

Supporting Information for

The “Electron Crystal” Behavior in Copper Chalcogenides Cu_2X ($\text{X}=\text{Se}, \text{S}$)

Yongxing Sun^{1,2,3}, Lili Xi^{1,3}, Jiong Yang^{1*}, Lihua Wu¹, Xun Shi³, Lidong Chen³, Jeffrey Snyder⁴, Jihui Yang⁵, and Wenqing Zhang^{1,2*}

¹*Materials Genome Institute, Shanghai University, Shanghai 200444, China*

²*School of Materials Science and Engineering, Shanghai University, Shanghai 200444, China*

³*State Key Laboratory of High Performance Ceramics and Superfine Microstructure, Shanghai Institute of Ceramics, Chinese Academy of Sciences, Shanghai 200050, China*

⁴*Department of Materials Science and Engineering, Northwestern University, Evanston, Illinois 60208, USA*

⁵*Material Science and Engineering Department, University of Washington, Seattle, WA 98195, USA*

This supporting information includes:

Fig. SI1. Band structures of S1, S2, and Gulay structures.

Fig. SI2. The pCOHP analysis for the Cu-Se bond in Cu_2Se with the antifluorite structure.

Fig. SI3. The band structure of S3 Cu_2Se with some Se vacancy.

Fig. SI4. Electronic band structures of $\text{Cu}_{1.92}\text{Se}$ in the low temperature S3 structure, with the Cu vacancy locating at different Cu sites.

Fig. SI5. Electronic band structures of Cu_2Se , $\text{Cu}_{1.96}\text{Se}$, $\text{Cu}_{1.93}\text{Se}$, and Cu_2Se with all Cu ions disordered.

Fig. SI6. Theoretical absolute electrical conductivities as a function of hole concentrations at 300K and 800K.

Fig. SI7. Wave functions at the valence band maxima for both LT and HT Cu_2Se .

Table SI1. Structural parameters and atomic coordinates for S1, S2, S3 and Gulay structures.

