

## **Theoretical insights into the $\omega$ -alkynylfuran cycloisomerisation catalyzed by Au/CeO<sub>2</sub>(111): the role of CeO<sub>2</sub>(111) support**

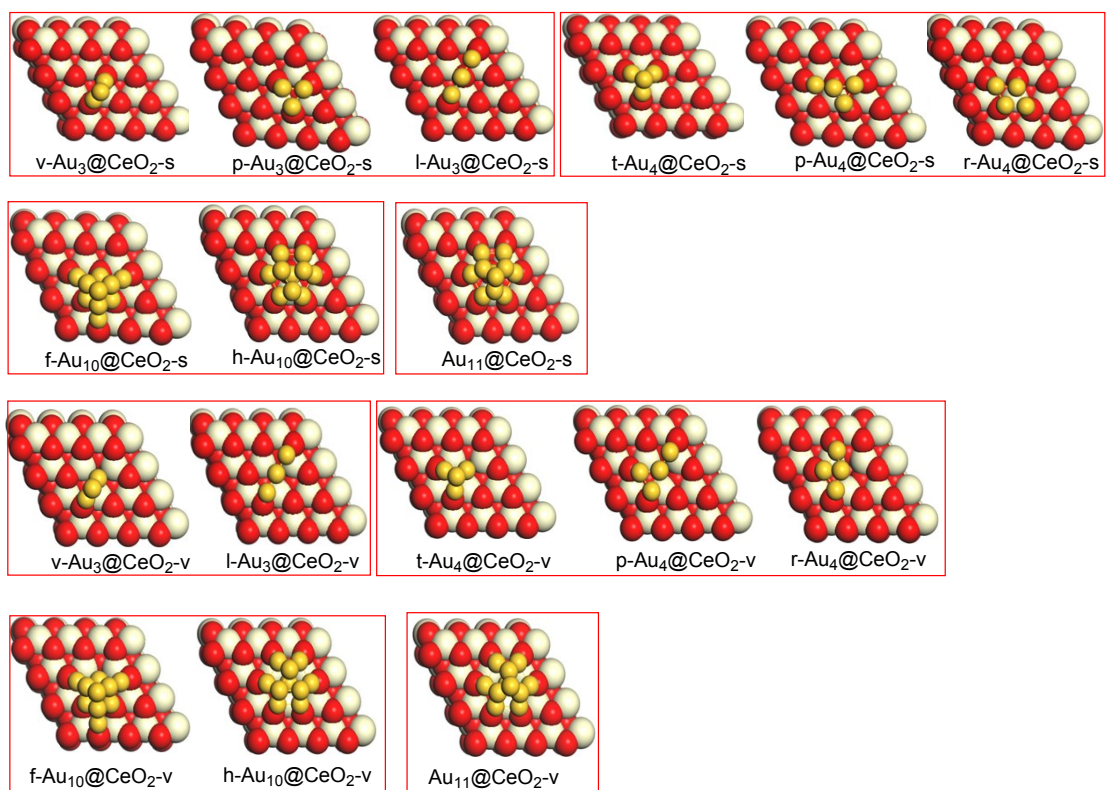
Yafei Luo,<sup>a</sup> Zhongzhu Chen,<sup>a</sup> Jin Zhang,<sup>b,\*</sup> Ying Tang,<sup>a</sup> Zhigang Xu<sup>a</sup> and Dianyong Tang<sup>a,b,\*</sup>

<sup>a</sup> *International Academy of Targeted Therapeutics and Innovation, Chongqing University of Arts and Sciences, Chongqing 402160 (P.R. China).*

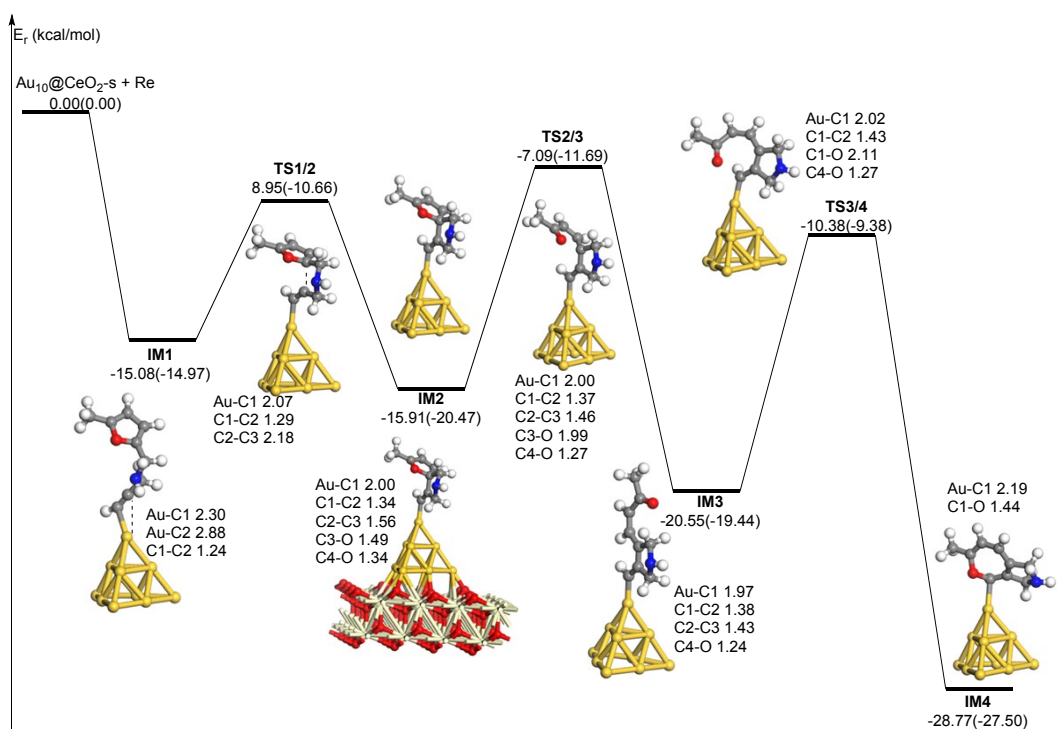
<sup>b</sup> *Chongqing Key Laboratory of Environmental Materials and Remediation Technologies and Research Institute for New Materials Technology, Chongqing University of Arts and Sciences, Chongqing 402160 (P.R. China).*

E-mail: Zhangjin@cqwu.edu.cn and tangdy2008@163.com

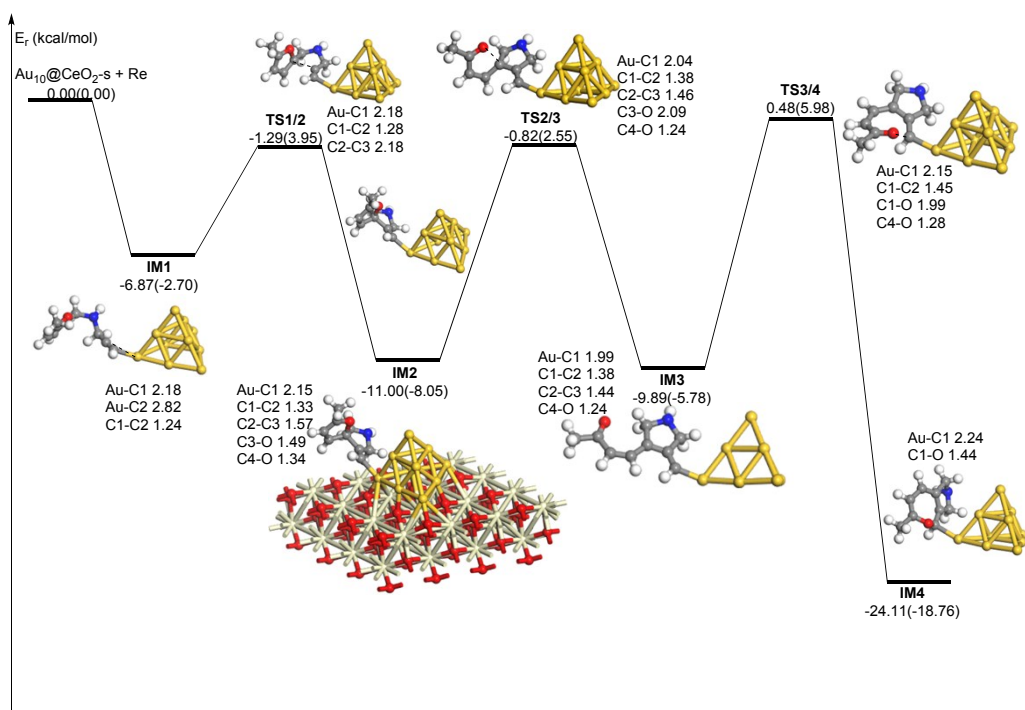
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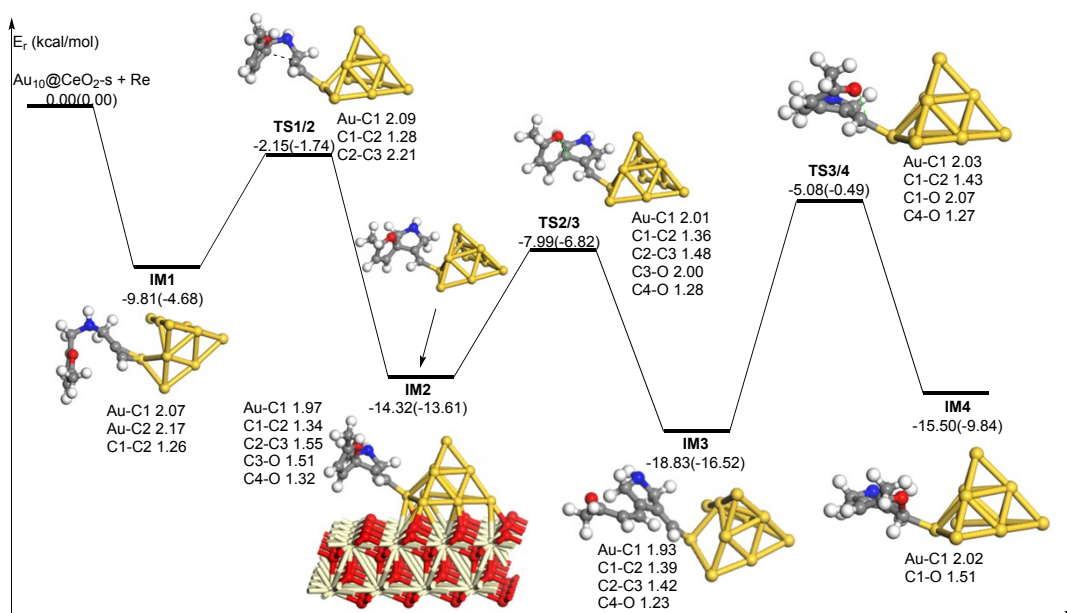
**Figure S1.** The optimized structures of Au<sub>3</sub>, Au<sub>4</sub>, Au<sub>10</sub>, and Au<sub>11</sub> supported on CeO<sub>2</sub>(111) slab.



