency is 1 Hz and the strain is 0.1%, 12mg/ml).

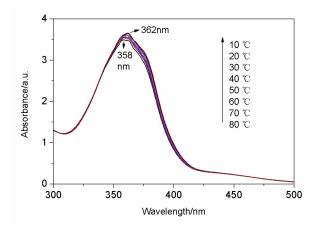


Fig. S5 Temperature-dependent UV-vis absorption spectra of E6 solution $(1 \times 10^{-3} \text{ mol/L})$ in DMF

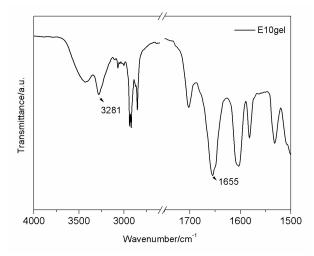
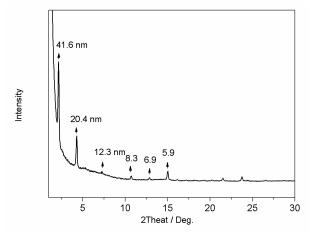
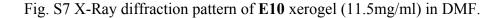


Fig. S6 FTIR spectra of E10 xerogel (11mg/mL) in DMF.





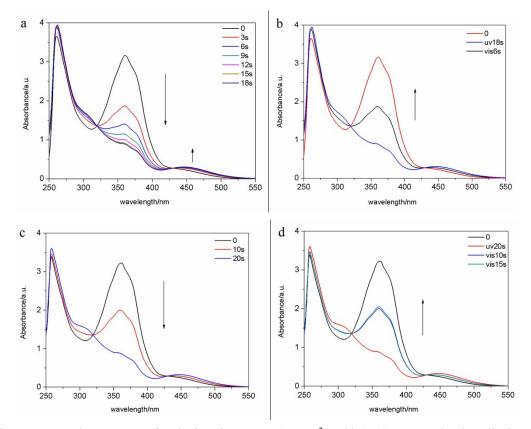


Fig. S8 UV-vis spectra of solution in DMF $(1 \times 10^{-3} \text{mol/L})$ (a) E5 under irradiation at UV light for 0-18s; (b) E5 under visible light irradiation for 6s; (c) E10 under irradiation at UV light for 20s; (d) E10 under visible light irradiation for 15s.

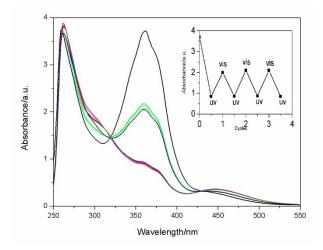


Fig. S9 UV-vis spectra of E6 solution in DMF $(1 \times 10^{-3} \text{mol/L})$ are irradiated multiple times by UV and vis light.

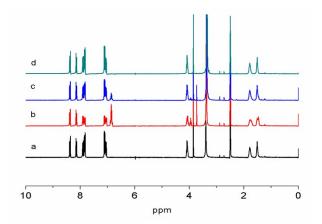


Fig. S10 Partial ¹H-NMR spectrum of **E6** upon alternating irradiation (a) none; (b) UV light for 1.5h; (c) vis light for 1.5h; (d) heat in DMSO-d6(1×10^{-2} mol/L).

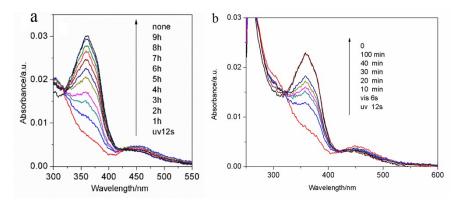


Fig. S11 The *trans*-azobenzene of E_6 (1×10⁻⁵mol/L) be recovered by place in darkness (a) after UV light irradiation and then placed in darkness for 9h; (b) after UV light irradiation for 12s and vis light irradiation 6s, and then placed in darkness for 100min.

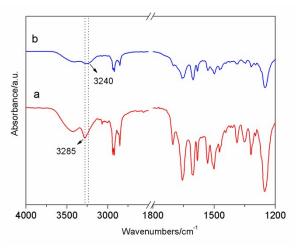


Fig. S12 FTIR spectra of (a) xerogel of **E10** gel in DMF; (b) the precipitate developed from the gel in DMF under the irradiation of 365 nm UV light.

