## Supporting information

## Direct Regioselective C-H Borylation of [5]Helicene

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## 1. General

Catalysts, reagents and solvents were obtained from commercial suppliers (Aldrich, TCI, Alfa Aesar, Strem Chemicals, Penta) and were used without further purification. Solvents for column chromatography were distilled prior to use. Argon was used as an inert gas. Silica gel $60(0.035-0.070 \mathrm{~mm})$ (Acros Organics) was used for column chromatography. TLC was performed on aluminum sheets with layer of Silica gel $60 \mathrm{~F}_{254}$ purchased from MERCK. Spots on TLC plates were detected by using UV lamp ( $\lambda=254 \mathrm{~nm}$ ) or basic permanganate solution. PTLC was performed on glass sheets with layer of Silica gel GF (Analtech).

NMR spectra were recorded on Varian VNMRS $300\left(v\left({ }^{1} \mathrm{H}\right)=299.94 \mathrm{MHz}, v\left({ }^{13} \mathrm{C}\right)=\right.$ 75.43 MHz), on Bruker Avance III ( 400 MHz ) $\left(v\left({ }^{1} \mathrm{H}\right)=400.13 \mathrm{MHz}, v\left({ }^{13} \mathrm{C}\right)=100.61 \mathrm{MHz}\right.$ ), or on Bruker Avance III ( 600 MHz ) $\left(v\left({ }^{1} \mathrm{H}\right)=600.17 \mathrm{MHz}, v\left({ }^{13} \mathrm{C}\right)=150.04 \mathrm{MHz}\right)$ in deuterated solvents and referenced to residual solvent peak $\left(\mathrm{CDCl}_{3}{ }^{1} \mathrm{H}: 7.26 \mathrm{ppm},{ }^{13} \mathrm{C}: 77\right.$ ppm). Chemical shifts are given in $\delta$-scale, coupling constants $J$ are given in Hz. Mass spectra were recorded on VG-Analytical ZAB-SEQ. All melting points are uncorrected and were determined on Kofler apparatus. HPLC separations were performed on a Gilson 321 H2 pumps with $25 \times 250 \mathrm{~mm}$ preparative column (Labio, Czech Republic), filled with BIOSPHER PSI 120 C 18 sorbent ( $7 \mu \mathrm{~m}$ mesh).
$[\operatorname{Ir}(\mathrm{OH}) \mathrm{cod}]_{2}$ was prepared according to the reported procedure. ${ }^{[1]}[5]$ Helicene $\mathbf{1}$ was prepared according to the reported procedure, ${ }^{[2]}$ and recrystallized before use. Iridium complexes $\mathbf{C 1}-\mathbf{C} 7$ were prepared according to reported procedure. ${ }^{[3]}$

## 2. Borylation of [5]helicene.

## First experiment and separation of products:

A flame dried 10 ml thick-wallet crimp reaction vial was charged with $[\mathrm{Ir}(\mathrm{OMe}) \operatorname{cod}]_{2}(3.3$ $\mathrm{mg}, 0.005 \mathrm{mmol})$, dtbpy ( $2.7 \mathrm{mg}, 0.01 \mathrm{mmol}$ ), $\mathrm{B}_{2} \operatorname{pin}_{2}(25 \mathrm{mg}, 0.1 \mathrm{mmol})$, and [5]helicene ( 28 $\mathrm{mg}, 0.1 \mathrm{mmol})$. Then cyclohexane ( 0.8 mL ) was added and argon was bubbled through the reaction mixture to remove $\mathrm{O}_{2}$ and reduce the volume to 0.5 mL and the vessel was crimpsealed with a septum cap under the flow of argon. Afterwards, the reaction mixture was stirred at $100{ }^{\circ} \mathrm{C}$ for 24 h . DCM ( 2 mL ) was added and the solution was filtered through a short pad of silica gel (washed with DCM 5 mL ). Volatiles were removed under reduced pressure and column chromatography of the residue (hexane/DCM from $100 / 0$ to $0 / 100$ ) afforded recovered [5]helicene (the first fraction, $22.1 \mathrm{mg}, 79 \%$ ), a mixture of 2a and $\mathbf{2 b}$ (the second fraction, $6.5 \mathrm{mg}, 16 \%$ ), and a trace of 2c (the third fraction, $\sim 1 \%$ ). The second fraction was subsequently separated by preparative nonaqueous reversed-phase HPLC (NARP). Separation of the second fraction ( MeCN ) afforded 1.6 mg of $\mathbf{2 a}(4 \%)$ and 4.8 mg of $\mathbf{2 b}$ (12\%).