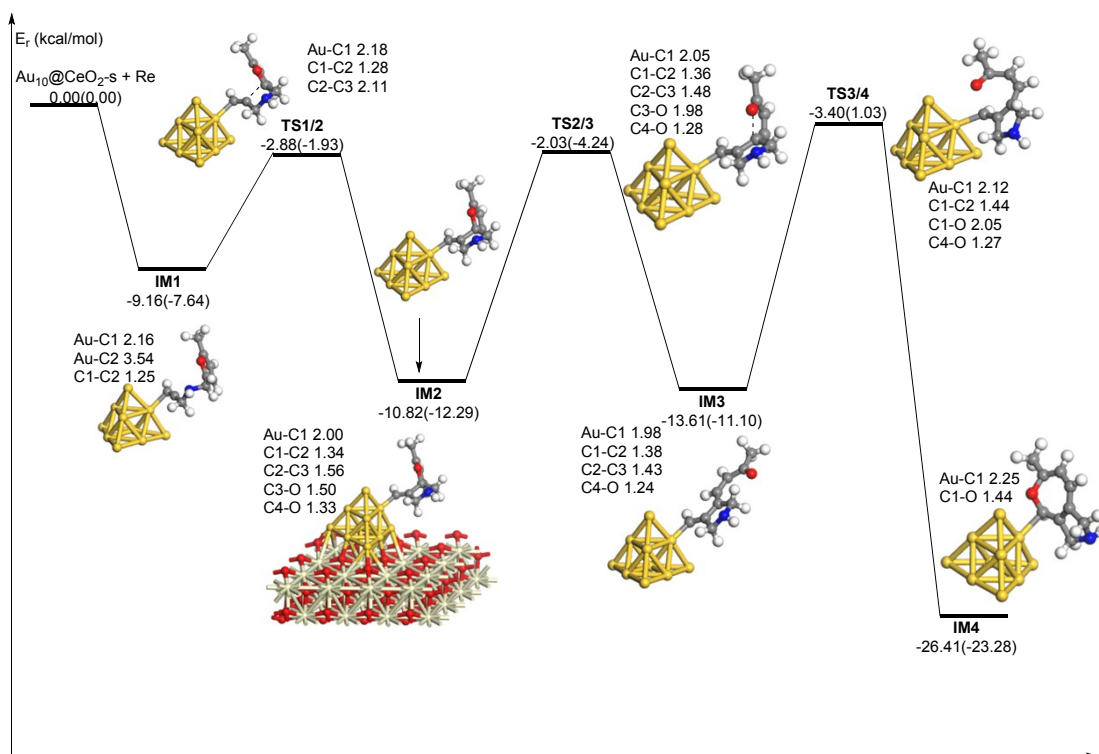
**Figure S2.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the atop site of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-s}$ .



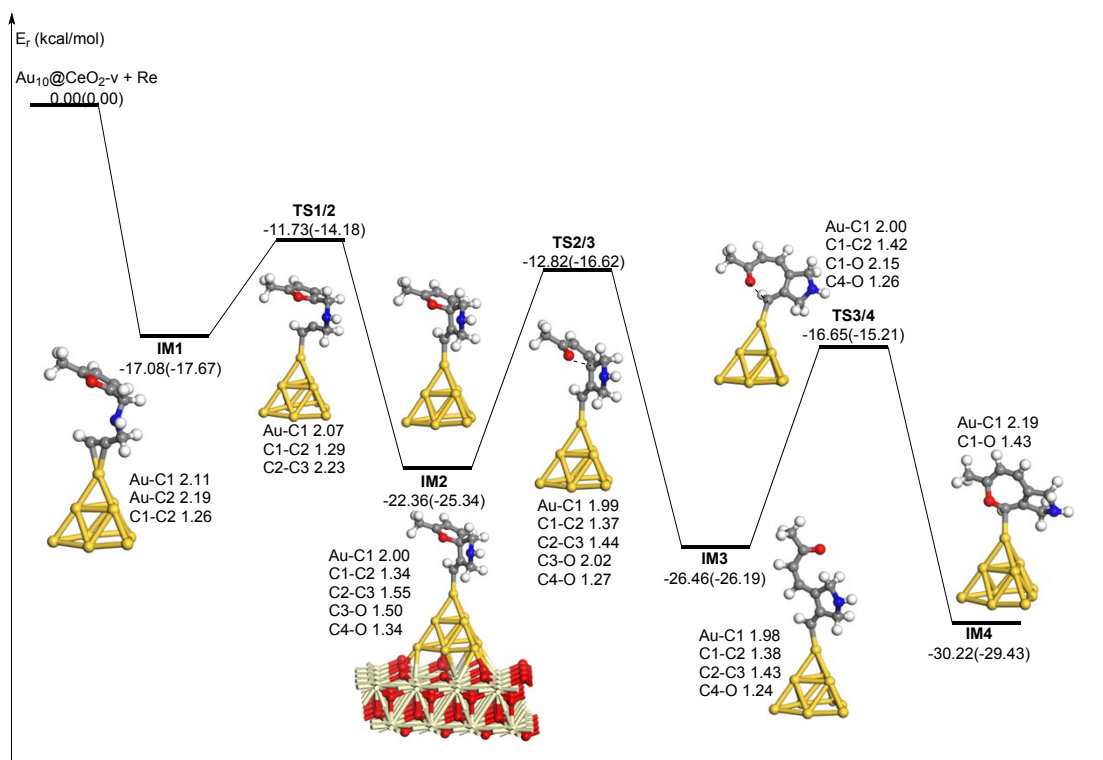
**Figure S3.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the interface-corner site of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-s}$ .



