In-situ Growth of Binder-free CNTs@Ni-Co-S Nanosheet Core/Shell Hybrids on Ni Mesh for High Energy Density

Asymmetric Supercapacitors

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Fig. S1. (a) TEM image of the CNTs grown on Ni mesh. (b) FESEM overview images for the CNTs@Ni-Co-S composites on Ni mesh. (c-d) TEM images for Ni-Co-S nanosheets.



Fig. S2. The XRD patterns for Ni mesh, Ni@CNTs and Ni@CNTs@Ni-Co-S.



Fig. S3. (a-c) SEM images of the CNTs grown on carbon cloth. (b) TEM image for the CNTs on carbon cloth.



Fig. S4. Nyquist plots of (a) Ni@CNTs@Ni-Co-S and Ni@Ni-Co-S (inset: EIS spectrum in high frequency), (b) Ni@CNTs@Ni-Co-S//CC@CNTs asymmetric supercapacitors.

The Nyquist plots of Ni@CNTs@Ni-Co-S and Ni@Ni-Co-S are illustrated in Fig. S4. As shown in Fig. S4a, the axis intercepts in the high frequency range of the Ni@CNTs@Ni-Co-S is much smaller than the Ni@Ni-Co-S, indicating

Ni@CNTs@Ni-Co-S has a much smaller internal resistance (Rs). In addition, it is obvious that the semicircle of the Ni@CNTs@Ni-Co-S is much smaller than the Ni@Ni-Co-S, indicating Ni@CNTs@Ni-Co-S has a much lower interfacial charge-transfer resistance (Rct). Furthermore, Ni@CNTs@Ni-Co-S exhibits a slightly more vertical line in the low frequency range, suggesting that Ni@CNTs@Ni-Co-S has a lower Warburg resistance (described as diffusive impedance of ions) than Ni@Ni-Co-S. All the evidence show that the "core/shell" design of CNTs@Ni-Co-S composites realize a lower resistance than Ni-Co-S.

Fig. S4b illustrates the impedance data of Ni@CNTs@Ni-Co-S//CC@CNTs asymmetric supercapacitors. It can be found that the device has a low Rs ($\approx 0.6 \Omega$), relatively high Rct and low Warburg resistance.

