

## Electronic Supplementary Information

### **Oxidation of Phosphinidene Oxide: Formation of Dioxaphosphirane Oxide with Oxygen Scrambling**

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## Experimental details

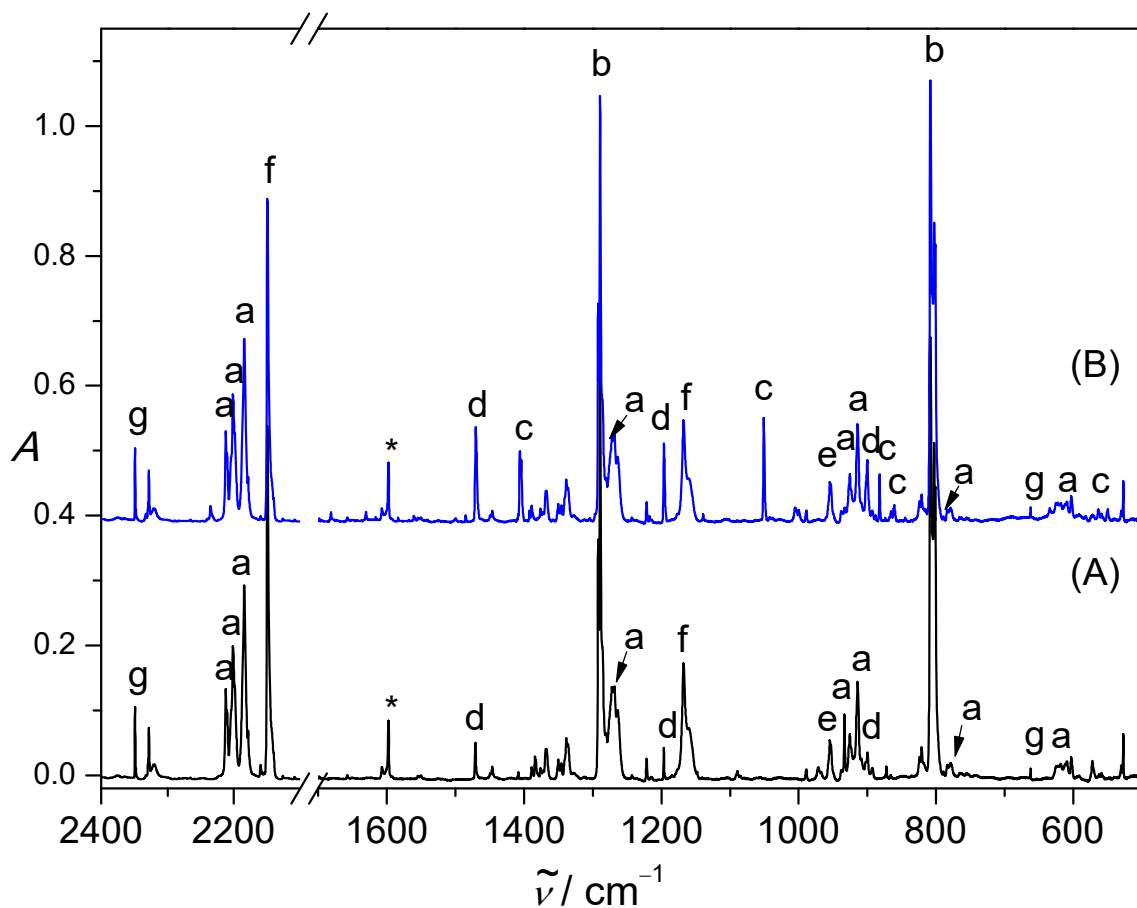
**Sample preparation.** Fluorophosphoryl diazide,  $\text{FP}(\text{O})(\text{N}_3)_2$ , was prepared and purified according to the published protocol.<sup>[1]</sup> Ar ( $\geq 99.999\%$ , Messer),  $\text{N}_2$  ( $\geq 99.999\%$ , Messer), Ar ( $\geq 99.999\%$ , Messer), Ne ( $\geq 99.999\%$ , Messer), and  $\text{O}_2$  ( $\geq 99.999\%$ , Messer) gases were used without further purification. For the labeling experiments,  $^{18}\text{O}_2$  (97 atom %, Aldrich) was used.

