

Supplementary Information

Efficient bifunctional vanadium doped Ni₃S₂ nanorod array for overall water splitting

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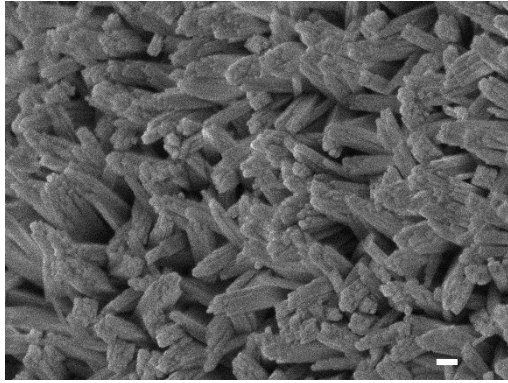


Fig. S1 SEM image of pristine $\text{Ni}_3\text{S}_2/\text{NF}$, showing morphology of nanorod array. Scale bar: 100 nm.

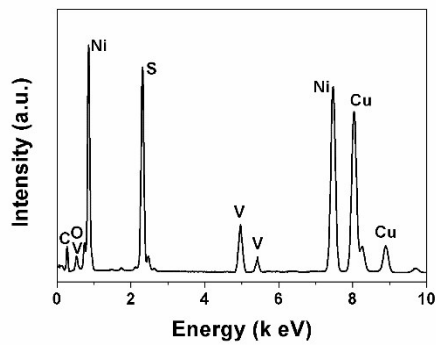


Fig. S2 EDX spectrum of $\text{V-Ni}_3\text{S}_2$ nanorod.

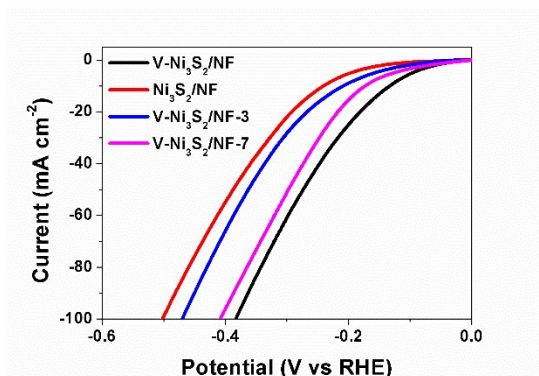


Fig. S3 The HER polarization curves of $\text{V-Ni}_3\text{S}_2/\text{NF}$, $\text{Ni}_3\text{S}_2/\text{NF}$, $\text{V-Ni}_3\text{S}_2/\text{NF-3}$, and $\text{V-Ni}_3\text{S}_2/\text{NF-7}$ in 1

M KOH.

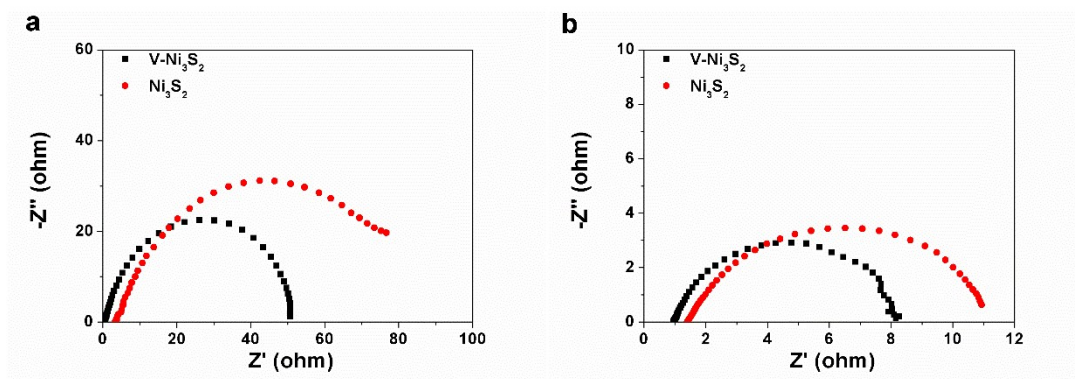


Fig. S4 The Nyquist plots of V- Ni_3S_2 nanorod/NF and Ni_3S_2 /NF electrodes obtained at (a) open circuit potential and (b) 40 mV for HER in 1 M KOH.

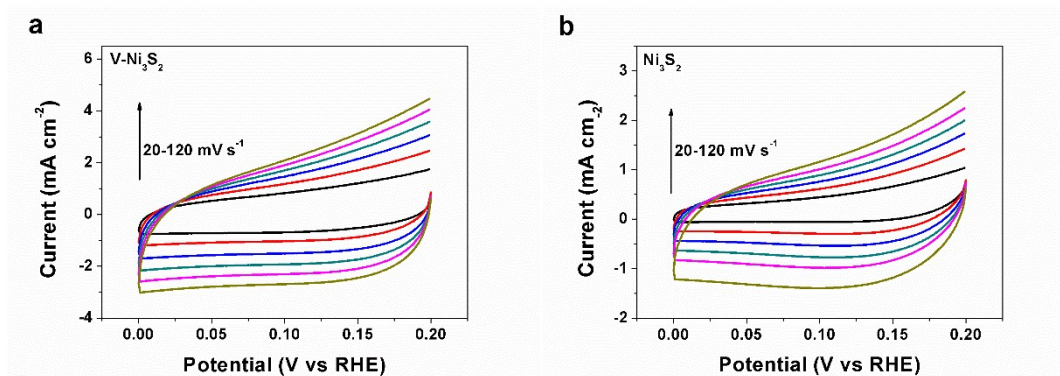


Fig. S5 CV curves of (a) V- Ni_3S_2 nanorod and (b) Ni_3S_2 obtained in 1 M KOH.

Table S1 Electrocatalytic performances of the designed V-Ni₃S₂ nanorod array electrode compared with the reported state-of-the-art bifunctional electrocatalysts for HER, OER, and overall water splitting in alkaline media.

Samples	η_{HER} at 10 mA cm ⁻² (mV)	η_{OER} at 10 mA cm ⁻² (mV)	Cell voltage for overall water splitting at 10 mA cm ⁻² (V)	Ref.
V-Ni ₃ S ₂ /NF	133	148	1.421	This work
(Fe,Co,Ni) ₉ S ₈ -MoS ₂ nanotube array	58	184	1.429	[1]
Ni ₂ P/Ni/NF	98	200	1.49	[2]
MoS ₂ -Ni ₃ S ₂ /NF	98	249	1.5	[3]
Co _{0.7} Fe _{0.3} P/CNT	76	243	1.5	[4]
MoS ₂ /Ni ₃ S ₂ @NF	110	218	1.56	[5]
NiCoP/N-rGO	115	310/40	1.57/20	[6]
NiFe LDH@NiCoP/NF	120	220	1.57	[7]
NiCoP/NF	32	280	1.58	[8]
Ni _{2-x} Co _x P	138	270	1.59	[9]
Co-Pi/CoP/Ti	68	310	1.6	[10]
NiCo ₂ S ₄ nanowire/NF	210	260	1.63	[11]
(Ni,Co) _{0.85} Se	169	287/20	1.65	[12]
Cu@CoFe LDH	171	240	1.681	[13]
NiFe LDH	210	240	1.7	[14]
Ni ₃ S ₂ nanosheet	223	260	1.76/13	[15]

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