

Supporting Information

Iron-nitrogen co-doped hierarchically mesoporous carbon spheres as highly efficient electrocatalysts for oxygen reduction reaction

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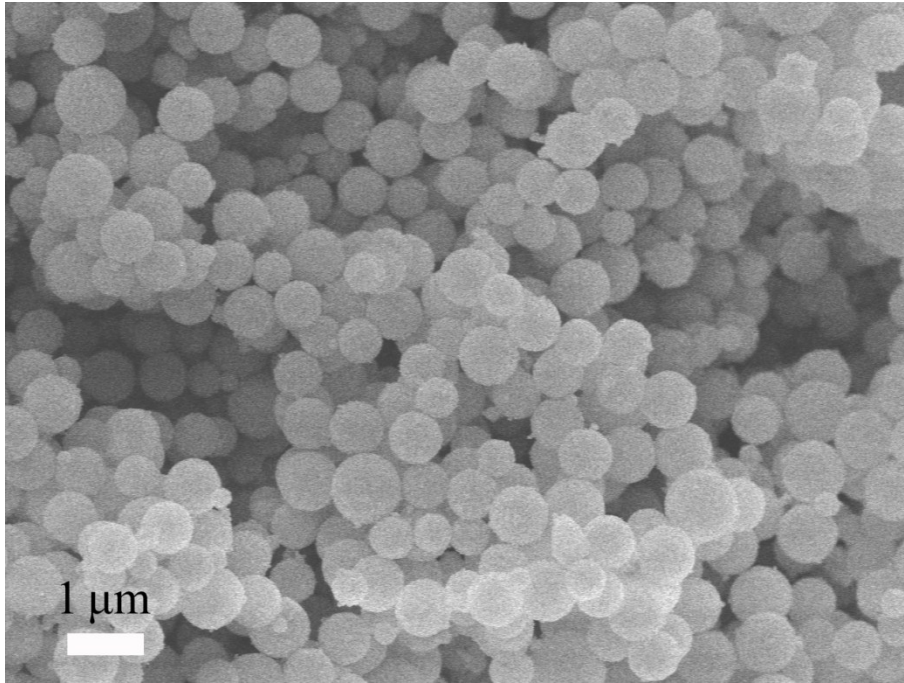


Figure S1 Typical FESEM image of hierarchically mesoporous silica template.

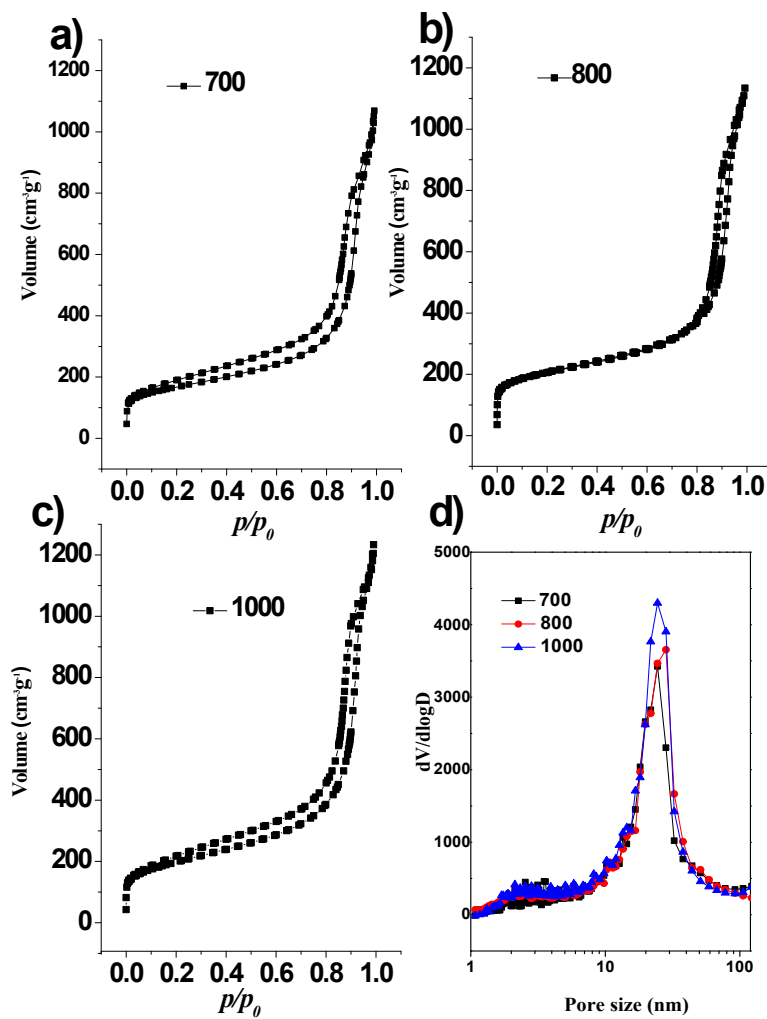


Figure S2. a, b, c) Nitrogen adsorption-desorption isotherms of Fe-N-CS-T obtained from different carbonization temperature of 700, 800 and 1000 °C, respectively. d) Their corresponding pore size distribution curves.

