

Effect of Nb doping in WO₃/ZrO₂ catalysts on gas phase dehydration of glycerol to form acrolein

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Supporting Information

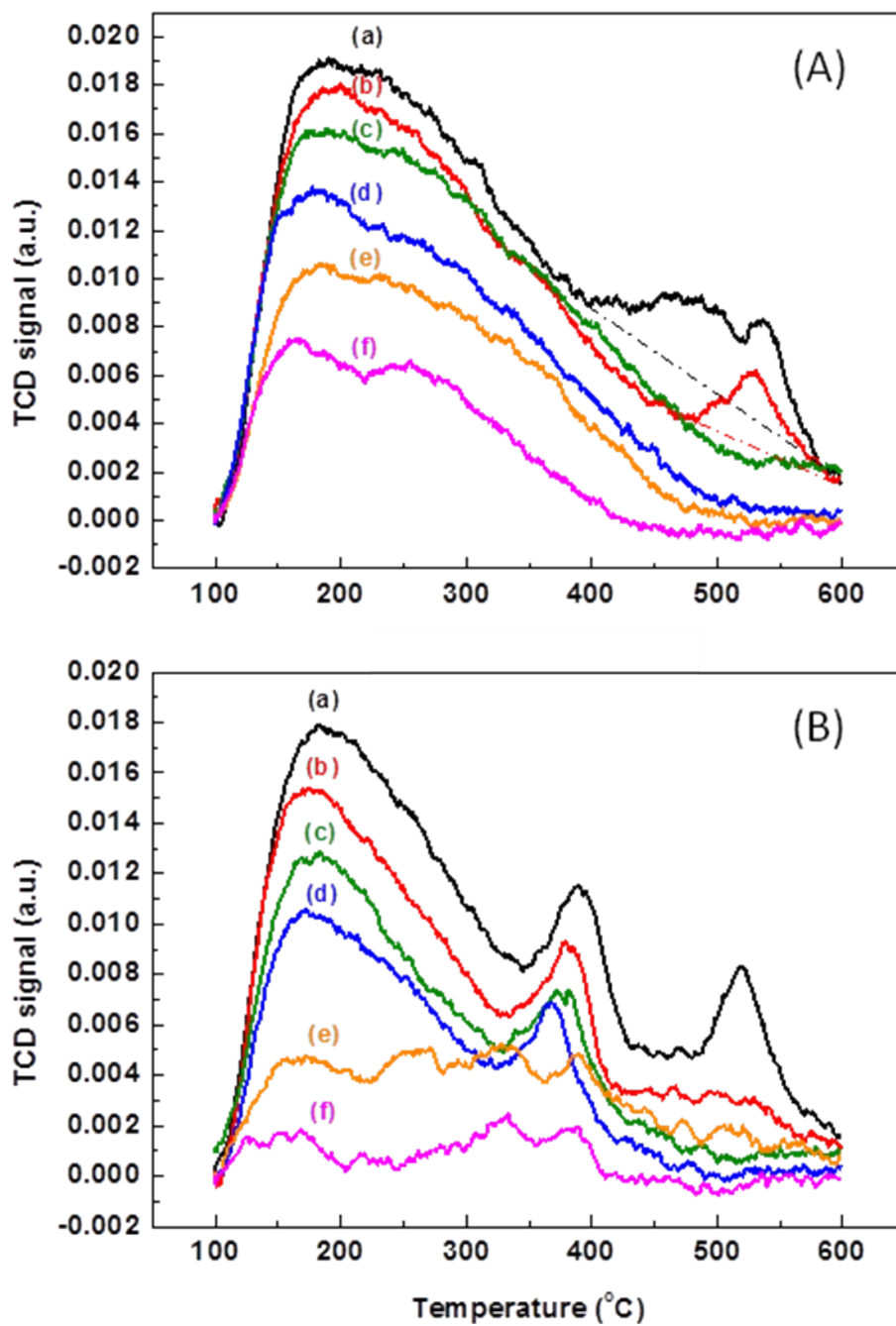


Fig. S1 (A) NH₃- and (B) CO₂-TPD thermograms of WO₃/ZrO₂ catalysts calcined at: (a) 400, (b) 450, (c) 500, (d) 600, (e) 700 and (f) 800°C.

