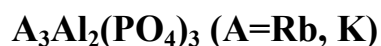


Supporting Information

**Experimental characterization and first principle calculation on  
linear and nonlinear optical properties of two orthophosphates**



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**Table S1.** Crystal data and structure refinements for Rb<sub>3</sub>Al<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> and K<sub>3</sub>Al<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>.

Empirical formula	Rb <sub>3</sub> Al <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub>	K <sub>3</sub> Al <sub>2</sub> (PO <sub>4</sub> ) <sub>3</sub>
Temperature	296(2) K	
Wavelength	0.71073 Å	
Crystal system	Orthorhombic	
Space group	<i>Cmc</i> 2 <sub>1</sub>	<i>Pna</i> 2 <sub>1</sub>
Formula weight	595.28	456.17
<i>a</i> (Å)	17.145(6)	8.671(3)
<i>b</i> (Å)	8.622(2)	16.906(5)
<i>c</i> (Å)	8.796(3)	8.445(2)
Z, Volume (Å <sup>3</sup> )	4, 1300.3(7)	4, 1238.0(6)
$\rho_{\text{Calcd}}$ (Mg/m <sup>3</sup> )	3.041	2.447
$\mu$ (mm)	11.806	1.689
F(000)	1112	896
R(int)	0.1303	0.0523
Goodness-of-fit on F <sup>2</sup>	0.943	1.081
Final R indices [ $F_o^2 > 2\sigma(F_o^2)$ ] <sup>a</sup>	R1 = 0.0497, wR2=0.0776	R1 =0.0387, wR2=0.0691
R indices (all data) <sup>a</sup>	R1 = 0.0725, wR2=0.0878	R1= 0.0489, wR2=0.0766
Absolute structure parameter	0.017(17)	0.31(8)
Largest diff. peak and hole (e <sup>-</sup> Å <sup>-3</sup> )	1.675 and 1.482	0.513 and 0.524

$$^a R_1 = \frac{\sum ||F_o| - |F_c||}{\sum |F_o|} \text{ and } wR_2 = \left[ \frac{\sum w(F_o^2 - F_c^2)^2}{\sum w F_o^4} \right]^{1/2} \text{ for } F_o^2 > 2\sigma(F_o^2)$$

**Table S2a.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for  $\text{Rb}_3\text{Al}_2(\text{PO}_4)_3$ .  $U_{\text{eq}}$  is defined as one-third of the trace of the orthogonalized  $U_{ij}$  tensor.

Atom	x	y	z	U(eq)
Rb(1)	3848(1)	10462(1)	7384(2)	23(1)
Rb(2)	5000	3551(2)	4468(3)	46(1)
P(1)	5000	7655(5)	5174(4)	15(1)
P(2)	7052(2)	3295(3)	4245(3)	13(1)
Al(1)	6674(2)	6528(4)	5840(3)	14(1)
O(1)	7897(4)	3037(8)	4841(9)	22(2)
O(2)	5000	9392(11)	5219(13)	23(2)
O(3)	6727(4)	4775(10)	4936(13)	36(2)
O(4)	7158(4)	3631(9)	2538(10)	34(2)
O(5)	5000	6988(14)	3616(13)	34(3)
O(6)	6523(4)	1964(9)	4534(9)	29(2)
O(7)	5713(4)	7034(9)	6122(8)	24(2)

**Table S2b.** Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for  $\text{K}_3\text{Al}_2(\text{PO}_4)_3$ .  $U_{\text{eq}}$  is defined as one-third of the trace of the orthogonalized  $U_{ij}$  tensor.

Atom	x	y	z	U(eq)
K(1)	-437(2)	6423(1)	92(2)	24(1)
K(2)	131(2)	6013(1)	4858(2)	24(1)
K(3)	3252(2)	4910(1)	7097(2)	36(1)
P(1)	6599(2)	7134(1)	3184(2)	13(1)
P(2)	2966(2)	6884(1)	8111(2)	14(1)
P(3)	2394(2)	4971(1)	2395(3)	17(1)
Al(1)	3514(2)	6604(1)	1758(2)	13(1)
Al(2)	6309(2)	6688(1)	6663(2)	14(1)
O(1)	5823(5)	7068(3)	4817(6)	17(1)
O(2)	5307(5)	7044(3)	1937(6)	18(1)
O(3)	7161(5)	8001(3)	3064(6)	18(1)
O(4)	3703(5)	5599(3)	2081(6)	24(1)
O(5)	7860(5)	6550(3)	2912(6)	21(1)
O(6)	2835(5)	6904(3)	9915(6)	25(1)
O(7)	898(5)	5275(3)	1754(7)	28(1)
O(8)	2929(6)	4246(3)	1400(6)	25(1)
O(9)	2369(6)	4764(3)	4119(7)	30(1)
O(10)	1815(6)	6340(3)	7423(7)	34(1)
O(11)	4633(5)	6644(3)	7749(6)	34(2)
O(12)	2749(5)	7740(3)	7548(7)	27(1)

**Table S3a.** Selected bond distances (Å) and angles (deg) for Rb<sub>3</sub>Al<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>.