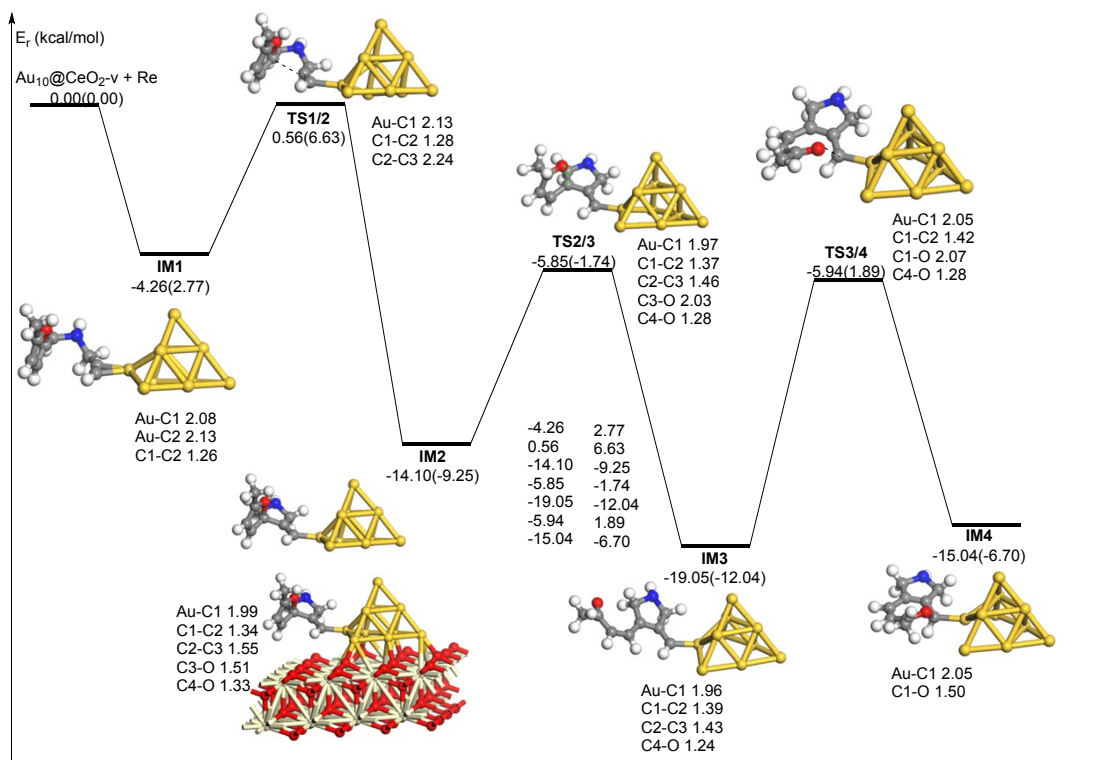
**Figure S4.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the interface-edge site of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-s}$ .



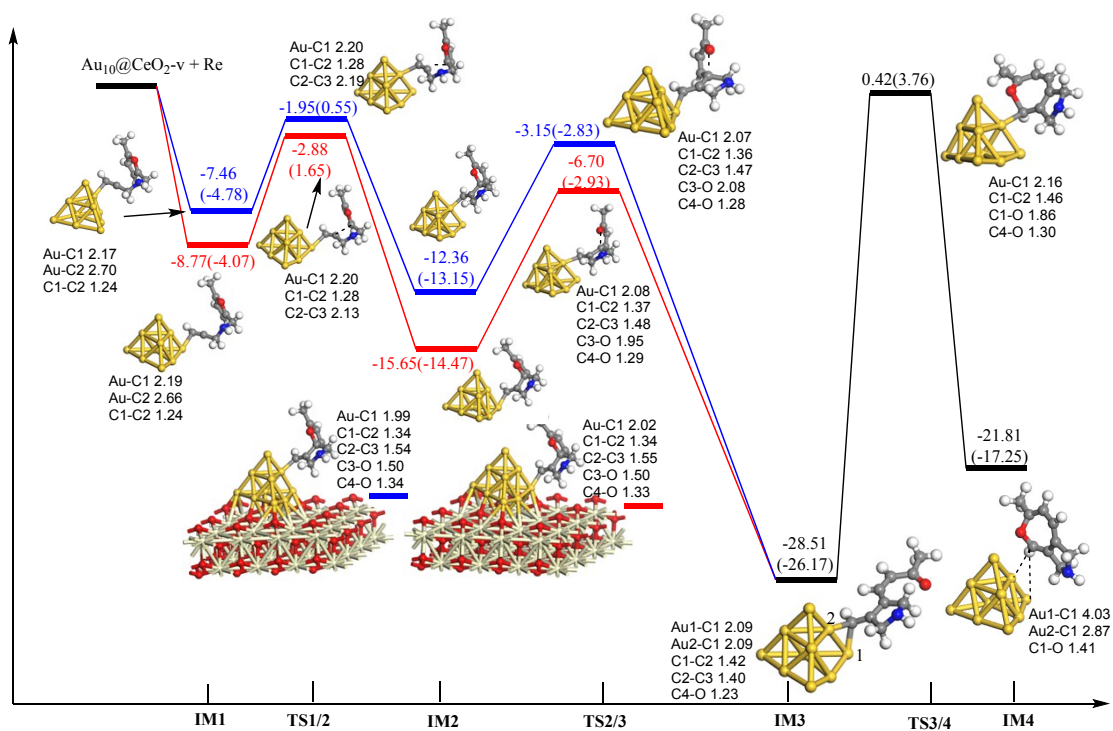
**Figure S5.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the edge site of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-s}$ .



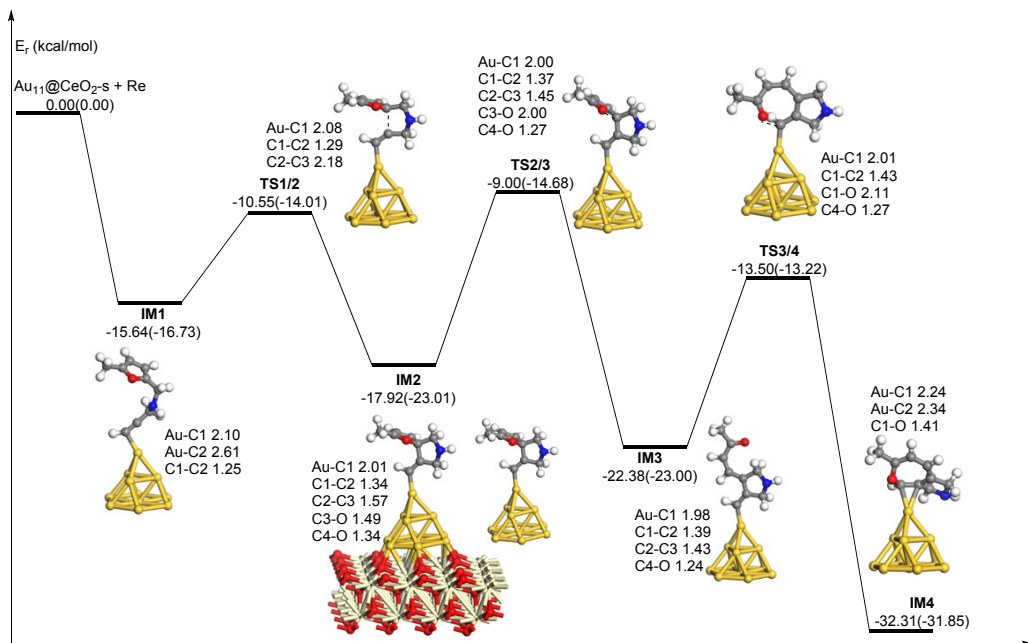
**Figure S6.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the atop site of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-v}$ .



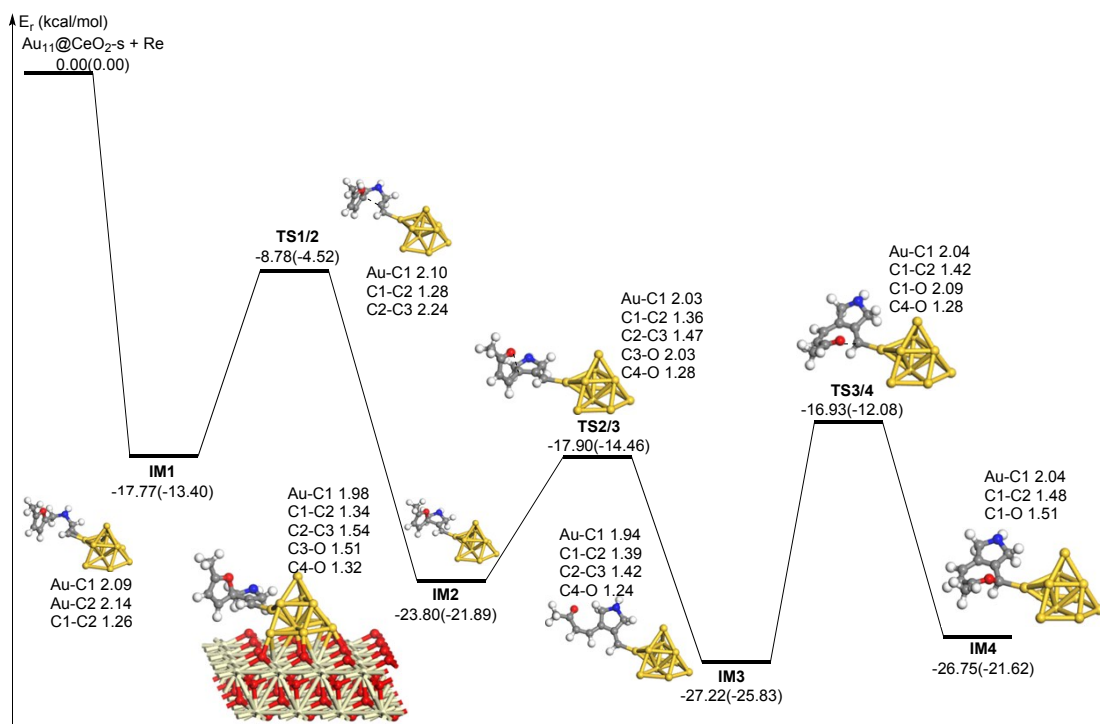
**Figure S7.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the interface-edge site of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-v}$ .