Table S1. BET surface areas, total pore volumes and pore size distribution of Fe-N-CS-TObtained from different carbonization temperature.

catalysts	T (°C) ^a	S _{BET} (m ² /g) ^b	V _{tot} (cm ³ /g) ^c	d _{meso} (nm) ^d
Fe-N-CS-700	700	602	1.33	24
Fe-N-CS-800	800	724	1.51	28
Fe-N-CS-900	900	758	1.59	28
Fe-N-CS-1000	1000	695	1.63	24
N-CS-900	900	429	1.18	22
Fe-N-CS/SiO ₂ -900	900	302	0.29	--

^a Carbonization temperature.

^b BET specific surface areas obtained from N₂ adsorption isotherm in the range of P/P₀ = 0.05-0.3.

^c Total pore volume was obtained at P/P₀ of 0.95.

^dPrimary mesopore diameter calculated from BJH method.

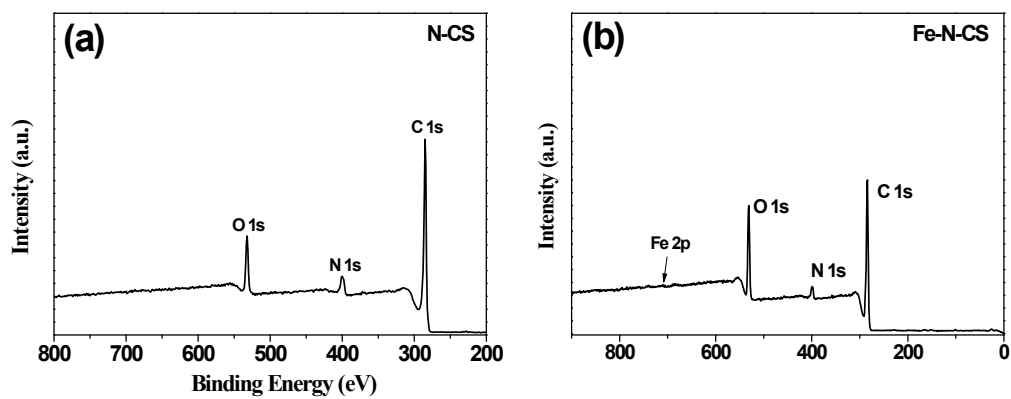


Figure S3 XPS survey of (a) N-CS-900 and (b) Fe-N-CS-900.

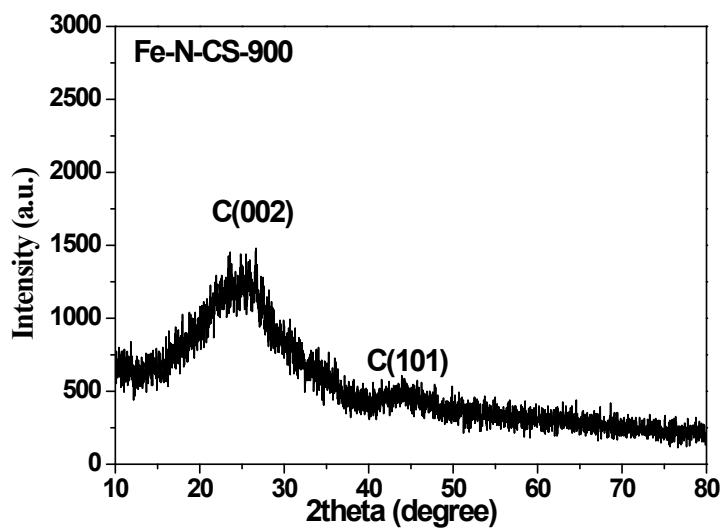


Figure S4 XRD pattern of Fe-N-CS-900 sample.

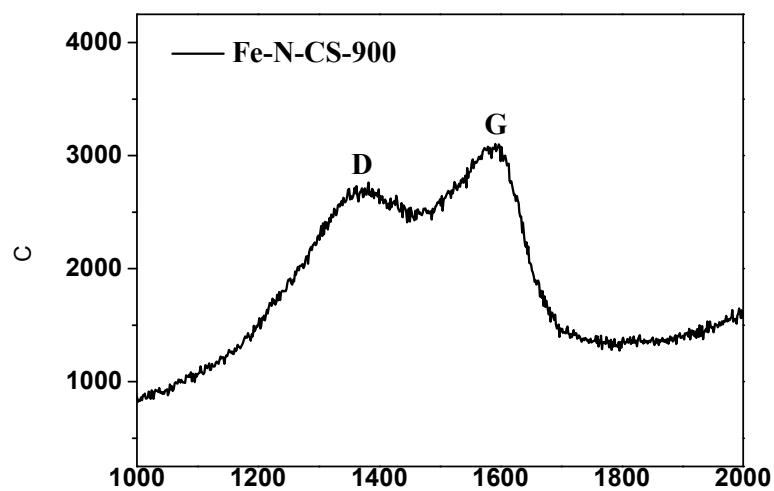


Figure S5 Raman spectrum of Fe-N-CS-900 sample.

