

## **Electronic Supplementary Information**

### **2D MoS<sub>2</sub>/Polyaniline Heterostructures with Enlarged Interlayer Spacing for Superior Lithium and Sodium Storage**

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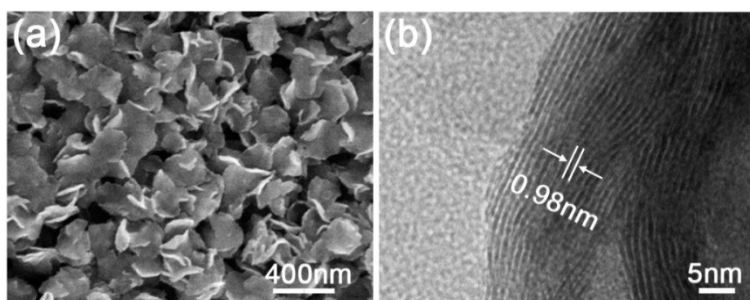
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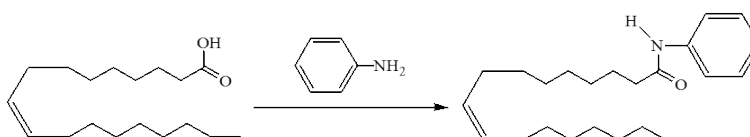
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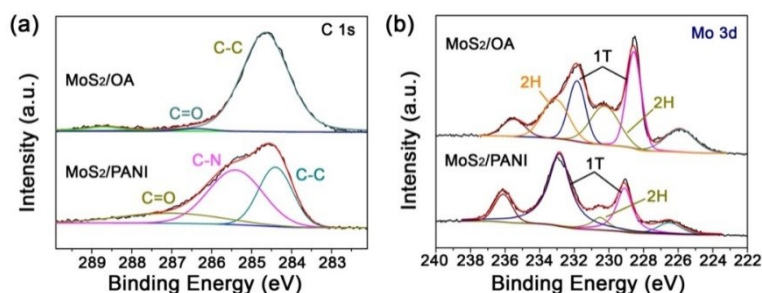
## 1. Figures



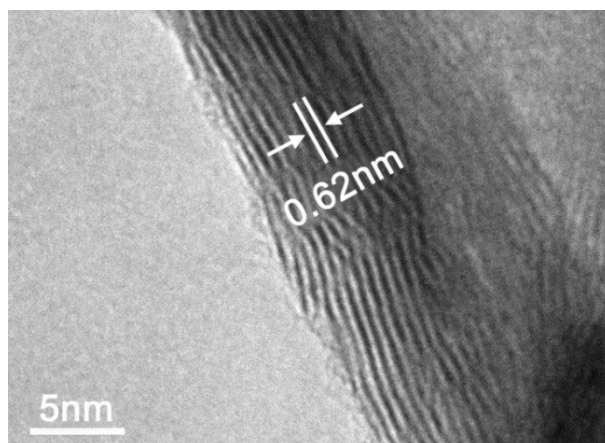
**Fig. S1** (a) SEM and (b) TEM images of the MoS<sub>2</sub>/OA nanosheets.



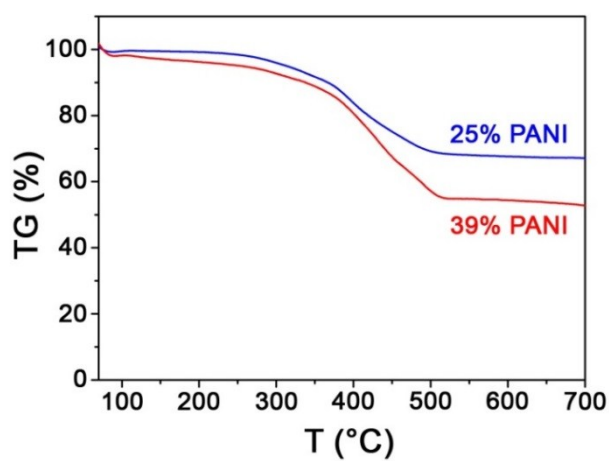
**Fig. S2** The amidation reaction equation of OA and aniline.