Rb(1)-O(2)	2.894(9)	O(1)#3-Rb(1)-O(1)#6	117.9(2)
Rb(1)-O(6)#1	2.891(8)	O(4)#3-Rb(1)-O(1)#6	74.5(2)
Rb(1)-O(6)#2	2.892(8)	O(5)#4-Rb(1)-O(1)#6	159.7(2)
Rb(1)-O(1)#3	3.002(7)	O(2)#4-Rb(1)-O(1)#6	145.1(2)
Rb(1)-O(4)#3	3.004(8)	O(7)#5-Rb(1)-O(1)#6	48.68(18)
Rb(1)-O(5)#4	3.148(10)	O(6)#5-Rb(2)-O(6)	124.7(3)
Rb(1)-O(2)#4	3.184(9)	O(6)#5-Rb(2)-O(5)	117.04(15)
Rb(1)-O(7)#5	3.245(7)	O(6)-Rb(2)-O(5)	117.04(15)
Rb(1)-O(1)#6	3.469(7)	O(6)#5-Rb(2)-O(3)	168.3(2)
Rb(2)-O(6)#5	2.949(8)	O(6)-Rb(2)-O(3)	47.5(2)
Rb(2)-O(6)	2.949(8)	O(5)-Rb(2)-O(3)	73.10(18)
Rb(2)-O(5)	3.057(12)	O(6)#5-Rb(2)-O(3)#5	47.5(2)
Rb(2)-O(3)	3.170(8)	O(6)-Rb(2)-O(3)#5	168.3(2)
Rb(2)-O(3)#5	3.170(8)	O(5)-Rb(2)-O(3)#5	73.10(18)
Rb(2)-O(7)#7	3.227(8)	O(3)-Rb(2)-O(3)#5	138.2(3)
Rb(2)-O(7)#8	3.227(8)	O(6)#5-Rb(2)-O(7)#7	67.0(2)
P(1)-O(5)	1.487(11)	O(6)-Rb(2)-O(7)#7	106.3(2)
P(1)-O(2)	1.498(10)	O(5)-Rb(2)-O(7)#7	85.9(2)
P(1)-O(7)	1.574(7)	O(3)-Rb(2)-O(7)#7	121.6(2)
P(1)-O(7)#5	1.574(7)	O(3)#5-Rb(2)-O(7)#7	79.4(2)
P(2)-O(6)	1.485(8)	O(6)#5-Rb(2)-O(7)#8	106.3(2)
P(2)-O(3)	1.519(9)	O(6)-Rb(2)-O(7)#8	67.0(2)
P(2)-O(4)	1.540(9)	O(5)-Rb(2)-O(7)#8	85.9(2)
P(2)-O(1)	1.555(8)	O(3)-Rb(2)-O(7)#8	79.4(2)
Al(1)-O(3)	1.710(9)	O(3)#5-Rb(2)-O(7)#8	121.6(2)
Al(1)-O(4)#9	1.715(9)	O(7)#7-Rb(2)-O(7)#8	44.5(3)
Al(1)-O(7)	1.722(8)	O(6)#5-Rb(2)-O(7)#5	94.55(18)
Al(1)-O(1)#10	1.734(8)	O(6)-Rb(2)-O(7)#5	133.5(2)
O(2)-Rb(1)-O(6)#1	110.5(3)	O(5)-Rb(2)-O(7)#5	44.0(2)
O(2)-Rb(1)-O(6)#2	73.9(2)	O(3)-Rb(2)-O(7)#5	89.26(19)
O(6)#1-Rb(1)-O(6)#2	147.4(3)	O(3)#5-Rb(2)-O(7)#5	49.0(2)
O(2)-Rb(1)-O(1)#3	169.1(2)	O(7)#7-Rb(2)-O(7)#5	112.03(18)
O(6)#1-Rb(1)-O(1)#3	73.8(2)	O(7)#8-Rb(2)-O(7)#5	129.48(17)
O(6)#2-Rb(1)-O(1)#3	108.1(2)	O(6)#5-Rb(2)-O(7)	133.5(2)
O(2)-Rb(1)-O(4)#3	140.4(2)	O(6)-Rb(2)-O(7)	94.55(18)
O(6)#1-Rb(1)-O(4)#3	86.9(2)	O(5)-Rb(2)-O(7)	44.0(2)
O(6)#2-Rb(1)-O(4)#3	73.2(2)	O(3)-Rb(2)-O(7)	49.0(2)
O(1)#3-Rb(1)-O(4)#3	48.0(2)	O(3)#5-Rb(2)-O(7)	89.26(19)
O(2)-Rb(1)-O(5)#4	91.2(2)	O(7)#7-Rb(2)-O(7)	129.48(17)
O(6)#1-Rb(1)-O(5)#4	114.7(3)	O(7)#8-Rb(2)-O(7)	112.03(18)
O(6)#2-Rb(1)-O(5)#4	97.1(3)	O(7)#5-Rb(2)-O(7)	40.2(2)
O(1)#3-Rb(1)-O(5)#4	78.0(2)	O(5)-P(1)-O(2)	114.3(7)

