## Electronic Supporting Information

## Reconstructing ZnO quantum dot assembled tubular structures from nanotubes within graphene matrix *via* ongoing pulverization towards high-

## performance lithium storage<sup>+</sup>

Zihua Li,<sup>a,c</sup> Xiao Yu,<sup>a,c</sup> Yong Liu,<sup>\*,a,c</sup> Wenxia Zhao,<sup>b</sup> Hao Zhang,<sup>b</sup> Ruimei Xu,<sup>b</sup> Donghai Wang<sup>c</sup> and Hui Shen<sup>c</sup>

<sup>a</sup>School of Materials Science and Engineering, State Key Laboratory of Optoelectronic Materials and Technologies, Sun Yat-sen University, Guangzhou 510275, China <sup>b</sup>Instrumental Analysis & Research Center, Sun Yat-sen University, Guangzhou 510275, China <sup>c</sup>Institute for Solar Energy System, Guangdong Provincial Key Laboratory of Photovoltaics Technologies, Sun Yat-sen University, Guangzhou 510275, China *E-mail: liuyong7@mail.sysu.edu.cn* 



**Fig. S1** FESEM images and corresponding histogram showing the morphological evolution of ZnO nanotubes by adding different concentration of GO nanosheets in reaction solution. (a) and (b) 0.17 g L<sup>-1</sup>, (c) 0.12 g L<sup>-1</sup>, (d) 0.06 g L<sup>-1</sup>, (e) 0 g L<sup>-1</sup>, and (f) The histogram showing relations of average diameters of ZnO nanotubes summarized in above FESEM images and concentration of GO nanosheets in the reaction.



**Fig. S2** FESEM images of (a) as-synthesized graphene oxide nanosheets and (b) graphene nanosheets obtained by after reduction heat treatments at 700 °C for 2 h under 5% hydrogen in nitrogen atmosphere.



Fig. S3 XRD patterns of post-annealed graphene-wrapped ZnO nanotubes and bare ZnO nanotubes.



Fig. S4 The TGA curve of the post-annealed graphene-wrapped ZnO.



Fig. S5  $N_2$  adsorption-desorption isotherm of the post-annealed graphene-wrapped ZnO, bare ZnO and re-aggregated graphene.



**Fig. S6** (a) XRD pattern and (b) SAED of the graphene-wrapped ZnO electrode after 1000 discharge/charge cycles at a current density of 2000 mA  $g^{-1}$ .



**Fig. S7** FESEM images of graphene-wrapped ZnO electrodes after (a) 2, (b) 100, (c) 400 and (d) 700 discharge/charge cycles at a current density of 2000 mA  $g^{-1}$ .



Fig. S8 TEM image of the graphene-wrapped ZnO electrode after 100 discharge/charge cycles at a current density of 2000 mA  $g^{-1}$ .



Fig. S9 (a) The TEM image, (b) HRTEM image and (c) SAED of the graphene-wrapped ZnO electrode after 400 discharge/charge cycles at a current density of 2000 mA  $g^{-1}$ .



**Fig. S10** (a) The TEM image, (b) HRTEM image and (c) SAED of the graphene-wrapped ZnO electrode after 400 discharge/charge cycles at a current density of 2000 mA g<sup>-1</sup>.



**Fig. S11** TEM images of the graphene-wrapped ZnO quntom dots assembled tubular structure after completion of 1000 discharge/charge cycles at a current density of 2000 mA g<sup>-1</sup>. (a) TEM and (b) HRTEM images. (c) and (d) Magnified HRTEM images corresponded to region I and II outlined by dashed line in (b).



**Fig. S12** TEM images of the graphene-wrapped ZnO quntom dots assembled tubular structure after completion of 1000 discharge/charge cycles at a current density of 2000 mA g<sup>-1</sup>. (a) TEM and (b) HRTEM images. (c), (d) and (e) Magnified HRTEM images corresponded to region I, II and III outlined by dashed line in (b).



**Fig. S13** FESEM images the cycled bare ZnO electrode after (a) and (b) 2 discharge/charge cycles, (c) and 100 discharge/charge cycles, and (e) and (f) 1000 discharge/charge cycles at a current density of 2000 mA  $g^{-1}$ .



**Fig. S14** The equivalent circuit used to fit the Nyquist plots of the graphene-wrapped ZnO and bare ZnO electrodes, In this model,  $R_e$  represents the internal resistance of cells, and  $R_f$  and CPE<sub>1</sub> are associated with the resistance and constant phase element of SEI film, respectively.  $R_{ct}$  and CPE<sub>2</sub> depict the charge transfer resistance and constant phase element of the electrode/electrolyte interface, respectively. Meanwhile,  $Z_W$  is the Warburg impedance.

Materials	Morphology	Discharged capacity (mAh g <sup>-1</sup> )	Cycles	Current density (mA g <sup>-1</sup> )	Ref.
ZnO	Hierarchical flower-like nanospheres	381	30	493.5	28
ZnO	Ultralong mesoporous nanowires	392	50	98.7	29
ZnO	Ultrathin nanotubes	386	50	494	30
ZnO	Mesoporous nanosheets	421	100	20	31
ZnO	Dandelion-like nanorod arrays	310	40		32
Au–ZnO	Hierarchical flower-like nanostructures	392	50	120	33
Al- ZnO	Nanoparticles	418		50	34
ZnO-Ag–C	Porous microspheres	729	200	100	5
ZnO/Graphene	Nanocrystals	~300	25	50	17
Al-ZnO-graphene	Aerogel composite	490	100	100	6
ZnO/MWCNT	Nanotube Nanocomposite	460	100	197.4	35
ZnO/Ketjenblack	Porous structure	538.4	100	100	36
ZnO/Graphene	Nanotubes	1058 747 683 891	200 200 200 1000	100 200 500 2000	Our work

**Table S1.** Summary of the electrochemical performance of various ZnO-based anode materials for lithium-ion batteries.

**Table S2.** Kinetic parameters of graphene-wrapped ZnO and bare ZnO electrodes before and after different galvanostatic discharging/charging cycles.

Graphene-wrapped	$R_{ct}(\Omega)$	Bare ZnO electrodes	$R_{ct}(\Omega)$		
ZnO electrodes					
Before cycling	26.4	Before cycling	36.7		
After 2 cycles	28.3				
After 50 cycles	214.7				
After 100 cycles	245.9	After 100 cycles	301.4		
After 400 cycles	38.1				
After 700 cycles	33.1				
After 1000 cycles	13.9	After 1000 cycles	313.2		