

## Supplementary Information

### Efficient production of acrylic acid by dehydration of lactic acid over $\text{BaSO}_4$ with crystal defects

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Table S1. Catalytic performance and carbon balance of all the catalysts <sup>a</sup>

Catalyst	WHSV (h <sup>-1</sup> )	Conversion (%)	Selectivity (%)				Carbon deposits	Carbon balance (%)
			AA	AD	AcOH	PA		
Commercial	2.5	29.8	28.8	41.2	8.2	4.5	0.3	83.0
Co-pre	2.5	69.5	65.8	23.2	5.1	1.9	0.3	96.3
Water/ult	2.5	70.1	69.7	18.0	3.9	1.5	0.3	93.4
EtOH/ult-0h-500	2.5	77.6	78.9	12.7	3.2	2.2	0.3	97.3
EtOH/ult-0h-500	5.0	66.1	76.7	16.4	2.9	2.0	0.2	98.2
EtOH/ult-1h-500	5.0	56.2	70.5	18.2	3.0	3.0	0.2	94.9
EtOH/ult-2h-500	5.0	52.6	65.7	17.2	3.2	2.7	0.2	89.0
EtOH/ult-24h-500	5.0	51.7	58.6	15.2	2.7	1.4	0.2	78.1
EtOH/ult-0h-350	5.0	72.9	66.4	17.4	3.1	1.0	0.3	88.2
EtOH/ult-0h-600	5.0	45.8	59.2	20.5	3.5	2.3	0.2	85.7
EtOH/ult-0h-700	5.0	28.2	59.4	21.3	6.6	3.8	0.1	91.2

<sup>a</sup> Conditions: reaction temperature 350 °C, feed flow rate: 1.2 mL/h, LA feedstock: 20 wt% in water, carrier gas N<sub>2</sub>: 30 mL/min.

Table S2. Catalytic performance at different reaction temperatures<sup>a</sup>

Reaction temperature (°C)	Conversion (%)	Selectivity (%)				Carbon balance (%)
		AA	AD	AcOH	PA	
300	51.6	33.5	33.6	1.7	1.2	70.1
350	77.6	78.9	12.7	3.2	2.2	97.3
400	93.4	67.7	18.0	5.6	4.8	96.4

<sup>a</sup> Conditions: EtOH/ult-0h-500, WHSV = 2.5 h<sup>-1</sup>, feed flow rate: 1.2 mL/h, LA feedstock: 20 wt% in water, carrier gas N<sub>2</sub>: 30 mL/min, TOS = 4-9 h.

Table S3. Catalytic performance at different WHSV<sup>a</sup>

WHSV (h <sup>-1</sup> )	Conversion (%)	Selectivity (%)				Carbon balance (%)
		AA	AD	AcOH	PA	
0.8	97.5	77.7	13.5	1.2	1.0	94.0
1.3	88.9	78.8	14.6	3.5	1.9	99.3
2.5	77.6	78.9	12.7	3.2	2.2	97.3
5.0	66.1	76.7	16.4	2.9	2.0	98.2

<sup>a</sup> Conditions: EtOH/ult-0h-500, reaction temperature 350 °C, feed flow rate: 1.2 mL/h, LA feedstock: 20 wt% in water, carrier gas N<sub>2</sub>: 30 mL/min, TOS = 4-9 h.