## General procedures for the Table 1 and Table SI-1:

A flame dried 10 ml thick-wallet crimp reaction vial was charged with $[\operatorname{Ir}(\mathrm{OMe}) \operatorname{cod}]_{2}(3.3$ $\mathrm{mg}, 0.005 \mathrm{mmol})$ or $[\mathrm{Ir}(\mathrm{OH}) \mathrm{cod}]_{2}(3.2 \mathrm{mg}, 0.005 \mathrm{mmol})$, ligand ( 0.01 mmol ), B $\mathrm{B}_{2} \mathrm{pin}_{2}(25 \mathrm{mg}$, 0.1 mmol ), and [5]helicene ( $56 \mathrm{mg}, 0.2 \mathrm{mmol}$ ). Then solvent ( 0.8 mL ) was added and argon was bubbled through the reaction mixture to remove $\mathrm{O}_{2}$ and reduce the volume to 0.5 mL and the vessel was crimp-sealed with a septum cap under the flow of argon. Afterwards, the reaction mixture was stirred and heated in an aluminum heating block at $120^{\circ} \mathrm{C}$ for 24 hours. (Entry 7, microwave reactor was used). After cooling, volatiles were removed under reduced pressure, internal standard added (mesitylene) and the ratio of $\mathbf{2 a} \mathbf{2} \mathbf{2 b}$ was calculated by integration of aromatic region of ${ }^{1} \mathrm{H}$ NMR spectra of reaction mixture. Solitary singlet at $\delta=$ 9.03 ppm belongs to $\mathbf{2 a}$ and solitary dd at $\delta=7.64 \mathrm{ppm}$ belongs to $\mathbf{2 b}$.

## General procedure for the Table 2 and Table SI-2:

A flame dried 10 ml thick-wallet crimp reaction vial was charged with catalyst $\mathbf{C}(0.01$ $\mathrm{mmol}), \mathrm{B}_{2} \mathrm{pin}_{2}(25 \mathrm{mg}, 0.1 \mathrm{mmol})$, and [5]helicene ( $56 \mathrm{mg}, 0.2 \mathrm{mmol}$ ). Then solvent ( 0.8 mL ) was added and argon was bubbled through the reaction mixture to remove $\mathrm{O}_{2}$ and reduce the volume to 0.5 mL and the vessel was crimp-sealed with a septum cap under the flow of argon. Afterwards, the reaction mixture was stirred and heated in an aluminum heating block at 120 ${ }^{\circ} \mathrm{C}$ for 24 hours. After cooling, volatiles were removed under reduced pressure, internal standard added (mesitylene) and the ratio of $\mathbf{2 a}: \mathbf{2 b}$ was calculated by integration of aromatic region of ${ }^{1} \mathrm{H}$ NMR spectra of reaction mixture. Solitary singlet at $\delta=9.03 \mathrm{ppm}$ belongs to $\mathbf{2 a}$ and solitary dd at $\delta=7.64 \mathrm{ppm}$ belongs to $\mathbf{2 b}$.