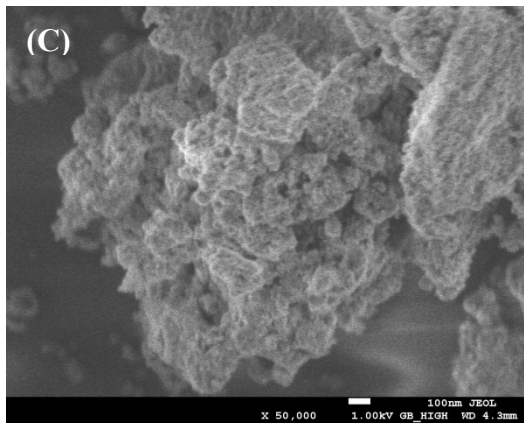
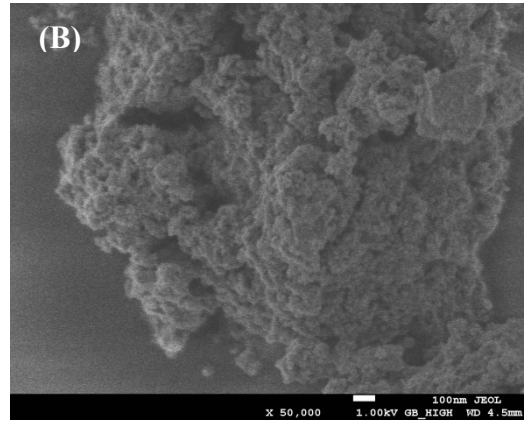
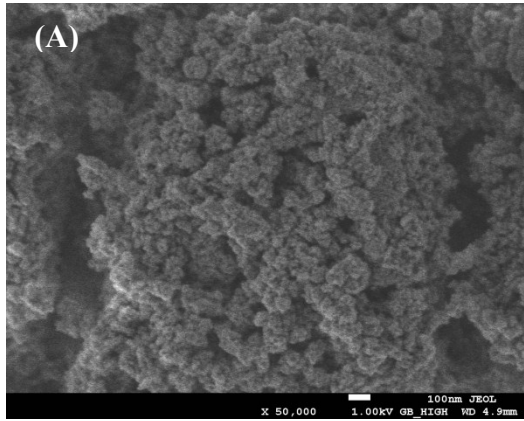


Fig. S2 The SEM photographs of (A) $\text{WO}_3/\text{ZrO}_2\text{-450}$ (B) $1\%\text{NbWO}_x/\text{ZrO}_2\text{-450}$ and (C) $3\%\text{NbWO}_x/\text{ZrO}_2\text{-450}$.

