

## Electronic Supplementary Information

### Calculations for the Grüneisen parameter in Cu<sub>8</sub>GeSe<sub>6</sub>:

The Grüneisen parameter is estimated by<sup>1</sup>

$$\kappa_L = A \frac{M_{avg} \Theta_D^3 \delta}{\gamma^2 n^{2/3} T} , \quad (1)$$

where  $\kappa_L$  is the lattice thermal conductivity,  $M_{avg}$  is the average mass per atom,  $\delta$  is the average volume per atom,  $n$  is the number of atoms in the primitive unit cell ( $n = 90$  for Cu<sub>8</sub>GeSe<sub>6</sub>), and  $A$  is a collection of physical constants ( $A = 3.1 \times 10^{-6}$  if  $\kappa$  is in W m<sup>-1</sup> K<sup>-1</sup>,  $M_{avg}$  is in amu, and  $\delta$  is in Angstroms).

**Table S1** Chemical composition of Cu<sub>8</sub>GeSe<sub>6</sub>, Cu<sub>8</sub>GeSe<sub>5.1</sub>Te<sub>0.9</sub> and Cu<sub>7.6</sub>Ag<sub>0.4</sub>GeSe<sub>5.1</sub>Te<sub>0.9</sub> measured by EDS at 10 randomly selected locations.

Nominal composition: Cu <sub>8</sub> GeSe <sub>6</sub>										
Actual composition: Cu <sub>7.95</sub> GeSe <sub>5.98</sub>										
elements	Atomic percent concentration (normalized to 100%)									
Cu	53.28	53.35	53.05	53.38	53.21	53.19	53.46	53.26	53.32	52.92
Ge	6.49	6.81	6.59	6.75	6.53	6.74	6.82	6.77	6.72	6.75
Se	40.24	40.02	40.36	39.87	40.26	40.08	39.72	39.96	39.96	40.29
Nominal composition: Cu <sub>8</sub> GeSe <sub>5.1</sub> Te <sub>0.9</sub>										
Actual composition: Cu <sub>7.86</sub> GeSe <sub>5.03</sub> Te <sub>0.88</sub>										
elements	Atomic percent concentration (normalized to 100%)									
Cu	52.99	54.19	53.74	53.56	52.12	53.48	52.98	53.23	53.11	53.09
Ge	6.94	6.70	6.98	7.00	6.60	6.62	6.82	6.60	6.59	6.83
Se	34.22	33.14	33.47	33.45	35.24	33.90	34.63	33.87	34.57	33.92
Te	5.85	5.98	5.81	5.99	6.04	6.01	5.98	6.30	5.73	6.15
Nominal composition: Cu <sub>7.6</sub> Ag <sub>0.4</sub> GeSe <sub>5.1</sub> Te <sub>0.9</sub>										
Actual composition: Cu <sub>7.45</sub> Ag <sub>0.37</sub> GeSe <sub>5.04</sub> Te <sub>0.86</sub>										
elements	Atomic percent concentration (normalized to 100%)									
Cu	50.33	50.44	50.30	50.39	50.87	50.58	50.63	50.72	50.70	50.66
Ag	2.65	2.55	2.44	2.43	2.49	2.50	2.51	2.58	2.48	2.51
Ge	6.78	6.71	6.86	6.75	6.73	6.86	6.73	6.84	6.68	6.92
Se	34.47	34.53	34.35	34.37	34.10	34.19	34.25	34.00	34.19	34.00
Te	5.78	5.78	5.80	5.82	5.81	5.86	5.88	5.87	5.96	5.92

**Table S2** Parameters used to fit the heat capacity ( $C_P$ ) data of Cu<sub>8</sub>GeSe<sub>6</sub>, Cu<sub>8</sub>GeSe<sub>5.1</sub>Te<sub>0.9</sub>, and Cu<sub>7.6</sub>Ag<sub>0.4</sub>GeSe<sub>5.1</sub>Te<sub>0.9</sub> by using a Debye and two Einstein modes.  $R^2$  stands for the correlation coefficient. The Debye temperature ( $\Theta_D$ ) is derived by the equation  $\Theta_D = (12\pi^4 pR/5\beta)^{1/3}$ , where  $R$  is the gas constant and  $p$  is the number of atoms in each molecule ( $p = 15$  for our samples).

