

Supplementary Information

Kinetic and mechanism study on gas phase reactions of ozone with a series of *cis*-3-hexenyl esters

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SUMMARY:

11 pages, including 12 figures and 1 tables.

Figure Captions:

Fig. S1 Schematic of the experimental setup and instrumentation

Fig.S2 One example of comparative experiments in the reaction of O₃ with *cis*-3-hexenyl acetate with and without cyclohexane as OH scavenger.

Fig. S3 Pseudo-first-order plots for O₃ reactions with different concentrations of Isoprene (in molecule cm⁻³).

Fig. S4 Pseudo-first-order plots for O₃ reactions with different concentrations of *cis*-3-Hexenyl Formate (in molecule cm⁻³).

Fig. S5 Pseudo-first-order plots for O₃ reactions with different concentrations of *cis*-3-Hexenyl propionate (in molecule cm⁻³).

Fig. S6 Pseudo-first-order plots for O₃ reactions with different concentrations of *cis*-3-Hexenyl Butyrate (in molecule cm⁻³).

Fig. S7 The structures for the reactants, transition states and products along the *cis*-3-hexenyl acetate + O₃ reaction profile.

Fig. S8 The structures for the reactants, transition states and products along the *cis*-3-hexenyl propionate + O₃ reaction profile.

Fig. S9 The structures for the reactants, transition states and products along the *cis*-3-hexenyl butyrate + O₃ reaction profile.

Fig. S10 BHandHLYP/6-311+G(d,p) energy profile (in kJ mol⁻¹) for the *cis*-3-hexenyl acetate ozonolysis.

Fig. S11 BHandHLYP/6-311+G(d,p) energy profile (in kJ mol⁻¹) for the *cis*-3-hexenyl propionate ozonolysis.

Fig. S12 BHandHLYP/6-311+G(d,p) energy profile (in kJ mol⁻¹) for the *cis*-3-hexenyl butyrate ozonolysis.

Fig.S1

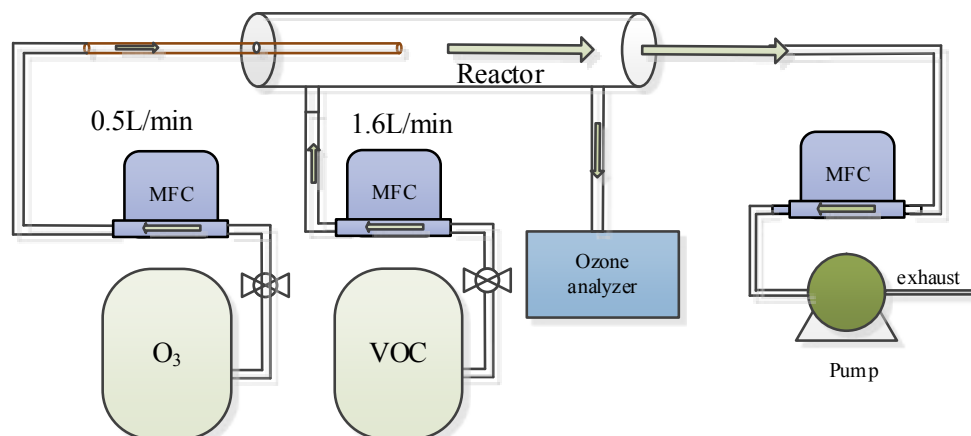
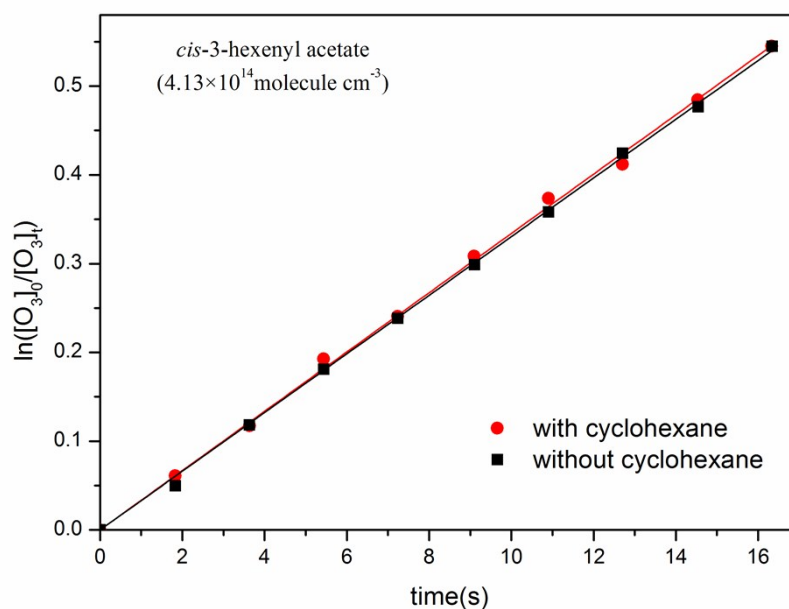