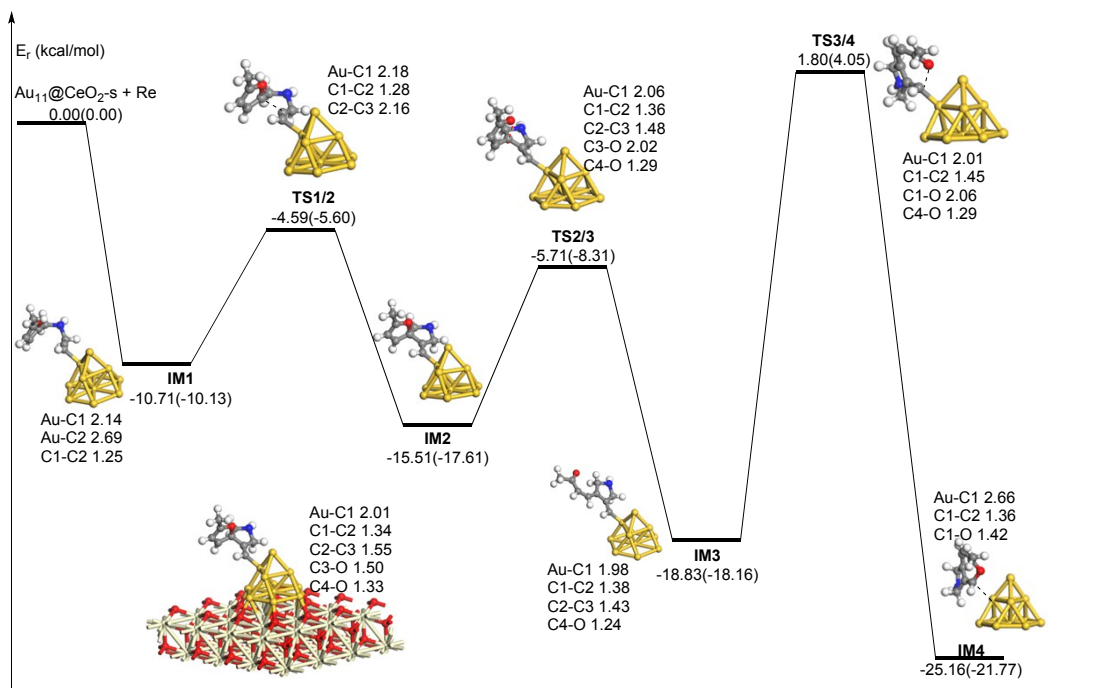
**Figure S8.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the interface-corner and edge sites of  $\text{Au}_{10}/\text{CeO}_2(111)\text{-v}$ .



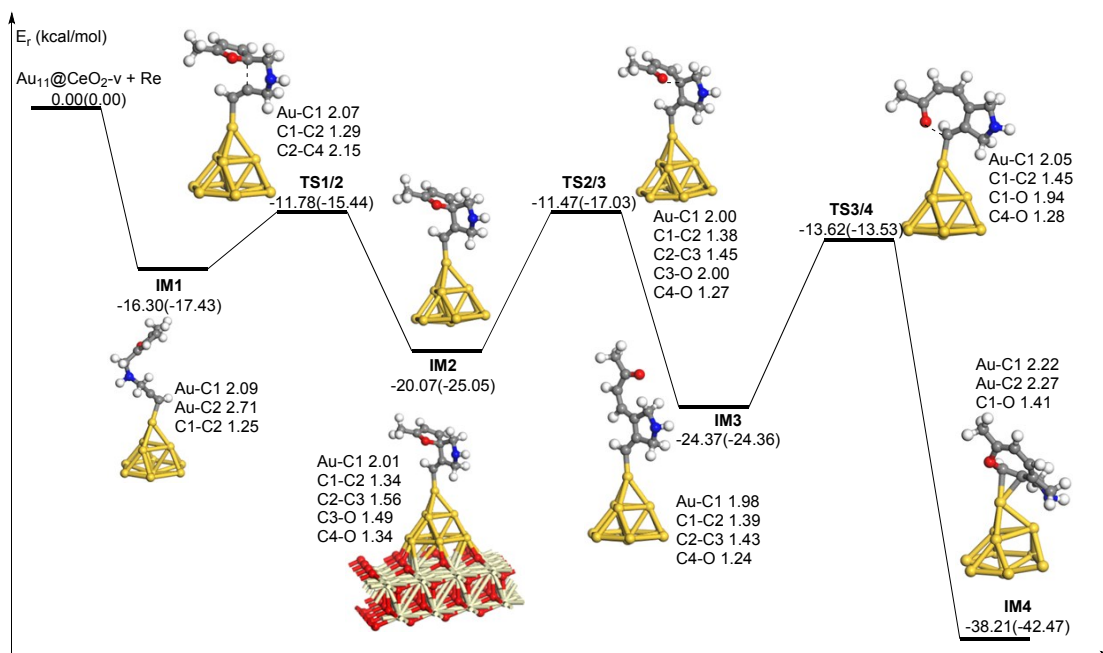
**Figure S9.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the atop site of  $\text{Au}_{11}/\text{CeO}_2(111)\text{-s}$ .



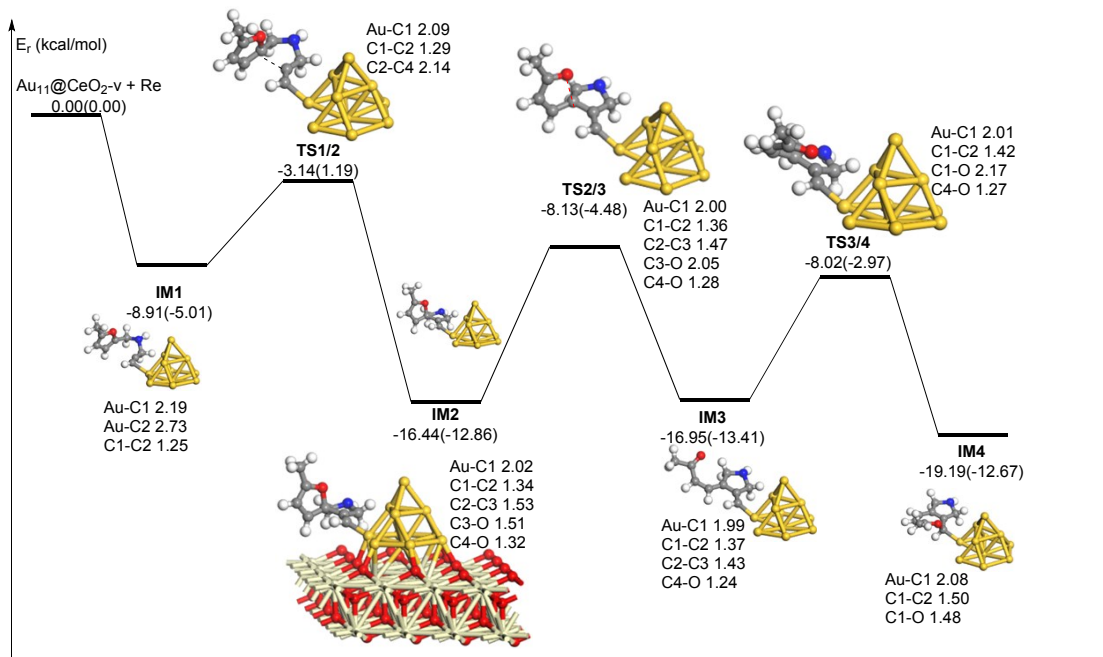
**Figure S10.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the interface site of  $\text{Au}_{11}/\text{CeO}_2(111)\text{-s}$ .



**Figure S11.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the edge site of  $\text{Au}_{11}/\text{CeO}_2(111)\text{-s}$ .