Fitting parameters	Cu <sub>8</sub> GeSe <sub>6</sub>	Cu <sub>8</sub> GeSe <sub>5.1</sub> Te <sub>0.9</sub>	Cu <sub>7.6</sub> Ag <sub>0.4</sub> GeSe <sub>5.1</sub> Te <sub>0.9</sub>
$\varphi$ (10 <sup>-3</sup> Jmol <sup>-1</sup> K <sup>-2</sup> )	26.68±2.79	49.76±3.01	42.01±2.79
$\beta$ (10 <sup>-4</sup> Jmol <sup>-1</sup> K <sup>-4</sup> )	10.5±0.332	11.1±0.332	12.1±0.318
$A$ (Jmol <sup>-1</sup> K <sup>-1</sup> )	16.335±0.597	14.094±0.642	14.196±0.548
$\Theta_E$ (K)	36.426±0.419	35.953±0.515	33.812±0.363

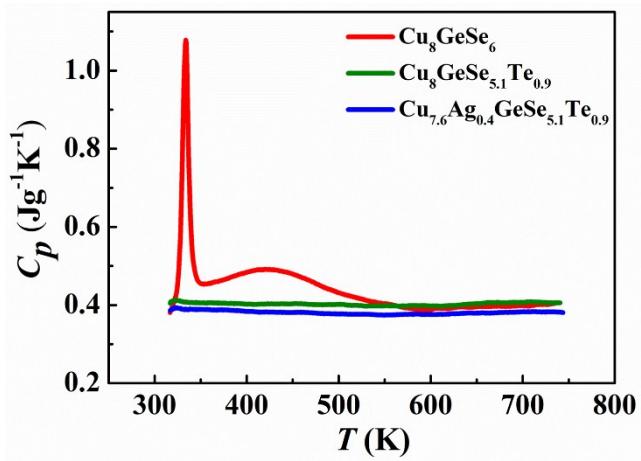
$B$ (Jmol <sup>-1</sup> K <sup>-1</sup> )	78.851±1.697	77.774±1.584	79.023±1.457
$\theta_{E2}$ (K)	82.110±0.920	79.854±0.921	76.170±0.765
$\theta_D$ (K)	302.81	297.3	288.83
$R^2$	0.99994	0.99988	0.99986
$\chi^2$	0.0000585	0.000111	0.000148

**Table S3** Parameters used to fit the heat capacity ( $C_P$ ) data of Cu<sub>8</sub>GeSe<sub>6</sub> by using different methods (one Debye and one Einstein modes, one Debye and two Einstein modes, one Debye and three Einstein modes).

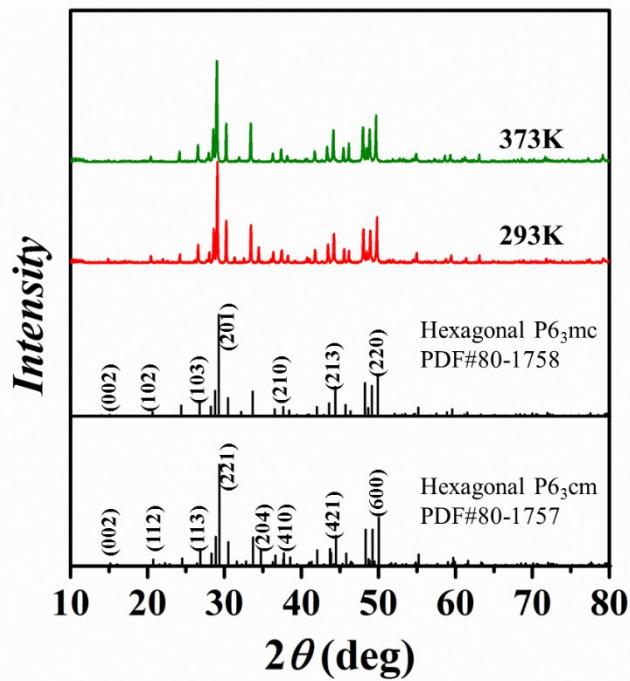
Fitting parameters	One Debye+ One Einstein	One Debye+ Three Einstein
$\varphi$ (10 <sup>-3</sup> Jmol <sup>-1</sup> K <sup>-2</sup> )	80.43±14.62	11.38±3.76
$\beta$ (10 <sup>-4</sup> Jmol <sup>-1</sup> K <sup>-4</sup> )	19.2±0.822	8.7±0.385
$A$ (Jmol <sup>-1</sup> K <sup>-1</sup> )	46.088±2.303	4.202±1.399
$\theta_{E1}$ (K)	53.908±1.084	26.722±2.021
$B$ (Jmol <sup>-1</sup> K <sup>-1</sup> )		24.552±1.844
$\theta_{E2}$ (K)		47.957±2.773
$C$ (Jmol <sup>-1</sup> K <sup>-1</sup> )		82.919±1.792
$\theta_{E3}$ (K)		92.645±2.267
$\theta_D$ (K)	247.63	322.40
$R^2$	0.99618	0.99998
$\chi^2$	0.00362	0.0000173

