

Supporting Information

Synthesis of All-Inorganic CsPb₂Br₅ Perovskite and Determination of Its Luminescence Mechanism

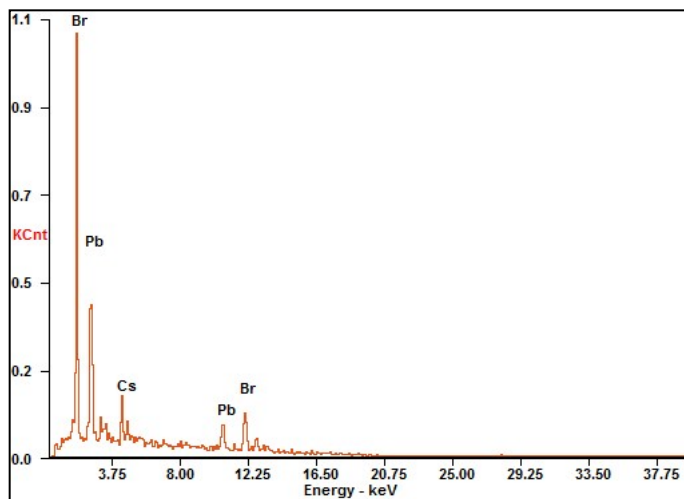
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<i>Element</i>	<i>Wt%</i>	<i>At%</i>
<i>BrL</i>	45.15	65.40
<i>PbM</i>	42.16	23.55
<i>CsL</i>	12.69	11.05

Figure S1. Energy dispersive X-spectroscopy of CsPb₂Br₅. The ratio of each element is listed in the table.

Table S1. Total energy of unit cell for each reactant and the resultant chemical equation:



The total energy of each materials is calculated based on the optimized structure using DFT. The change of the total energy is obtained by the difference between the products and the reactants.

	CsOH	PbBr₂	CsPb₂Br₅	Pb(OH)Br
Total energy (Et)	-386.66121413Ry	-1032.20505769Ry	-2505.07266819Ry	-978.53862569Ry

$$\Delta E_t = E_{t_{\text{CsPb}_2\text{Br}_5}} + E_{t_{\text{Pb(OH)Br}}} - E_{t_{\text{CsOH}}} - 3E_{t_{\text{PbBr}_2}} = -0.33507975\text{Ry} \approx -4.559 \text{ eV}$$

Movie S1. Movie S1.mp4