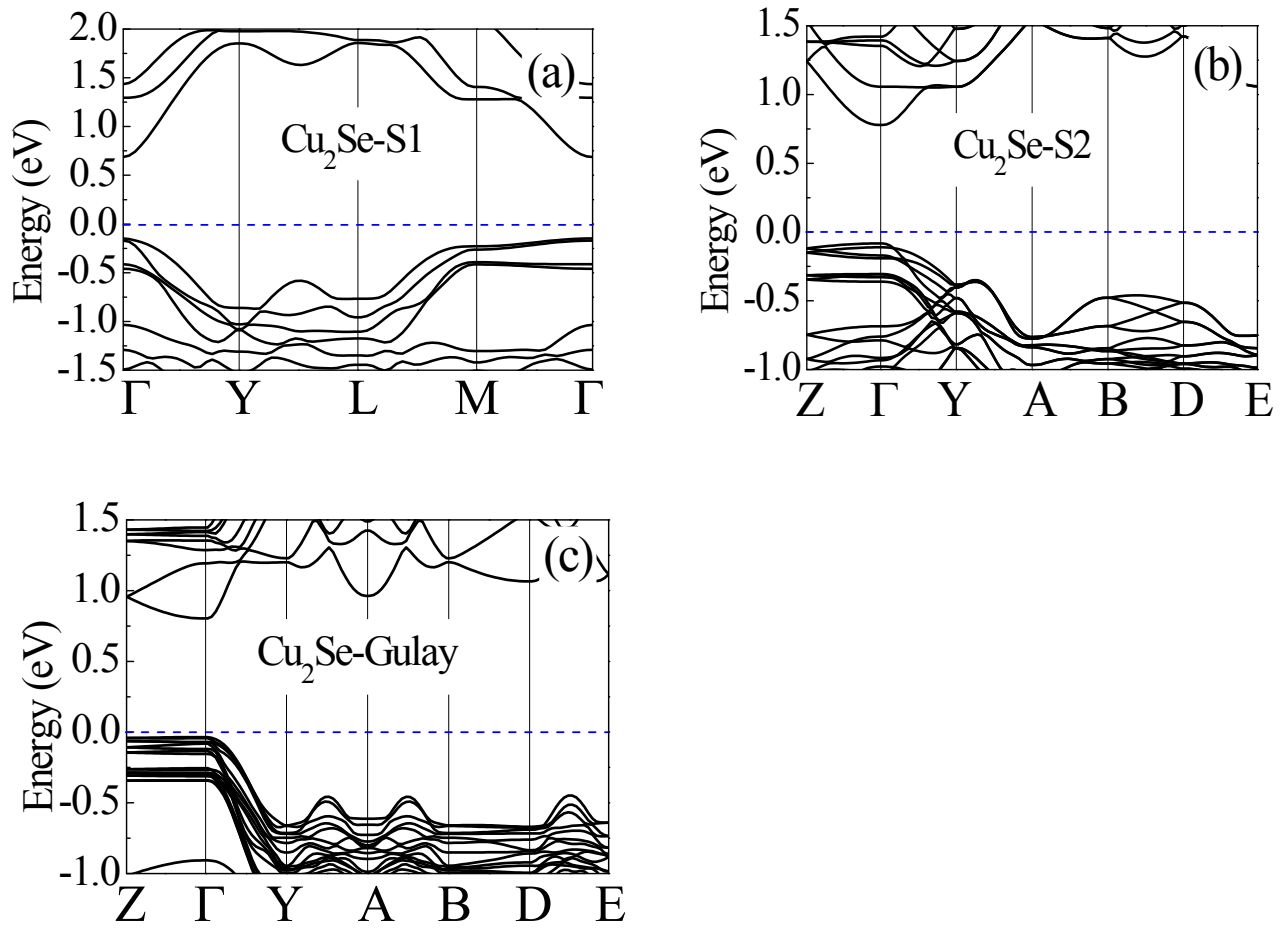


Figure S11. Electronic band structures for Cu_2Se in the LT (a) S1, (b) S2, and (c) Gulay structures.

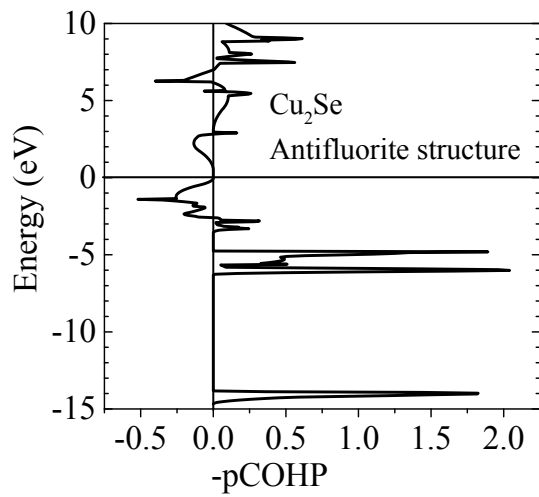


Figure S12. The pCOHP analysis for the Cu-Se bond in Cu_2Se with the antifluorite structure.

