

Enhanced deep-red emission from Mn⁴⁺/Mg²⁺ co-doped CaGdAlO₄ phosphors for plant cultivation

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Fig. S7 Raman spectra of un-doped CaGdAlO₄ host.

Table S1 Crystallographic data of *x* mol% Mn⁴⁺/Mg²⁺ co-doped CaGdAlO₄ phosphors (*x* = 0.2, 0.5, 1, 1.5, and 2) from Rietveld refinement.

Table S2 Emission peaks and quantum efficiency for reported Mn⁴⁺ doped phosphors.

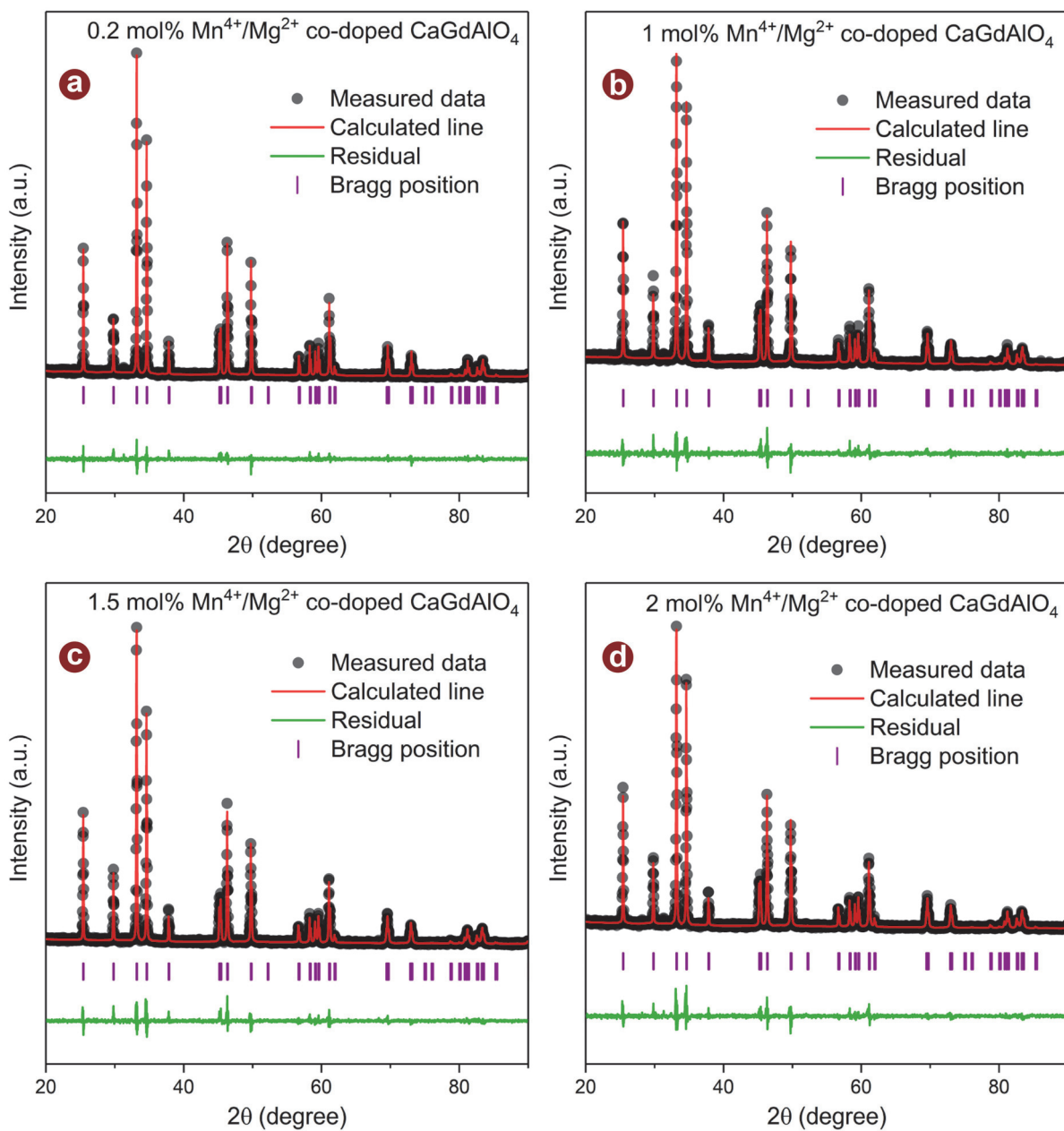


Fig. S1 (a)-(d) XRD Rietveld refinement results for x mol% $\text{Mn}^{4+}/\text{Mg}^{2+}$ co-doped CaGdAlO_4 phosphors ($x = 0.2, 1, 1.5, 2$).

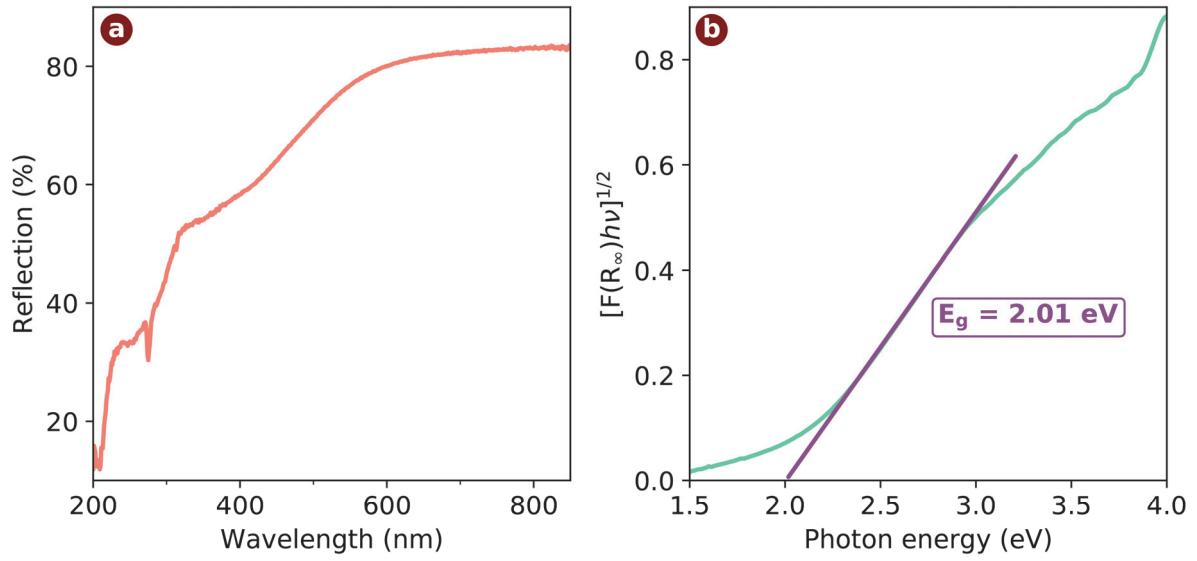


Fig. S2 (a) Diffuse reflection spectra of undoped CaGdAlO₄ phosphor. (b) The dependence of $[F(R_{\infty})hv]^{1/2}$ on the photon energy hv .

The energy gap (E_g) of CaGdAlO₄ can be evaluated by the following equation:

$$[F(R_{\infty})hv]^n = A(hv - E_g) \quad (1)$$

where hv denotes photon energy, A is a proportional constant, n equals to 1/2 for an indirect allowed transition. $F(R_{\infty})$ is the Kubelka-Munk function, which can be expressed as:

$$[F(R_{\infty})] = (1 - R)^2/2R$$

where R is the reflection coefficient. $[F(R_{\infty})hv]^{1/2}$ is plotted against hv according to Taus method, from which the energy gap is determined to be 2.01 eV.