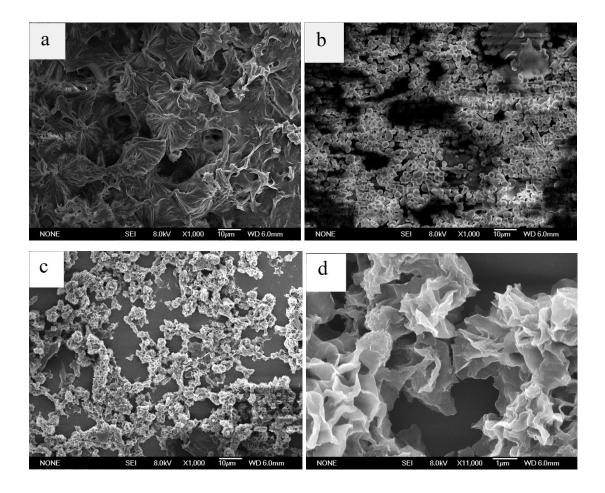
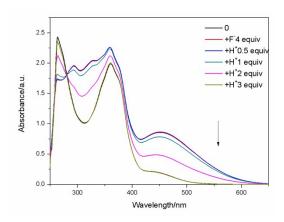


Fig. S13 SEM images of **E10**: (a) the precipitate from organogel in DMF under irradiation by 365 nm UV light for 9h; (b) the solution of organogel in DMF under irradiation by 365 nm UV light for 9h. (c) (d) the solution after visible light irradiation. (11.5 mg/mL)



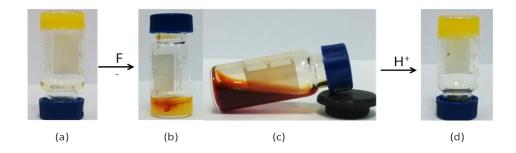


Fig. S14 (a) **E6** organogel (8mg/ml, in DMF); (b) immediately after addition of solid TBAF (10equiv); (c) after 1 min; (d) addition of H^+ .

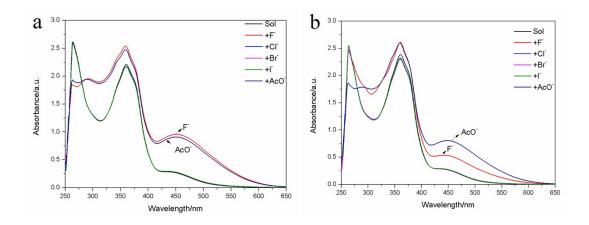
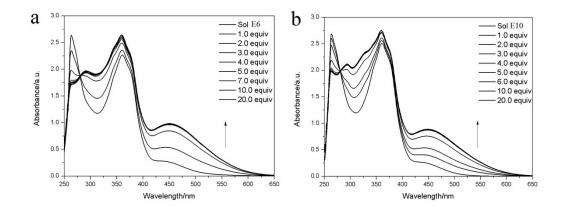


Fig. S15 UV-vis spectra of (a) E5; (b) E10 $(1 \times 10^{-4} \text{mol/L})$ in DMF upon addition of 10 equiv of various anions (F⁻, Cl⁻, Br⁻, I⁻, AcO⁻).



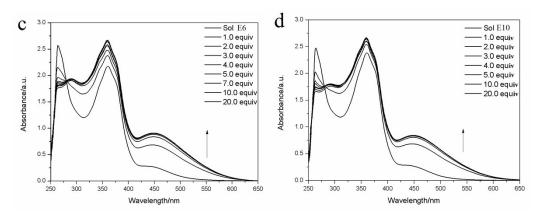


Fig. S16 UV-vis absorption spectra of **E6 and E10** solution in the presence of $F^-(a, b)$ and AcO⁻ (c, d) in DMF (1×10⁻⁴mol/L).

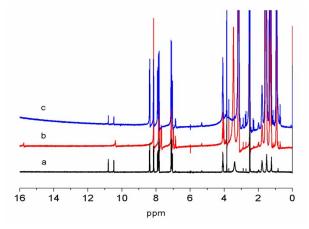


Fig. S17 Partial ¹H-NMR spectra of **E6** in the addition of (a) none; (b) 5 equiv F⁻; (c) 10 equiv CF₃COOH in DMSO- $d6(1 \times 10^{-2}M)$.

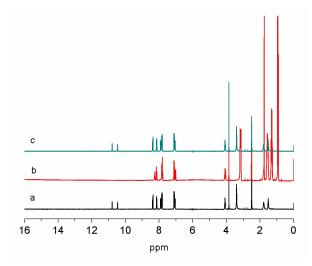


Fig. S18 Partial ¹H-NMR spectra of **E6** in the addition of (a) none; (b) 3 equiv AcO⁻; (c) 10 equiv CF₃COOH in DMSO-*d6*(1×10^{-2} mol/L).

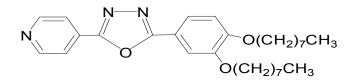


Fig. S19. Molecular structure of compound 4-poxd-B8.