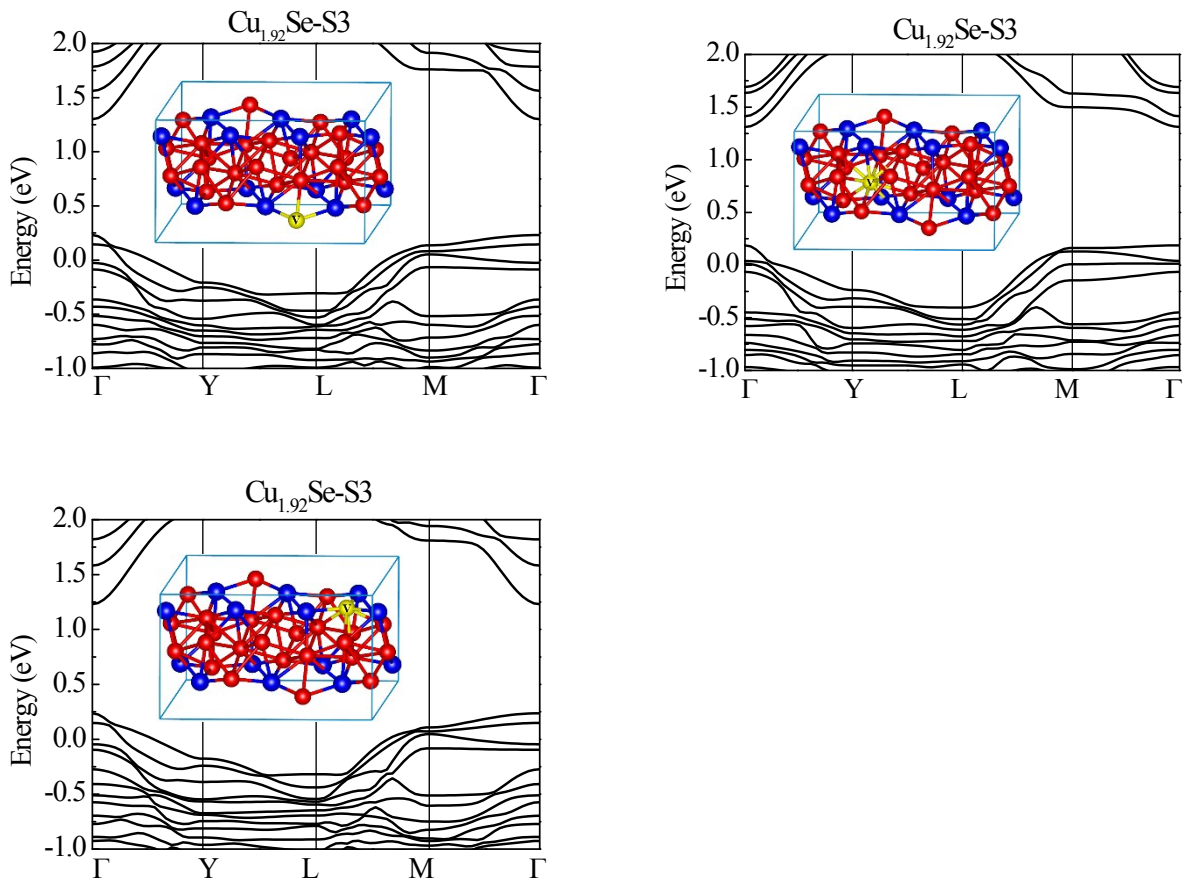


Figure SI3. Electronic band structures of $\text{Cu}_{1.92}\text{Se}$ in the low temperature S3 structure, with the Cu vacancy (denoted as “V”) locating at different Cu sites. The first panel is used in the main text.

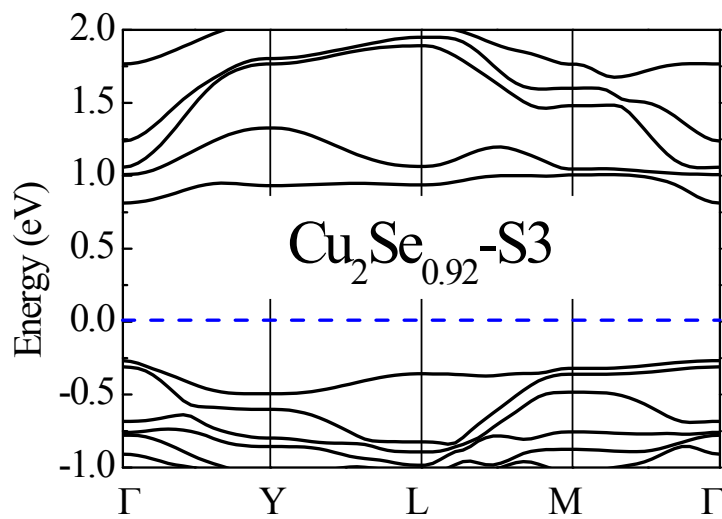


Figure SI4. The band structure of S3 Cu_2Se with Se vacancy.