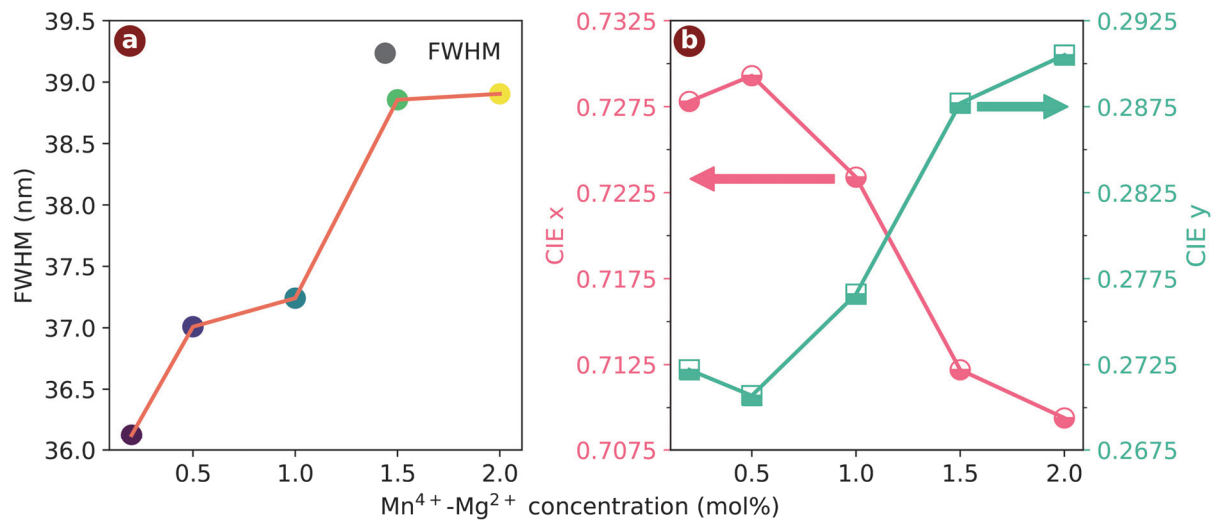


Fig. S3 (a) FWHM dependent plot on Mn⁴⁺/Mg²⁺ doping concentration. (b) CIE chromaticity coordinates (x, y) as a function of Mn⁴⁺/Mg²⁺ doping concentration.