## Matrix IR spectra

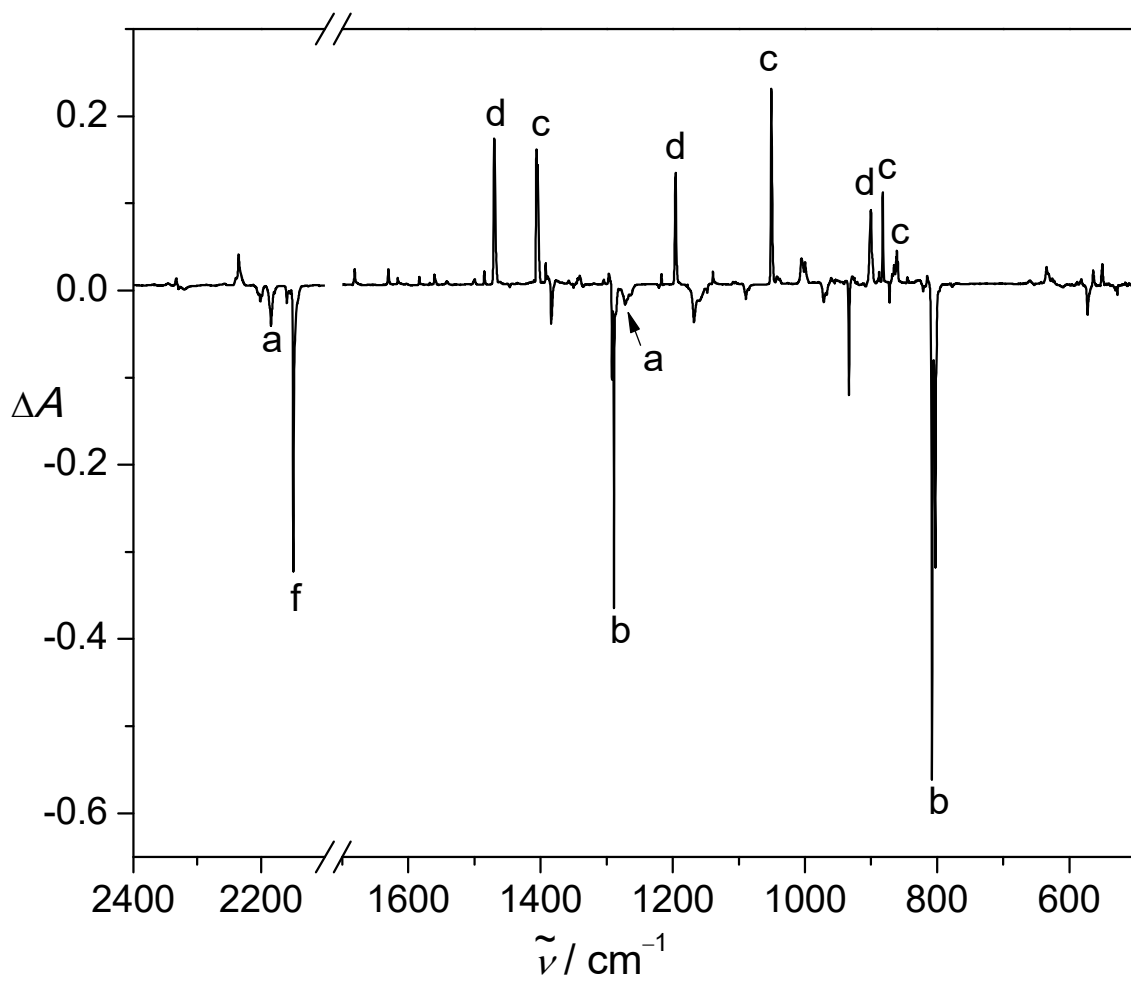
Matrix IR spectra were recorded on a FT-IR spectrometer (Bruker 70V) in a reflectance mode using a transfer optic. A KBr beam splitter and wide band MCT detector were used in the mid-IR region (4000–400  $\text{cm}^{-1}$ ). Typically, 200 scans at a resolution of 0.5  $\text{cm}^{-1}$  were co-added for each spectrum. Fluorophosphoryl diazide was mixed by passing a flow of dilution gas (Ar, Ne, or  $\text{N}_2$ ) through a U-trap ( $-55\text{ }^\circ\text{C}$ ) containing ca. 20 mg of the  $\text{FP}(\text{O})(\text{N}_3)_2$ . Then the mixture (estimated 1:50:1000, azide:  $\text{O}_2$ : noble gas) was passed through an aluminum oxide furnace (o.d. 2.0 mm, i.d. 1.0 mm), deposited (2  $\text{mmol h}^{-1}$ ) in a high vacuum ( $\sim 10^{-6}$  pa) onto the Rh-plated Cu block matrix support (10K) using a closed-cycle helium cryostat (Sumitomo Heavy Industries, SRDK-408D2-F50H) inside the vacuum chamber. Temperatures at the second stage of the cold head were controlled and monitored using a LakeShore 335 digital cryogenic temperature controller a Silicon Diode (DT-670). The voltage and current used in the pyrolysis experiments were 5.4 V and 3.96 A. Photolysis experiments were performed using an ArF excimer laser (Gamlaser EX5/250, 5 mJ, 3Hz), a  $\text{Nd}^{3+}$ :YAG laser (MPL-F-266, 10 mW), and an UV flashlight (Boyu T648, 30 W).

