

**CuI nanoparticles Catalyzed Synthesis of Tetracyclic Heterocycles
benzo[*e*]benzo[4,5]imidazo[1,2-*c*][1,3]thiazin-6-imine by S_NAr-Type C-S, C-N
Bond Formation with isothiocyanatobenzenes and Benzimidazoles**

Xiaolong Guo, Luyao Wang*, Jing Hu and Mengmeng Zhang

Department of Chemistry, Capital Normal University, Beijing 100048, People's Republic of China

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1. Crystal data of 3b

Table S1 Data collection and structure refinement of 3b (CCDC No. 1818722).

Compound	3b
Empirical formula	C ₂₁ H ₁₅ N ₃ S
Formula weight	341.42
Crystal system	monoclinic
Space group	C12/c1
T(K)	299
Λ (Å)	1.54184
A (Å)	23.3793(3)
B (Å)	7.23290(10)
C (Å)	19.5343(3)
α (°)	90
β (°)	92.6530(10)
γ (°)	90
V(Å ³)	3299.71(8)
Z	8
D _{calc} (mg/m ³)	1.375
F(000)	1424
Crystal size (mm ³)	0.4 x 0.2 x 0.2
θ range for data collection (deg.)	3.785 to 66.565
Data/restraints/parameters	2882 / 0 / 226
Goodness-of-fit on F ²	1.094
Final R ₁ /wR ₂ indices [I > 2 σ (I)]	0.0471 / 0.1295
R indices (all data)	0.0500
Largest diff. peak and hole(A ⁻³)	0.199 and -0.491

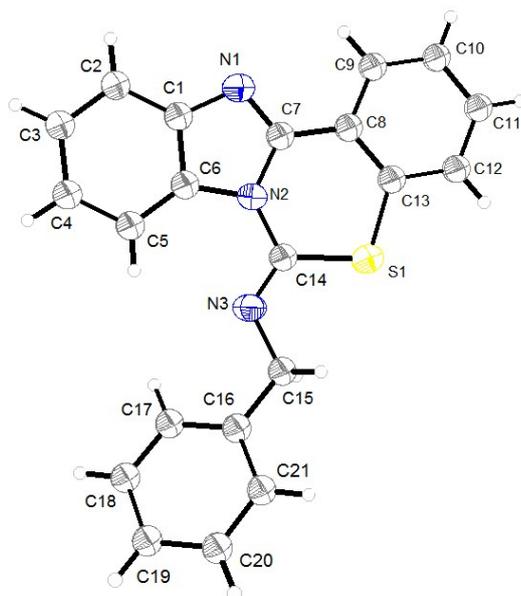


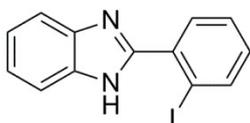
Figure S1. ORTEP plot of 3b showing thermal ellipsoids at the 50% probability level

2. General Procedures

General Procedure for the Preparation of Substituted 2-(2-Halophenyl)-1H-benzo[d]imidazoles (1)¹

To a flask were added substituted 1,2-diaminobenzene (1.08 g, 10 mmol), 2-halobenzoic acid (10 mmol) and PPA (20 g), and the reaction mixture was heated at 170 °C in an oil bath for 7 h. After the reaction was complete, the reaction mixture was slowly poured into ice water, and the resulting mixture was basified with solid NaOH and NaHCO₃. Once the solution had reached pH 8-10, the precipitate was isolated by filtration, washed with cold water and recrystallized from hot aqueous ethanol. The solid was then dried to afford the corresponding 2-(2-halophenyl)-1H-benzo[d]imidazole.

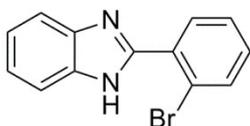
2-(2-iodophenyl)-1H-benzo[d]imidazole



¹H NMR (600 MHz, dmsO) δ 12.70 (s, 1H), 8.06 (dd, J = 8.0, 1.0 Hz, 1H), 7.70 (d, J = 7.8 Hz, 1H), 7.64 – 7.60 (m, 1H), 7.55 (ddd, J = 10.8, 8.5, 4.5 Hz, 2H), 7.30 – 7.19 (m, 3H).

