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Electronic Supplementary Information

Broadening the Sunlight Response Region with Carbon Dot Sensitized TiO₂ as Support for Pt Catalyst in Methanol Oxidation Reaction

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Fig. S1 Particle size histogram of CDs.



Fig. S2 SEM images of Pt3/CD2/TiO₂: (a) the section used for EDX mapping and (b) a magnified image.

Quantum Yield Calculations

The quantum yield of carbon dots (Φ_{CD}) was calculated by comparing their photoluminescence intensities ($\lambda_{ex} = 340$ nm) and the absorbance values (at 340 nm) with reference to quinine sulphate (QS). Five concentrations of each of CDs and QS were prepared by dissolving them in distilled water (refractive index (η) = 1.33) and 0.1 M H₂SO₄ (η = 1.33) respectively. The absorbance of all the samples was kept less than 0.1 at 340 nm. The quantum yield of QS (Φ_{QS}) reported in literature is 0.54. The data of integrated PL intensity *vs*. absorbance was plotted (**Fig. S3**) to determine the slopes of CDs (m_{CD}) and QS (m_{QS}). The value of Φ_{CD} was then calculated in accordance too the following equation:

$$\Phi_{CD} = \Phi_{QS} (m_{CD} / m_{QS}) (\eta_{CD}^2 / \eta_{QS}^2)$$



Fig. S3 Photoluminescence vs. absorbance plots of CD and QS.

The quantum yield of CD is found to be 29%

Composite	$ m R_{S}/ \Omega$	R_{CT}/Ω	$ m R_{S}/ \Omega$	R_{CT}/Ω
	(Light)		(Dark)	
Pt2.5/CD2/TiO ₂	15.06	24.44	15.71	28.13
Pt3/CD2/TiO ₂	12.15	17.29	12.73	21.48
Pt3.5/CD2/TiO ₂	12.98	19.26	13.48	23.63
Pt4/CD2/TiO ₂	13.77	22.01	14.34	25.92

Table S1. EIS parameters of the $Pty/CD2/TiO_2$ composites recorded under light and dark conditions.