## Quantum chemical calculation methods and additional results

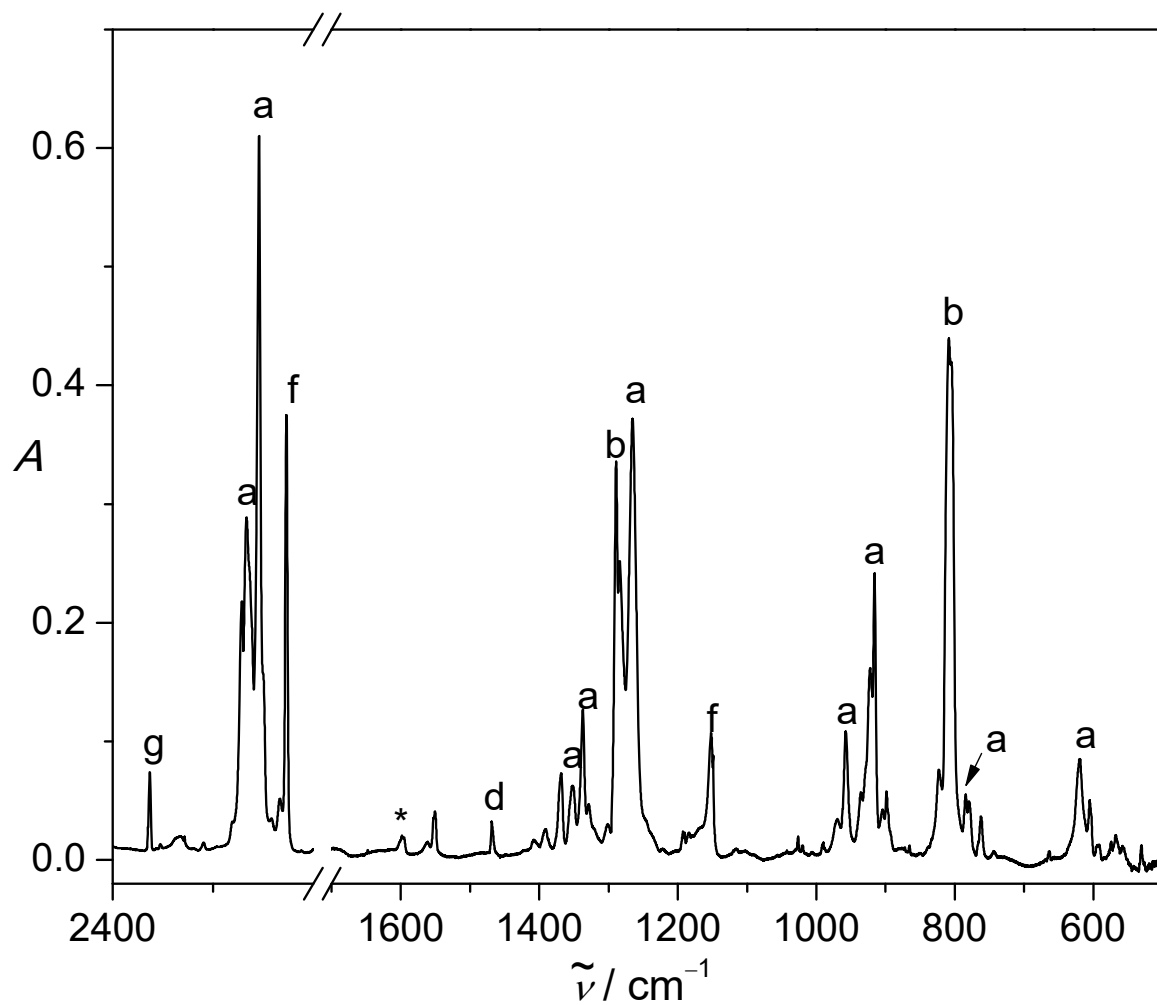
Structures and IR frequencies of stationary points and transient states were calculated using the DFT-B3LYP<sup>[2]</sup> and ab initio CCSD(T)<sup>[3]</sup> methods with the 6-311+G(3df)<sup>[4]</sup> and cc-pVTZ<sup>[5]</sup> basis sets, respectively. Local minima were confirmed by vibrational frequency analysis, and transition states were further confirmed by intrinsic reaction coordinate (IRC) calculations.<sup>[6]</sup> Time-dependent (TD) DFT (B3LYP/6-311+G(3df)) calculations<sup>[7]</sup> were performed for the prediction of UV-Vis transitions. These computations were performed using the Gaussian 09 software package.<sup>[8]</sup> The CCSD(T) computations were performed using the CFOUR<sup>[9]</sup> packages. The minimum energy crossing point (MECP) is located at the B3LYP/6-311+G(3df) level using the Newton-Lagrange method, which was introduced by Koga and Morokuma.<sup>[12]</sup> The algorithm is based on the minimization of the Lagrangian function  $L(\mathbf{R}, \lambda) = E_1(\mathbf{R}) - \lambda[E_1(\mathbf{R}) - E_2(\mathbf{R})]$ , where  $\mathbf{R}$  is nuclear coordinates,  $E_1(\mathbf{R})$  and  $E_2(\mathbf{R})$  are the energies of the states presently considered as functions of  $\mathbf{R}$ , and  $\lambda$  is Lagrange multiplier. The energies, energy gradients, and Hessian matrices of both the two states need to be calculated, and the minimum point is found on the seam of intersection. These calculations are treated using a homemade program LookForMECP (version 1.0). This program can be obtained from the authors upon request. The early version of this program has been used successfully to search the MECP.<sup>[11]</sup>