Fig. S2



Comparative experiments with and without cyclohexane as OH scavenger were conducted, and no significant difference was found in the rate constants. Fig.S2 gives one example of comparative experiments in the reaction of O_3 with 4.13×10^{14} molecule cm^{-3} *cis*-3-hexenyl acetate. As can be seen, regardless of the addition of excess cyclohexane (6.36×10^{15} molecule cm^{-3}) or not, the slope values of the lines (k') could be considered to be the same within the error.

Fig. S3

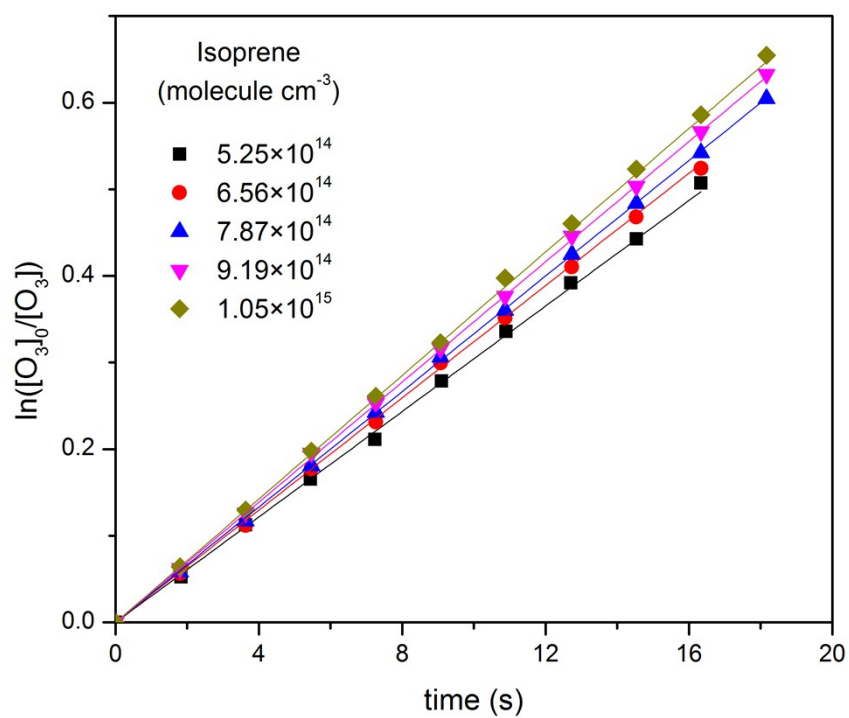


Fig. S4

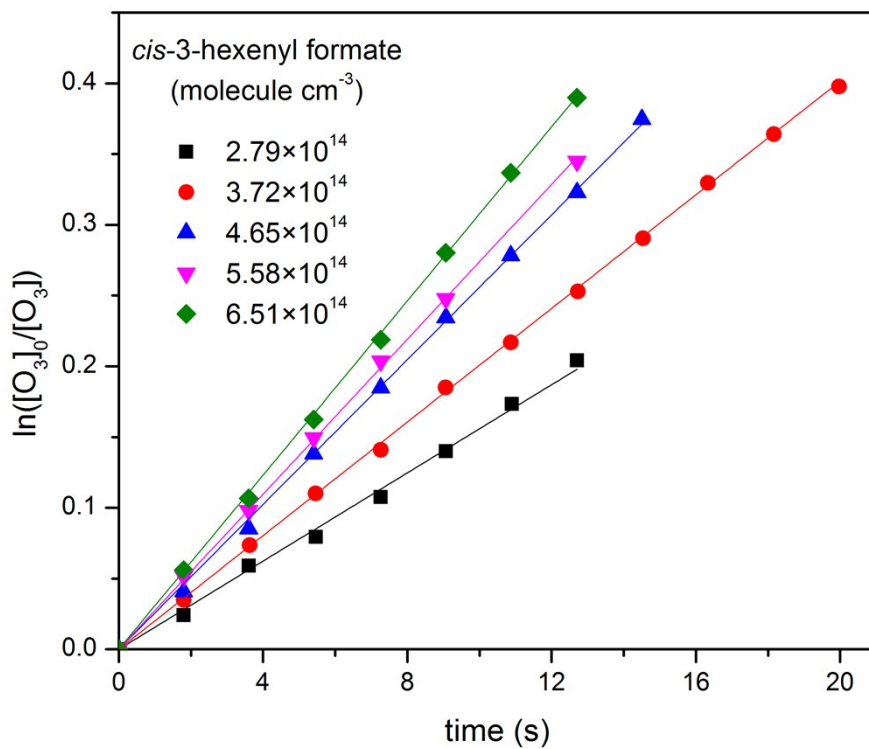


Fig. S5

