1	Supplementary Information
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3	A new strategy for sensitive electrochemical determination of
4	nitrophenol isomers using $\beta$ -cyclodextrin derivatives functionalized
5	silicon carbide
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14	S1. Materials and methods
15	The experimental details are provided in supplementary information.
16	S1.1. Chemicals
17	SiC was purchased from Nanjing Aipurei Nano-Material Company (Nanjing,
18	China). SH- $\beta$ -CD and NH <sub>2</sub> - $\beta$ -CD were purchased from Shandong Binzhou Zhiyuan
19	Bio-Technology Co., Ltd (Shandong, China). Gold chloride (HAuCl <sub>4</sub> · 3H <sub>2</sub> O, 99%),
20	N-(3-dimethylaminopropyl)-N'-ethylcarbodiimidehydrochloride (EDC) and N-
21	hydroxysuccinimide (NHS) were obtained from Sigma Chemical Co. (St. Louis, MO,
22	USA). 3-aminopropyltriethoxysilane (APTES, 98%), o-NP, and p-NP were obtained

from Aladdin Chemical Co. (Southern California, USA). N,N-dimethylformamide
(DMF) was purchased from Tianjin Zhiyuan Chemical Reagent Co., Ltd. Sodium
borohydride (NaBH<sub>4</sub>, 98%) was purchased from Tianjin great chemical reagent
factory. All other reagents were of analytical grade. All aqueous solutions were
prepared with deionized water (DW, 18 MΩ cm).

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### 29 S1.2. Apparatus

The morphologies of the prepared samples were characterized by a JEM 2100 30 transmission electron microscopy (TEM, Japan). Fourier transform infrared (FTIR) 31 study was performed over the wavenumber, range of 4000–400 cm<sup>-1</sup> by a Thermo 32 Fisher SCIENTIFIC Nicolet IS10 (Massachusetts, USA) FTIR impact 410 33 spectrophotometer using KBr pellets. Thermogravimetric analysis (TGA) was carried 34 out on a Q50 TGA (TA Instruments, New Castle, USA), from 25 to 800 °C in argon 35 at a heating rate of 5 °C min<sup>-1</sup>. Electrochemical impedance spectroscopy (EIS), cyclic 36 voltammetry (CV), and differential pulse voltammetery (DPV) experiments were 37 performed with a CHI 660E Electrochemical Workstation from Shanghai Chen-hua 38 Instrument (Shanghai, China) and conducted using a three-electrode system, with the 39 modified GCE as working electrode, a platinum wire as the counter electrode, a 40 saturated calomel electrode (SCE) as the reference electrode. 41

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# 43 S1.3. Synthesis of carboxyl silicon carbide (CSiC)

44 CSiC was synthesized according to a previous study.<sup>1</sup> Firstly, SiC-OH was

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prepared through a hydrothermal reaction of glucose. Briefly, 50 mg of SiC and 200 45 mg of glucose were dispersed in 60 mL of DW via sonication, the suspension was 46 transferred to an autoclave and kept at 180 °C for 8 h. After reaction, the autoclave 47 was cooled naturally. The suspension was centrifuged and washed with DW and 48 alcohol for 3 times, respectively. SiC-OH was obtained by freeze-drying. Secondly, 49 the obtained SiC-OH was dispersed in 30 mL of ethanol via sonication, and then 1.0 50 mL of APTES was added to the suspension with vigorous stirring. After stirring for 51 12 h, the resulting mixture was cleaned with ethanol for 3 times. SiC-NH<sub>2</sub> was 52 obtained by freeze-drying. Thirdly, a dispersion of SiC-NH<sub>2</sub> in 20 mL DMF was 53 added dropwise to a flask containing 20 mL of 0.1 M succide anhydride in DMF. The 54 mixture was stirred for 24 h, the resulting SiC nanoparticles with carboxylic-function 55 groups at their surface (CSiC) were washed with DMF, alcohol, and DW for 3 times. 56 Finally, CSiC was obtained by freeze-drying. 57

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### 59 S1.4. Preparation of Au@CSiC-SH/NH<sub>2</sub>-CD composite material