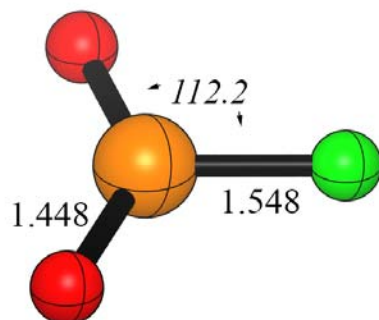
**Figure S1.** A) IR spectrum of the flash vacuum pyrolysis products of a mixture of  $\text{FP(=O)(N}_3)_2$ ,  $\text{O}_2$ , and  $\text{N}_2$  (ca. 1:50:1000) at 10 K. B) IR spectrum of the matrix after 266 nm laser photolysis (50 minutes) at 10 K. The IR bands of  $\text{FP(=O)(N}_3)_2$  (a),  $\text{FP=O}$  (b), cyclic  $\text{FP(=O)(O}_2)$  (c),  $\text{FP(=O)}_2$  (d),  $\text{F}_3\text{P=O}$  (e),  $\text{HN}_3$  (f),  $\text{CO}_2$  (g), and  $\text{H}_2\text{O}$  (\*) are labeled.



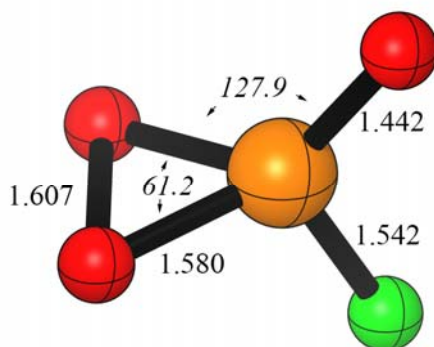
**Figure S2.** IR difference spectrum showing of the change of the matrix containing the flash vacuum pyrolysis products of a mixture of  $\text{FP(=O)(N}_3)_2$ ,  $\text{O}_2$ , and  $\text{N}_2$  (ca. 1:50:1000) upon a 266 nm laser photolysis.



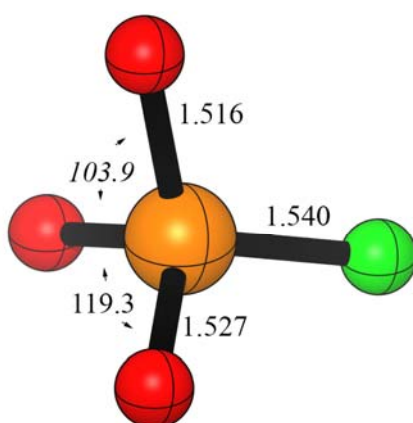
**Figure S3.** IR spectrum of the flash vacuum pyrolysis products of a mixture of  $\text{FP}(=\text{O})(\text{N}_3)_2$ ,  $\text{O}_2$ , and Ne (ca. 1:50:1000) at 10 K.