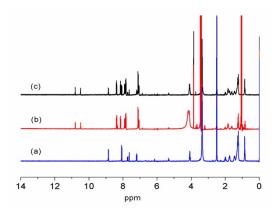


Fig. S20 Partial ¹H-NMR spectra of (a) **4-poxd-B8**, (b) **E5**, (c) two-components (**4-poxd-B8** : **E5=1:1**) in DMSO-*d6* (3×10^{-3} mol/L).

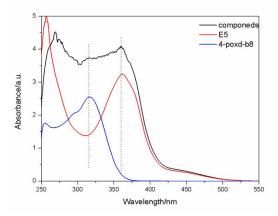
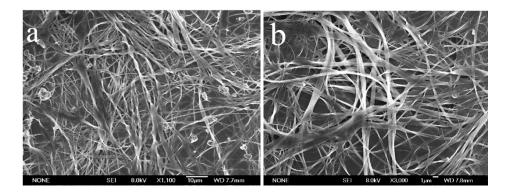


Fig. S21 UV-vis absorption spectra of 4-poxd-B8 \times E5 and two-components (4-poxd-B8 : E5=1:1) in DMSO (1×10⁻⁴mol/L).



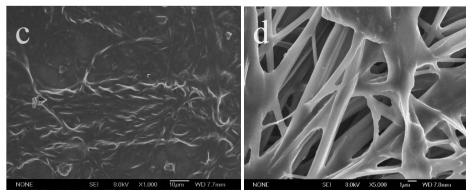


Fig. S22 SEM images of (a) **4-poxd-B8** xerogels in DMSO (5mg/ml); (b) twocomponents xerogels in DMSO (1:1); (c) two-components in the addition of 4 equiv F^- ; (d) two-components of addition 4 equiv CF₃COOH under (c).

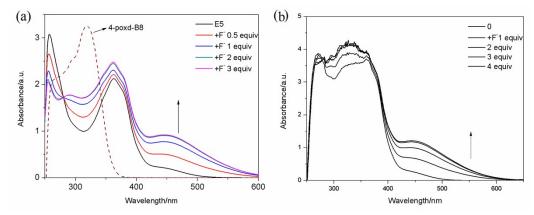


Fig. S23 UV-vis absorption spectra of (a) **4-poxd-B8** and **E5** solution in the DMSO $(1 \times 10^{-4} \text{ mol/L})$ in the presence of 0-3 equiv F⁻; (b) two-components (**4-poxd-B8** : **E5=1:1**) solution in the DMSO $(1 \times 10^{-4} \text{ mol/L}, 1:1)$ in the presence of 0-4 equiv F⁻.

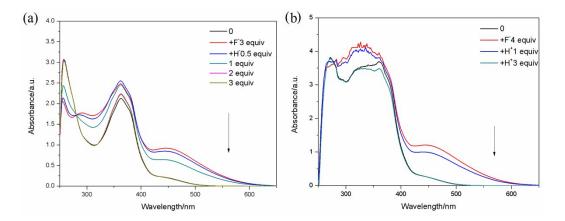


Fig. S24 UV-vis absorption spectra of (a) **E5** solution treated with 3equiv F⁻, then added 0.5-3 equiv CF₃COOH; (b) two-components (**4-poxd-B8** : **E5=1:1**) solution in the DMSO (1×10^{-4} mol/L,1:1) in the presence of 4 equiv F⁻, then added 1-3 equiv

CF₃COOH.



Fig. S25 (a) Organogel **E6** (8mg/ml, in DMF); (b) after addition of solid NaOH (20equiv) for 6h; (c) addition of HCl (20 equiv).

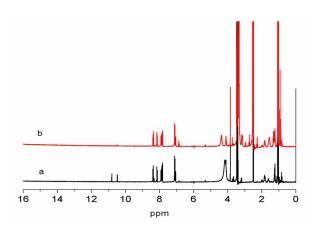


Fig. S26 Partial ¹H-NMR spectra of (a) E5; (b) addition 20 equiv NaOH.

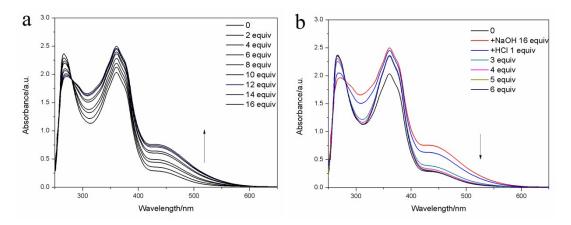


Fig. S27 UV-vis absorption spectra of E10 in the presence of (a) NaOH (0-16equiv.); (b) HCl in DMF (1×10^{-4} mol/L).