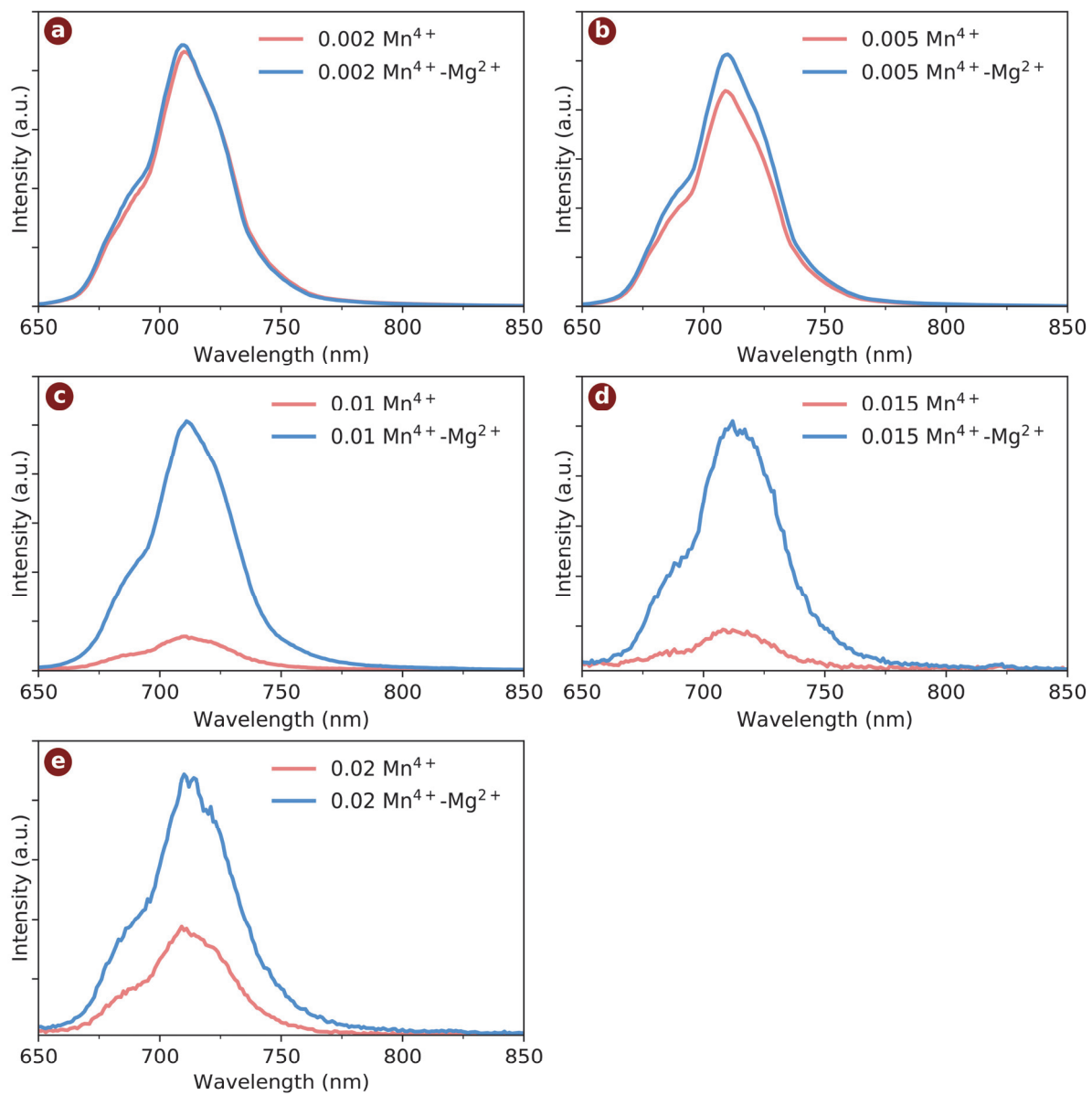


Fig. S4 (a)-(e) A comparison of red emission from Mn^{4+} ions with and without Mg^{2+} ions in CaGdAlO_4 host.