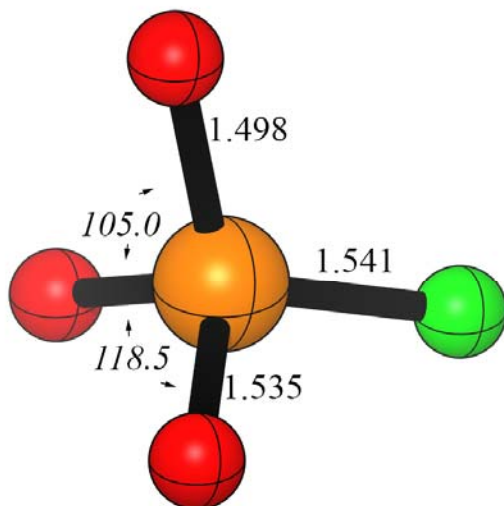
**Figure S4.** RB3LYP/6-311+G(3df) calculated bond lengths (Å) and angle (°) of singlet FPO<sub>2</sub>.



**Figure S5.** RB3LYP/6-311+G(3df) calculated bond lengths (Å) and angle (°) of close-shell singlet FPO<sub>3</sub>.



**Figure S6.** UB3LYP/6-311+G(3df) calculated bond lengths (Å) and angle (°) of triplet FPO<sub>3</sub>.



**Figure S7.** UB3LYP/6-311+G(3df) (guess=mix) calculated bond lengths (Å) and angle (°) of open-shell singlet FPO<sub>3</sub>.



**Table S1.** Calculated IR frequencies ( $\text{cm}^{-1}$ ) and intensities ( $\text{km mol}^{-1}$ , in parentheses) of  $\text{FPO}_n$  ( $n=2, 3$ ) species at B3LYP/6-311+G(3df) level.

Triplet-FP(=O)(O) <sub>2</sub>	FOP(=O) <sub>2</sub>	FOP=O	FP(= <sup>16</sup> O) <sup>16</sup> O <sup>16</sup> O-syn	FP(= <sup>16</sup> O) <sup>18</sup> O <sup>18</sup> O-syn	FP(= <sup>16</sup> O) <sup>16</sup> O <sup>16</sup> O-anti	FP(= <sup>16</sup> O) <sup>18</sup> O <sup>18</sup> O-anti
1077.2 (129)	1493.4 (147)	1303.7 (89)	1315.2 (114)	1314.2 (113)	1316.5 (122)	1316.4 (122)
897.6 (105)	1179.6 (80)	875.0 (24)	1039.6 (133)	982.7 (113)	989.6 (108)	937.6 (75)
860.3 (130)	942.9 (9)	682.5 (72)	881.1 (182)	879.2 (187)	901.0 (164)	896.1 (182)
844.3 (49)	792.0 (82)	495.3 (41)	621.8 (31)	596.1 (26)	621.0 (28)	598.5 (24)
422.4 (28)	469.2 (38)	244.9 (10)	463.4 (5)	450.6 (6)	479.5 (7)	464.2 (7)
371.9 (0)	409.4 (51)	181.8 (2)	349.2 (19)	344.2 (19)	352.9 (24)	348.6 (24)
324.7 (3)	390.3 (19)		266.9 (41)	256.3 (43)	230.8 (46)	223.4 (48)
292.3 (6)	231.9 (3)		189.1 (49)	186.0 (46)	191.2 (25)	186.9 (28)
246.7 (1)	166.2 (1)		162.8 (9)	157.9 (7)	165.2 (35)	161.0 (27)

**Table S2.** Calculated IR frequencies ( $\text{cm}^{-1}$ ) and intensities ( $\text{km mol}^{-1}$ , in parentheses) of close shell singlet FP(=O)(O<sub>2</sub>) isotopomers.