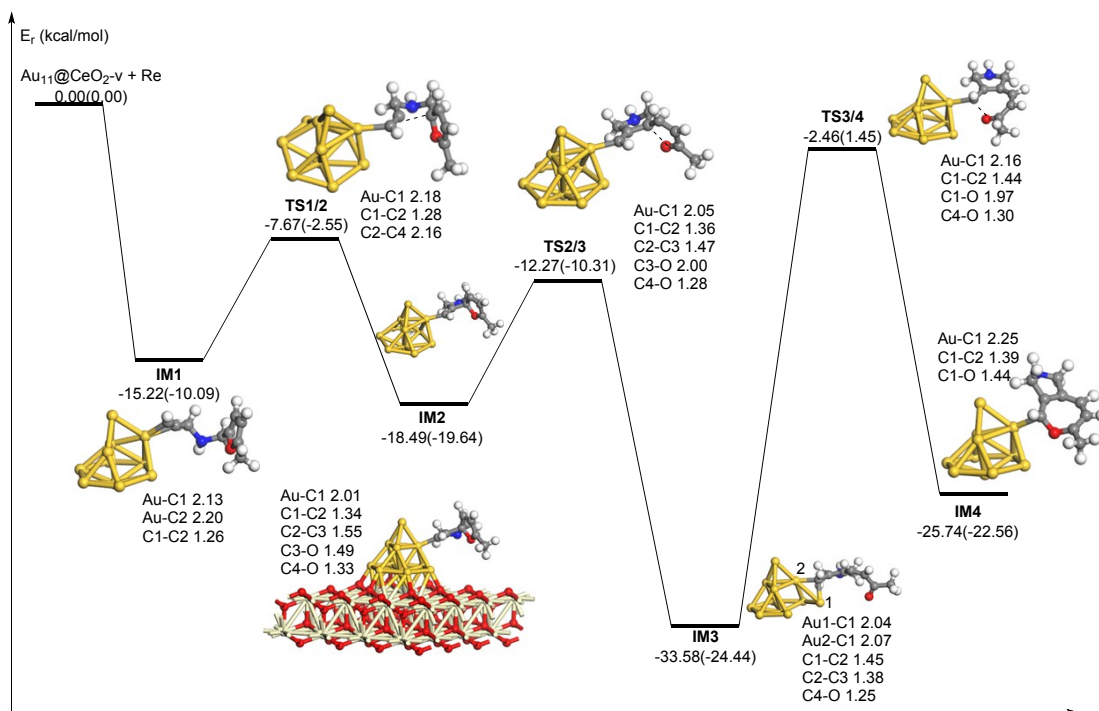


**Figure S12.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the atop site of  $\text{Au}_{11}/\text{CeO}_2(111)\text{-v}$ .

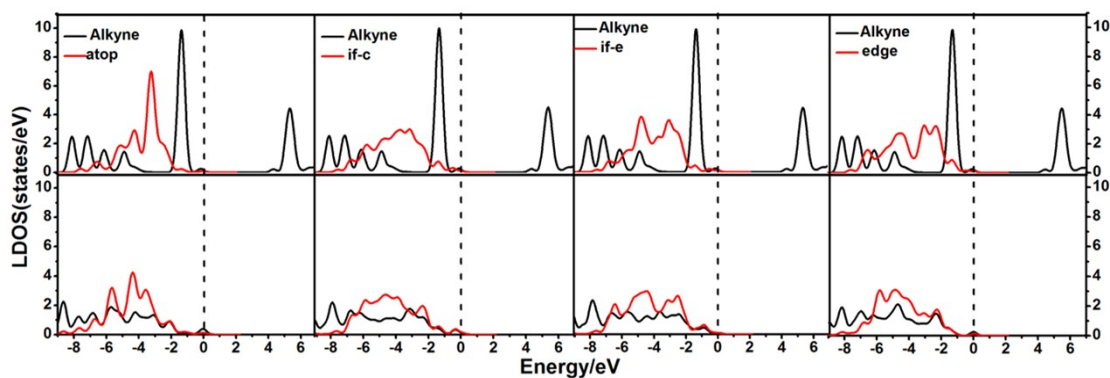


**Figure S13.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the interface site of  $\text{Au}_{11}/\text{CeO}_2(111)\text{-v}$ .

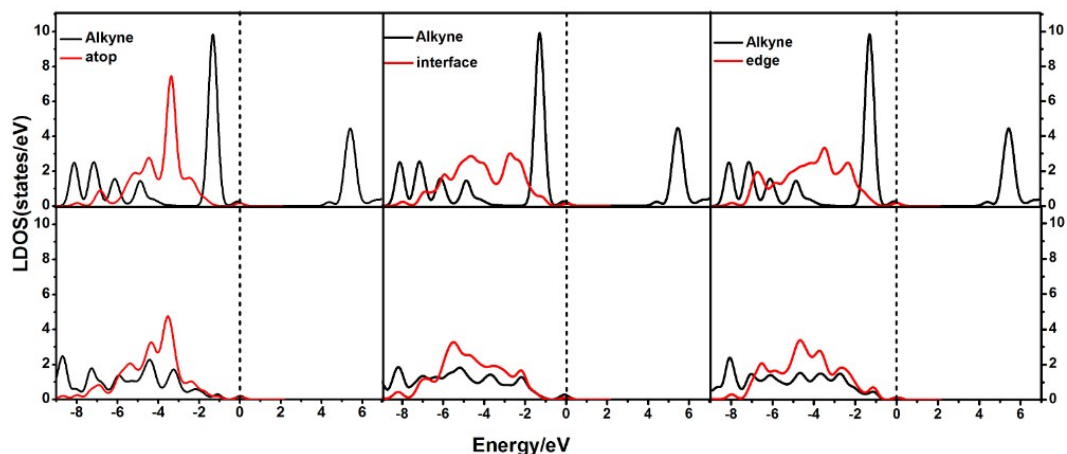




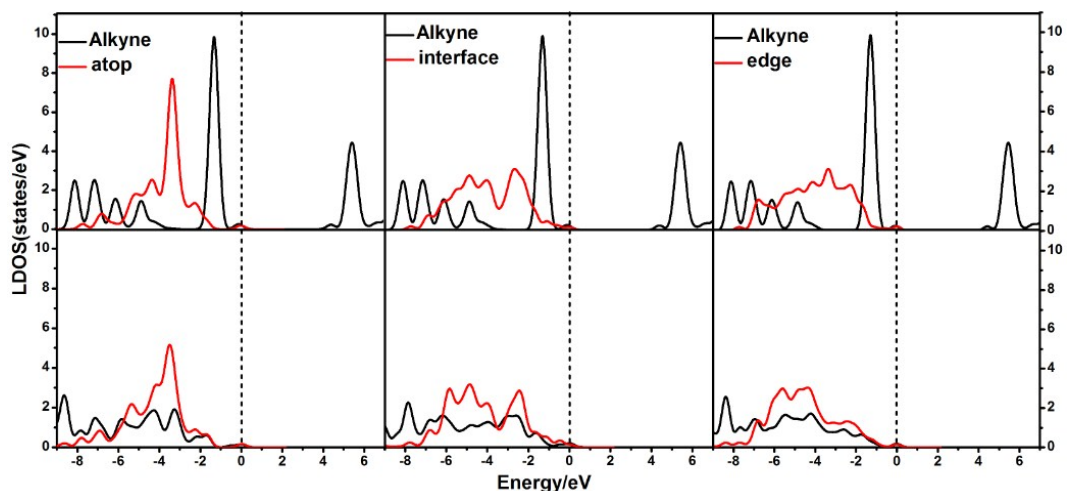
**Figure S14.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the edge site of  $\text{Au}_{11}/\text{CeO}_2(111)\text{-v}$ .



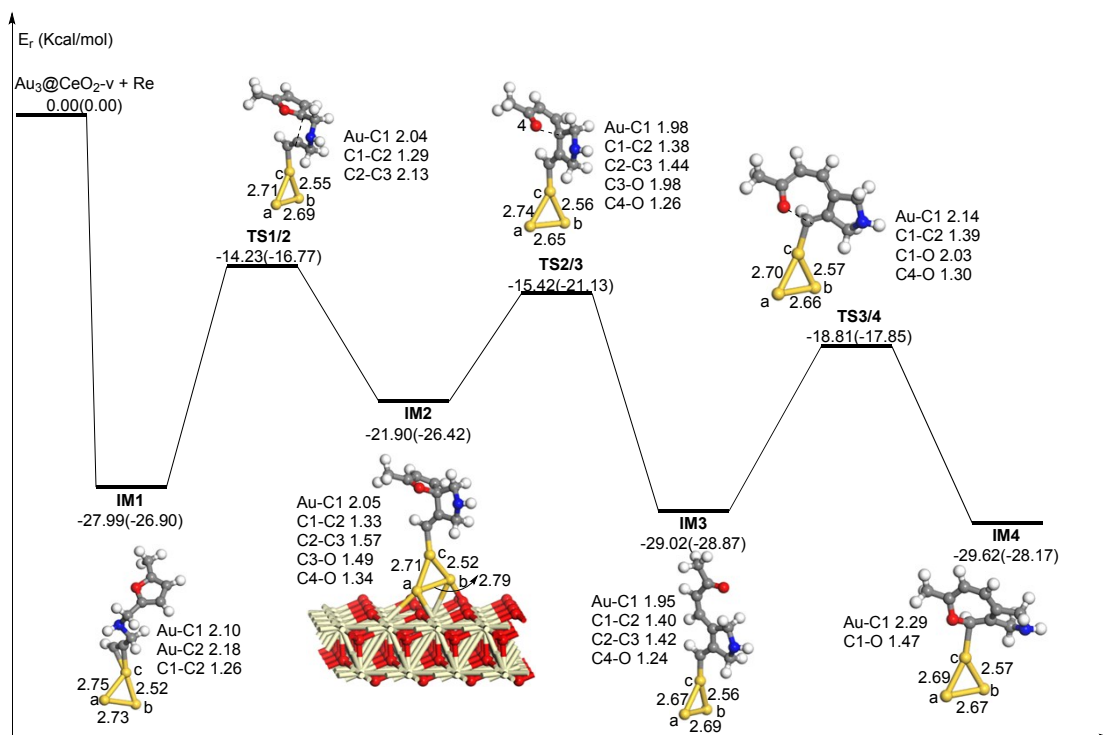
**Figure S15.** The local density of states (LDOS) projected on triplet bonds on the  $\text{Au}_{10}/\text{CeO}_2(111)\text{-sto}$  for the adsorption of the substrate at the different sites, together with the d-projected LDOS of the  $\text{Au}_{10}/\text{CeO}_2(111)\text{-v}$  of the Au atom at different sites. Free C-C triplet bond and  $\text{CeO}_2(111)$  (top), adsorption (bottom). The Fermi level is set to zero. (if-c=interface-corner, if-e=interface-edge)