**Table S4** Parameters used to fit the lattice thermal conductivity ( $\kappa_L$ ) of Cu<sub>7.6</sub>Ag<sub>0.4</sub>GeSe<sub>5.1</sub>Te<sub>0.9</sub>.

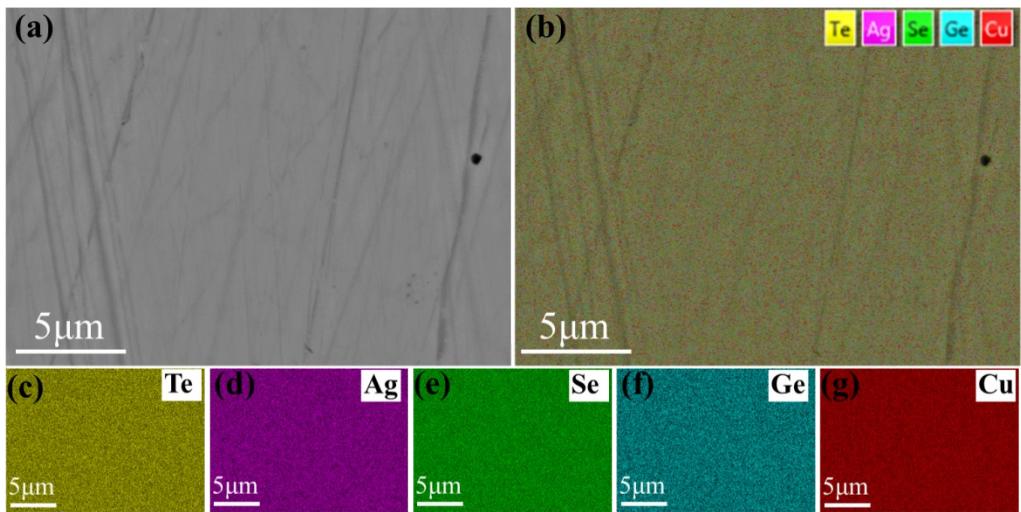
Fitting parameters	Cu <sub>7.6</sub> Ag <sub>0.4</sub> GeSe <sub>5.1</sub> Te <sub>0.9</sub>
$L$ (μm)	0.634
$A$ (10 <sup>-41</sup> s <sup>3</sup> )	14.304
$B$ (10 <sup>-17</sup> s K <sup>-1</sup> )	2.8848
$C_1$ (10 <sup>36</sup> s <sup>-3</sup> )	0.1206
$\omega_1$ (THz)	3.442
$C_2$ (10 <sup>36</sup> s <sup>-3</sup> )	19.55
$\omega_2$ (THz)	6.889
$R^2$	0.99800
$\chi^2$	0.005987