FP(= <sup>16</sup> O)( <sup>16</sup> O <sup>16</sup> O)			FP(= <sup>16</sup> O)( <sup>18</sup> O <sup>18</sup> O)		FP(= <sup>18</sup> O)( <sup>16</sup> O <sup>18</sup> O)		FP(= <sup>18</sup> O)( <sup>18</sup> O <sup>18</sup> O)	
CCSD(T)/cc-pVTZ	CCSD(T)/cc-pVTZ	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)
$\nu_{\text{anharm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$
1402.3 (166)	1424.2 (178)	1420.2 (205)	1416.6 (201)	1378.8 (203)	1376.6 (201)			
1044.4 (168)	1063.2 (176)	1062.2 (147)	1036.0 (168)	1045.1 (160)	1032.4 (171)			
874.6 (42)	885.1 (48)	879.0 (78)	865.0 (58)	868.4 (62)	858.1 (51)			
847.9 (15)	860.6 (15)	841.0 (15)	805.1 (16)	819.6 (16)	805.0 (16)			
553.2 (25)	567.5 (28)	629.7 (31)	602.3 (30)	613.6 (29)	599.7 (29)			
430.4 (41)	433.8 (41)	434.4 (40)	427.8 (38)	426.2 (38)	422.7 (37)			
411.4 (33)	413.9 (33)	412.5 (32)	407.1 (30)	400.5 (30)	396.9 (30)			
399.4 (13)	402.6 (13)	399.9 (14)	393.7 (14)	392.2 (13)	390.1 (12)			
265.9 (0)	269.1 (0)	268.9 (0)	257.8 (0)	261.2 (1)	255.8 (1)			

**Table S3.** Calculated IR frequencies ( $\text{cm}^{-1}$ ) and intensities ( $\text{km mol}^{-1}$ , in parentheses) of close shell singlet  $\text{FP}(\text{O})_2$  isotopomers.

FP( $=^{16}\text{O}$ )( $=^{16}\text{O}$ )		FP( $=^{16}\text{O}$ )( $=^{18}\text{O}$ )		FP( $=^{18}\text{O}$ )( $=^{18}\text{O}$ )
CCSD(T)/cc-pVTZ	CCSD(T)/cc-pVTZ	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)	B3LYP/6-311+G(3df)
$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$	$\nu_{\text{harm}}$
1457.9 (129)	1478.2 (135)	1478.4 (176)	1457.1 (173)	1435.0 (170)
1177.3 (97)	1193.0 (101)	1193.3 (101)	1166.3 (109)	1145.9 (116)
889.4 (98)	899.2 (100)	873.9 (110)	867.2 (102)	862.5 (93)
431.6 (36)	434.0 (37)	434.2 (37)	428.3 (31)	421.8 (53)
426.5 (19)	428.9 (20)	429.6 (55)	425.6 (54)	416.9 (34)
421.9 (50)	425.7 (51)	426.5 (18)	418.0 (22)	415.9 (17)

**Table S4.** Calculated vertical transition energies (nm) and oscillator strength (in parentheses) at TD-B3LYP/6-311+G(3df) level.

FP=O		FP(=O) <sub>2</sub>		cyclic FP(=O)(O) <sub>2</sub>		FP(=O)OO	
energy (nm)	oscillator strength	energy (nm)	oscillator strength	energy (nm)	oscillator strength	energy (nm)	oscillator strength
264	0.0181	225	0.0033	373	0.0006	1410	0.0003
212	0.0156	180	0.0411	254	0.0038	402	0.1140
180	0.0226	163	0.0020	187	0.0006	311	0.0022
151	0.1010	160	0.0041	185	0.0001	258	0.0191
147	0.1078	157	0.0504	176	0.0018	203	0.0067
143	0.0100	140	0.2121	173	0.0027	192	0.0506
142	0.0742	134	0.0095	165	0.0236	186	0.0020
138	0.0241			159	0.0110	184	0.0204
134	0.0022			141	0.0171	180	0.0231
132	0.0372					178	0.0215

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**Calculated structures, atomic coordinates (in Angstroms), and energies (in Hartrees) for all optimized species at B3LYP/6-311+G(3df)**

**Closed-shell singlet FP(=O)(O)<sub>2</sub>**

8	-1.113460	1.210398	0.000000
8	-0.102777	-1.178031	0.803827
8	-0.102777	-1.178031	-0.803827
15	-0.102777	0.181925	0.000000
9	1.343751	0.715160	0.000000

Zero-point correction=	0.014461 (Hartree/Particle)
Thermal correction to Energy=	0.018944
Thermal correction to Enthalpy=	0.019888
Thermal correction to Gibbs Free Energy=	-0.013431
Sum of electronic and zero-point Energies=	-666.983003
Sum of electronic and thermal Energies=	-666.978520
Sum of electronic and thermal Enthalpies=	-666.977576
Sum of electronic and thermal Free Energies=	-667.010895

**Triplet FP(=O)(O)<sub>2</sub>**

8	-0.544825	1.399871	-0.000000
8	-0.544825	-0.601343	1.318148
8	-0.544825	-0.601343	-1.318148
15	-0.031208	-0.026889	0.000000
9	1.504880	-0.130461	0.000000