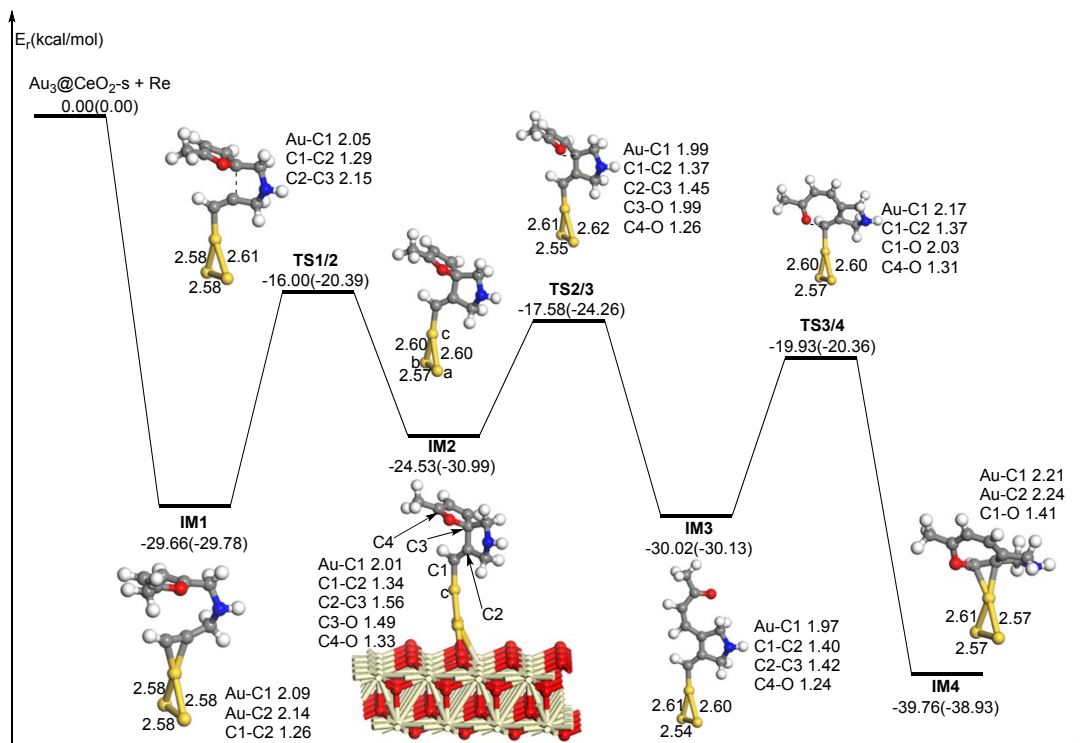
**Figure S16.** The local density of states (LDOS) projected on triplet bonds on the  $\text{Au}_{11}/\text{CeO}_2(111)$  for the adsorption of the substrate at the different sites, together with the d-projected LDOS of the  $\text{Au}_{11}/\text{CeO}_2(111)$ -s of the Au atom at different sites. Free C-C triplet bond and  $\text{CeO}_2(111)$  (top), adsorption (bottom). The Fermi level is set to zero.



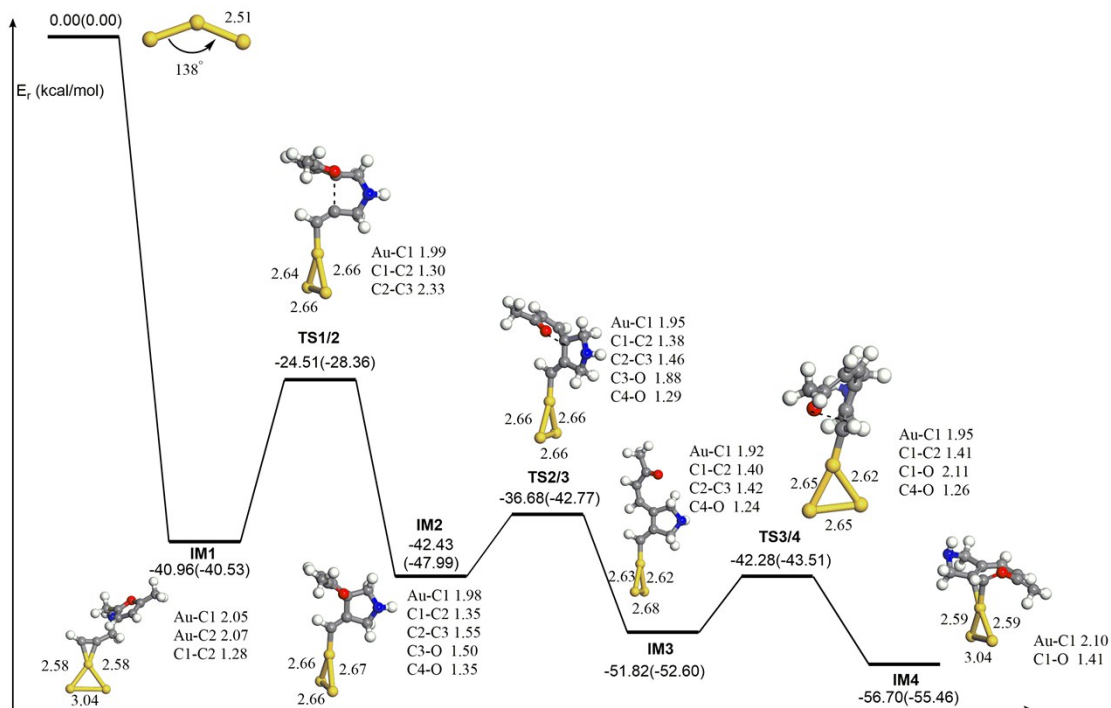
**Figure S17.** The local density of states (LDOS) projected on triplet bonds on the  $\text{Au}_{11}/\text{CeO}_2(111)$  for the adsorption of the substrate at the different sites, together with the d-projected LDOS of the  $\text{Au}_{11}/\text{CeO}_2(111)$ -v of the Au atom at different sites. Free C-C triplet bond and  $\text{CeO}_2(111)$  (top), adsorption (bottom). The Fermi level is set to zero.



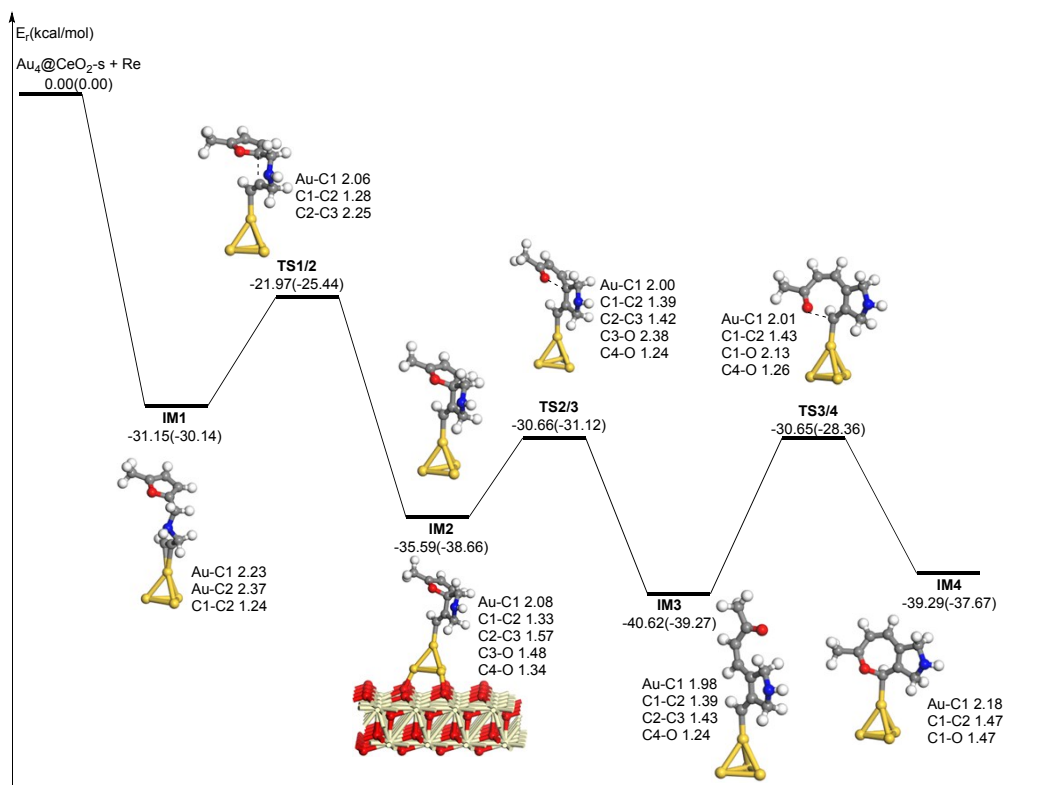
**Figure S18.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by  $\text{Au}_3$  cluster on the stoichiometric  $\text{CeO}_2(111)(\text{Au}_3/\text{CeO}_2(111)\text{-s})$ .