O(4)#3-Rb(1)-O(5)#4	114.1(2)	O(5)-P(1)-O(7)	110.9(4)
O(2)-Rb(1)-O(2)#4	96.01(8)	O(2)-P(1)-O(7)	109.0(4)
O(6)#1-Rb(1)-O(2)#4	69.7(2)	O(5)-P(1)-O(7)#5	110.9(4)
O(6)#2-Rb(1)-O(2)#4	142.9(2)	O(2)-P(1)-O(7)#5	109.0(4)
O(1)#3-Rb(1)-O(2)#4	75.9(2)	O(7)-P(1)-O(7)#5	102.0(6)
O(4)#3-Rb(1)-O(2)#4	123.5(2)	O(6)-P(2)-O(3)	110.9(5)
O(5)#4-Rb(1)-O(2)#4	46.6(3)	O(6)-P(2)-O(4)	112.6(5)
O(2)-Rb(1)-O(7)#5	47.7(2)	O(3)-P(2)-O(4)	106.0(5)
O(6)#1-Rb(1)-O(7)#5	67.4(2)	O(6)-P(2)-O(1)	113.6(5)
O(6)#2-Rb(1)-O(7)#5	99.4(2)	O(3)-P(2)-O(1)	109.1(4)
O(1)#3-Rb(1)-O(7)#5	139.90(19)	O(4)-P(2)-O(1)	104.2(4)
O(4)#3-Rb(1)-O(7)#5	118.4(2)	O(3)-Al(1)-O(4)#9	108.0(5)
O(5)#4-Rb(1)-O(7)#5	127.5(2)	O(3)-Al(1)-O(7)	110.0(4)
O(2)#4-Rb(1)-O(7)#5	99.2(2)	O(4)#9-Al(1)-O(7)	111.0(4)
O(2)-Rb(1)-O(1)#6	72.8(2)	O(3)-Al(1)-O(1)#10	113.9(5)
O(6)#1-Rb(1)-O(1)#6	83.23(18)	O(4)#9-Al(1)-O(1)#10	107.2(4)
O(6)#2-Rb(1)-O(1)#6	66.9(2)	O(7)-Al(1)-O(1)#10	106.8(4)

- #1  $-x+1, -y+1, z+1/2$     #2  $-x+1, y+1, z$   
 #3  $x-1/2, -y+3/2, z+1/2$     #4  $-x+1, -y+2, z+1/2$   
 #5  $-x+1, y, z$     #6  $x-1/2, y+1/2, z$     #7  $-x+1, -y+1, z-1/2$   
 #8  $x, -y+1, z-1/2$     #9  $x, -y+1, z+1/2$     #10  $-x+3/2, y+1/2, z$   
 #11  $x, -y+2, z-1/2$     #12  $-x+1, -y+2, z-1/2$   
 #13  $x+1/2, -y+3/2, z-1/2$     #14  $-x+1, y-1, z$   
 #15  $x+1/2, y-1/2, z$     #16  $-x+3/2, y-1/2, z$

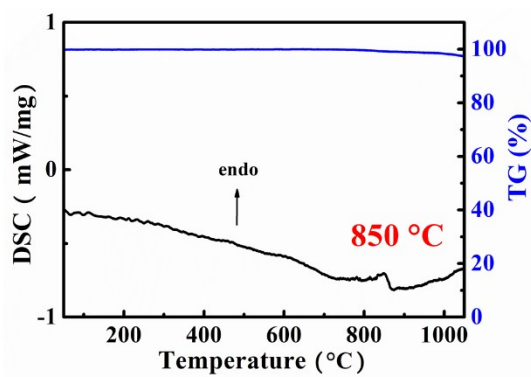
**Table S3b.** Selected bond distances (Å) and angles (deg) for K<sub>3</sub>Al<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>.