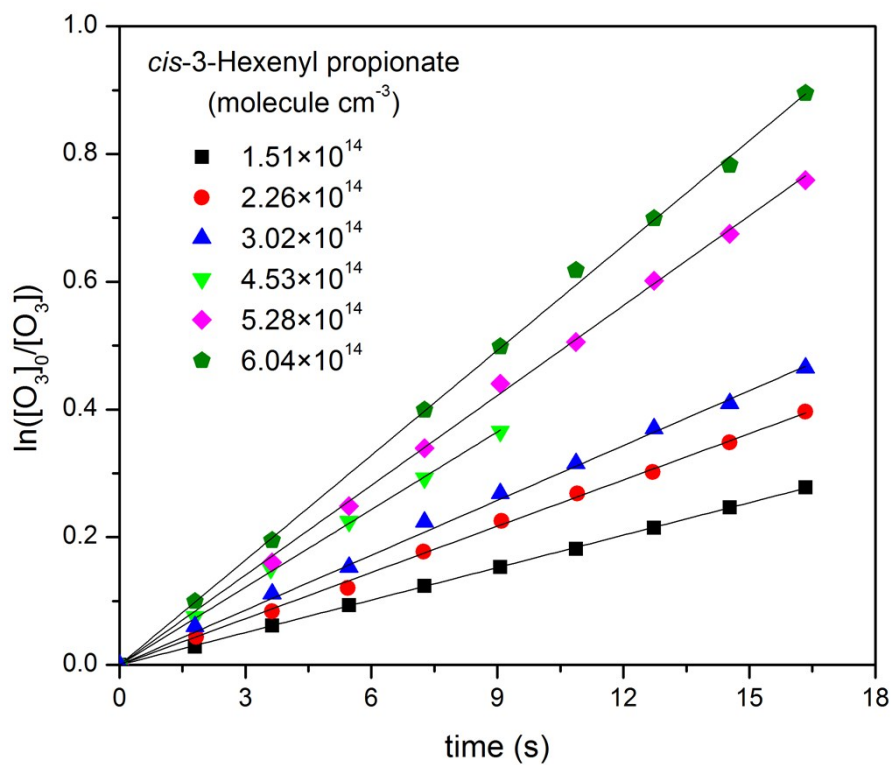


Fig.S6

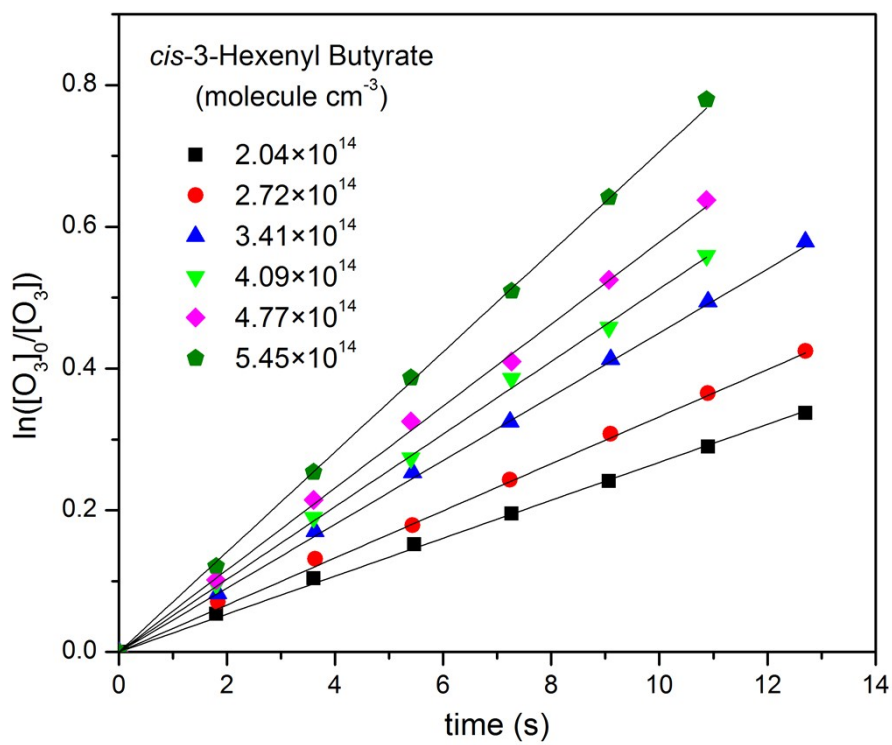


Fig. S7

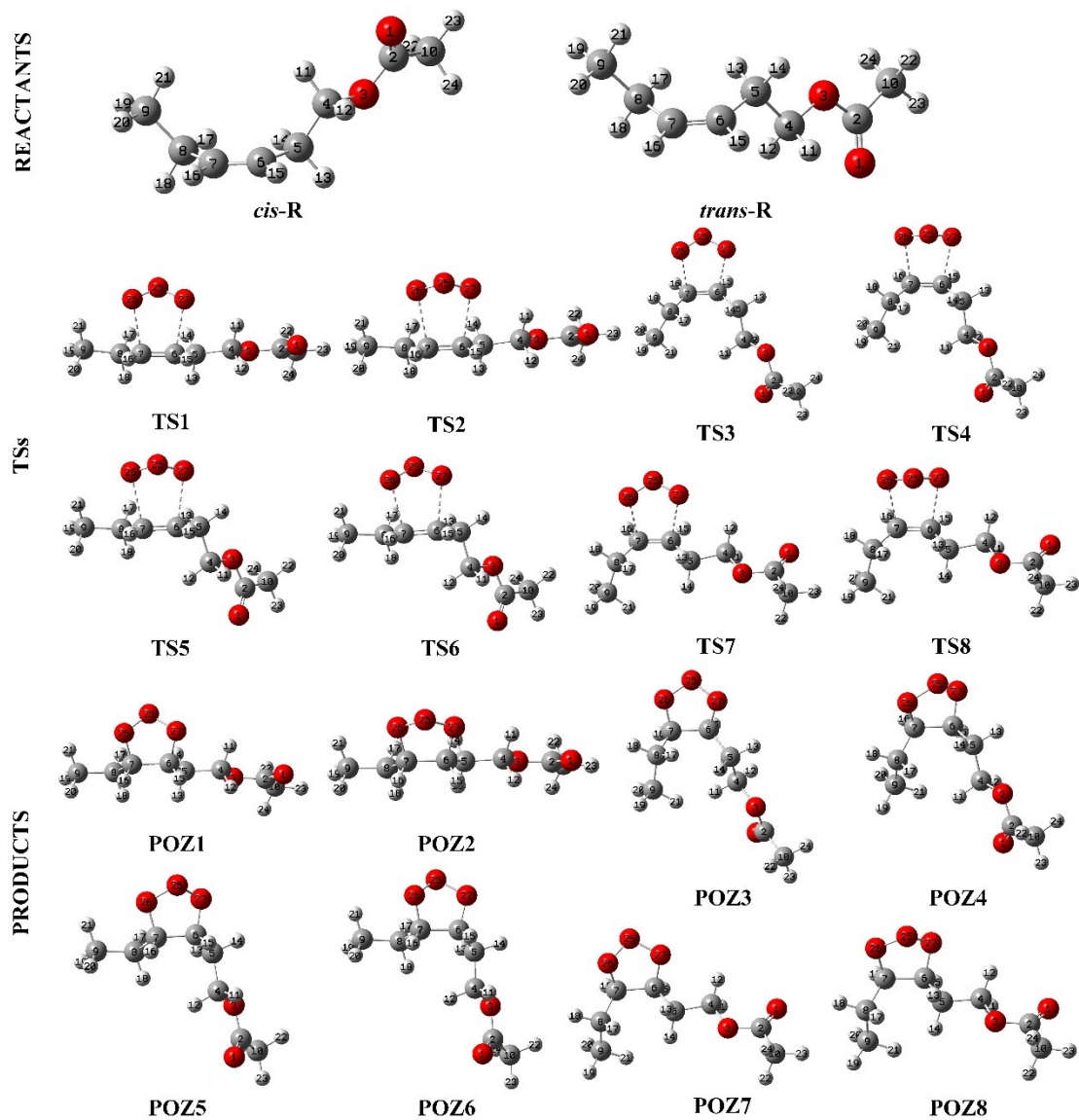


Fig.S8

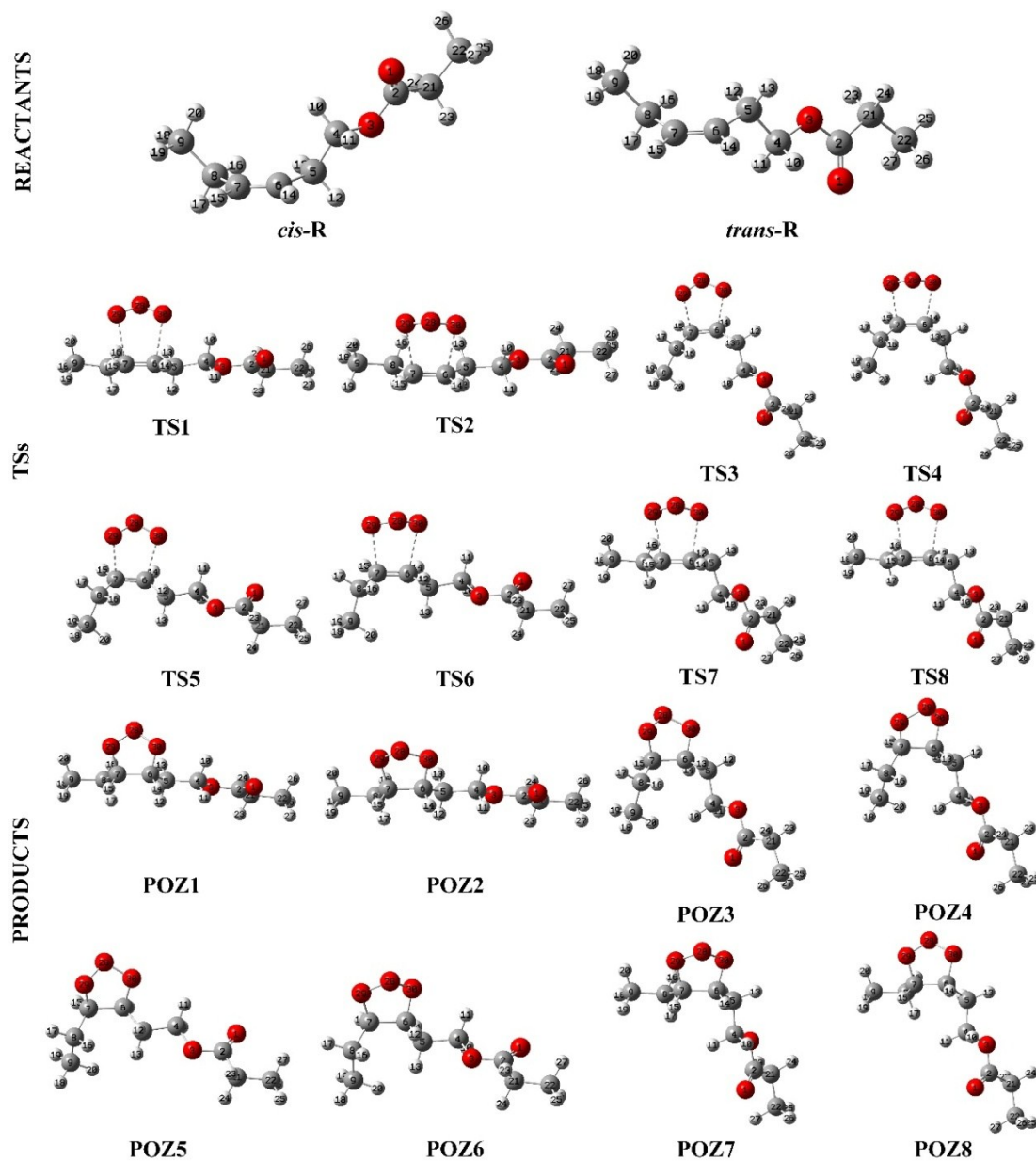


Fig.S9