Table SI-1. Ir-catalyzed borylation of $\mathbf{1}$ under various conditions.



| Entry | Ligand ${ }^{[\mathrm{ab]}}$ | Solvent ${ }^{[\mathrm{b}]}$ | Temp. $\left[{ }^{\circ} \mathrm{C}\right]$ | Time $[\mathrm{h}]$ | Ratio 2a/2b | Yield 2a + 2b ${ }^{[\mathrm{c}]}$ |
| :--- | :--- | :--- | :--- | :---: | :---: | :---: |
| 1 | dtbpy | cyclohexane | 23 | 16 | n.d. | trace |
| 2 | dtbpy | cyclohexane | 50 | 16 | n.d. | $\sim 5$ |
| 3 | dtbpy | cyclohexane | 80 | 16 | n.d. | $\sim 10$ |
| 4 | dtbpy | cyclohexane | 100 | 24 | $1 / 3$ | 16 |
| 5 | dtbpy | cyclohexane | 120 | 24 | $1 / 3$ | 26 |
| 6 | dtbpy | 1,4 -dioxane | 120 | 24 | n.d. | 0 |
| 7 | dtbpy | $\mathrm{Bu}_{2} \mathrm{O}$ | 120 | 24 | $1 / 3.6$ | 20 |
| 8 | dtbpy | EtOAc | 120 | 24 | $1 / 3.2$ | 19 |
| 9 | dtbpy | 2-Me-THF | 120 | 24 | n.d. | 0 |
| 10 | dtbpy | mesitylene | 120 | 24 | $1 / 3.1$ | 9 |
| 11 | dtbpy | MTBE | 120 | 24 | $1 / 3.5$ | 21 |
| 12 | dtbpy | MTBE | $120^{[d]}$ | 2 | n.d. | $--[$ e] |
| 13 | dtbpy | octane | 120 | 24 | $1 / 3.1$ | 25 |
| 14 | tmphen | cyclohexane | 100 | 24 | $1 / 5$ | 45 |
| 15 | tmphen | cyclohexane | 120 | 24 | $1 / 5$ | 56 |
|  |  |  |  |  |  |  |
| 16 | L1 | cyclohexane | 120 | 24 | $1 / 5$ | 14 |
| 17 | L2 | cyclohexane | 120 | 24 | $1 / 4.7$ | 21 |
| 18 | L3 | cyclohexane | 120 | 24 | $1 / 6$ | 18 |
| 19 | L4 | cyclohexane | 120 | 24 | $1 / 5.7$ | 14 |

${ }^{[a]}$ dtbpy: 4,4'-di-tert-butyl-2,2'-dipyridyl combined with $[\operatorname{Ir}(\mathrm{OMe}) \text { cod }]_{2}$; tmphen: 3,4,7,8-tetramethyl-1,10phenanthroline combined with $[\operatorname{Ir}(\mathrm{OMe}) \operatorname{cod}]_{2} ; \mathbf{L 1}, \mathbf{L 2}, \mathbf{L 3}$ and $\mathbf{L 4}$ combined with $[\operatorname{Ir}(\mathrm{OH}) \operatorname{cod}]_{2} .{ }^{[b]}$ MTBE Methyl tert-butyl ether. ${ }^{[c]}$ Combined yield based on 1 eq. of 1. ${ }^{[d]}$ Microwave irradiation. ${ }^{[e]}$ Complex mixture.

Table SI-2. Ir-catalyzed borylation of $\mathbf{1}$ under various conditions.




| $R^{1}=M e, R^{2}=R^{3}=H$ | C2 |
| :--- | :--- |
| $R^{1}=R^{2}=H, R^{3}=O M e$ | C3 |
| $R^{1}=R^{2}=H, R^{3}=P h$ | $C 4$ |
| $R^{1}=R^{3}=H, R^{2}=p-C_{3}-C_{6} H_{4}$ | C5 |
| $R^{1}=R^{2}=R^{3}=H$ | C6 |
| $R^{1}=H, R^{2}=R^{3}=M e$ | $C 7$ |