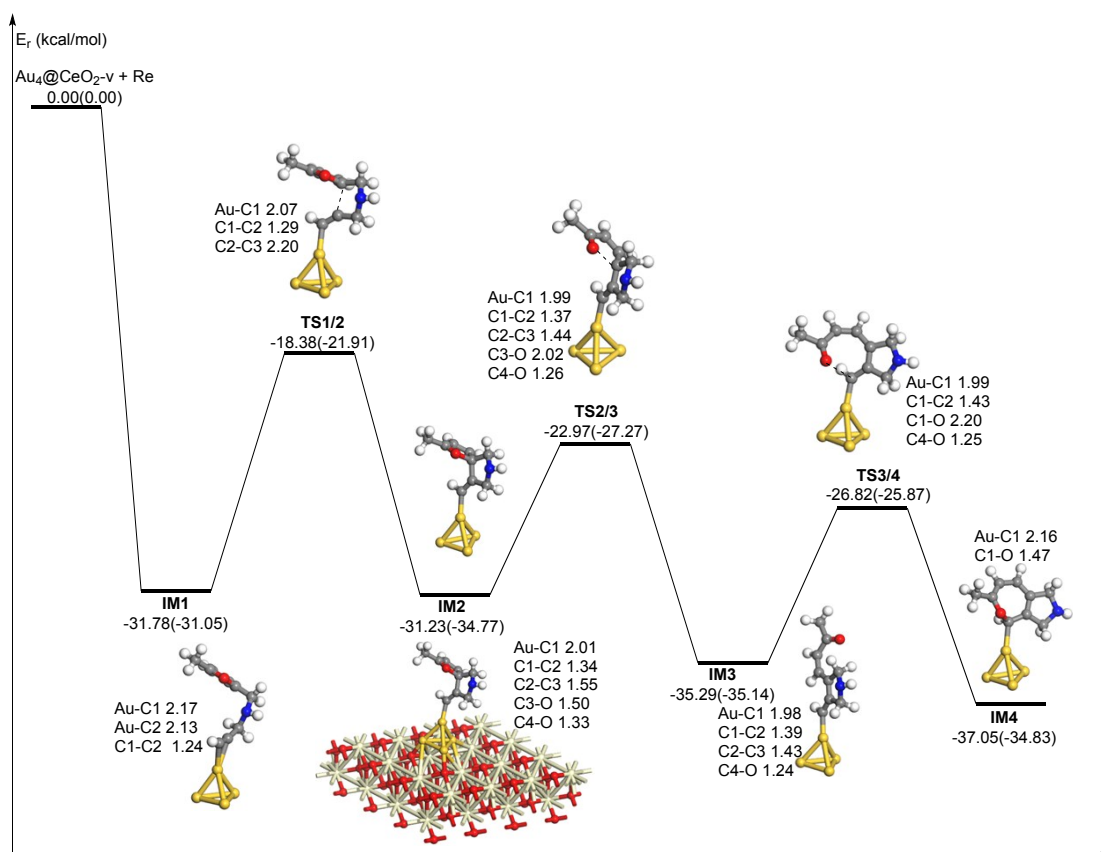
**Figure S19.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by  $\text{Au}_3$  cluster on the vacancy  $\text{CeO}_2(111)(\text{Au}_3/\text{CeO}_2(111)\text{-v})$ .



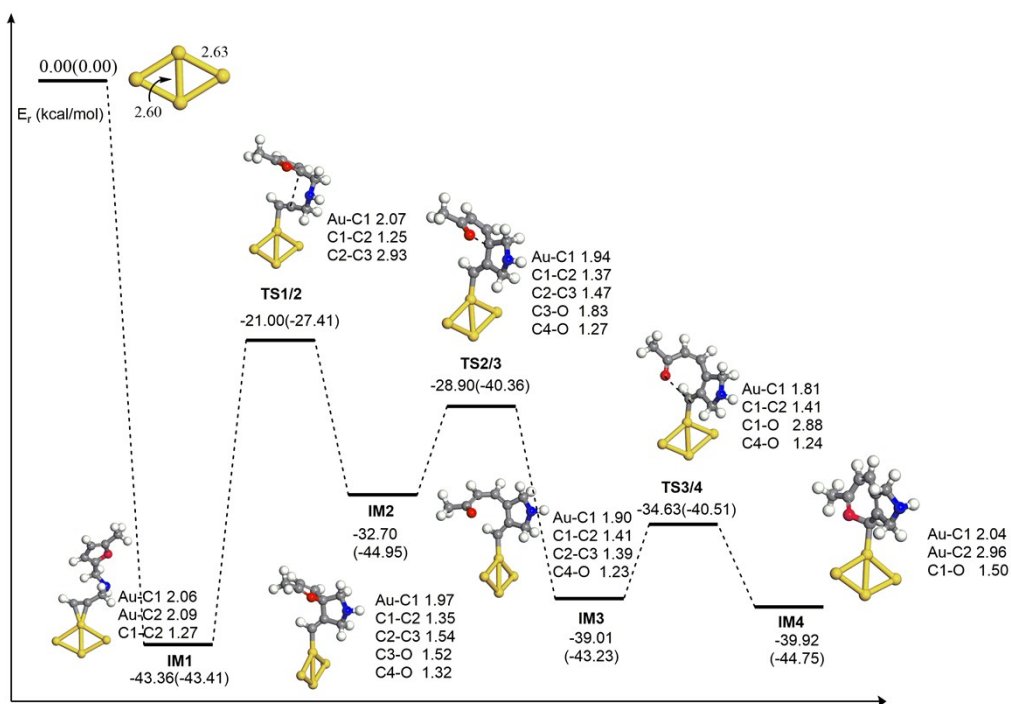
**Figure S20.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by the free  $\text{Au}_3$  cluster.



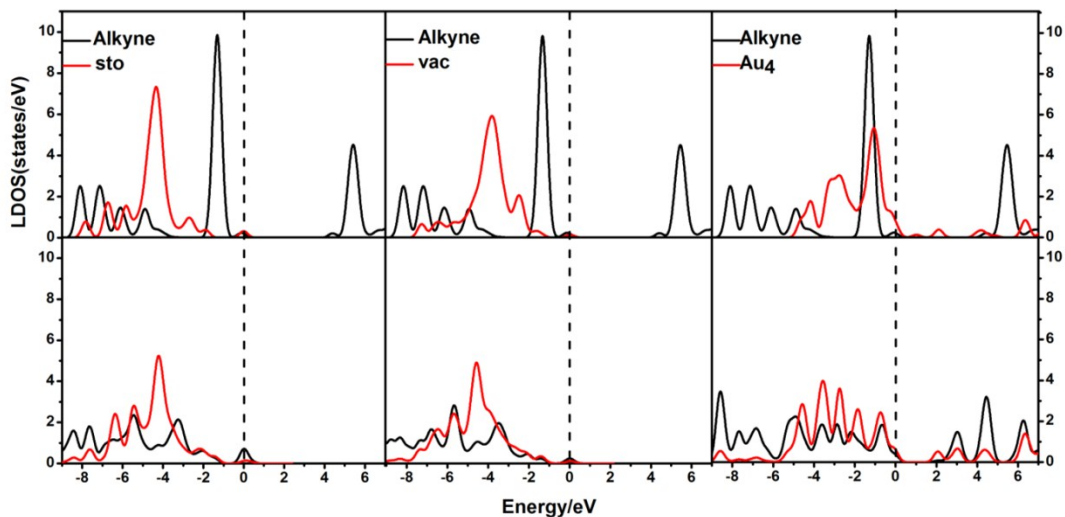
**Figure S21.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by  $\text{Au}_4/\text{CeO}_2(111)\text{-s}$ .