K(1)-O(7)	2.660(5)	O(10)-K(2)-O(5)#2	144.77(17)
K(1)-O(9)#1	2.741(5)	O(10)-K(2)-O(7)#6	82.72(17)
K(1)-O(5)#2	2.811(5)	O(5)#2-K(2)-O(7)#6	111.59(15)
K(1)-O(6)#3	2.955(5)	O(10)-K(2)-O(3)#5	88.43(17)
K(1)-O(10)#3	2.985(6)	O(5)#2-K(2)-O(3)#5	86.06(14)
K(1)-O(12)#4	3.015(5)	O(7)#6-K(2)-O(3)#5	159.57(14)
K(1)-O(2)#5	3.093(5)	O(10)-K(2)-O(9)	87.69(16)
K(1)-O(6)#4	3.204(5)	O(5)#2-K(2)-O(9)	126.17(17)
K(2)-O(10)	2.669(6)	O(7)#6-K(2)-O(9)	77.07(14)
K(2)-O(5)#2	2.721(5)	O(3)#5-K(2)-O(9)	84.23(14)
K(2)-O(7)#6	2.847(5)	O(10)-K(2)-O(7)	132.73(16)
K(2)-O(3)#5	2.858(5)	O(5)#2-K(2)-O(7)	76.69(14)
K(2)-O(9)	2.935(6)	O(7)#6-K(2)-O(7)	104.15(8)
K(2)-O(7)	2.978(6)	O(3)#5-K(2)-O(7)	68.91(14)
K(2)-O(8)#6	2.988(5)	O(9)-K(2)-O(7)	50.49(14)
K(2)-O(1)#5	3.298(5)	O(10)-K(2)-O(8)#6	99.35(16)
K(3)-O(9)	2.641(6)	O(5)#2-K(2)-O(8)#6	70.68(14)
K(3)-O(10)	2.734(6)	O(7)#6-K(2)-O(8)#6	50.55(12)
K(3)-O(5)#7	2.737(5)	O(3)#5-K(2)-O(8)#6	149.69(13)
K(3)-O(4)#7	2.777(5)	O(9)-K(2)-O(8)#6	125.06(14)
K(3)-O(11)	3.215(6)	O(7)-K(2)-O(8)#6	121.36(14)
P(1)-O(5)	1.491(4)	O(10)-K(2)-O(1)#5	72.89(14)
P(1)-O(1)	1.538(5)	O(5)#2-K(2)-O(1)#5	78.31(13)
P(1)-O(2)	1.545(5)	O(7)#6-K(2)-O(1)#5	144.62(15)
P(1)-O(3)	1.548(4)	O(3)#5-K(2)-O(1)#5	46.30(12)
P(2)-O(10)	1.477(5)	O(9)-K(2)-O(1)#5	125.79(14)
P(2)-O(6)	1.529(6)	O(7)-K(2)-O(1)#5	111.22(14)
P(2)-O(11)	1.532(5)	O(8)#6-K(2)-O(1)#5	108.07(13)
P(2)-O(12)	1.535(5)	O(9)-K(3)-O(10)	92.62(17)
P(3)-O(7)	1.496(5)	O(9)-K(3)-O(5)#7	93.07(16)
P(3)-O(9)	1.498(6)	O(10)-K(3)-O(5)#7	127.69(17)
P(3)-O(8)	1.558(5)	O(9)-K(3)-O(4)#7	104.02(18)
P(3)-O(4)	1.576(5)	O(10)-K(3)-O(4)#7	135.04(16)
Al(1)-O(4)	1.730(5)	O(5)#7-K(3)-O(4)#7	93.29(14)
Al(1)-O(2)	1.730(5)	O(9)-K(3)-O(11)	110.85(15)
Al(1)-O(6)#3	1.739(6)	O(10)-K(3)-O(11)	49.15(14)
Al(1)-O(3)#5	1.743(5)	O(5)#7-K(3)-O(11)	155.57(16)
Al(2)-O(11)	1.720(5)	O(4)#7-K(3)-O(11)	85.92(13)
Al(2)-O(8)#7	1.725(5)	O(5)-P(1)-O(1)	114.3(3)
Al(2)-O(1)	1.738(5)	O(5)-P(1)-O(2)	111.2(3)
Al(2)-O(12)#8	1.747(5)	O(1)-P(1)-O(2)	106.6(3)
O(7)-K(1)-O(9)#1	83.68(16)	O(5)-P(1)-O(3)	112.7(3)