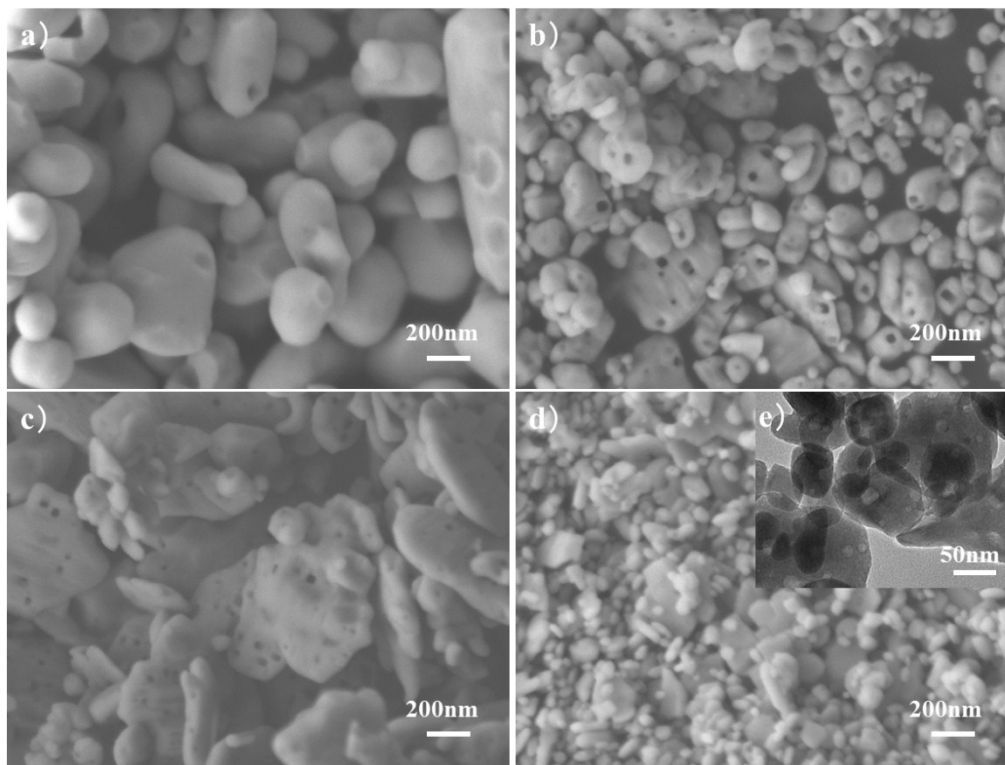


Figure S1. SEM images of BaSO<sub>4</sub> catalysts prepared with different methods  
(a) Commercial, (b) Co-pre, (c) Water/ult, (d) EtOH/ult-0h-500, and (e) TEM image of EtOH/ult-0h-500

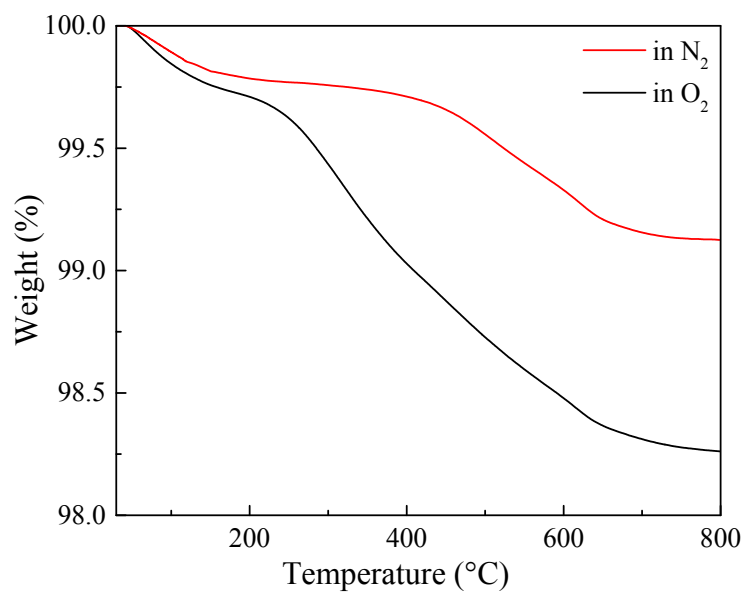


Figure S2. TGA profiles of EtOH/ult-350 in N<sub>2</sub> and O<sub>2</sub> atmosphere

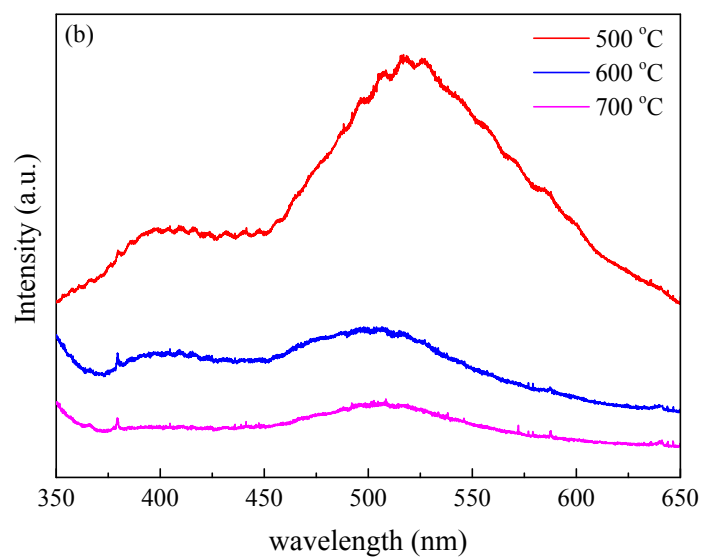
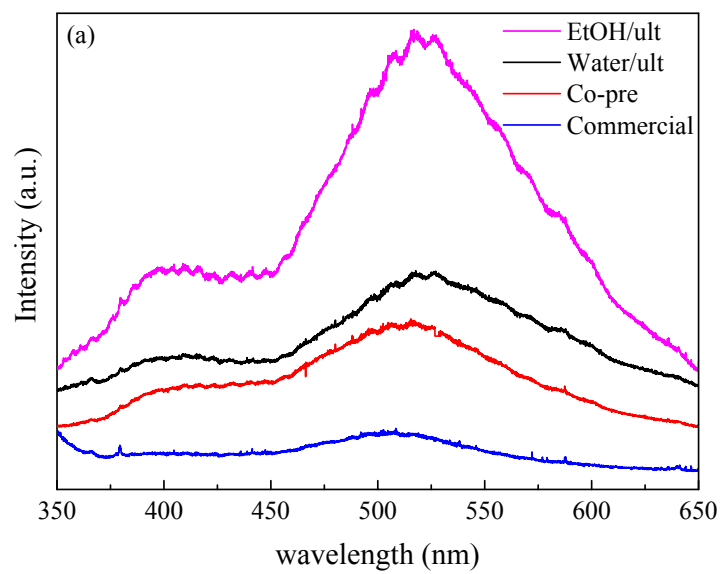