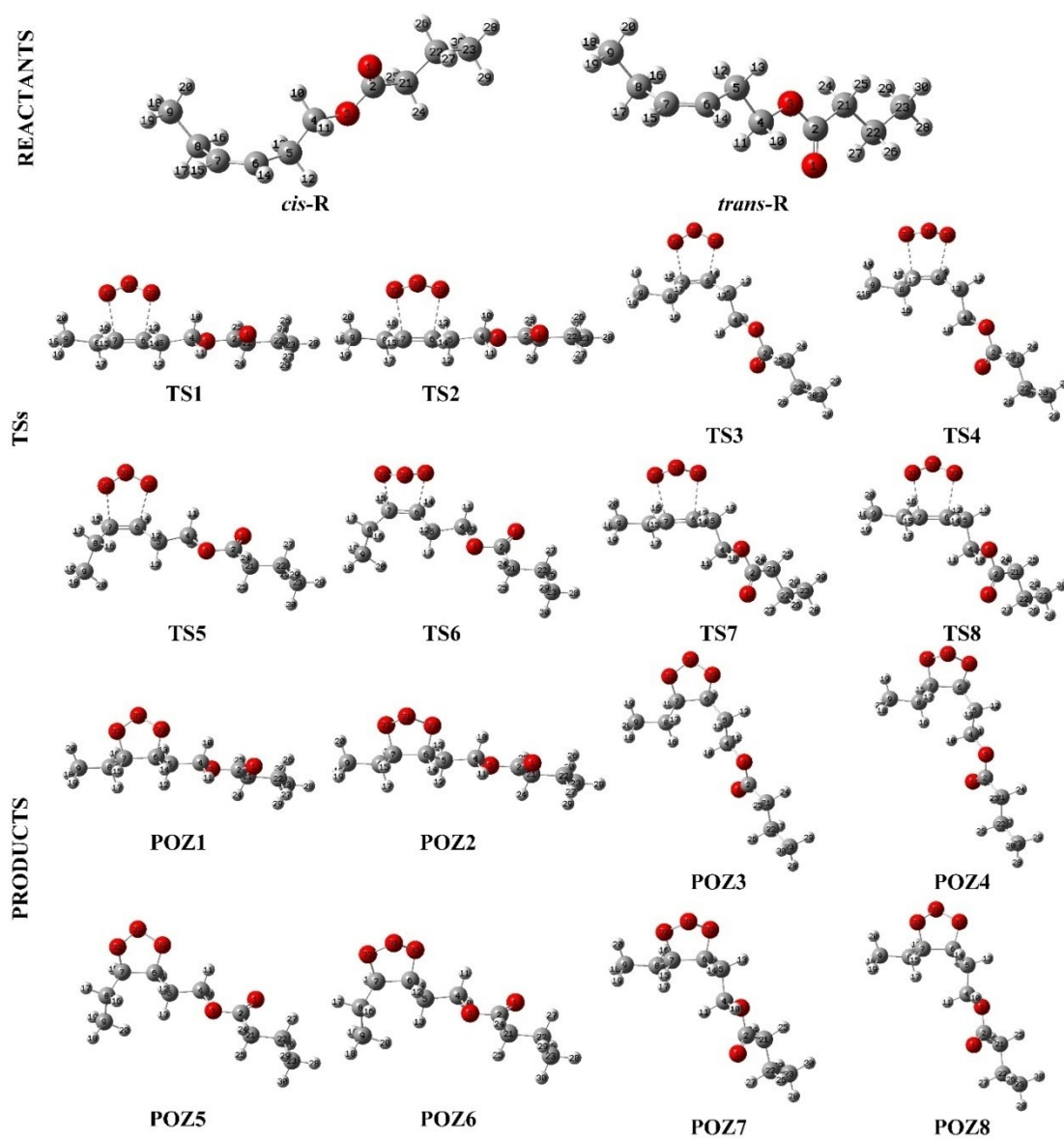


Fig.S10

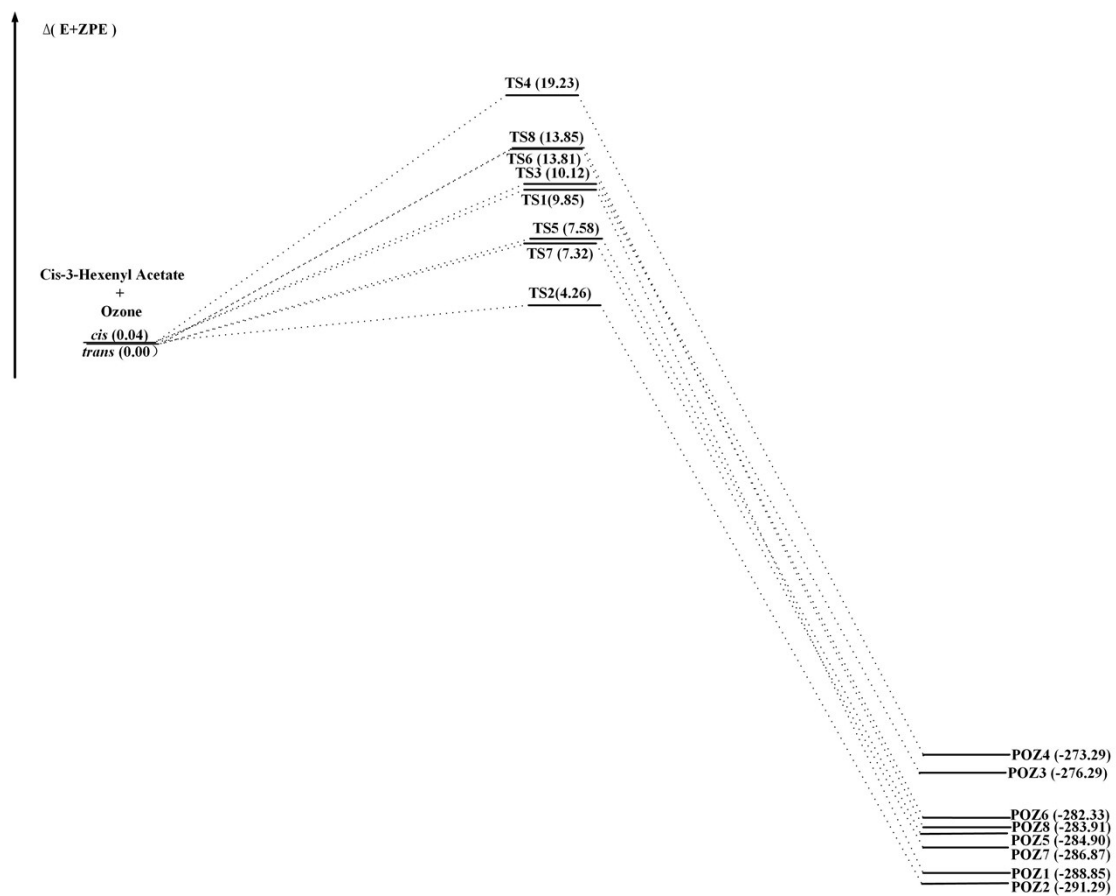


Fig.S11

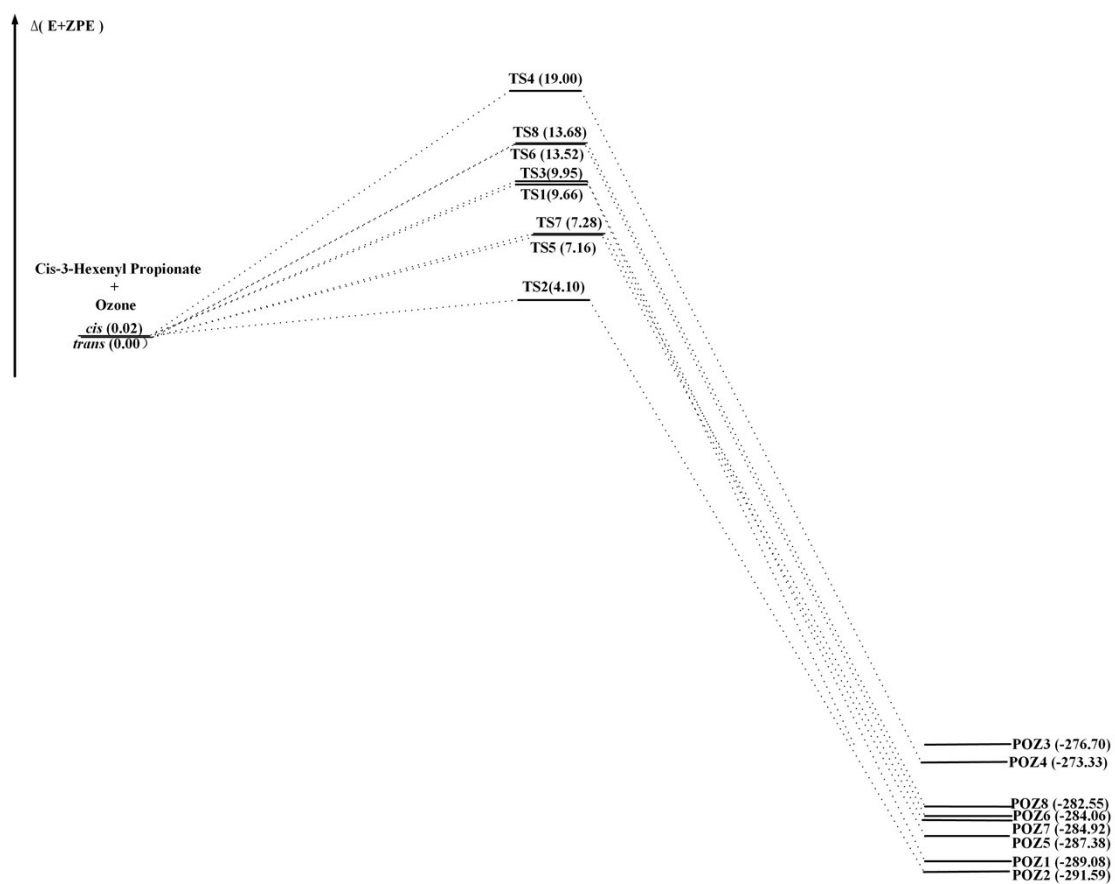


Fig.S12

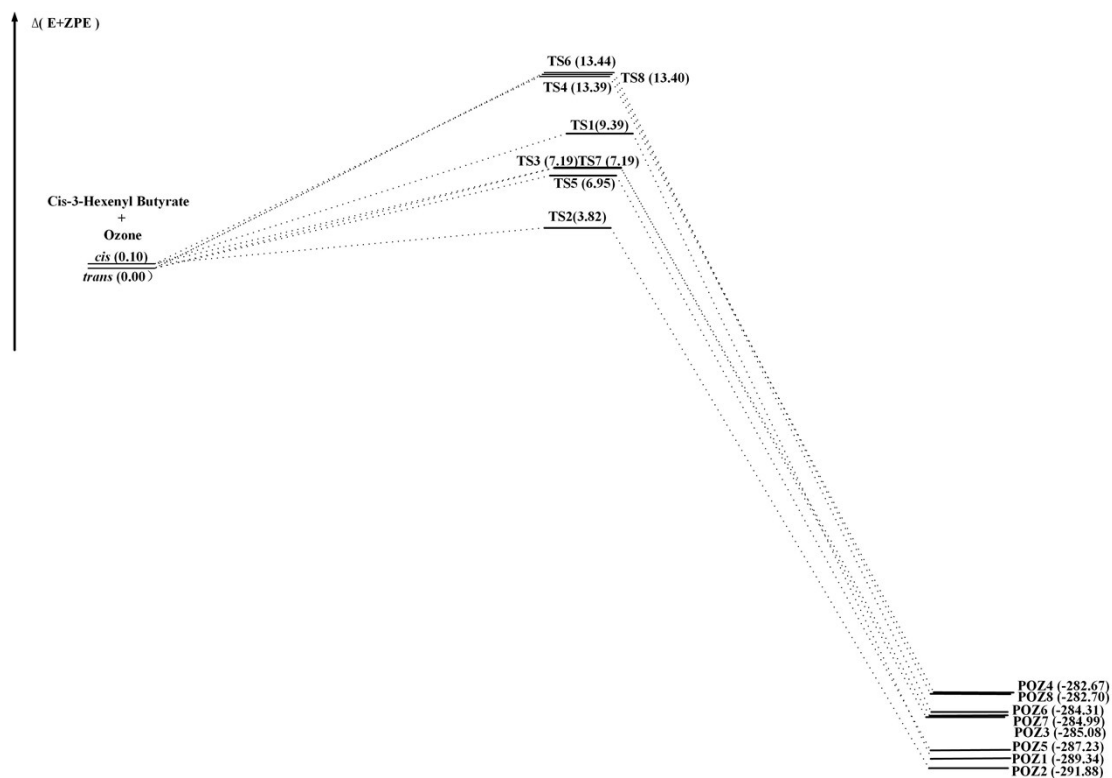


Table S1 The HOMO energies of the unsaturated esters and the parent hexene calculated at HF/6-31G** level of theory.

Compounds	HOMO energies
cis-3-hexenyl formate	-0.35465 Hartree (-9.65052 eV)
cis-3-hexenyl acetate	-0.35205 Hartree (-9.57977 eV)
cis-3-hexenyl propionate	-0.35174 Hartree (-9.57133 eV)
cis-3-hexenyl butyrate	-0.35159 Hartree (-9.56725 eV)
cis-3-hexene	-0.34030 Hartree (-9.26003 eV)