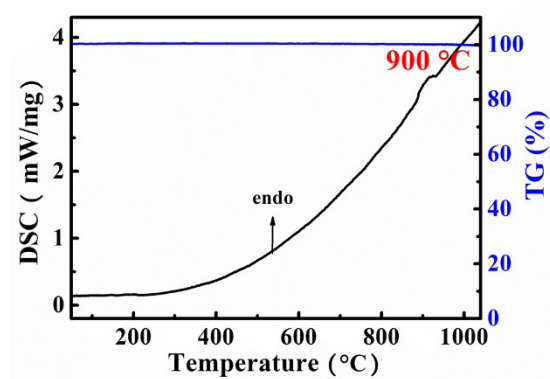
O(7)-K(1)-O(5)#2	80.62(16)	O(1)-P(1)-O(3)	105.3(3)
O(9)#1-K(1)-O(5)#2	89.35(15)	O(2)-P(1)-O(3)	106.1(3)
O(7)-K(1)-O(6)#3	79.03(15)	O(10)-P(2)-O(6)	110.9(3)
O(9)#1-K(1)-O(6)#3	140.67(16)	O(10)-P(2)-O(11)	113.2(3)
O(5)#2-K(1)-O(6)#3	121.74(15)	O(6)-P(2)-O(11)	105.9(3)
O(7)-K(1)-O(10)#3	94.57(16)	O(10)-P(2)-O(12)	112.5(3)
O(9)#1-K(1)-O(10)#3	97.99(16)	O(6)-P(2)-O(12)	106.2(3)
O(5)#2-K(1)-O(10)#3	170.77(15)	O(11)-P(2)-O(12)	107.6(3)
O(6)#3-K(1)-O(10)#3	49.24(14)	O(7)-P(3)-O(9)	114.8(4)
O(7)-K(1)-O(12)#4	160.98(16)	O(7)-P(3)-O(8)	109.5(3)
O(9)#1-K(1)-O(12)#4	79.12(15)	O(9)-P(3)-O(8)	110.1(3)
O(5)#2-K(1)-O(12)#4	107.08(15)	O(7)-P(3)-O(4)	109.4(3)
O(6)#3-K(1)-O(12)#4	109.62(15)	O(9)-P(3)-O(4)	109.3(3)
O(10)#3-K(1)-O(12)#4	79.94(15)	O(8)-P(3)-O(4)	103.0(3)
O(7)-K(1)-O(2)#5	104.76(15)	O(4)-Al(1)-O(2)	108.8(2)
O(9)#1-K(1)-O(2)#5	153.17(15)	O(4)-Al(1)-O(6)#3	117.4(3)
O(5)#2-K(1)-O(2)#5	67.57(13)	O(2)-Al(1)-O(6)#3	104.9(3)
O(6)#3-K(1)-O(2)#5	66.07(13)	O(4)-Al(1)-O(3)#5	109.9(2)
O(10)#3-K(1)-O(2)#5	106.48(14)	O(2)-Al(1)-O(3)#5	112.6(2)
O(12)#4-K(1)-O(2)#5	94.27(14)	O(6)#3-Al(1)-O(3)#5	103.1(3)
O(7)-K(1)-O(6)#4	150.67(17)	O(11)-Al(2)-O(8)#7	110.6(3)
O(9)#1-K(1)-O(6)#4	110.27(15)	O(11)-Al(2)-O(1)	106.8(2)
O(5)#2-K(1)-O(6)#4	74.11(13)	O(8)#7-Al(2)-O(1)	108.4(3)
O(6)#3-K(1)-O(6)#4	101.75(9)	O(11)-Al(2)-O(12)#8	113.6(3)
O(10)#3-K(1)-O(6)#4	108.20(15)	O(8)#7-Al(2)-O(12)#8	106.7(3)
O(12)#4-K(1)-O(6)#4	46.26(14)	O(1)-Al(2)-O(12)#8	110.6(3)
O(2)#5-K(1)-O(6)#4	51.76(12)		

#1  $-x, -y+1, z-1/2$     #2  $x-1, y, z$     #3  $x, y, z-1$   
 #4  $x-1/2, -y+3/2, z-1$     #5  $x-1/2, -y+3/2, z$   
 #6  $-x, -y+1, z+1/2$     #7  $-x+1, -y+1, z+1/2$   
 #8  $x+1/2, -y+3/2, z$     #9  $-x+1, -y+1, z-1/2$   
 #10  $x+1, y, z$     #11  $x, y, z+1$     #12  $x+1/2, -y+3/2, z+1$