| Entry | Catalyst ${ }^{[a]}$ | Solvent ${ }^{[b]}$ | Temp. [ $\left.{ }^{\circ} \mathrm{C}\right]$ | Time [h] | Ratio 2a/2b | Yield $\mathbf{2 a}+\mathbf{2 b}^{[\mathrm{c}]}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | C1 | cyclohexane | 120 | 24 | 1/5.4 | 60 |
| 2 | C2 | cyclohexane | 120 | 24 | 1/3 | 66 |
| 3 | C3 | cyclohexane | 120 | 24 | 1/4.3 | 51 |
| 4 | C4 | cyclohexane | 120 | 24 | 1/3.3 | 49 |
| 5 | C5 | cyclohexane | 120 | 24 | 1/2.3 | 42 |
| 6 | C6 | cyclohexane | 120 | 24 | 1/6 | 52 |
| 7 | C7 | cyclohexane | 120 | 24 | 1/8 | 89 |
| 8 | C7 | cyclohexane | 100 | 24 | 1/8 | 54 |
| 9 | C7 | 2-Me-THF | 120 | 24 | 1/5 | 38 |
| 10 | C7 | mesitylene | 120 | 24 | n.d. | 15 |
| 11 | C7 | Acetonitrile | 120 | 24 | n.d. | trace |
| 12 | C7 | EtOAc | 120 | 24 | 1/3.2 | 20 |
| 13 | C7 | 1,4-dioxane | 120 | 24 | 1/5 | $\sim 7$ |
| 14 | C7 | $\mathrm{Bu}_{2} \mathrm{O}$ | 120 | 24 | 1/7 | 31 |
| 15 | C7 | CPME | 120 | 24 | 1/8 | 21 |
| 16 | tmphen/[ $[\mathrm{Ir}(\mathrm{Cl}) \text { cod }]_{2}$ | cyclohexane | 120 | 24 | 1/6.4 | 70 |
| 17 | tmphen/[Ir(cod)2]BF4 | cyclohexane | 120 | 24 | 1/5 | 69 |
| 18 | dtbpy/ $[\mathrm{Ir}(\mathrm{Cl}) \mathrm{cod}]_{2}$ | cyclohexane | 120 | 24 | 1/4.7 | 80 |

2-(Dibenzo[c,g]phenanthren-9-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (2a).
A yellowish amorphous solid: m.p. $153.1{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 600 MHz , Chloroform-d) $\delta 9.03$ (s, 1H), 8.47 (d, $J=8.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.96-7.92(\mathrm{~m}, 3 \mathrm{H})$, $7.91(\mathrm{~s}, 2 \mathrm{H}), 7.89-7.86(\mathrm{~m}, 4 \mathrm{H}), 7.49(\mathrm{ddd}, J=7.9,6.7,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 7.21$ (ddd, $J=8.3,6.9,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.26(\mathrm{~d}, J=14.9 \mathrm{~Hz}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(151$ $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 136.96,134.30,132.64,132.29,132.14,130.94,130.88$, 129.90 , 128.69, 127.79, 127.65, 127.54, 127.37, 127.26, 127.21, 127.05 (2C), 126.88, 126.18, 126.04, 124.21, 83.65 (2C), 25.24 (2C), 24.14 (2C); EI-MS $\mathrm{m} / \mathrm{z}(\%) 404\left(\mathrm{M}^{+}, 21\right), 320$ (15), 303 (20), 276 (100), 138 (8); HRMS (EI+) calcd for $\mathrm{C}_{28} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{~B} 404.1948$, found 404.1951.

2-(Dibenzo[c,g]phenanthren-8-yl)-4,4,5,5-tetramethyl-1,3,2-dioxaborolane (2b).