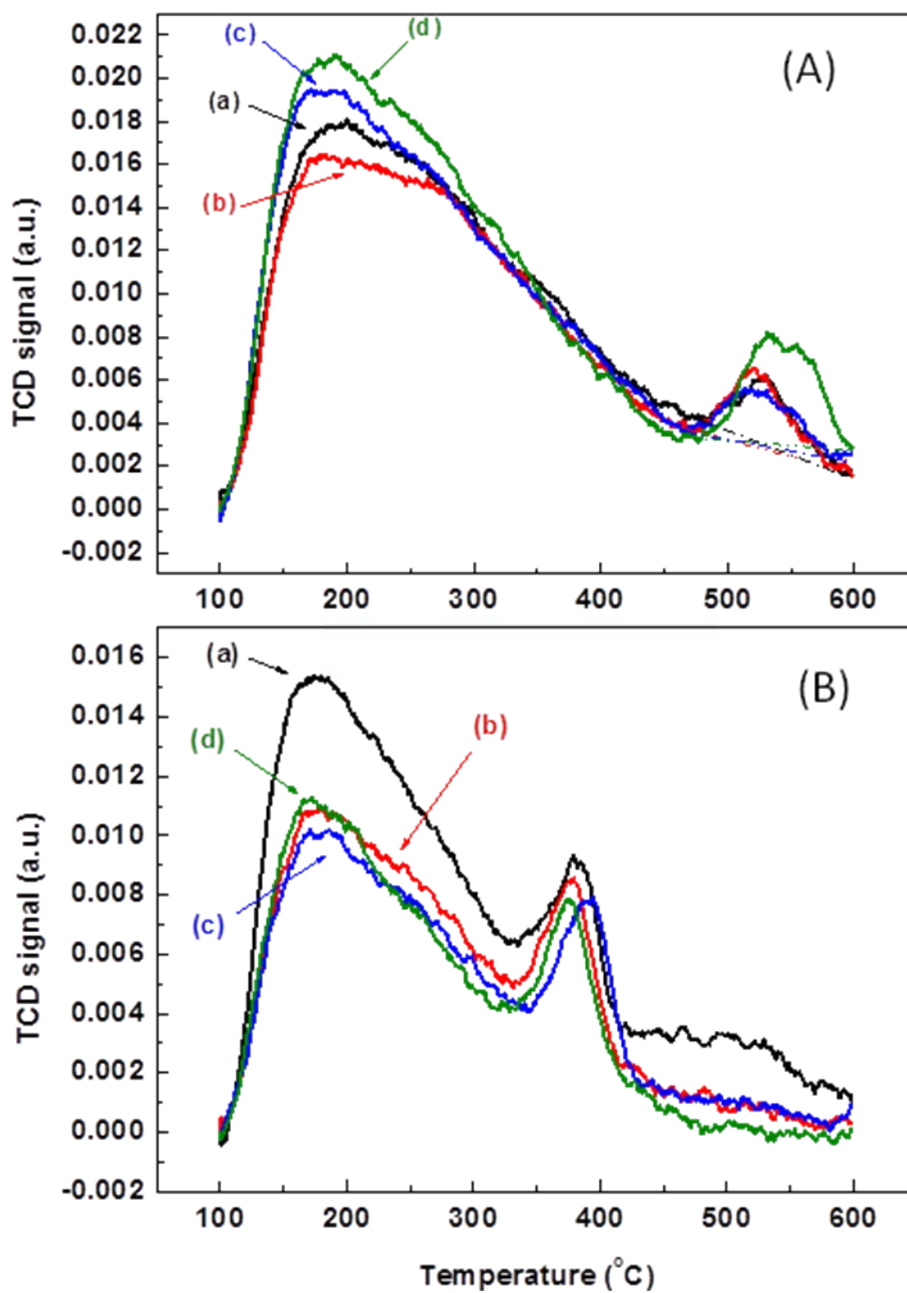


Fig. S3 (A) NH₃- and (B) CO₂-TPD thermograms of (a) WO₃/ZrO₂-450, and those doped with (b) 1% Nb₂O₅, (c) 3% Nb₂O₅, and (d) 5% Nb₂O₅.

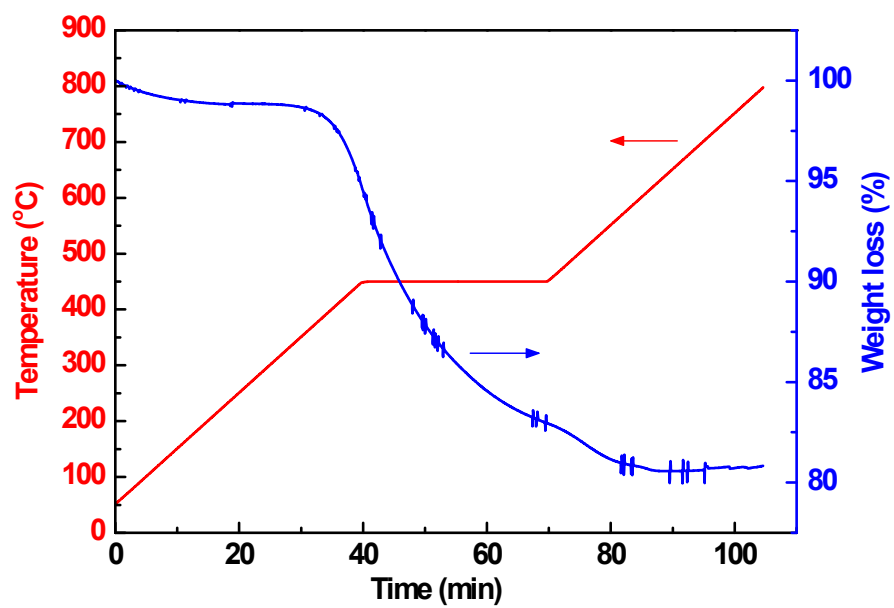


Fig. S4 Thermogravimetric analysis of 3%NbWO_x/ZrO₂-450 under air after catalytic reaction.

