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Supporting Information

Novel and effective strategy of dual bis(trifluoromethylsulfonyl)imide imidazolium ionic liquid immobilized on periodic mesoporous organosilica-mediated greener cycloaddition of carbon dioxide to epoxides

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Scheme S1 Preparation of PMO@IL-NTf₂.

Sample	Si (wt.%)	C (wt.%)	O (wt.%)	F (wt.%)
РМО	51.32	1.79	47.29	-
PMO@IL-NTf ₂ (0.5)	46.42	6.31	42.75	0.16
PMO@IL-NTf ₂ (1.0)	39.27	13.71	36.09	2.51
PMO@IL-NTf ₂ (1.5)	28.34	17.95	31.27	4.38

Table S1 EDX elemental composition of PMO and PMO@IL-NTf $_2$

Table S2 BET surface area and pore volume of the samples

Sample	$S_{BET} (m^2/g)$	$V(cm^{3}/g)$	D (nm)
РМО	539.26	0.82	5.80
PMO@IL-NTf ₂ (1.0)	316.14	0.51	5.57

Table S3 Elemental analysis of the fresh and reused PMO@IL-NTf₂(1.0)

Catalyst	C (wt.%)	Si (wt.%)	F (wt.%)
Fresh catalyst	14.65	41.32	2.54
Second reused catalyst	14.58	41.22	2.49
Fourth reused catalyst	14.52	41.17	2.43
Sixth reused catalyst	14.41	41.09	2.36



Fig. S1 UV–Vis spectras of PMO (a), PMO@IL-NTf₂(0.5) (b), PMO@IL-NTf₂(1.0) (c), and PMO@IL-NTf₂(1.5) (d).



Fig. S2 TG/DSC curves of PMO (a), PMO@IL-NTf₂(0.5) (b), PMO@IL-NTf₂(1.0) (c), and PMO@IL-NTf₂(1.5) (d).



Fig. S3 N₂ adsorption-desorption isotherms and pore size distribution of PMO and PMO@IL-NTf₂(1.0).





Fig. S4 XPS spectra of PMO@IL-NTf₂(1.0) (a) survey of the catalyst, (b) Si2p, (c) O1s, (d) C1s, (e) S2p, (f) N1s, (g) F1s.



Fig. S5 FT-IR spectra of the fresh and reused PMO@IL-NTf₂(1.0).



Fig. S6. PXRD patterns of PMO@IL-NTf₂(1.0) before and after reaction.