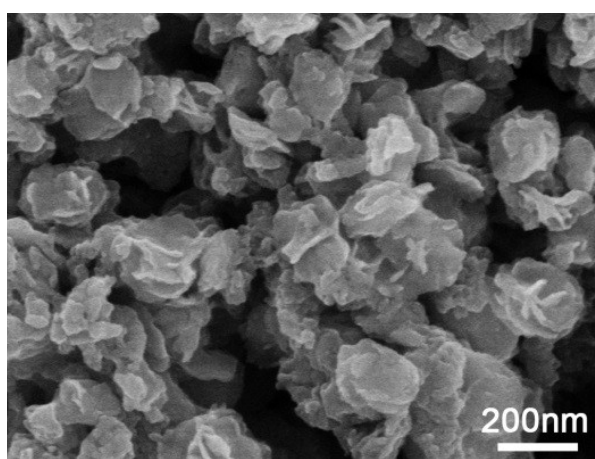
**Fig. S3** XPS spectra of MoS<sub>2</sub>/OA and MoS<sub>2</sub>/PANI for (a) C and (b) Mo. As depicted, after the reaction with PANI, the peak corresponding to C=O at 286.4 eV shifts to 287 eV, conforming the amidation process between oleic acid and aniline. For Mo 3d spectra, the MoS<sub>2</sub>-OA shows large double peaks at 228.6 eV (3d<sub>5/2</sub>) and 231.8 eV (3d<sub>3/2</sub>) and small double peaks at 230.2 eV (3d<sub>5/2</sub>) and 233.1 eV (3d<sub>3/2</sub>) as an evidence for the formation of MoS<sub>2</sub> consisting of the 1T and 2H phases. As for MoS<sub>2</sub>/PANI, the peaks indicate an increasing of 1T phase and a decreasing of 2H phase of MoS<sub>2</sub>, suggesting a nearly monolayer MoS<sub>2</sub> nanosheet, which is mainly benefit form the PANI layer in the MoS<sub>2</sub> interlayer effectivley avoiding their stacking/restacking. The peak Mo<sup>6+</sup> 3d<sub>5/2</sub> at 235.5 eV emerges because of partial oxidation during fabrication process. Moreover, both C and Mo peaks show a slight shift to high energy regions, indicating an electronic interaction between MoS<sub>2</sub> and PANI.



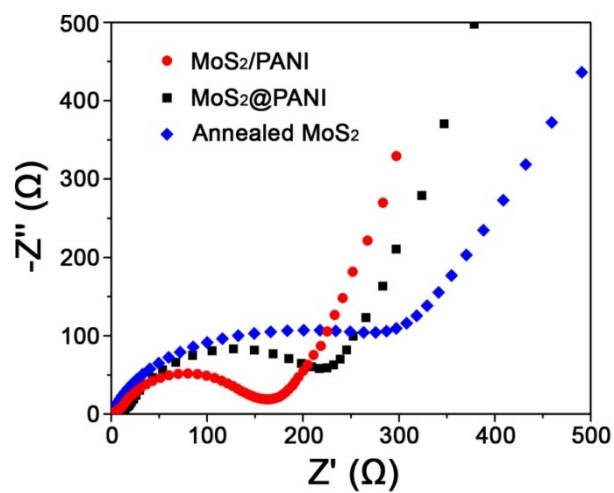
**Fig. S4** TEM image of the annealed MoS<sub>2</sub> nanosheets.



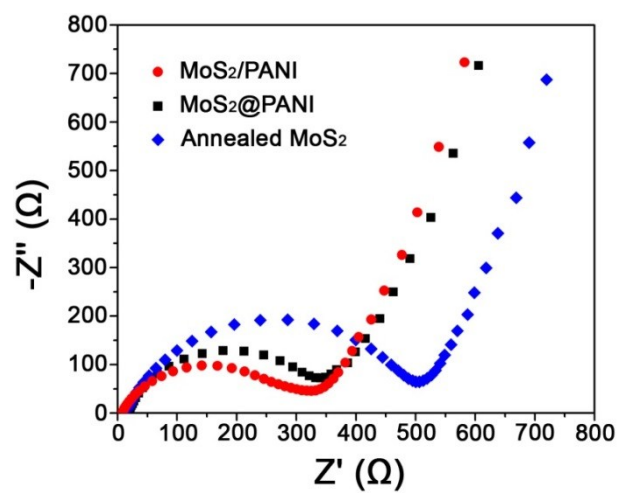
**Fig. S5** TG curves of the MoS<sub>2</sub>/PANI hybrids with different PANI content.



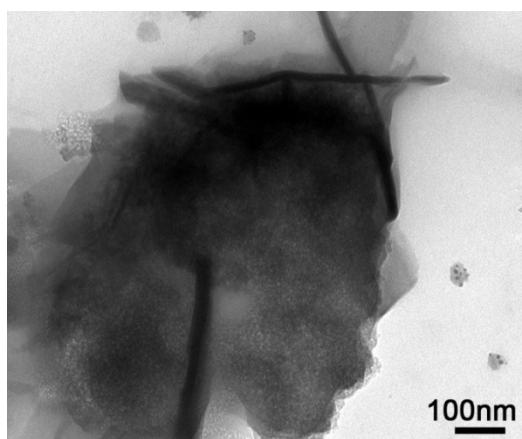
**Fig. S6** SEM image of the MoS<sub>2</sub>/PANI hybrids with 39% PANI content.



**Fig. S7** Nyquist plots of the MoS<sub>2</sub>/PANI hybrids, the MoS<sub>2</sub>@PANI hybrids and the annealed MoS<sub>2</sub> nanosheets in LIBs.



**Fig. S8** Nyquist plots of the MoS<sub>2</sub>/PANI hybrids, the MoS<sub>2</sub>@PANI hybrids and the annealed MoS<sub>2</sub> nanosheets in SIBs.