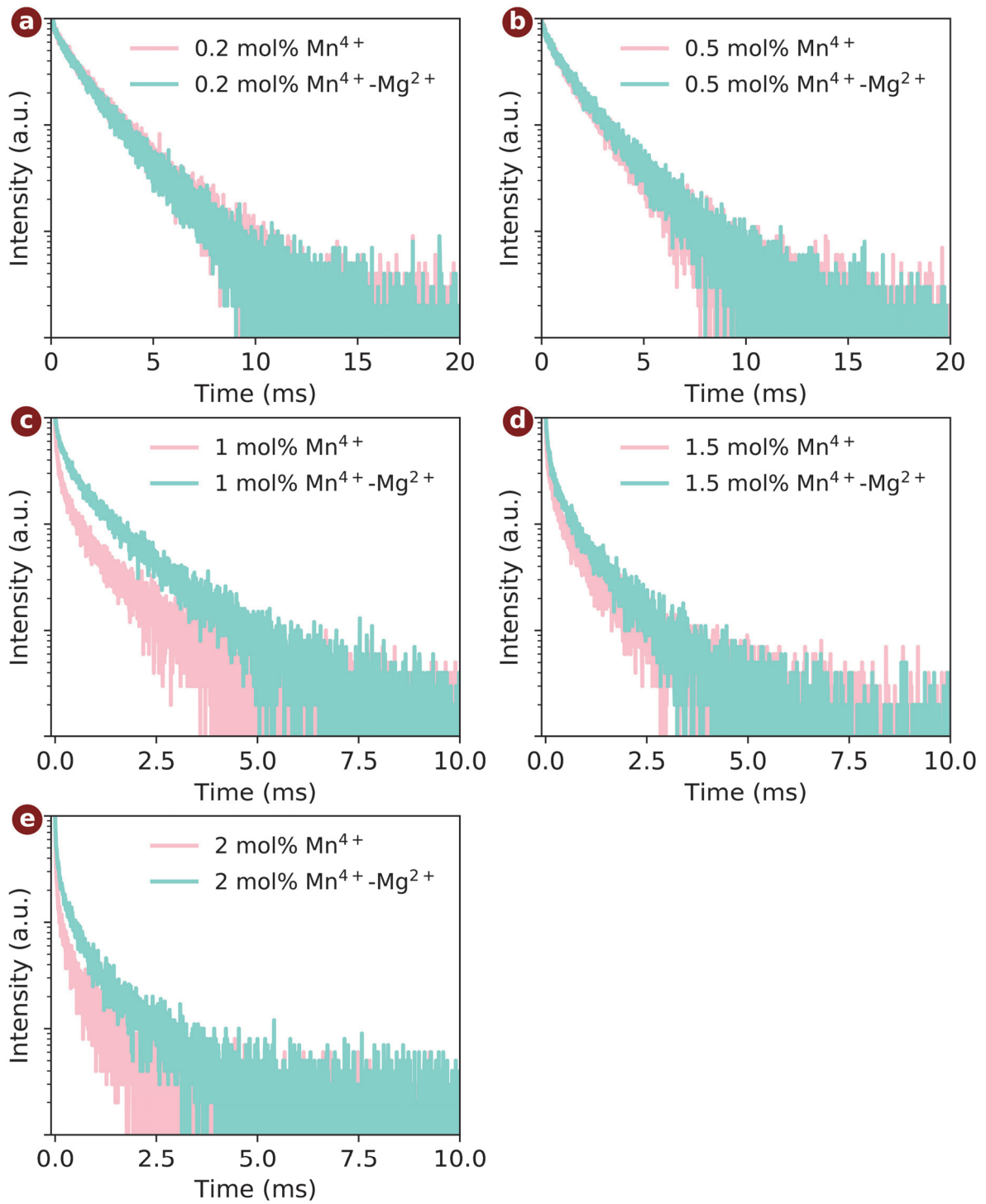


Fig. S5 (a)-(e) A comparison of decay time at 712 nm in the presence and absence of Mg²⁺ ions in CaGdAlO₄ host.

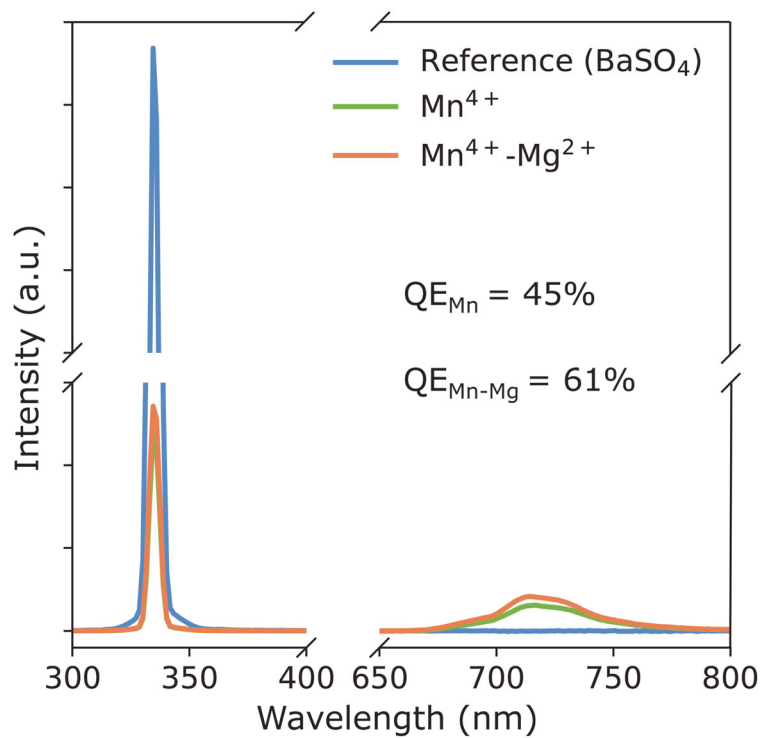


Fig. S6 Excitation spectra of BaSO₄ (in blue) and the emission spectra of 0.5 mol% Mn⁴⁺ singly doped (in green) and 0.5 mol% Mn⁴⁺/Mg²⁺ co-doped (in red) phosphors under 335 nm excitation collected by an integrating sphere.