Zero-point correction=	0.012162 (Hartree/Particle)
Thermal correction to Energy=	0.017141
Thermal correction to Enthalpy=	0.018085
Thermal correction to Gibbs Free Energy=	-0.017203
Sum of electronic and zero-point Energies=	-666.979570
Sum of electronic and thermal Energies=	-666.974591
Sum of electronic and thermal Enthalpies=	-666.973647
Sum of electronic and thermal Free Energies=	-667.008935

**Open-shell singlet FP(=O)(O)<sub>2</sub>**

8	-0.540132	1.403194	0.000000
8	-0.540132	-0.609496	1.319140
8	-0.540132	-0.609496	-1.319140
15	-0.035349	-0.007571	0.000000
9	1.499266	-0.151117	-0.000000

Zero-point correction=	0.012248 (Hartree/Particle)
Thermal correction to Energy=	0.017196

Thermal correction to Enthalpy=	0.018140
Thermal correction to Gibbs Free Energy=	-0.016044
Sum of electronic and zero-point Energies=	-666.978341
Sum of electronic and thermal Energies=	-666.973393
Sum of electronic and thermal Enthalpies=	-666.972448
Sum of electronic and thermal Free Energies=	-667.006633

#### MECP FP(=O)(O)<sub>2</sub>

8	1.183563	-1.085882	0.000228
8	-1.078575	-0.141007	-0.994142
8	-1.078046	-0.140089	0.994579
15	0.126312	-0.088388	-0.000111
9	0.654420	1.362405	-0.000406

Zero-point correction=	0.011823 (Hartree/Particle)
Thermal correction to Energy=	0.016412
Thermal correction to Enthalpy=	0.017356
Thermal correction to Gibbs Free Energy=	-0.017310
Sum of electronic and zero-point Energies=	-666.964092
Sum of electronic and thermal Energies=	-666.959503
Sum of electronic and thermal Enthalpies=	-666.958559
Sum of electronic and thermal Free Energies=	-666.993225

#### Syn FP(=O)OO

15	-0.425458	0.115139	0.255378
8	1.006756	-0.616456	0.019409
8	-0.777785	1.460762	-0.172238
8	2.108809	0.114242	-0.113884
9	-1.368929	-1.043941	-0.188552

Zero-point correction=	0.012049 (Hartree/Particle)
Thermal correction to Energy=	0.017408
Thermal correction to Enthalpy=	0.018352
Thermal correction to Gibbs Free Energy=	-0.017069
Sum of electronic and zero-point Energies=	-666.881066
Sum of electronic and thermal Energies=	-666.875707
Sum of electronic and thermal Enthalpies=	-666.874763
Sum of electronic and thermal Free Energies=	-666.910184

#### Anti FP(=O)OO

15	-0.494140	-0.067796	-0.225640
8	0.942653	-0.756059	-0.015662
8	-1.714259	-0.715027	0.234180
8	2.060712	-0.025491	0.114594

9	-0.322305	1.443284	0.079967
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Zero-point correction=			0.011955 (Hartree/Particle)
Thermal correction to Energy=			0.017343
Thermal correction to Enthalpy=			0.018287
Thermal correction to Gibbs Free Energy=			-0.017159
Sum of electronic and zero-point Energies=			-666.880310
Sum of electronic and thermal Energies=			-666.874922
Sum of electronic and thermal Enthalpies=			-666.873978
Sum of electronic and thermal Free Energies=			-666.909424

**FP(=O)<sub>2</sub>**

15	0.000000	0.000000	0.129076
8	0.000000	1.340792	0.677144
8	-0.000000	-1.340792	0.677144
9	-0.000000	-0.000000	-1.418937

Zero-point correction=			0.011017 (Hartree/Particle)
Thermal correction to Energy=			0.014775
Thermal correction to Enthalpy=			0.015719
Thermal correction to Gibbs Free Energy=			-0.015063
Sum of electronic and zero-point Energies=			-591.823031
Sum of electronic and thermal Energies=			-591.819273
Sum of electronic and thermal Enthalpies=			-591.818329
Sum of electronic and thermal Free Energies=			-591.849111

**FOP=O**

8	-0.853151	-0.655815	-0.000040
15	0.787366	-0.467878	0.000021
8	1.226781	0.923741	-0.000022
9	-1.644393	0.541640	0.000020