Figure S3. Photoluminescence (PL) spectra of EtOH/ult (a) prepared with different methods and (b) calcined at different temperatures

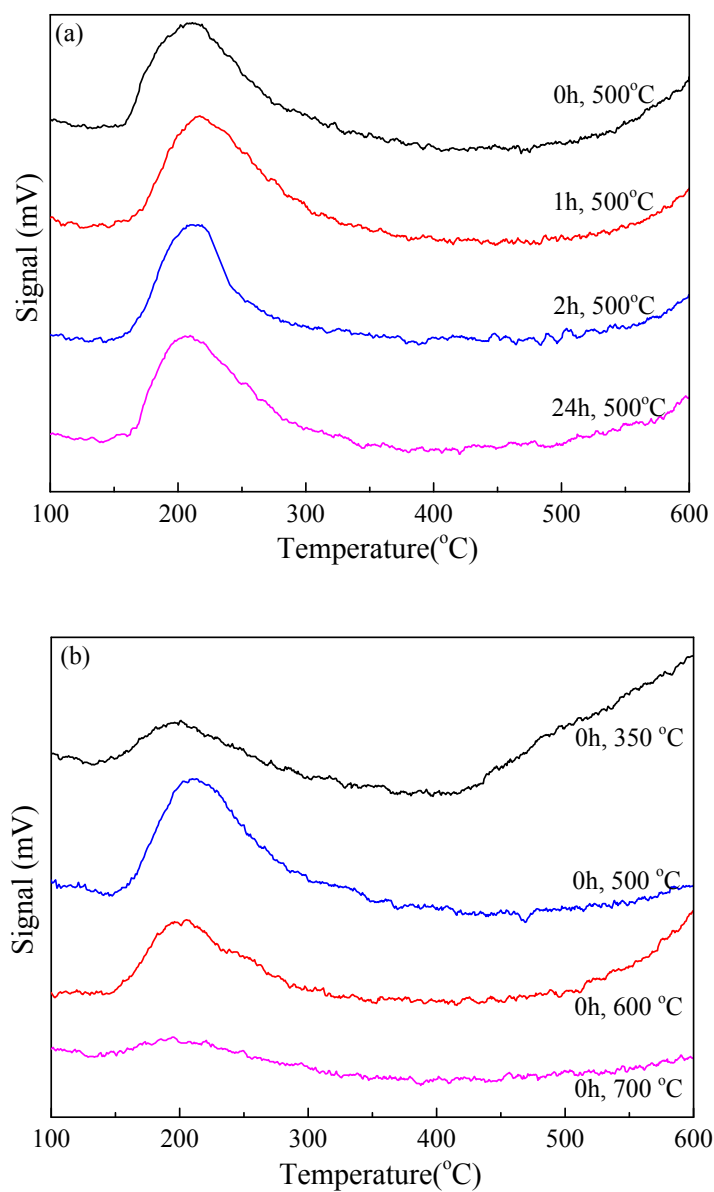


Figure S4. NH<sub>3</sub>-TPD profiles of BaSO<sub>4</sub> catalysts (a) with different aging time and (b) calcined at different temperature