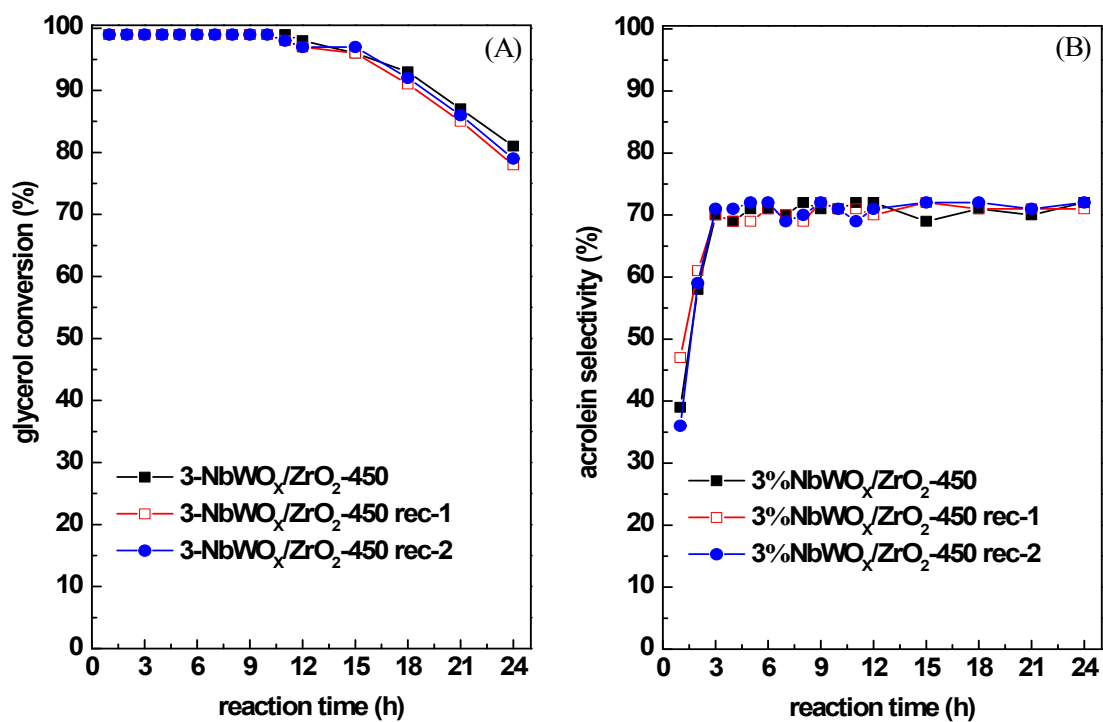


Fig. S5 Evolution of (A) glycerol conversion and (B) selectivity to acrolein versus TOS at 290°C over fresh and regenerated 3%NbWO_x/ZrO₂-450 catalyst. 0.3 g catalyst, GHSV = 1117 h⁻¹, 60 mL/min N₂ flow rate.

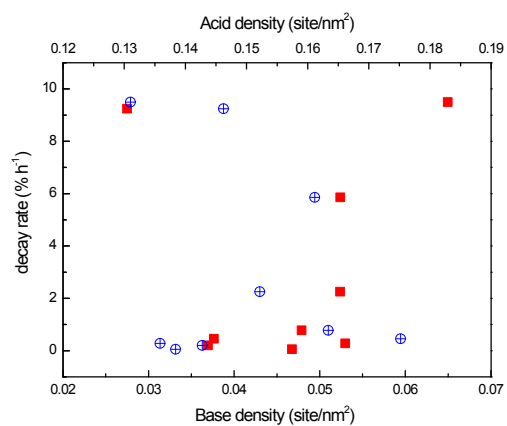
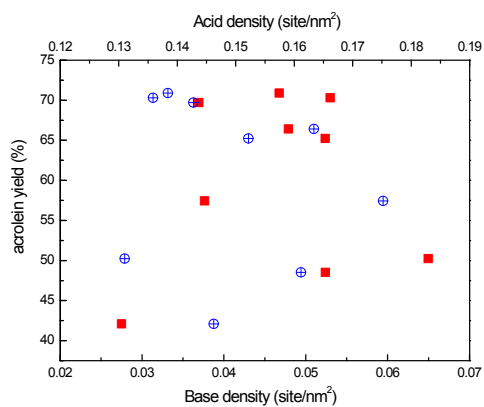
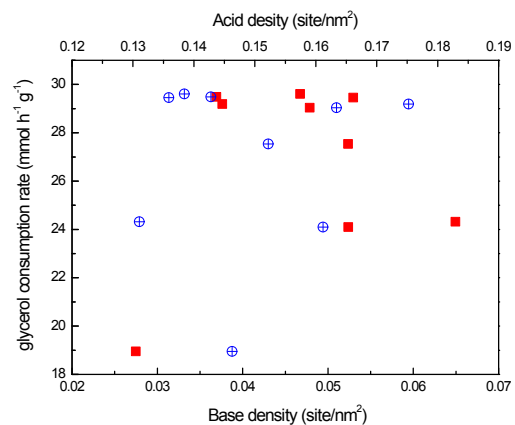


Fig. S6 Correlations of (A) specific glycerol consumption rates (average in 3–12 h TOS), (B) acrolein yield, and (C) decay rate with acid (■) and base (⊕) densities on

WO₃/ZrO₂ calcined at different temperatures and WO_x/ZrO₂-450 doped with 1–5% Nb₂O₅.