Electronic Supplementary Information (ESI) for

One-step topological preparation of carbon doped and coated TiO₂ hollow nanocubes and synergistically enhancing visible photodegradation activity

Chengjiang Zhang^a, Amin Cao^a, Lianqing Chen^{a,b*}, Kangle Lv^a, Tsunghsueh Wu^b, Kejian Deng^a

^a Key Laboratory of Catalysis and Materials Science of the State Ethnic Affairs Commission &
Ministry of Education, South-Central University for Nationalities, Wuhan, 430074, P. R. China.
^b Department of Chemistry, University of Wisconsin-Platteville, Platteville, 53818, United States

Corresponding Author:

Lianqing Chen, Tel. & Fax. +86-27-67842752

E-mail: chenlia@uwplatt.edu, lqchen@mail.scuec.edu.cn (L.Q Chen)

1. EXPERIMENTAL SECTION



Scheme S1. Formation of 7-hydroxycoumarin in the reaction of coumarin with hydroxyl radical.

2. RESULTS AND DISCUSSION SECTION



Fig. S1. TEM images of as-synthesized catalysts at different reaction time: (a) TiOF₂, (b) and (c) TiO₂-HNBs, (d)

T-12, (e) T-18, (f) T-24, (g) T-36, (h) T-48.



Fig. S2. TEM images of as-prepared catalysts with different mass ratios of $Glu/TiOF_2$ at 24 h: (a) T24-0.05, (b) T24-0.1, (c) T24-0.15, (d) T24-0.2, (e) T24-0.25, (f) HRTEM image of T24-0.1.



Fig. S3. The scan EDS maps of sample T24-0.15.



Fig. S4. FT-IR diagram of the catalyst (A Glu/TiOF₂ = 0.1/1 at 200 °C different reaction time, B Glu/TiOF₂ at different values reaction at 200 °C for 24 h).



Fig. S5. Raman spectra of as-prepared C@TiO2-HNBs.



Fig. S6. DRS diagram of as-prepared C@TiO₂-HNBs (A With Glu/TiOF₂=0.1/1 at different reaction time; B With different Glu/TiOF₂ mass ratio reacted at 200 °C for 24 h; C the plot of transformed Kubelka-Munk function *versus* the photon for TiO₂ and T24-0.15 samples).