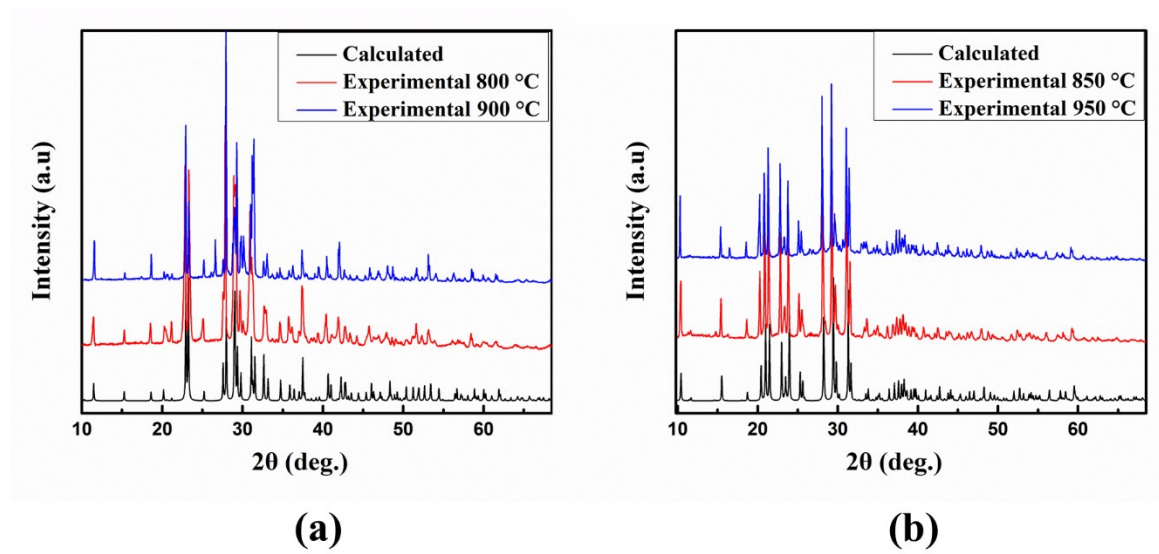


(a)



(b)

**Fig. S1.** TG-DSC curves of (a)  $\text{Rb}_3\text{Al}_2(\text{PO}_4)_3$  and (b)  $\text{K}_3\text{Al}_2(\text{PO}_4)_3$ .



**Fig. S2.** Experimental and calculated XRD patterns of (a)  $\text{Rb}_3\text{Al}_2(\text{PO}_4)_3$  and (b)  $\text{K}_3\text{Al}_2(\text{PO}_4)_3$ .

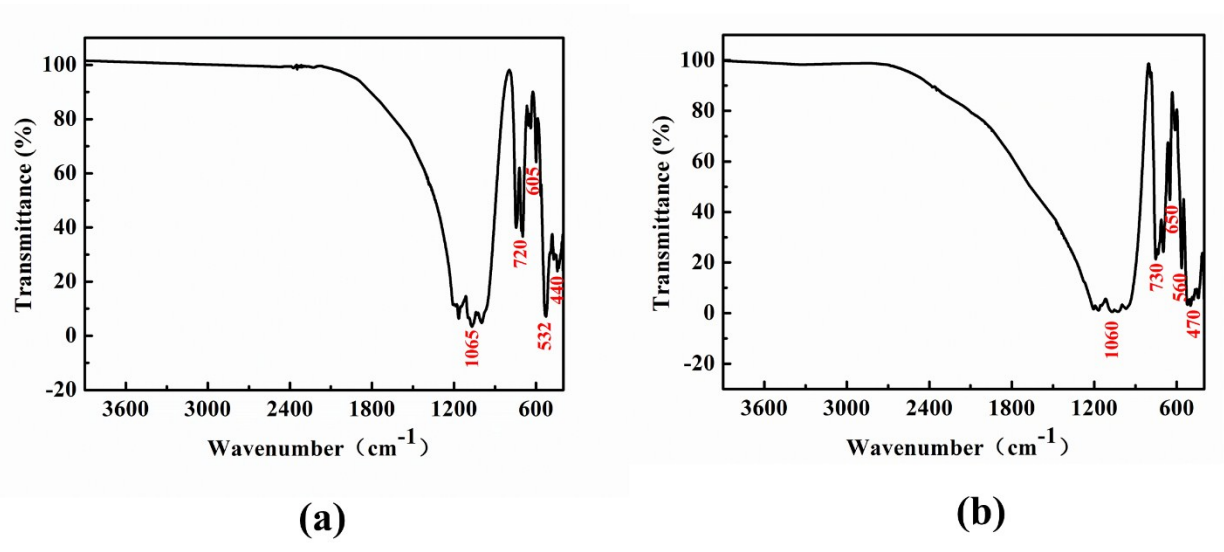
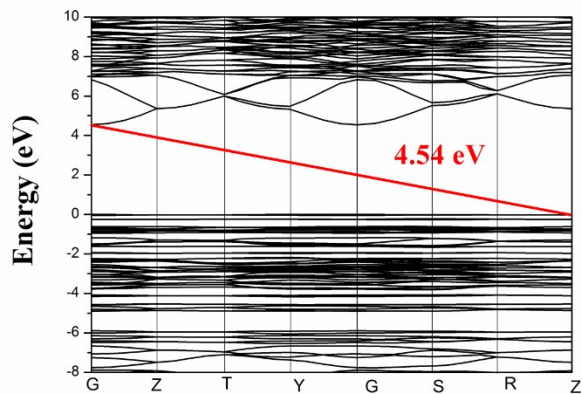
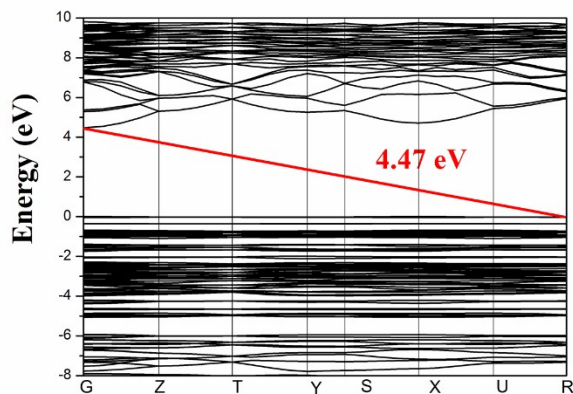


