Electronic Supplementary Information

Structure and Abnormal Photoluminescence of a Red-Emitting LiMgBO₃:

Mn²⁺ phosphor

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	Li1	Li2
01	0.2052	0.2808
02	0.3258	0.2554
03	0.3293	0.3171
	0.1464	0.0963
	0.0247	0.0373
$\sum s$	1.0314	0.9869

Table S1 Li(1)–O, Li(2)–O bonds valence of LiMgBO₃

Table S2 Fractional Atomic Coordinates, Equivalent Isotropic Displacement Parameters (Å²) and Occupancy for

LiMgBO₃: 0.04Mn²⁺

	site	Х	у	Z	Ueq	occupancy		
LiMgBO ₃ : 0.04Mn ²⁺								
Li1	8f	0.1968(11)	0.5104(7)	0.1509(3)	0.0299(6)	0.553(7)		
Mn1	8f	0.1968(11)	0.5104(7)	0.1509(3)	0.0299(6)	0.022(7)		
Li2	8f	0.1664(14)	0.4911(8)	0.1092(5)	0.0200(1)	0.407(4)		
Mn2	8f	0.1664(14)	0.4911(8)	0.1092(5)	0.0200(1)	0.016(4)		
Mg	8f	0.1560(2)	0.1668(1)	0.1244(1)	0.0236(3)	1.0000		
В	8f	0.1613(6)	0.8271(3)	0.1287(2)	0.0465(5)	1.0000		
01	8f	0.4042(2)	0.3330(1)	0.0869(1)	0.0267(6)	1.0000		
02	8f	0.2893(2)	0.6930(1)	0.1635(1)	0.0446(9)	1.0000		
03	8f	0.3087(2)	0.9627(1)	0.1282(1)	0.0520(6)	1.0000		



Figure S1 Excitation spectra of LiMgBO₃: 0.04Mn²⁺ monitored at 700 nm and 762 nm.



Figure S2 The orbital-decomposed DOS spectra of Mn ion in LiMgBO₃: 0.04Mn. The energy reference is set as the top of electron occupation. The upper and lower rows are corresponding to majority spin (spin-up) and minority spin (spin-down), respectively.



Figure S3 Temperature-dependent XRD patterns of LiMgBO₃: 0.04Mn²⁺.