

Fig. S1 FESEM image of TiO₂ NRs on a carbon paper

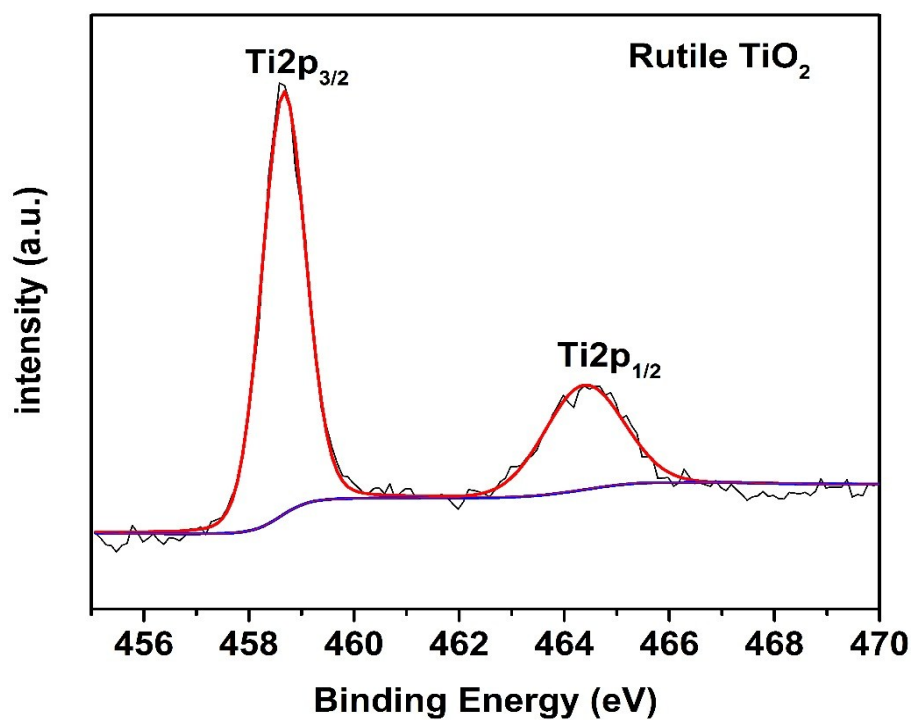


Fig. S2 Ti 2p XPS spectrum of rutile TiO₂.