A colorless crystals: m.p. $201.8{ }^{\circ} \mathrm{C}(\mathrm{DCM} / \mathrm{MeOH}) ;{ }^{1} \mathrm{H}$ NMR $(600 \mathrm{MHz}$, Chloroform-d) $\delta 8.51-8.45(\mathrm{~m}, 3 \mathrm{H}), 8.00-7.83(\mathrm{~m}, 7 \mathrm{H}), 7.64(\mathrm{dd}, J=8.4$, $1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.51$ (ddd, $J=7.9,6.8,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 7.28$ (ddd, $J=8.4,7.1,1.3$ $\mathrm{Hz}, 1 \mathrm{H}), 1.42(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 12 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(151 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 135.77$, 132.89, 132.69, 132.56, 132.20, 131.80, 130.68, 129.33, 129.09, 128.06, 127.92, 127.77, 127.65, 127.51, 127.18 (2C), 126.79, 126.28, 126.26 (2C), 124.52, 83.96 (2C), 25.00 (2C), 24.90 (2C); EI-MS m/z (\%) 404 ( $\mathrm{M}^{+}, 38$ ), 302 (27), 276 (100), 138 (6); HRMS (EI+) calcd for $\mathrm{C}_{28} \mathrm{H}_{25} \mathrm{O}_{2} \mathrm{~B} 404.1948$, found 404.1945.

8,13-Bis(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)dibenzo[c,g]phenanthrene (2c).
A yellowish amorphous solid: m.p. $183.8{ }^{\circ} \mathrm{C} ;{ }^{1} \mathrm{H}$ NMR ( 300 MHz ,
 Chloroform- $d$ ) $\delta 8.48-8.42(\mathrm{~m}, 2 \mathrm{H}), 8.42$ (d, $J=8.5 \mathrm{~Hz}, 2 \mathrm{H}), 7.96$ (d, $J$ $=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.90-7.83(\mathrm{~m}, 4 \mathrm{H}), 7.62(\mathrm{dd}, J=8.5,1.4 \mathrm{~Hz}, 2 \mathrm{H}), 1.42$ $(\mathrm{d}, J=3.4 \mathrm{~Hz}, 24 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{cdcl}_{3}$ ) $\delta 135.64$ (2C), 132.83 (2C), 131.80 (2C), 129.48 (4C), 128.20 (2C), 127.96 (4C), 127.60 (2C), 126.17 (2C), 83.95 (4C), 24.96 (8C; +ESI-MS m/z (\%) 553 (M + Na, 100), $531(\mathrm{M}+\mathrm{H}, 30)$; HRMS (+ESI) calcd for $\mathrm{C}_{34} \mathrm{H}_{37} \mathrm{O}_{4} \mathrm{~B}_{2} 531.28725$, found 531.28726.

## 5. Cross-coupling reactions of 2a or 2b with aryl iodides.

To a solution of $\mathbf{2 a}$ or $\mathbf{2 b}(20.2 \mathrm{mg}, 0.05 \mathrm{mmol})$ in toluene $/ \mathrm{H}_{2} \mathrm{O}(10 / 1,1.5 \mathrm{~mL})$ was added XPhos-Pd-G3 ( $2.2 \mathrm{mg}, 5 \mathrm{~mol} \%$ ), $\mathrm{K}_{3} \mathrm{PO}_{4}(21.3 \mathrm{mg}, 200 \mathrm{~mol} \%)$ and an aryl iodide ( 0.065 mmol ) and the reaction mixture was stirred at $23^{\circ} \mathrm{C}$ for 16 h . All volatiles were evaporated, the residue redissolved in DCM and separated by preparative TLC (hexane/DCM 10:1) to give the corresponding product $\mathbf{3}$ or 4 .