**Fig. S9** TEM image of the 2D MoS<sub>2</sub>/PANI nanosheets after 100 cycles in LIBs.

## 2. Tables

**Table S1.** Comparison of LIBs performances of MoS<sub>2</sub>-based anode materials.

<b>Materials</b>	<b>Interlayer spacing</b>	<b>LIBs performance</b>	<b>Ref.</b>
Mesoporous MoS <sub>2</sub>	0.66 nm	903, 845, 795 and 748 mAh g <sup>-1</sup> at 100, 500, 1000 and 2000 mA g <sup>-1</sup>	[1]
MoS <sub>2</sub> nanoplates	0.69 nm	912 mAh g <sup>-1</sup> at 1062 mA g <sup>-1</sup>	[1]
3D hierarchical MoS <sub>2</sub> /polyaniline	0.64 nm	798 mA h g <sup>-1</sup> at 100 mA g <sup>-1</sup> , 439 mA h g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[3]
MoS <sub>2</sub> /polyaniline nanowires	0.61 nm	1006 at 200 mA g <sup>-1</sup> , 320 mA h g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[4]
MoS <sub>2</sub> @C nanotubes	0.64 nm	1326.9, 1074.2, 993.1, and 929 mA h g <sup>-1</sup> at 100, 500, 1000, and 2000 mA g <sup>-1</sup>	[5]
MoS <sub>2</sub> nanoplates	Single-layer	1095, 986 mAh g <sup>-1</sup> at 500, 1000 mA g <sup>-1</sup>	[6]
3D MoS <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub> nanohybrid	Single-layer	1183, 1019 and 910 mA h g <sup>-1</sup> at 100, 500 and 1000 mA g <sup>-1</sup>	[7]
MoS <sub>2</sub> @CMK-3	0.65 nm	893, 773, 713 and 591 mAh g <sup>-1</sup> at 100, 500, 1000 and 2000 mA g <sup>-1</sup>	[8]
3D MoS <sub>2</sub> @porous carbon nanosheet	0.65 nm	1060, 950, 880 and 710 mA h g <sup>-1</sup> at 200, 500, 1000 and 2000 mA g <sup>-1</sup>	[9]
3D MoS <sub>2</sub> nanospheres	0.71 nm	1184.8, 882.7, 601.5, and 353.6 mAh g <sup>-1</sup> at 100, 500, 1000 and 2000 mA g <sup>-1</sup>	[10]
PANI/MoS <sub>2</sub> hybrids	1.08 nm	1319, 1285, 1247, 1152, 1077, 1029 and 885 mA h g <sup>-1</sup> at 100, 200, 500, 1000, 1500, 2000 and 4000 mA g <sup>-1</sup>	This work

**Table S2.** Comparison of SIBs performance of MoS<sub>2</sub>-based anode materials.

<b>Materials</b>	<b>Interlayer spacing</b>	<b>SIBs performance</b>	<b>Ref.</b>
MoS <sub>2</sub> @C nanotubes	0.64 nm	610, 560, 430 mA h g <sup>-1</sup> at 50, 100 and 1000 mA g <sup>-1</sup>	[5]
MoS <sub>2</sub> nanoplates	Singlelayer	854, 623 mA h g <sup>-1</sup> at 100 and 1000 mA g <sup>-1</sup>	[6]
TiO <sub>2</sub> -B/MoS <sub>2</sub> nanowires	0.64 nm	214 mA h g <sup>-1</sup> at 20 mA g <sup>-1</sup> , 173 mA h g <sup>-1</sup> at 100 mA g <sup>-1</sup> , 77 mA h g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[11]
Exfoliated MoS <sub>2</sub> nanosheets	0.638 nm	500 mA h g <sup>-1</sup> , 305 mA h g <sup>-1</sup> at 40 mA g <sup>-1</sup> , 320 mA g <sup>-1</sup>	[12]
3D MoS <sub>2</sub> -graphene microspheres	0.69 nm	427 mA h g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[13]
MoS <sub>2</sub> /C nanofibers	0.64 nm	400.6 mA h g <sup>-1</sup> at 50 mA g <sup>-1</sup> , 369.7 mA h g <sup>-1</sup> at 100 mA g <sup>-1</sup> , 246.5 mA h g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[14]
MoS <sub>2</sub> /C nanospheres	Not Mentioned	520 mA h g <sup>-1</sup> at 67 mA g <sup>-1</sup> , 390 mA h g <sup>-1</sup> at 1340 mA g <sup>-1</sup>	[15]
MoS <sub>2</sub> @C paper	0.62 nm	446, 205 mA h g <sup>-1</sup> at 20, 1000 mA g <sup>-1</sup>	[16]
HfO <sub>2</sub> -coated MoS <sub>2</sub> nanosheet	0.62 nm	613 mAh g <sup>-1</sup> at 100 mA g <sup>-1</sup> , 347 mAh g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[17]
MoS <sub>2</sub> nanoflowers	0.69 nm	200 mAh g <sup>-1</sup> at 1000 mA g <sup>-1</sup>	[18]
PANI/MoS <sub>2</sub> hybrids	1.08 nm	734, 634, 584, 510, 465 and 391 mA h g <sup>-1</sup> at 20, 50, 100, 300, 600 and 1000 mA g <sup>-1</sup>	This work

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**Movies S1-S4.**