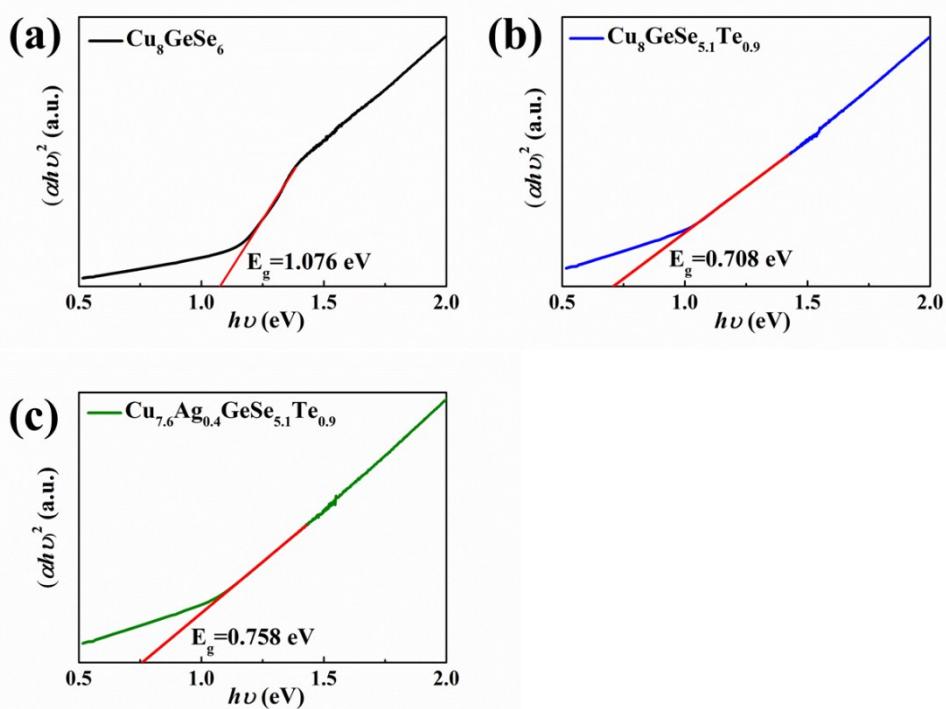
**Fig. S1** Temperature dependence of the heat capacity for  $\text{Cu}_8\text{GeSe}_6$ ,  $\text{Cu}_8\text{GeSe}_{5.1}\text{Te}_{0.9}$  and  $\text{Cu}_{7.6}\text{Ag}_{0.4}\text{GeSe}_{5.1}\text{Te}_{0.9}$  at constant pressure.



**Fig. S2** XRD patterns for  $\text{Cu}_8\text{GeSe}_6$  at 293K and 373K.



**Fig. S3** Scanning electron microscopy (SEM) images of  $\text{Cu}_{7.6}\text{Ag}_{0.4}\text{GeSe}_{5.1}\text{Te}_{0.9}$ . (a) Backscattered electron imaging map, (b) all elements, (c) Te, (d) Ag, (e) Se, (f) Ge, and (g) Cu mappings.



**Fig. S4**  $(\alpha h\nu)^2$  as a function of photon energy  $h\nu$ . An extrapolation of the spectra (red lines) is used to estimate optical band gap values for  $\text{Cu}_8\text{GeSe}_6$ ,  $\text{Cu}_8\text{GeSe}_{5.1}\text{Te}_{0.9}$  and  $\text{Cu}_{7.6}\text{Ag}_{0.4}\text{GeSe}_{5.1}\text{Te}_{0.9}$ .

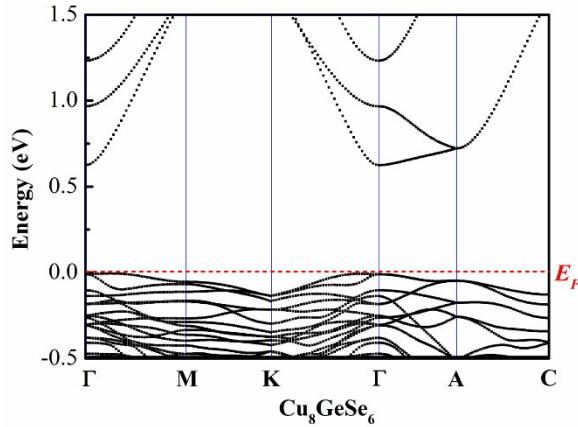


Fig. S5 Calculated band structure of low temperature phase  $\text{Cu}_8\text{GeSe}_6$  using revised PBE + 6U method. The Fermi level is set to zero. The calculated band structure indicates a direct transition. The CBM is mainly dominated by Se-4s and 4p bands, and the VBM is mainly from the Cu-3d and Se-4p bands, which indicates a direct allowed transition.