Ni-Co-S based electrode	Potential range	Specific capacitance	Rate capability	Ref.	
CNTs@Ni-Co-S nanosheet	-0.2 V ~ 0.6 V	$-0.2 \ V \sim 0.6 \ V \qquad 222 \ mAh \ g^{-1} \qquad 193 \ mAh \ g^{-1}$		this work	
core/shell arrays	(vs. SCE)	at 4 A g ⁻¹	at 50 A g ⁻¹	uns work	
	0V~0.565V	180 mAh g ⁻¹	139 mAh g ⁻¹ (77.3%)	1	
urchin-like $N_1Co_2S_4$	(vs. Hg/HgO)	at 1 A g ⁻¹	at 20 A g ⁻¹		
NiCo ₂ S ₄ nanosheets on	0V~0.5V	202 mAh g ⁻¹	106 mAh g ⁻¹ (52.4%)	2	
graphene	(vs. Ag/ AgCl) at 3 A g ⁻¹		at 20 A g ⁻¹	2	
NiCo ₂ S ₄ porous nanotubes	-0.1V~0.5V	-0.1V~0.5V 152 mAh g ⁻¹ 76 mAh g vs. Hg/HgO) at 0.2 A g ⁻¹ at 5 A		3	
	(vs. Hg/HgO)				
	0V~0.55V	7 366 mAh g ⁻¹ 248 mAh g ⁻¹ (6			
$N_1Co_2S_4$ nanotube arrays	(vs. Hg/HgO)	at 5 mA cm ⁻²	at 150 mA cm ⁻²	4	
	-0.2V~0.6V	197 mAh g ⁻¹	178 mAh g ⁻¹ (90.6%)		
N1-Co-S nanosheet arrays	(vs. Ag/ AgCl)	at 5 A g ⁻¹	at 100 A g ⁻¹	5	
CoNi ₂ S ₄ /graphene	0V~0.38V	212 mAh g ⁻¹	110 mAh g ⁻¹ (52.1%)	6	
nanocomposite	(vs. SCE)	at 1 A g ⁻¹	at 20 A g ⁻¹	0	
CoNi ₂ S ₄ nanosheet arrays	0V~0.45V	363 mAh g ⁻¹	284 mAh g ⁻¹ (78.1%)) 7	
	(vs. SCE)	at 5 mA cm ⁻²	at 50 mA cm ⁻²		
	0V~0.45V	0V~0.45V 302 mAh g ⁻¹ 147 r			
NI-Co sulfide nanowires	(vs. Ag/ AgCl)	at 2.5 mA cm ⁻²	at 30 mA cm^{-2}	0	
NixCo _{3-x} S ₄ hollow	0V~0.5V	124 mAh g ⁻¹	81 mAh g ⁻¹ (65.4%)	0	
nanoprisms	(vs. SCE)	at 1 A g ⁻¹	at 20 A g ⁻¹	,	
core-shell NiCo ₂ S ₄	0V~0.5V	271 mAh g ⁻¹	215 mAh g ⁻¹ (79.4%)	10	
nanostructures	(vs. Hg/HgO)	at 1 mA cm ⁻²	at 20 mA cm ⁻²		
carbon@NiCo2S4 nanorods	0V~0.45V	182 mAh g ⁻¹	158 mAh g ⁻¹ (86.7%)		
	(vs. Ag/ AgCl)	(vs. Ag/ AgCl) at 1 A g ⁻¹		11	
Ni-Co-S ball-in-ball	-0.1V~0.55V	158 mAh g ⁻¹	108 mAh g ⁻¹ (68.1%)	12	
hollow spheres	(vs. SCE)	at 1 A g ⁻¹	at 20 A g ⁻¹	12	
carbon-NiCo ₂ S ₄ nanosheet	-0.2V~0.8V	368 mAh g ⁻¹	146 mAh g ⁻¹ (39.6%)	13	
arrays	(vs. SCE)	at 2 mA cm ⁻²	at 200 mA cm ⁻²	15	
NiCo ₂ S ₄ mesoporous	0V~0.5V	103 mAh g ⁻¹	86 mAh g ⁻¹ (83.3%)	14	
nanosheets	(vs. Hg/HgO)	at 1 A g ⁻¹	at 20 A g ⁻¹	11	
NiCo ₂ S ₄ nanoparticles on	-0.2V~0.4V	190 mAh g ⁻¹	129 mAh g ⁻¹ (67.9%)	15	
graphene	(vs. Ag/ AgCl)	at 1 A g ⁻¹	at 40 A g ⁻¹		
NiCo ₂ S ₄ flaky arrays	-0.1V~0.5V	284 mAh g ⁻¹	145 mAh g ⁻¹ (51.1%)	16	
	(vs. SCE)	vs. SCE) $at 1 A g^{-1}$ at		10	
NiCo ₂ S ₄ /Ni(OH) ₂ core-shell	-0.2V~0.6V	338 mAh g ⁻¹	200 mAh g ⁻¹ (59.3%)	17	
nanotube arrays	(vs. Hg/HgO)	at 1 mA cm ⁻²	at 20 mA cm ⁻²	± /	
hollow Ni _x Co _{9-x} S ₈	0V~0.45V	176 mAh g ⁻¹	73 mAh g ⁻¹ (41.3%)) 18	
urchins@N-doped carbon	(vs. Ag/ AgCl)	at 2 A g ⁻¹	at 8 A g ⁻¹		

Table S1. Electrochemical properties for nickel cobalt sulfide-based supercapacitors reported in recent years.