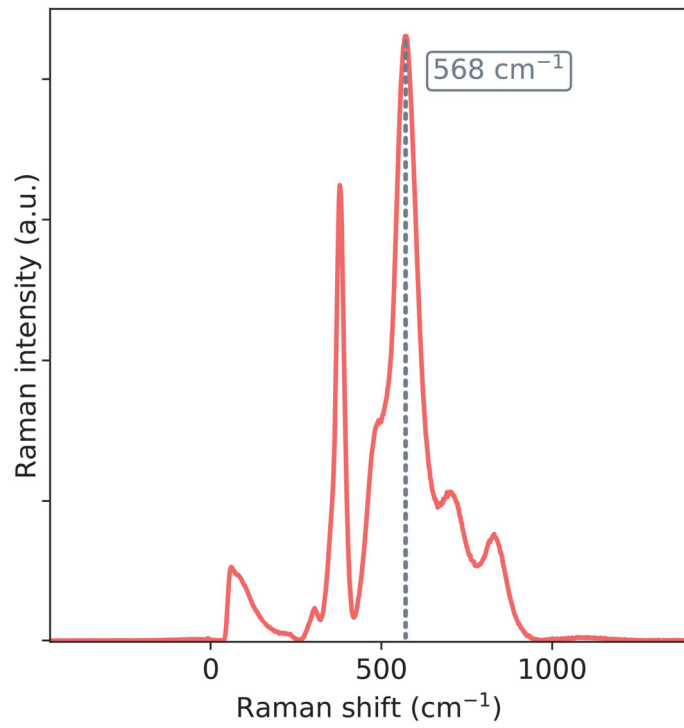


Fig. S7 Raman spectra of un-doped CaGdAlO₄ host.

Table S1 Crystallographic data of x mol% $\text{Mn}^{4+}/\text{Mg}^{2+}$ co-doped CaGdAlO_4 phosphors ($x = 0.2, 0.5, 1, 1.5, \text{ and } 2$) from Rietveld refinement.

Sample	0.2% $\text{Mn}^{4+}/\text{Mg}^{2+}$	0.5% $\text{Mn}^{4+}/\text{Mg}^{2+}$	1% $\text{Mn}^{4+}/\text{Mg}^{2+}$	1.5% $\text{Mn}^{4+}/\text{Mg}^{2+}$	2% $\text{Mn}^{4+}/\text{Mg}^{2+}$
Crystal system	Tetragonal	Tetragonal	Tetragonal	Tetragonal	Tetragonal
Space group	<i>I4/mmm</i>	<i>I4/mmm</i>	<i>I4/mmm</i>	<i>I4/mmm</i>	<i>I4/mmm</i>
a (Å)	3.6613	3.6624	3.6637	3.6642	3.6657
b (Å)	3.6613	3.6624	3.6637	3.6642	3.6657
c (Å)	11.9890	11.9858	11.9857	11.9891	11.9916
V (Å ³)	160.71	160.76	160.88	160.97	161.13
α (deg)	90	90	90	90	90
β (deg)	90	90	90	90	90
γ (deg)	90	90	90	90	90
χ^2	2.85	2.51	2.03	2.27	2.33
R_p (%)	4.74	4.41	4.31	4.39	4.66
R_{wp} (%)	5.93	5.76	5.79	5.85	5.91

Table S2 Emission peaks and quantum efficiency for reported Mn⁴⁺ doped phosphors.

Phosphors	Peak wavelength (nm)	Quantum efficiency (%)	Reference
La(MgTi) _{1/2} O ₃	708	27.2	12
Gd ₂ ZnTiO ₆	705	39.7	54
Ba ₂ TiGe ₂ O ₈	666	35.6	55
Li ₂ Mg ₃ SnO ₆	670	36.3	56
La ₂ MgTiO ₆	710	58.7	57
NaMgLaTeO ₆	703	57.4	40
Sr ₃ NaSbO ₆	695	56.2	58
Ca ₂ LaSbO ₆	685	52.2	59
BaLaMgNbO ₆	700	52.0	60
CaGdAlO ₄	712	61.0	This work