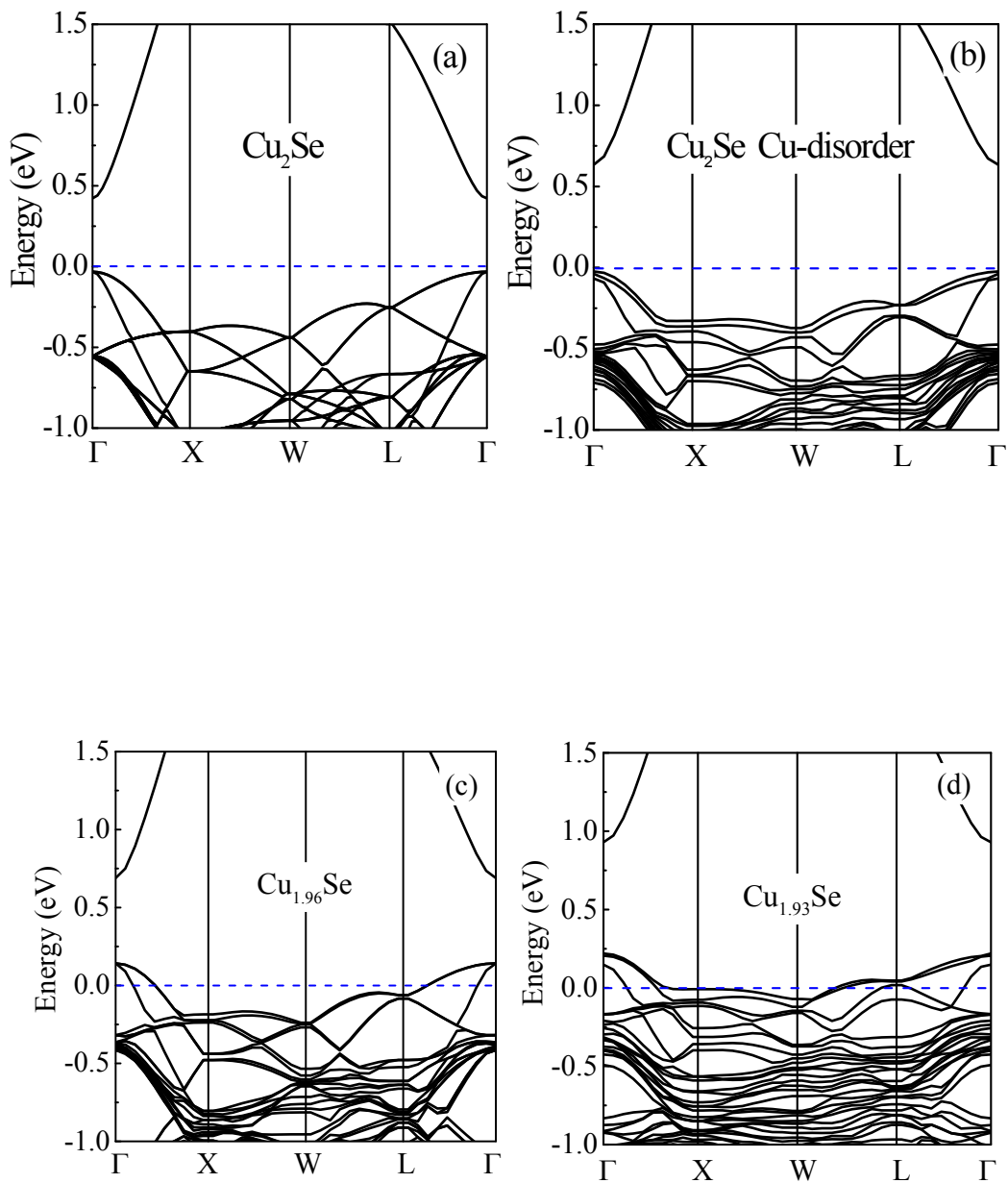


Figure SI5. Electronic band structures of (a) Cu_2Se , (b) Cu_2Se with all Cu ions disordered, (c) $\text{Cu}_{1.96}\text{Se}$ and (d) $\text{Cu}_{1.93}\text{Se}$. All the calculated structures are based on the high temperature antifluorite structure ($\text{Cu}_{54}\text{Se}_{27}$). Dotted lines denote the respective Fermi-levels.