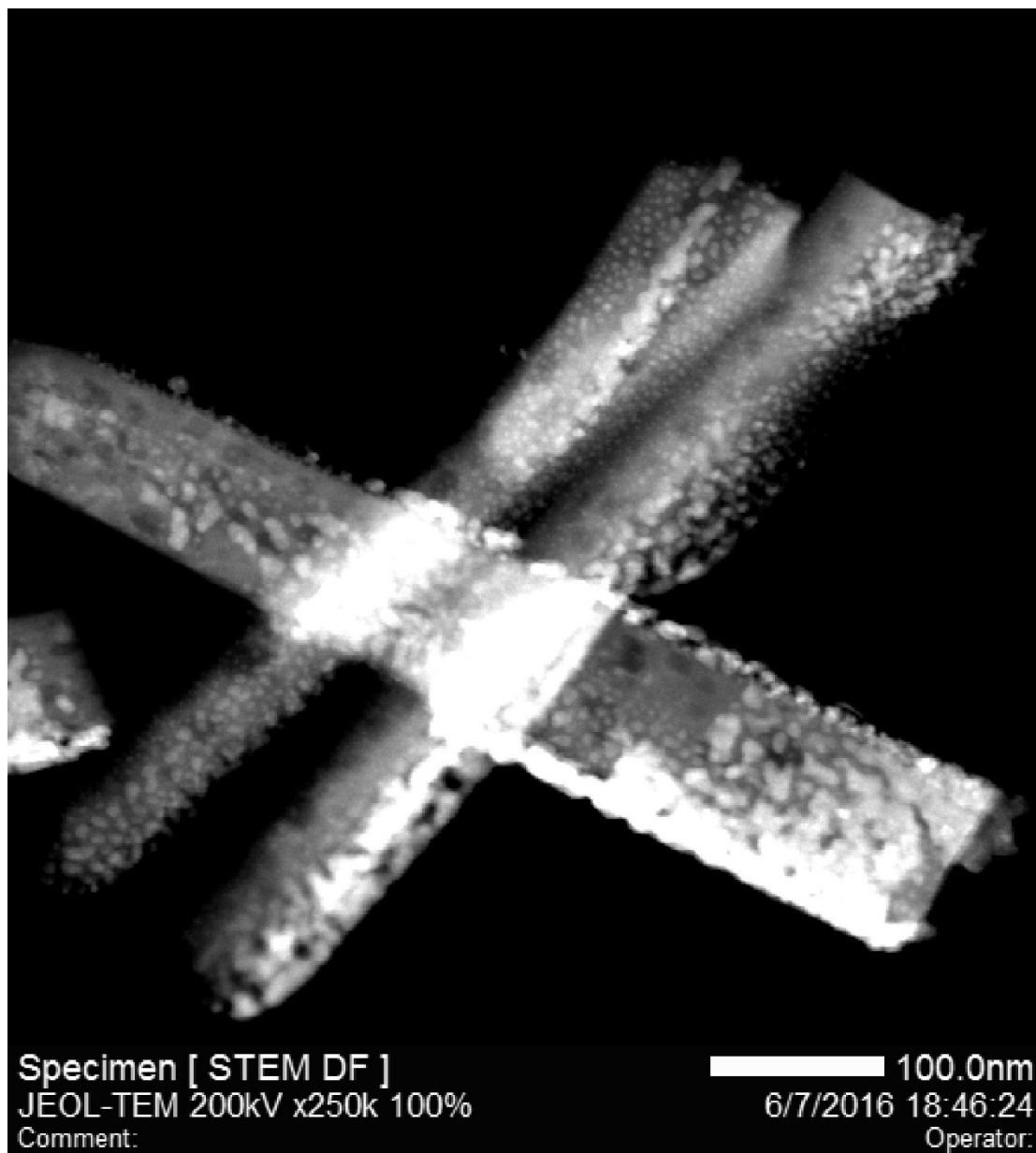


Fig. S3 STEM of PdNi nanoparticles on a TiO₂ nanorods

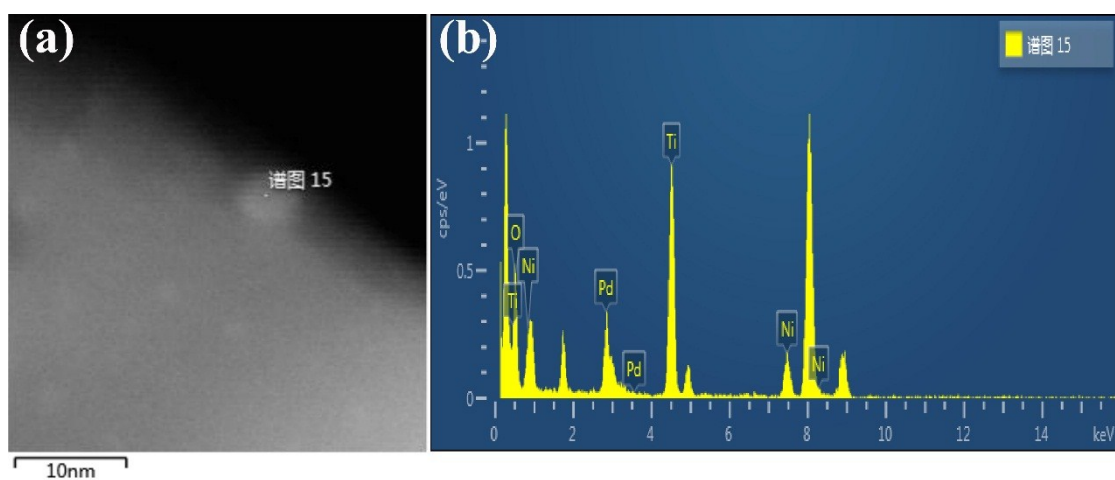


Fig. S4 STEM image (a), and EDX spectrum (b) of the PdNi-TiO₂ NRs

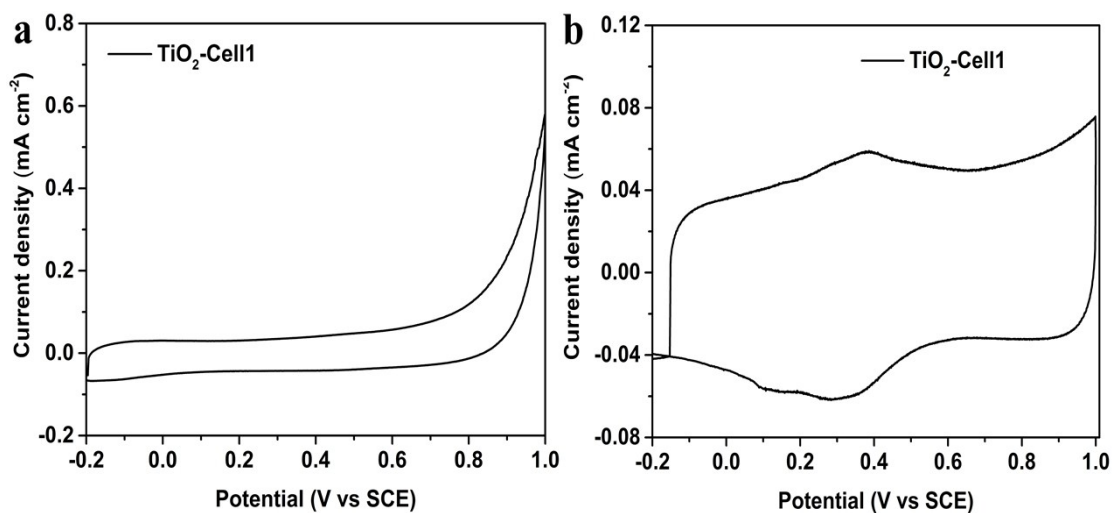


Fig. S5 CVs of TiO_2 sample in 0.5 M H_2SO_4 (a) and 0.5 M H_2SO_4 containing 0.5 M HCOOH at the scan rate of 50 mV s^{-1} .

Table S1 Comparison of the mass activity of the Pd₂Ni₃-TiO₂ with previous reported Pd-based catalysts.

Samples	Test conditions	ECSA (m ² g ⁻¹ _{Pd})	Mass activity (mA mg ⁻¹ _{Pd})	Ref.
Pd ₂ Ni ₃ -TiO ₂	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	43.6	753.1	This work
Pd/NS-G	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	83.4	501.8	S1 ¹
Pd ₆ Co/3DG	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	51.0	430.8	S2 ²
PdNi/C	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	---	556.7	S3 ³
PdNi/C	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	---	396	S4 ⁴
PdCu/C	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	---	654	S4 ⁴
Pd@graphene	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	56.0	89.5	S5 ⁵
Porous Pd ₅₇ Ni ₄₃	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	58.4 mC mg ⁻¹	~800	S6 ⁶
Pd-NF/C	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 20 mV s ⁻¹	2.47	16.4	S7 ⁷
Pd ₁ Ni ₁ -NNs/RGO	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	98.2	604.3	S8 ⁸
AP - Pd/GN	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	72.72	446.3	S9 ⁹
Pd-DNA@Graphene	0.5 M H ₂ SO ₄ + 0.5 M HCOOH; 50 mV s ⁻¹	147.1	140.1	S10 ¹⁰

References:

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