Supporting information for:

Bimetallic Gold Core - Silver Shell Nanorods Performance

for Surface Enhanced Raman Spectroscopy

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1- XPS data^{S1}

A Thermo Scientific Model K-Alpha x-ray photoelectron spectroscopy (XPS) instrument (Thermo Scientific LLC, Madison, WI) using monochromatic Al K α radiation (1486.7 eV) generated XPS data for the assessment of the silver layer formation and chemical composition for four classes of AuNR/Ag. The spot sizes of x-ray used in this study were 200 or 400 μ m. The electrons emitted from the sample surfaces were analyzed with a 128 multi-channel hemispherical electron energy analyzer. The analyzer pass energies for survey spectra and high-resolution (narrow scan) spectra were set to 200 eV and 50 eV, respectively. The base vacuum pressure maintained in the analysis chamber was typically ~1×10⁻⁹ mBar.

Several drops of the thiolated polyethylene glycol- (HS-PEG) coated samples were dried on glass microscope slides and mounted to the sample platen using copper clips. The food gun, utilizing a combination of low energy electrons and argon ions, was turned on for charge neutralization during all spectra collection, causing the vacuum in the analysis chamber to rise to $\sim 2 \times 10^{-7}$ mBar during the measurements. The Thermo Scientific Avantage XPS software package® was used to collect and analyze all spectra. All spectra were charge-corrected by shifting the lowest energy C1s peak to the value of 284.8 eV, corresponding to C–C or C–H bonds confirmed by the pure metallic Au4f_{7/2} peak at 84.0 eV. XPS peaks were fitted by mixed Gaussian/Lorentzian peaks after Shirley-Smart background subtraction. In our XPS studies, the binding energies of C1s, Au4f_{7/2}, Ag3d_{5/2}, O1s, N1s, and S2p_{3/2}, were investigated.

XPS data analysis was conducted to analyze the surface chemical compositions of (0, 1, 2, 3,

and 4 nm) silver-coated gold nanorods. Table S1 and Figure S1 reveal that as the silver layer thickness increased, the atomic percentage amount of the silver increased. However, increasing the amount of silver on the outermost layer of the nanorods decreased the signal from the gold core gradually, which confirms the formation and thickness of the silver layer.

Table S1: XPS atomic percentage of elemental chemical composition of the gold nanorod (AuNR\Ag(0)) and four classes of silver-coated gold SERS substrates.

| Silver layer thickness (nm) | Ag atomic % (Ag3d _{5/2}) | Au atomic % (Au4f _{7/2}) |
|-----------------------------|------------------------------------|------------------------------------|
| | | |
| AuNR\Ag(0) | 0 | 0.54 |
| AuNR\Ag(1) | 2.63 | 3.08 |
| AuNR\Ag(2) | 2.61 | 1.88 |
| AuNR\Ag(3) | 3.94 | 1 |
| AuNR\Ag(4) | 8.13 | 0.42 |



Figure S1: Atomic percentage of silver-coated gold nanorods.



Figure S2 and Figure S2: AuNRAg(0) (Ag = 0 nm)



Figure S3 and Figure S3: AuNR\Ag(1) (Ag = 1 nm)



Figure S4 and Figure S4: AuNR\Ag(2) (Ag = 2 nm)



Figure S5 and Figure S5: AuNR\Ag(3) (Ag = 3 nm)



Figure S6 and Figure S6: AuNR\Ag(4) (Ag = 4 nm)

2- Zeta Potential

Zeta potential for the following samples were conducted using ZetaView®, from Electrophoresis and Brownian Motion Video Analysis Laser Scattering Microscopy.

| Sample | Zeta Potential (mV) | Number of traced particles |
|-----------------|---------------------|----------------------------|
| AuNR-Ag(0) | -44.69 | 552 |
| AuNR-Ag(1) | 43.55 | 676 |
| AuNR-Ag(2) | 40.59 | 661 |
| AuNR-Ag(3) | 32.28 | 446 |
| AuNR-Ag(3) | 45.94 | 418 |
| AuNR-Ag(2)-PATP | -28.76 | 312 |

 Table S2: Zeta potential analysis of nanorods samples

Following images are the analysis report for each kind:

1) AuNR-Ag(0)



2) AuNR-Ag(1)



3) AuNR-Ag(2)



4) AuNR-Ag(3)



5) AuNR-Ag(4)



6) AuNR-Ag(2)-PATP



3- TEM images



4- SERS experimental error bars



5- STEM images

 $AuNR \setminus Ag(1)$



 $AuNR \setminus Ag(2)$



AuNRAg(3)



AuNR Ag(4)

| Bright | Au | Ag |
|----------|-------|-------|
| | | |
| 20 nm | 20 nm | 20 nm |
| BF+Au+Ag | BF+Au | BF+Ag |
| | | |
| | | |
| 20 nm | 20 nm | 20 nm |

References

(S1): Fumiya, W.; Zeid, A. N.; Takumi, H.; Masatoshi, M.; Minoru, N.; Alexandru, S. B. X-ray photoelectron spectroscopy and transmission electron microscopy analysis of silver-coated gold nanorods designed for bionanotechnology applications. *Nanotechnology* 2017, *28* (2), 025704.