**Figure S22.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by  $\text{Au}_4/\text{CeO}_2(111)\text{-v}$ .



**Figure S23.** The optimized structures and energy profile of cycloisomerisation of  $\omega$ -alkynylfuran catalyzed by free  $\text{Au}_4$  cluster.

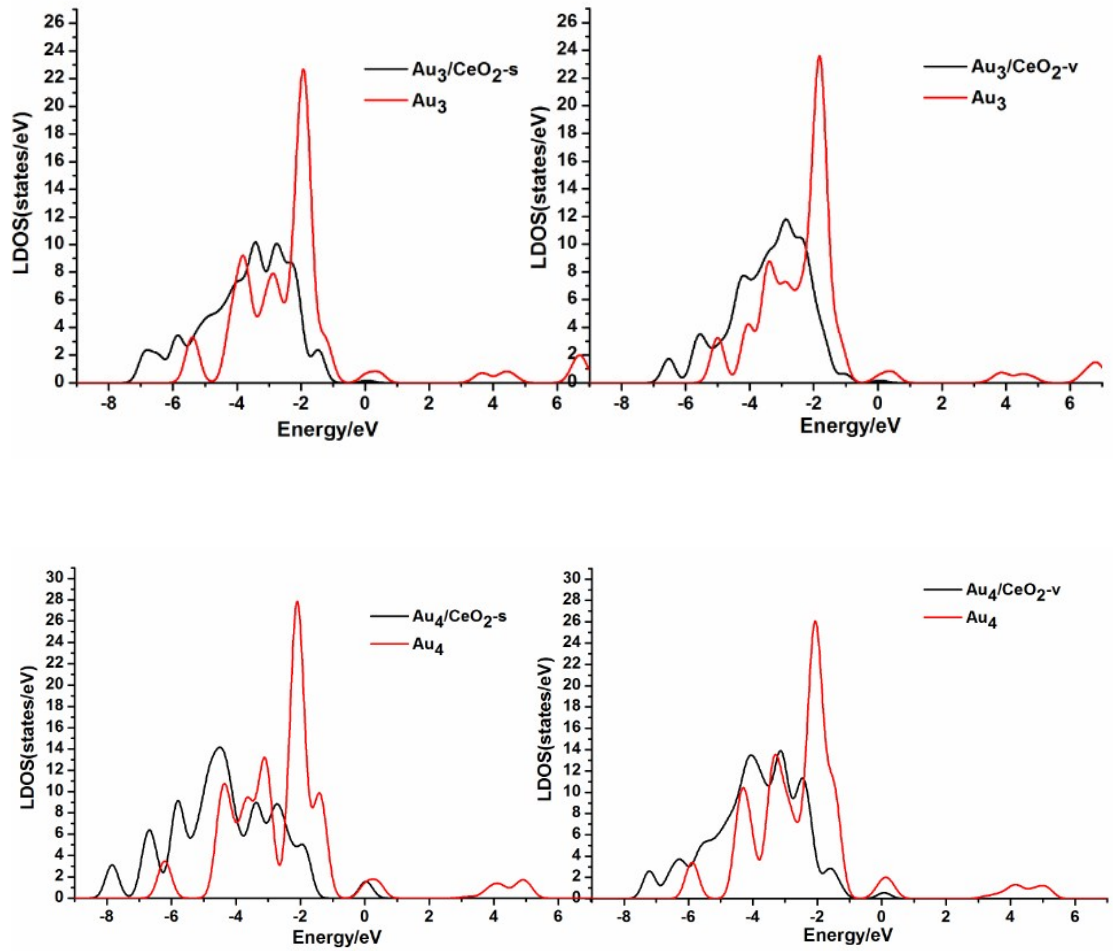


**Figure S24.** The local density of states (LDOS) projected on triplet bonds on the  $\text{Au}_4/\text{CeO}_2(111)$  for the adsorption of the substrate at the different sites, together with the d-projected LDOS of the  $\text{Au}_4/\text{CeO}_2(111)$  (stoichiometry and vacancy) of the Au atom at different sites. Free C-C triplet bond and  $\text{CeO}_2(111)$  (top), adsorption (bottom). The Fermi level is set to zero.

**Table S1.** Adsorption energy ( $\Delta E_{\text{ads}}$ , kcal/mol) of the  $\omega$ -alkynylfuran cycloisomerisation catalyzed by the  $\text{Au}_n$  and  $\text{Au}_n/\text{CeO}_2$ .

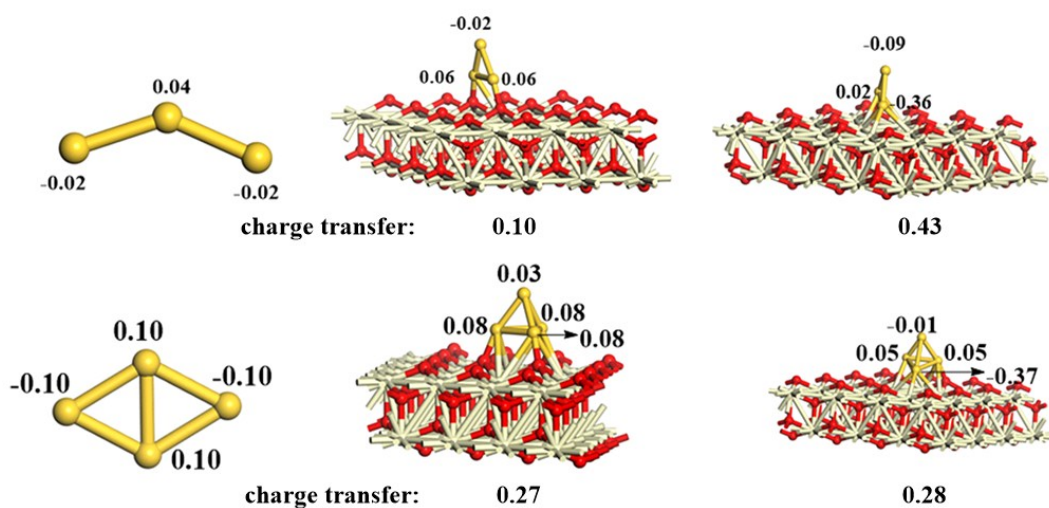
Catalysts	$\Delta E_{\text{ads}}$	
	initial	<sup>a</sup> treated
$\text{Au}_3@ \text{CeO}_2(111)\text{-s}$	-29.66	-42.58
$\text{Au}_3@ \text{CeO}_2(111)\text{-v}$	-27.99	-40.44
$\text{Au}_4@ \text{CeO}_2(111)\text{-s}$	-31.15	-49.87
$\text{Au}_4@ \text{CeO}_2(111)\text{-v}$	-31.78	-49.92
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-s}$ (atop)	-15.08	-33.20
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-s}$ (interface-corner)	-6.87	-21.19
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-s}$ (interface-edge)	-9.81	-14.51
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-s}$ (edge)	-9.16	-18.16
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-v}$ (atop)	-17.08	-36.31
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-v}$ (interface-corner)	-8.77	-21.08
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-v}$ (interface-edge)	-4.26	-28.63
$\text{Au}_{10}@ \text{CeO}_2(111)\text{-v}$ (edge)	-7.46	-17.17
$\text{Au}_{11}@ \text{CeO}_2(111)\text{-s}$ (atop)	-15.64	-27.38
$\text{Au}_{11}@ \text{CeO}_2(111)\text{-s}$ (interface)	-17.77	-34.43
$\text{Au}_{11}@ \text{CeO}_2(111)\text{-s}$ (edge)	-10.71	-19.97
$\text{Au}_{11}@ \text{CeO}_2(111)\text{-v}$ (atop)	-16.30	-28.56
$\text{Au}_{11}@ \text{CeO}_2(111)\text{-v}$ (interface)	-8.91	-19.10
$\text{Au}_{11}@ \text{CeO}_2(111)\text{-v}$ (edge)	-15.22	-34.26

<sup>a</sup> The corresponding values are obtained via removing the  $\text{CeO}_2(111)$  surface.



**Figure S25.** The local density of states (LDOS) projected on the d-projected LDOS of the Au<sub>3</sub>/Au<sub>4</sub>/CeO<sub>2</sub>(111) (stoichiometry and vacancy) of the Au atom and those of free Au<sub>3</sub> and Au<sub>4</sub> clusters which obtained via removing the CeO<sub>2</sub>(111) surface.





**Figure S26.** The Mulliken charges of Au atoms ( $|e|$ ) for the free Au<sub>3</sub>/Au<sub>4</sub> cluster (left), Au<sub>3</sub>/Au<sub>4</sub>/CeO<sub>2</sub> (stoichiometric, middle) and Au<sub>3</sub>/Au<sub>4</sub>/CeO<sub>2</sub> (stoichiometric, right).