Fig. S3. IR spectra of (a) Rb<sub>3</sub>Al<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub> and (b) K<sub>3</sub>Al<sub>2</sub>(PO<sub>4</sub>)<sub>3</sub>.

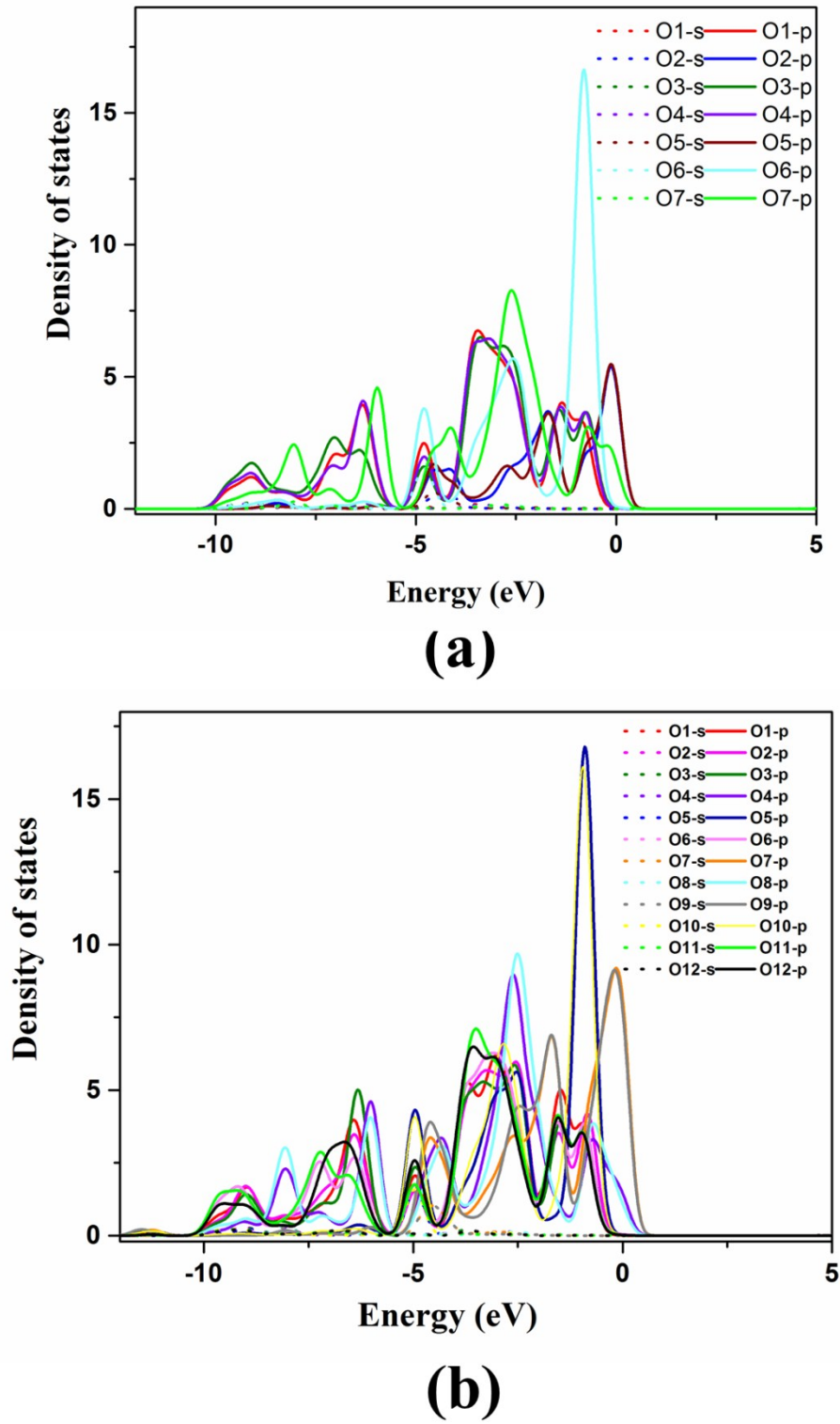


(a)

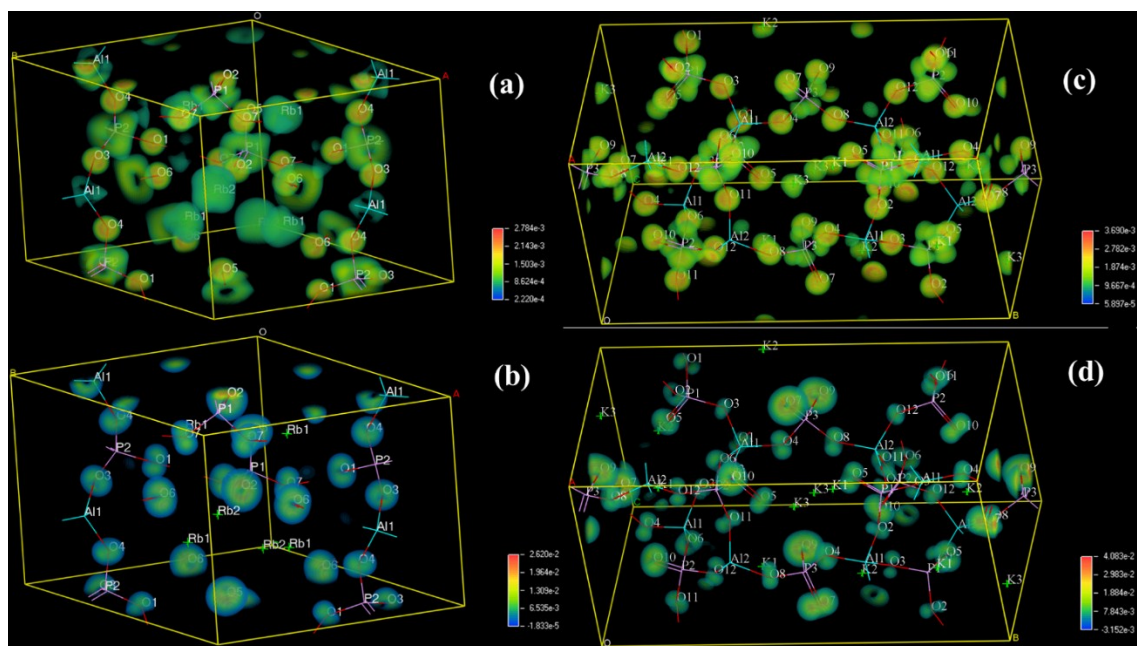


(b)

**Fig. S4.** Band structures of (a)  $\text{Rb}_3\text{Al}_2(\text{PO}_4)_3$  and (b)  $\text{K}_3\text{Al}_2(\text{PO}_4)_3$ .



**Fig. S5.** The detailed PDOSs of the O atoms in (a)  $\text{Rb}_3\text{Al}_2(\text{PO}_4)_3$  and (b)  $\text{K}_3\text{Al}_2(\text{PO}_4)_3$ .



**Fig. S6.** SHG density of VE processes of (a) unoccupied states and (b) occupied states for  $\text{Rb}_3\text{Al}_2(\text{PO}_4)_3$ ; (c) unoccupied states and (d) occupied states for  $\text{K}_3\text{Al}_2(\text{PO}_4)_3$ .