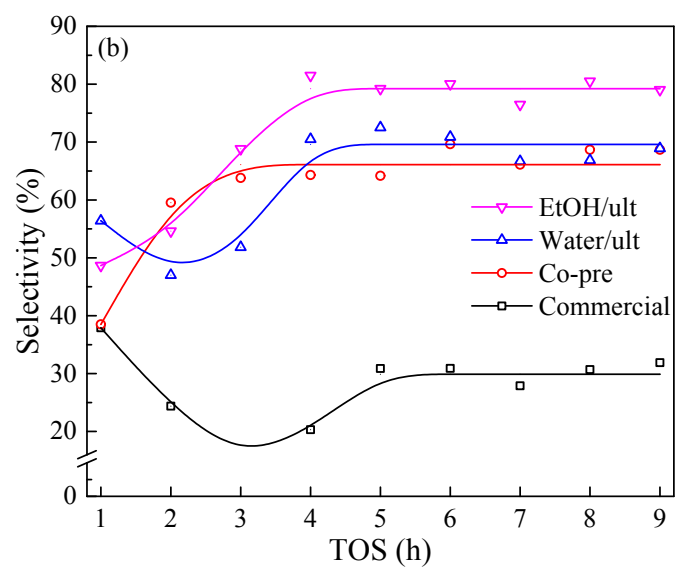
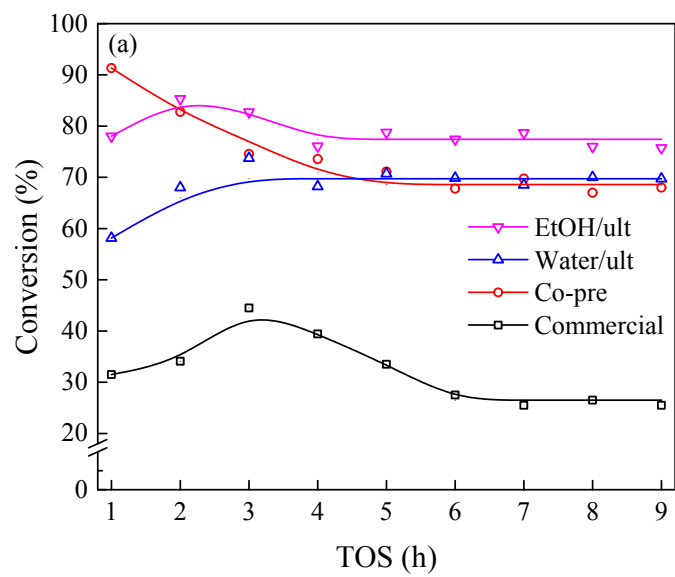


Figure S5. Performance of BaSO<sub>4</sub> catalysts prepared with different methods.

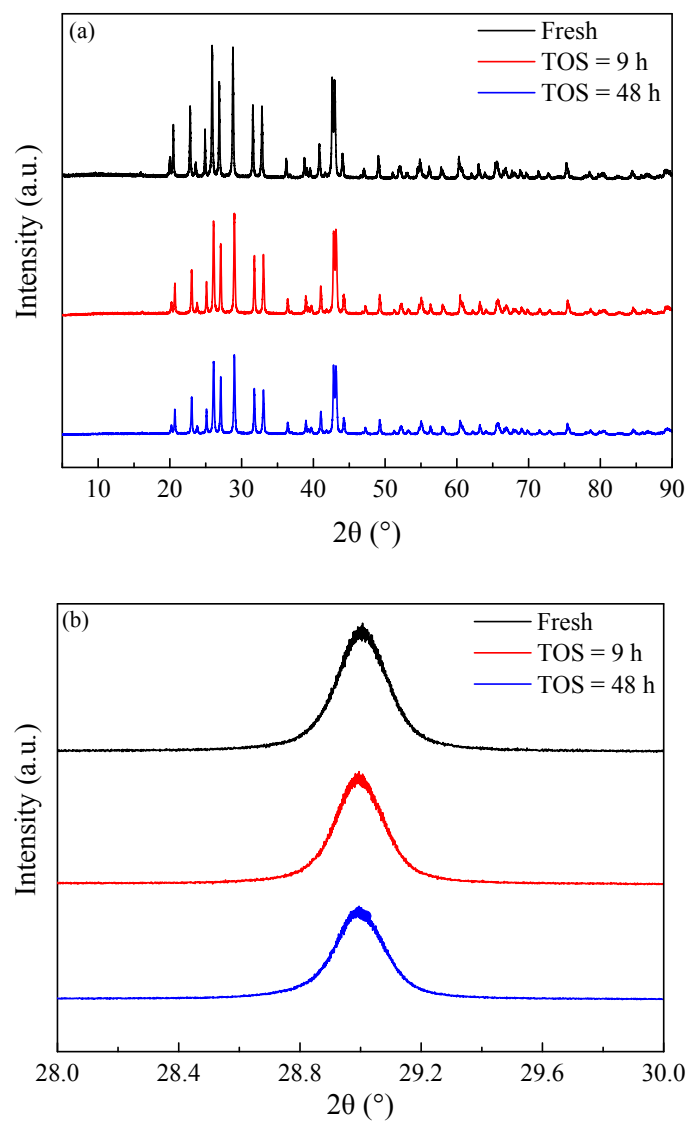


Figure S6. (a) XRD patterns and (b) enlarged (121) diffraction peaks of the fresh and spent EtOH/ult-0h-500 catalysts.