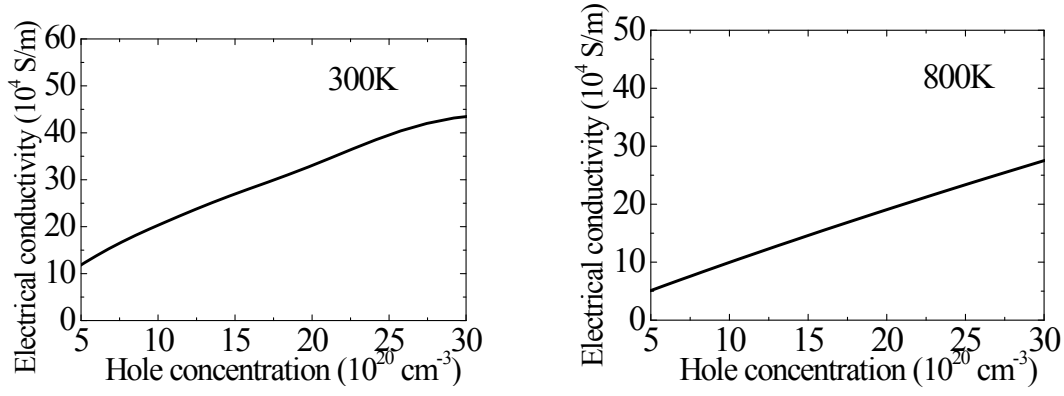


Figure S16. The theoretical absolute electrical conductivities as a function of hole concentrations at 300K and 800K.

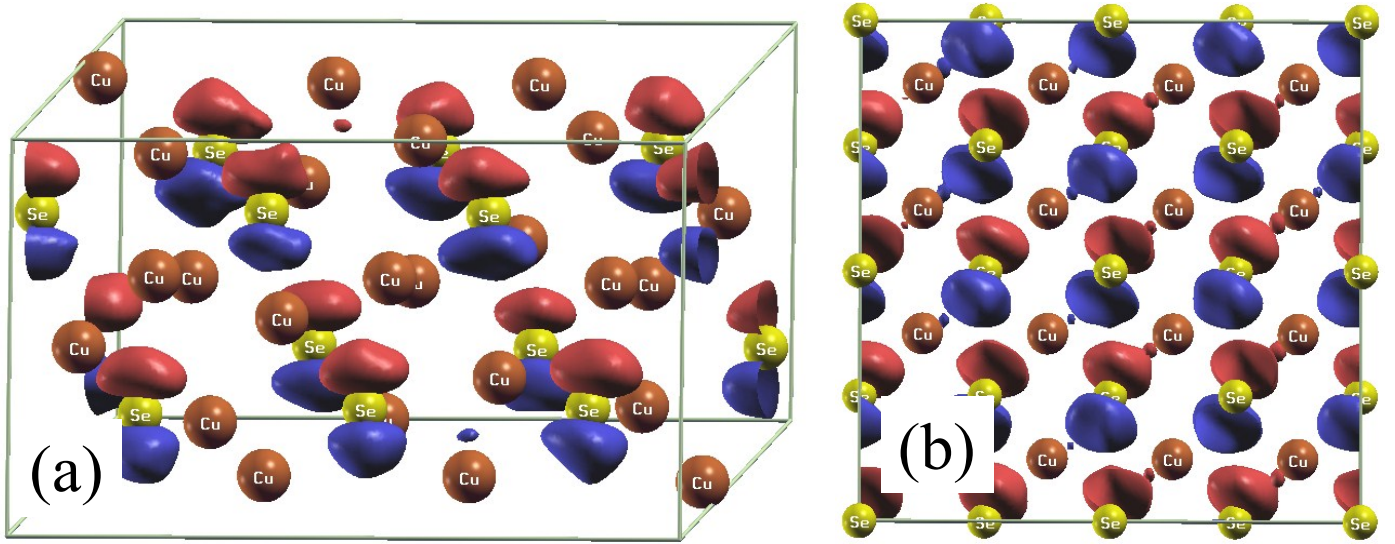


Figure S17. The electron wave functions at the valence band maxima for (a) the low temperature phase (S3 structure) and (b) the high temperature phase. The colors indicate the signs of the wave functions.