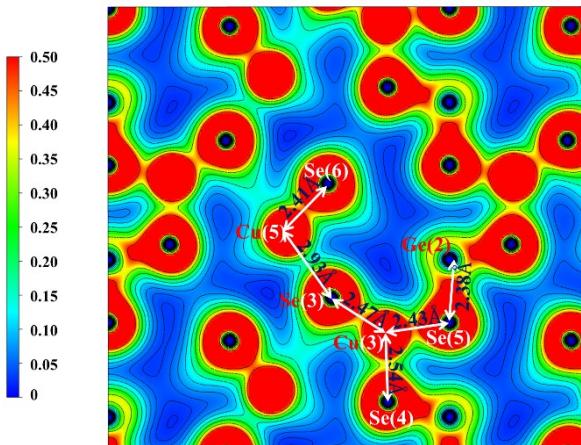


Fig. S6 Total charge density map of low-temperature phase  $\text{Cu}_8\text{GeSe}_6$ . The plot is on the  $\text{Se}(6)$ - $\text{Cu}(5)$ - $\text{Se}(3)$  plane. The unit is  $e/\text{\AA}^3$ . The contour lines are plotted from 0 to  $0.50 e/\text{\AA}^3$  with  $0.05 e/\text{\AA}^3$  intervals.

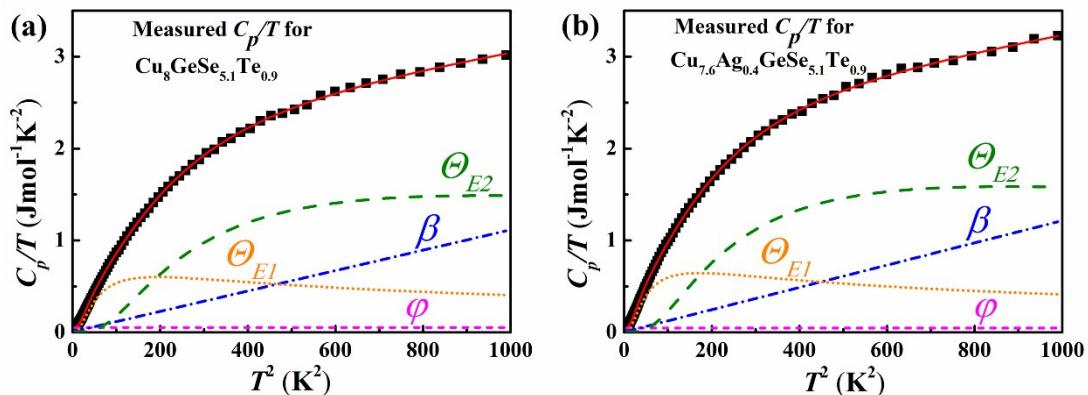
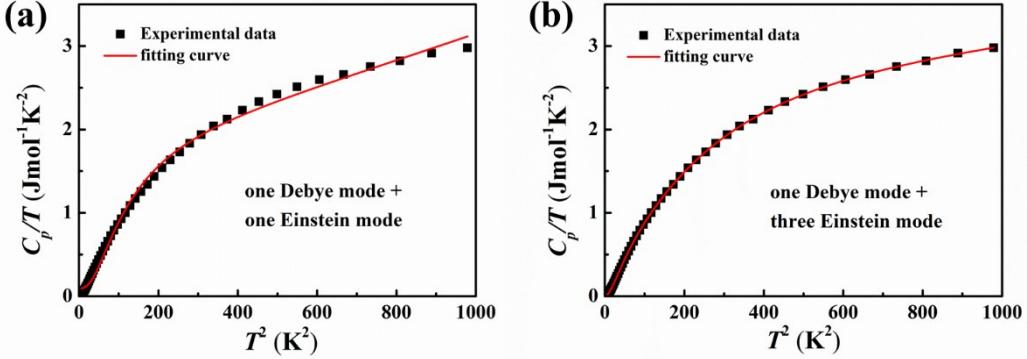


Fig. S7 Measured  $C_p/T$  versus  $T^2$  for  $\text{Cu}_8\text{GeSe}_{5.1}\text{Te}_{0.9}$  and  $\text{Cu}_{7.6}\text{Ag}_{0.4}\text{GeSe}_{5.1}\text{Te}_{0.9}$  (black

squires). The red solid line represents the fitted curve by using one Debye mode and two Einstein modes. The other lines represent the electronic term  $\varphi$ , Debye term  $\beta$ , and two Einstein terms,  $\Theta_{E1}$  and  $\Theta_{E2}$ .



**Fig. S8** Measured  $C_p/T$  versus  $T^2$  for  $\text{Cu}_8\text{GeSe}_6$  (black squares). The red solid line represents the fitted curve by using different methods, (a) one Debye and one Einstein mode, (b) one Debye and three Einstein modes.

## References

- 1 D. T. Morelli, V. Jovovic, J. P. Heremans, *Phys. Rev. Lett.* 2008, **101**, 035901.