9-(p-Tolyl)dibenzo[c,g|phenanthrene (3a). Yield 16.4 mg (89\%); a yellowish amorphous solid: m.p. $215.6^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , Chloroform- $d$ ) $\delta 8.78$ (s, 1H), 8.71 (d, $J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.97-8.02(\mathrm{~m}, 2 \mathrm{H}), 7.93(\mathrm{dd}, J=8.5,3.3 \mathrm{~Hz}$, $2 \mathrm{H}), 7.89-7.84(\mathrm{~m}, 4 \mathrm{H}), 7.77(\mathrm{dd}, J=8.3,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.59-7.52(\mathrm{~m}$, $1 \mathrm{H}), 7.33-7.37(\mathrm{~m}, 3 \mathrm{H}), 7.13(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 2.33(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 138.13,136.87$ (2C), 132.72, 132.52, 132.27, 131.62, 131.01, 130.89, 129.36 (2C), 128.82, 128.32, 128.06, 127.46, 127.28 (2C), 127.15 (2C), 127.07 (2C), 126.97, 126.40, 126.23, 126.16, 125.48, 124.61, 21.02; EI-MS m/z (\%) 368 ( $\mathrm{M}^{+}$, 100), 350 (23), 337 (13), 276 (89), 263 (14); HRMS (EI+) calcd for $\mathrm{C}_{29} \mathrm{H}_{20} 368.1565$, found 368.1567.

9-(4-Methoxyphenyl)dibenzo[c,g]phenanthrene (3b). Yield 17.7 mg (92\%); a yellowish amorphous solid: m.p. $176.2{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR ( 600 MHz , Chloroform-d) $\delta$ 8.74 (d, $J=1.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.71$ (d, $J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.99(\mathrm{~d}, J=8.3 \mathrm{~Hz}$, $1 \mathrm{H}), 7.98$ (dd, $J=7.9,1.3 \mathrm{~Hz}, 1 \mathrm{H}) 7.93$ (d, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 7.89-7.84$ (m, 4H), 7.74 (dd, $J=8.3,1.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.57 (ddd, $J=8.0,6.8,1.1 \mathrm{~Hz}$, $1 \mathrm{H}), 7.38(\mathrm{dm}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.35(\mathrm{ddd}, J=8.3,6.8,1.4 \mathrm{~Hz}, 1 \mathrm{H})$, $6.85(\mathrm{dm}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 3.80(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.05,136.55$, $133.61,132.72,132.54,132.26,131.40,131.03,130.87,128.84,128.33,128.26$ (2C), 128.09, 127.44, 127.29, 127.28, 127.16, 127.12, 127.06, 126.66, 126.44, 126.16, 126.10, 125.31, 124.56, 114.09, 55.29; EI-MS m/z (\%) 384 (M ${ }^{+}$, 100), 351 (12), 337 (17), 276 (95); HRMS (EI+) calcd for $\mathrm{C}_{29} \mathrm{H}_{20} \mathrm{O} 384.1514$, found 368.1512 .

8-(p-Tolyl)dibenzo[c,g]phenanthrene (4a). Yield $16.6 \mathrm{mg}(90 \%)$; a colorless crystals: m.p.
 $198.8{ }^{\circ} \mathrm{C}(\mathrm{DCM} / \mathrm{MeOH}) ;{ }^{1} \mathrm{H}$ NMR ( 400 MHz , Chloroform- $d$ ) $\delta 8.60$ (dt, $J$ $=8.4,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.57(\mathrm{dd}, J=8.8,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.16(\mathrm{~d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H})$, $8.03-7.85(\mathrm{~m}, 7 \mathrm{H}), 7.70(\mathrm{dm}, J=8.2 \mathrm{~Hz}, 2 \mathrm{H}), 7.61-7.50(\mathrm{~m}, 2 \mathrm{H}), 7.39$ $-7.28(\mathrm{~m}, 3 \mathrm{H}), 2.45(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 138.59, $137.68,137.26,132.99,132.60,132.32,132.24,130.79,129.80,129.61$ (2C), 129.43, 129.02, 127.81, 127.69, 127.48, 127.23, 127.14, 127.03 (2C), 126.98, 126.96, 126.70, 126.35, 126.25, 125.17, 124.40, 123.49, 21.15; EI-MS m/z (\%) $368\left(\mathrm{M}^{+}, 100\right), 350$ (24), 337 (20), 276 (41), 263 (30); HRMS (EI+) calcd for $\mathrm{C}_{29} \mathrm{H}_{20}$ 368.1565 , found 368.1563 .