Table S11. Structural parameters and atomic coordinates for S1, S2, S3, and Gulay structures.

Structure	Space group	Lattice parameters					
		a (Å)	b (Å)	c (Å)	α (°)	β (°)	γ (°)
S1	P-1 (2)	7.101	7.122	7.407	81.227	80.503	119.878
S2	C2/c (15)	7.120	12.333	14.364	90.000	100.333	90.000
S3	P-1 (2)	7.078	12.434	7.428	90.236	108.736	90.038
Gulay	C2/c (15)	7.092	12.331	29.418	90.000	106.969	90.000

Structure	atom	x/a	y/b	z/c	Structure	atom	x/a	y/b	z/c
S1	Cu1	0.3881	0.6669	0.6818	Gulay	Cu1	0.2991	0.4477	0.3299
	Cu2	0.0590	0.9998	0.6862		Cu2	0.9849	0.8059	0.4211
	Cu3	0.7674	0.3325	0.8149		Cu3	0.6352	0.1202	0.3308
	Cu4	0.0543	0.6948	0.4336		Cu4	0.3284	0.4829	0.4515
	Cu5	0.5997	0.9200	0.4397		Cu5	0.0216	0.4301	0.3898
	Cu6	0.2823	0.3842	0.4348		Cu6	0.6546	0.1430	0.4236
	Se7	0.5800	0.6645	0.2203		Cu7	0.7179	0.5154	0.3589
	Se8	0.2420	0.0038	0.2172		Cu8	0.9625	0.7834	0.2935
	Se9	0.9008	0.3264	0.2111		Cu9	0.2478	0.8189	0.3582
S2	Cu1	0.7804	0.9174	0.5853	Cu10	0.5642	0.7583	0.3897	
	Cu2	0.2861	0.7493	0.5937	Cu11	0.9461	0.0442	0.3587	
	Cu3	0.3064	0.0829	0.6492	Cu12	0.3451	0.2074	0.3914	
	Cu4	0.0164	0.9036	0.4674	Se13	0.9894	0.1423	0.4450	
	Cu5	0.5779	0.7834	0.4669	Se14	0.6242	0.7801	0.3046	
	Cu6	0.6178	0.0638	0.4676	Se15	0.9628	0.4451	0.3048	
	Se7	0.3572	0.9171	0.3566	Se16	0.3244	0.8181	0.4448	
	Se8	0.3650	0.2473	0.3604	Se17	0.6648	0.4802	0.4475	
	Se9	0.8675	0.0858	0.3607	Se18	0.3059	0.1211	0.3024	
S3	Cu1	0.5426	0.8375	0.6577					
	Cu2	0.5604	0.5133	0.6880					
	Cu3	0.5823	0.1735	0.7514					
	Cu4	0.1653	0.8502	0.4367					
	Cu5	0.1269	0.4631	0.4277					
	Cu6	0.1156	0.2062	0.4231					
	Cu7	0.0527	0.9828	0.6906					
	Cu8	0.1064	0.6594	0.8356					
	Cu9	0.0606	0.3295	0.6835					
	Cu10	0.6569	0.9694	0.4458					
	Cu11	0.5834	0.6694	0.4526					
	Cu12	0.7224	0.3400	0.4252					
	Se13	0.7604	0.8424	0.2349					

Se14	0.7425	0.5066	0.2227
Se15	0.7339	0.1729	0.2094
Se16	0.2486	0.9968	0.2164
Se17	0.2438	0.6554	0.2172
Se18	0.2444	0.3240	0.2049