Au@CSiC-SH/NH<sub>2</sub>-CD composite material was synthesized according to a
previous work <sup>1, 2</sup> with some modifications. Briefly, 5 mg of CSiC was dissolved in 5
ml of DMF, and 130 μL of chloroauric acid was added. Similarly, 10 mg of SH-CD
and 0.9 mg of sodium borohydride were dissolved in 5 mL of DMF, and then the
mixture of CSiC and chloroauric acid was added quickly dropwise to the above
solution under stirring for 6 h. The resulted suspension was centrifuged and washed
with DW for 3 times followed by freeze-drying to obtain Au@CSiC-SH-CD. The

Au@CSiC composite was prepared using the similar process of Au@CSiC-SH-CD in
the absence of SH-β-CD. Then 20 mg of Au@CSiC-SH-CD, the carboxyl groups on
which have been activated by 10 mL of EDC/NHS (0.5 mM) under stirring for 2 h,
was added into the above mixture and stirred for another 5 h. After that, the mixture
was centrifuged and washed with DW for 3 times. The final Au@CSiC-SH/NH<sub>2</sub>-CD
composite product was obtained by freeze-drying.

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## 74 S1.5. Fabrication of the modified electrodes

A glassy carbon electrode (GCE, 3 mm in diameter) was polished with 0.3 and 0.05 75 µm Al<sub>2</sub>O<sub>3</sub> powder respectively and subsequently sonicated in ethanol and DW to 76 remove the physically adsorbed substance and dried in air. The CSiC, Au@CSiC, 77 Au@CSiC-SHCD, or Au@CSiC-SH/NH2-CD was dissolved in DW respectively at a 78 concentration of 1.0 mg/mL with the aid of ultrasonic agitation for 20 min, resulting 79 in a homogeneous suspension. To prepare the materials modified electrode, 5  $\mu$ L 80 of the suspension was dropped onto the electrode surface and dried at room 81 temperature. The obtained electrode was noted as the CSiC/GCE, Au@CSiC/GCE, 82 Au@CSiC-SHCD/GCE, and Au@CSiC-SH/NH2-CD/GCE electrode. 83 84

## 85 S1.6. Electrochemical measurements

CV was carried out in 0.1 M PBS (pH 6.0) containing 1 mM o-NP and p-NP. DPV
was performed in 0.1 M PBS (pH 6.0) containing different concentrations of o-NP
and p-NP by scanning the potential from -0.2 to 1.2 V with a pulse amplitude of 0.05

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V and a pulse width of 0.05 s. EIS was recorded in the frequency range from 10<sup>-1</sup> to
10<sup>-5</sup> Hz with an amplitude of 5 mV using 2.0 mM [Fe(CN)<sub>6</sub>]<sup>3-/4-</sup> redox couple (1:1)
with 0.1 M KCl as supporting electrolyte. All the measurements were carried out at
room temperature.

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### 94 S1.7. Molecular docking

Crystal structure of  $\beta$ -CD (ID: IHEGEI) was obtained from Crystallography Open 95 Database (http://www.crystallography.net/). The structures of nitrophenol isomers were 96 constructed by the UCSF Chimera software. The  $\beta$ -CD and nitrophenol isomers 97 structures were fully optimized by the UCSF Chimera software. Hydrogen atoms 98 were added using the Dock Prep module. The GAFF force field and the AM1-BCC 99 charges were used. The optimized structures were used as a starting structure in the 100 docking study. For the molecular docking study, the molecular surface of the  $\beta$ -CD 101 molecule was generated using the DMS tool with a probe radius of 1.4 Å. The sphgen 102 module was applied to generate spheres surrounding the binding site. The Grid 103 module of DOCK6 was employed to generate Grid file which was used in scoring in 104 the subsequent docking procedure. The flexible docking method was utilized 105 producing 1000 different conformational orientations for the guest molecule. The 106 electrostatic interactions and van der Waals forces were calculated based on the grids. 107 Finally, clustering analysis with root-mean-square deviation (RMSD) threshold of 2.0 108 Å was performed to retain the best results. 109

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# 112 **References**

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