8-(4-Methoxyphenyl)dibenzo[c,g]phenanthrene (4b). Yield 18.1 mg (94\%); a yellowish
 crystals: m.p. $216.8{ }^{\circ} \mathrm{C}(\mathrm{DCM} / \mathrm{MeOH}) ;{ }^{1} \mathrm{H}$ NMR ( 600 MHz , Chloroform- $d$ ) $\delta 8.58(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.54(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.11$ $(\mathrm{d}, J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.00-7.85(\mathrm{~m}, 7 \mathrm{H}), 7.73(\mathrm{dm}, J=8.7 \mathrm{~Hz}, 2 \mathrm{H})$, $7.55-7.50(\mathrm{~m}, 2 \mathrm{H}), 7.30(\mathrm{ddd}, J=8.4,6.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.04(\mathrm{dm}, J=$ $8.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.89(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 159.33$, $138.27,133.10,133.04,132.60$, 132.33, 132.19, 130.80, 129.58, $129.45,129.01,128.24$ (2C), 127.82, 127.66, 127.48, 127.25, 127.09,
126.98, 126.96, 126.72, 126.36, 126.25, 124.79, 124.40, 123.32, 114.35 (2C), 55.39; EI-MS $\mathrm{m} / \mathrm{z}$ (\%) 384 ( $\mathrm{M}^{+}$, 100), 351 (23), 337 (25), 276 (45), 263 (26); HRMS (EI+) calcd for $\mathrm{C}_{29} \mathrm{H}_{20} \mathrm{O}$ 384.1514, found 368.1510.

8-(4-(Trifluoromethyl)phenyl)dibenzo[c,g]phenanthrene (4c). Yield 18.2 mg (86\%); a
 yellowish amorphous solid: m.p. $210.5{ }^{\circ} \mathrm{C}$; ${ }^{1} \mathrm{H}$ NMR $(600 \mathrm{MHz}$, Chloroform- $d$ ) $\delta 8.59(\mathrm{~d}, J=8.8 \mathrm{~Hz}, 1 \mathrm{H}), 8.55(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 8.16$ (d, $J=2.0 \mathrm{~Hz}, 1 \mathrm{H}), 8.00-7.84(\mathrm{~m}, 9 \mathrm{H}), 7.75(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.55$ (ddd, $J=8.0,6.8,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 7.51(\mathrm{dd}, J=8.8,2.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.31$ (ddd, $J=8.4,6.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}$ ); ${ }^{13} \mathrm{C}$ NMR ( $151 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ 144.11, 137.05 , $132.86,132.68,132.52,132.40,130.69,130.50,129.75,129.45$ (q, $J=$ 33 Hz ), 128.94, 127.92, 127.65, 127.59, 127.55, 127.44 (2C), 127.22, 127.11, 126.98, 126.80, 126.37, 126.34, 126.03, 125.79 (q, $J=3.8 \mathrm{~Hz}$ ), 124.49, 124.33 (q, $J=270 \mathrm{~Hz}$ ), 123.24; APCI-MS m/z (\%) 423 (M + H, 100); HRMS (APCI) calcd for $\mathrm{C}_{29} \mathrm{H}_{18} \mathrm{~F}_{3} 423.13551$, found 423.13562.