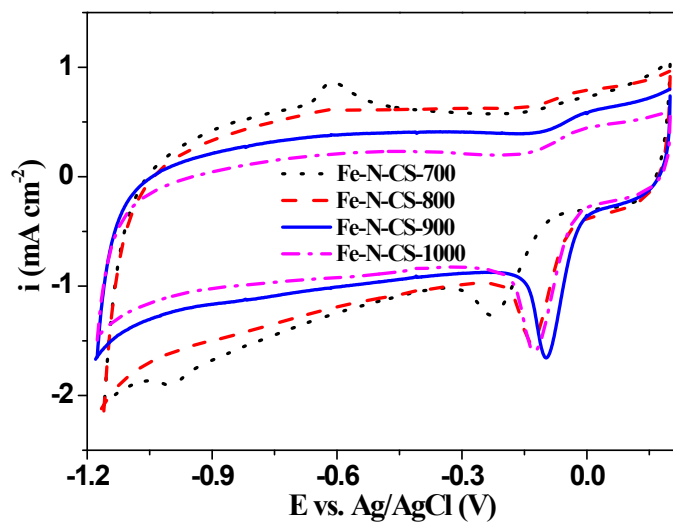


Figure S6 a, b, c) Cyclic voltammograms of Fe-N-CS-700, Fe-N-CS-800 and Fe-N-CS-900, Fe-N-CS-1000 samples on a glassy carbon rotating disk electrode in O_2 -saturated 0.1 M KOH with scan rate of 20 mV s^{-1} .

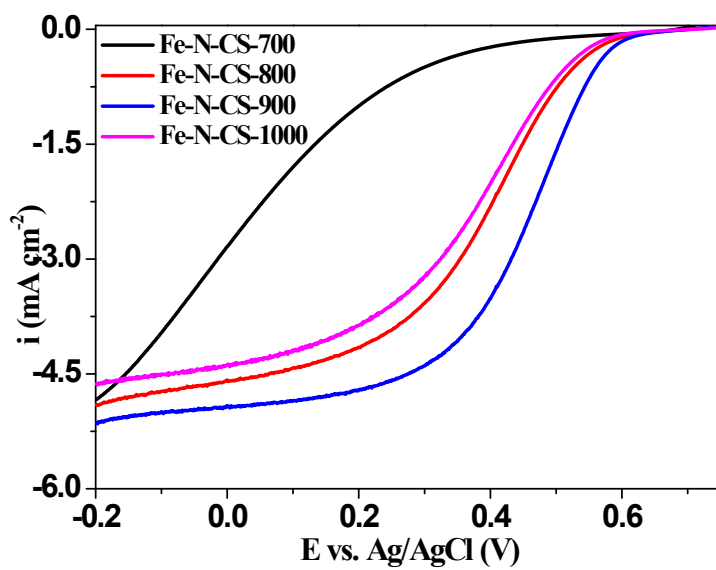


Figure S7 LSV curves of Fe-N-CS-700, Fe-N-CS-800, Fe-N-CS-900 and Fe-N-CS-1000 catalysts modified electrodes at rotation rate of 1600 rpm in O₂-saturated acidic media.

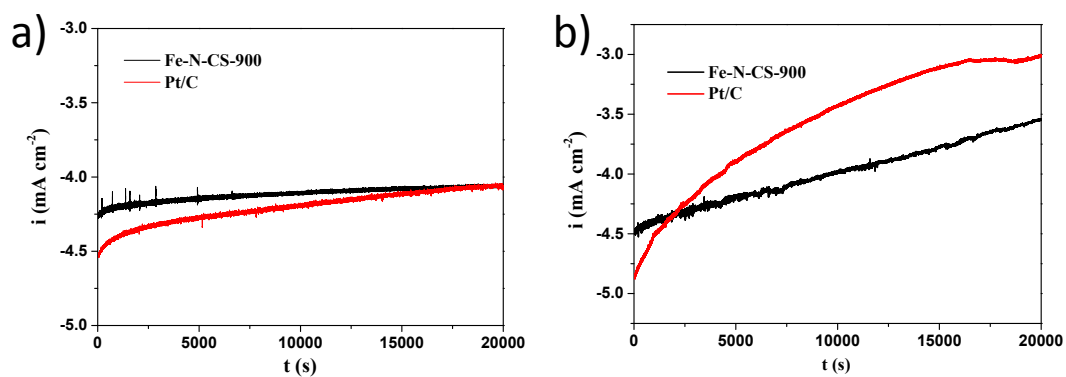


Figure S8 a) Chronoamperometric responses for the ORR on the Fe-N-CS-900 catalyst and commercial Pt/C catalyst modified electrodes in O₂-saturated alkaline media for 20000 s.

b) Chronoamperometric responses for the ORR on the Fe-N-CS-900 catalyst and commercial Pt/C catalyst modified electrodes in O₂-saturated acidic media for 20000 s.