Desitive electrode	Negative Highest		Maximum	Maximum	Dof	
rositive electrode	electrode	potential	energy density	power density	Kel.	
CNTs@Ni-Co-S	CNT	$1 \in \mathbf{V}$	49.2 Wh kg ⁻¹ (at	40 kW kg ⁻¹ (at	this work	
core/shell arrays	CNIS	1.0 V	800 W kg ⁻¹)	18.9 Wh kg ⁻¹)		
NiCo ₂ S ₄ nanotube	reduced	1.6 V	31.5 Wh kg ⁻¹ (at	2348.5 W kg ⁻¹	4	
arrays	grapnene oxide(RGO)		156.6 W kg ⁻¹)	(at 16.6 Wh kg ⁻¹)		
Ni-Co-S nanosheet	porous	1.0.17	60 Wh kg ⁻¹ (at	28.8 kW kg ⁻¹ (at	5	
arrays	graphene film	1.8 V	1.8 kW kg ⁻¹)	33 Wh kg ⁻¹)	5	
CoNi ₂ S ₄ nanosheet	active carbon	1 7 17	33.9 Wh kg ⁻¹ (at	2458 W kg ⁻¹ (at	7	
arrays	(AC)	1.7 V	409 W kg ⁻¹)	27.2 Wh kg ⁻¹)		
Ni–Co sulfide	10	1.0.17	25 Wh kg ⁻¹ (at	3.57 kW kg ⁻¹ (at	8	
nanowires	AC	1.8 V	447 W kg ⁻¹)	17.8 Wh kg ⁻¹)		
porous Ni-Co	DCO	1 (17	37.6 Wh kg ⁻¹ (at	23.25 kW kg ⁻¹	19	
sulphides	RGO	1.6 V	775 W kg ⁻¹)	(at 17.7 Wh kg ⁻¹)		
core-shell NiCo ₂ S ₄	1	1 (17	22.8 Wh kg ⁻¹ (at	2.47 kW kg ⁻¹ (at	10	
nanostructures	porous carbon	1.6 V	160 W kg ⁻¹)	10.6 Wh kg ⁻¹)		
2D porous Ni-Co	10	1.0.17	41.4 Wh kg ⁻¹ (at	4.8 kW kg ⁻¹ (at	20	
Sulfide	AC	1.8 V	414 W kg ⁻¹)	23.8 Wh kg ⁻¹)		
	FeOOH	1.6 V	45.9 Wh kg ⁻¹ (at	8.6 kW kg ⁻¹ (at	21	
$N_1Co_2S_4$ nanosheets	nanorods		1.7 kW kg ⁻¹)	19.9 Wh kg ⁻¹)		
	graphene/	1.6 V		10.01.001 1 ()	12	
N1-Co-S ball-in-ball	carbon		42.3 Wh kg^{-1} (at	10.2 kW kg ⁻¹ (at		
hollow spheres	spheres		476 W kg^{-1}	22.9 Wh kg^{-1})		
carbon-NiCo ₂ S ₄	10	1.0.17	68.82 Wh kg ⁻¹ (at	1.4 kW kg ⁻¹ (at	13	
nanosheet arrays	AC	1.8 V	47.83 W kg ⁻¹)	26.74 Wh kg ⁻¹)		
NiCo ₂ S ₄ mesoporous	1.0	1 (17	25.5 Wh kg ⁻¹ (at	8 kW kg ⁻¹ (at	14	
nanosheets	AC	1.6 V	334 W kg ⁻¹)	10.8 Wh kg ⁻¹)	14	
3D cauliflower-like			44.0 101 1 1 ()	161301 164		
NiCo ₂ S ₄	AC	1.6 V	44.8 Wh kg ⁻¹ (at	16 kW kg^{-1} (at	22	
architectures			401 W kg ⁻¹)	23.1 Wh kg^{-1}		
graphene@NiCo ₂ S ₄	1.0	1 7 17	68.5 Wh kg ⁻¹ (at	17 kW kg ⁻¹ (at	15	
nanoparticles	AC	1.7 V	850 W kg ⁻¹)	37.7 Wh kg ⁻¹)		
mesoporous NiCo ₂ S ₄	10	1 7 17	28.3 Wh kg ⁻¹ (at	9.8 kW kg ⁻¹ (at	23	
nanoparticles	AC	1.5 V	245 W kg ⁻¹)	6.8 Wh kg ⁻¹)	23	

Table S2. Energy densities and power densities for nickel cobalt sulfide-based ASCs in recent reports.

Current density (A g ⁻¹)	1	2	4	8	10	15	20	30	40	50
$E = I \int_{t=0}^{t=t} V(t) dt$ (Wh kg ⁻¹)	46.5	42.1	38.9	35.3	34.2	29.8	26.4	21.1	17.4	15.9
E=0.5C _s $\Delta V^{2}/3.6$ (Wh kg ⁻¹)	49.2	45.6	41.9	37.9	34.4	31.0	28.9	22.0	19.6	18.9

Table S3. Energy densities comparison calculated via two different methods

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