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Fig. S1 XRD patterns of as-prepared  $Y(OH)_{1.57}F_{1.43}$  under the different pH value.



Fig. S2 XRD patterns of V-YOF with different F<sup>-</sup> sources.



Fig. S3 PLE and PL spectra of V-YOF:0.05Eu3+ (a), V-YOF:0.05Tb3+ (b), V-YOF:0.05Tm3+ (c), V-YOF:0.05Dy3+ (d).



Figure. S4 PL excitation spectra of V-YOF: $Dy^{3+}$  and PL spectra of V-YOF: $Tm^{3+}$  Phosphors.



Fig. S5 Scatter plot of decay time of  $Tm^{3+}$  and energy transfer efficiency from  $Tm^{3+}$  to  $Dy^{3+}$ .



**Fig. S6** CIE chromaticity diagram for V-YOF: $0.05Ln^{3+}$  (Ln= Tm, Dy, Tb, Eu) synthesized under different conditions.

**Table. S1** CIE chromaticity coordinates (x, y) for V-YOF:0.05Ln<sup>3+</sup> (Ln= Tm, Dy, Tb,

Eu) synthesized under	different	conditions.
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Sample	pH=8	pH=8	pH=8	pH=10	Standard
CIE(x,y)	NaF	$\rm NH_4F$	KF	NaF	color coordinates
Condition					
V-YOF:0.05Tm <sup>3+</sup>	(0.153,0.049)	(0.155,0.062)	(0.158,0.079)	(0.152,0.046)	Blue (0.14,0.08)
V-YOF:0.05Dy <sup>3+</sup>	(0.307,0.350)	(0.278,0.310)	(0.295,0.336)	(0.304,0.348)	White (0.33,0.33)
V-YOF:0.05Tb <sup>3+</sup>	(0.255,0.590)	(0.255,0.589)	(0.255,0.589)	(0.257,0.595)	Green (0.21,0.71)
V-YOF:0.05Eu <sup>3+</sup>	(0.534,0.344)	(0.491,0.316)	(0.541,0.344)	(0.552,0.346)	Red (0.67,0.33)