Zero-point correction=			0.008619 (Hartree/Particle)
Thermal correction to Energy=			0.012953
Thermal correction to Enthalpy=			0.013897
Thermal correction to Gibbs Free Energy=			-0.018882
Sum of electronic and zero-point Energies=			-591.668245
Sum of electronic and thermal Energies=			-591.663911
Sum of electronic and thermal Enthalpies=			-591.662967
Sum of electronic and thermal Free Energies=			-591.695746

**FP=O**

15	0.000000	0.470675	0.000000
8	1.305190	-0.176320	0.000000

9	-1.160169	-0.627729	0.000000
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Zero-point correction=			0.005702 (Hartree/Particle)
Thermal correction to Energy=			0.008934
Thermal correction to Enthalpy=			0.009878
Thermal correction to Gibbs Free Energy=			-0.019551
Sum of electronic and zero-point Energies=			-516.563925
Sum of electronic and thermal Energies=			-516.560693
Sum of electronic and thermal Enthalpies=			-516.559749
Sum of electronic and thermal Free Energies=			-516.589178

**FOP(=O)<sub>2</sub>**

15	-0.000000	0.547008	0.000000
8	1.445863	0.578512	0.000000
8	-1.087448	1.505405	0.000000
8	-0.692991	-0.918600	0.000000
9	0.297401	-1.947517	0.000000

Zero-point correction=			0.013840 (Hartree/Particle)
Thermal correction to Energy=			0.018817
Thermal correction to Enthalpy=			0.019762
Thermal correction to Gibbs Free Energy=			-0.014859
Sum of electronic and zero-point Energies=			-666.929171
Sum of electronic and thermal Energies=			-666.924193
Sum of electronic and thermal Enthalpies=			-666.923249
Sum of electronic and thermal Free Energies=			-666.957870

**Singlet O<sub>2</sub>**

8	0.000000	0.000000	0.601398
8	0.000000	0.000000	-0.601398

Zero-point correction=			0.003722 (Hartree/Particle)
Thermal correction to Energy=			0.006085
Thermal correction to Enthalpy=			0.007030
Thermal correction to Gibbs Free Energy=			-0.015201
Sum of electronic and zero-point Energies=			-150.314432
Sum of electronic and thermal Energies=			-150.312068
Sum of electronic and thermal Enthalpies=			-150.311124
Sum of electronic and thermal Free Energies=			-150.333355

**TS1: FP=O to FP(=O)OO**

15	-0.651883	0.129675	0.402767
8	1.370318	-0.556685	0.033623
8	-0.983587	1.351995	-0.321359

8		2.260047	0.265502	-0.130949
9		-1.266220	-1.159069	-0.299114

Zero-point correction=	0.010681 (Hartree/Particle)
Thermal correction to Energy=	0.016285
Thermal correction to Enthalpy=	0.017229
Thermal correction to Gibbs Free Energy=	-0.019469
Sum of electronic and zero-point Energies=	-666.875876
Sum of electronic and thermal Energies=	-666.870273
Sum of electronic and thermal Enthalpies=	-666.869329
Sum of electronic and thermal Free Energies=	-666.906026

**TS2: FP(=O)OO to FP(=O)(O)<sub>2</sub>**

8	0	1.303921	0.947086	0.000000
8	0	-0.569774	-1.054901	0.000000
8	0	0.622365	-1.975586	0.000000
15	0	0.000000	0.336450	0.000000
9	0	-1.205789	1.291163	0.000000

Zero-point correction=	0.011575 (Hartree/Particle)
Thermal correction to Energy=	0.016566
Thermal correction to Enthalpy=	0.017510
Thermal correction to Gibbs Free Energy=	-0.017532
Sum of electronic and zero-point Energies=	-666.878351
Sum of electronic and thermal Energies=	-666.873360
Sum of electronic and thermal Enthalpies=	-666.872416
Sum of electronic and thermal Free Energies=	-666.907458

**TS3: FP(=O)O(O)<sub>2</sub> scrambling**

8		-0.539479	1.415338	-0.000000
8		-0.539479	-0.692402	1.245921
8		-0.539479	-0.692402	-1.245921
15		-0.040131	-0.009149	0.000000
9		1.505496	-0.011892	-0.000000

Zero-point correction=	0.012353 (Hartree/Particle)
Thermal correction to Energy=	0.017045
Thermal correction to Enthalpy=	0.017989
Thermal correction to Gibbs Free Energy=	-0.016492
Sum of electronic and zero-point Energies=	-666.950355
Sum of electronic and thermal Energies=	-666.945662
Sum of electronic and thermal Enthalpies=	-666.944718
Sum of electronic and thermal Free Energies=	-666.979199