## 6. X-ray data.

Crystallographic data for $\mathbf{2 b}$ compound were collected on Bruker D8 VENTURE Kappa Duo PHOTON 100 by $\mathrm{I} \mu \mathrm{S}$ with Mo K $\alpha 0.71073 \AA$ at a temperature of $120(2) \mathrm{K}$. The absorption corrections were carried on using multi-scan procedure.
The structures were solved by direct methods (XP) ${ }^{[4]}$ and refined by full matrix least squares based on $F^{2}$ (SHELXL2014). ${ }^{[5]}$ The hydrogen atoms on carbons were fixed into idealised positions (riding model) and assigned temperature factors either $H_{\text {iso }}(\mathrm{H})=1.2 \mathrm{U}_{\mathrm{eq}}$ (pivot atom) or $\operatorname{Hiso}(\mathrm{H})=1.5 \mathrm{U}_{\mathrm{eq}}$ for methyl moiety. The crystal was refined as non-merohedral twin. Ration of twin parts : 0.53:0.47.

Single crystals of 2b suitable for X-ray analysis were grown via methanol diffusion into a concentrated $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ solution of $\mathbf{2 b}$.

Crystal data for 2b: $\mathrm{C}_{28} \mathrm{H}_{25} \mathrm{BO}_{2}, M_{\mathrm{r}}=404.29$; Triclinic, $P-1$ (No 2), $a=8.0397$ (5) $\AA$, $b=$ 9.6270 (6) $\AA, c=14.3024$ (8) $\AA, \alpha=81.219(2)^{\circ}, \beta=79.289$ (2) ${ }^{\circ}, \gamma=86.038$ (2) $)^{\circ}, V=$ $1074.00(11) \AA^{3}, \mathrm{Z}=2, D_{\mathrm{x}}=1.250 \mathrm{Mg} \mathrm{m}^{-3}$, colourless prism of dimensions $0.54 \times 0.20 \times$ 0.18 mm , multi-scans absorption correction $\left(\mu=0.08 \mathrm{~mm}^{-1}\right) T_{\min }=0.94, T_{\max }=0.99$; a total of 25588 measured reflections ( $\theta_{\max }=27.5^{\circ}$ ), from which 4946 were unique ( $R_{\mathrm{int}}=0.045$ ) and 4261 observed according to the $I>2 \sigma(I)$ criterion. The refinement converged ( $\Delta / \sigma_{\max }=$ 0.001 ) to $R=0.042$ for observed reflections and $\mathrm{w} R\left(F^{2}\right)=0.105, G O F=1.06$ for 285 parameters and all 4946 reflections. The final difference map displayed no peaks of chemical significance $\left(\Delta \rho_{\max }=0.25, \Delta \rho_{\min }-0.21\right.$ e. $\AA^{-3}$ ).

CCDC 1587494 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.


Figure SI-1. View of the molecule of $\mathbf{2 b}$ with atom numbering scheme. The dispacement ellipsoids are drawn with $30 \%$ probability level.

## 7. References.

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## 8. $\quad{ }^{1} H$ and ${ }^{13} C$ NMR spectra.

${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{2 a}$

${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{2 a}$


2a





[^0]
## ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{2 b}$


${ }^{1} \mathrm{H}$ NMR spectrum of 2c


${ }^{13} \mathrm{C}$ NMR spectrum of 2c



## ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{3 a}$






${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{3 a}$


## ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{3 b}$


${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{3 b}$

${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{4 a}$

${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{4 a}$


## ${ }^{1} H$ NMR spectrum of $\mathbf{4 b}$



${ }^{13} \mathrm{C}$ NMR spectrum of $\mathbf{4 c}$

4c

 よ


$\begin{array}{llllllllllllllllllllllllllllllllll}133.0 & 132.5 & 132.0 & 131.5 & 131.0 & 130.5 & 130.0 & 129.5 & 129.0 & 128.5 & 128.0 & 127.5 & 127.0 & 126.5 & 126.0 & 125.5 & 125.0 & 124.5 & 124.0 & 123.5 & 123.0 & 122.5 & 122.0 & 121.5\end{array}$



[^0]:    ## 1

    $200 \quad 190$ 180 $170 \quad 160$ 150
    $130 \quad 1$

    110
    $\mathrm{f} 1(\mathrm{ppm})$
    90