¹³C NMR (151 MHz, dmsO) δ 152.54, 143.15, 139.61, 136.55, 134.32, 131.31, 131.16, 128.09, 122.51, 121.48, 119.10, 111.44, 97.34.

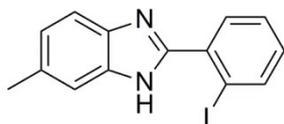
2-(2-bromophenyl)-1H-benzo[d]imidazole



¹H NMR (600 MHz, dmsO) δ 12.73 (s, 1H), 7.82 (dd, J = 8.1, 1.1 Hz, 1H), 7.77 (dd, J = 7.6, 1.7 Hz, 1H), 7.70 (d, J = 7.8 Hz, 1H), 7.59 – 7.52 (m, 2H), 7.49 – 7.44 (m, 1H), 7.24 (dt, J = 15.1, 6.7 Hz, 2H).

¹³C NMR (151 MHz, dmsO) δ 150.38, 143.20, 134.45, 133.36, 132.38, 132.22, 131.32, 127.75, 122.62, 121.56, 121.51, 119.08, 111.54.

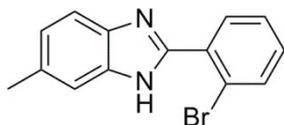
2-(2-iodophenyl)-6-methyl-1H-benzo[d]imidazole



¹H NMR (600 MHz, dmsO) δ 12.55 (s, 1H), 8.05 (d, J = 7.9 Hz, 1H), 7.60 (d, J = 7.4 Hz, 1H), 7.54 (t, J = 7.4 Hz, 1H), 7.49 (s, 1H), 7.38 (s, 1H), 7.26 (t, J = 7.6 Hz, 1H), 7.05 (d, J = 8.0 Hz, 1H), 2.44 (s, 3H).

¹³C NMR (151 MHz, dmsO) δ 166.18, 161.19, 160.49, 139.67, 136.66, 134.78, 131.31, 131.14, 130.01, 128.74, 128.14, 97.36, 21.33.

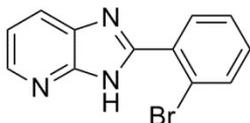
2-(2-bromophenyl)-6-methyl-1H-benzo[d]imidazole



^1H NMR (600 MHz, dmsO) δ 12.58 (s, 1H), 7.81 (dd, J = 8.0, 0.7 Hz, 1H), 7.75 (dd, J = 7.6, 1.6 Hz, 1H), 7.54 (td, J = 7.5, 0.9 Hz, 1H), 7.50 (d, J = 2.1 Hz, 1H), 7.45 (td, J = 7.8, 1.7 Hz, 1H), 7.39 (s, 1H), 7.06 (d, J = 8.1 Hz, 1H), 2.44 (s, 3H).

^{13}C NMR (151 MHz, dmsO) δ 150.00, 133.35, 132.46, 132.17, 131.19, 127.72, 123.52, 121.47, 21.29.

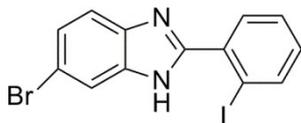
2-(2-bromophenyl)-3H-imidazo[4,5-b]pyridine



^1H NMR (600 MHz, dmsO) δ 13.27 (s, 1H), 8.40 (d, J = 4.1 Hz, 1H), 8.06 (d, J = 7.5 Hz, 1H), 7.84 (d, J = 7.9 Hz, 1H), 7.76 (dd, J = 7.5, 1.0 Hz, 1H), 7.58 (dd, J = 10.9, 4.1 Hz, 1H), 7.50 (td, J = 7.8, 1.6 Hz, 1H), 7.29 (dd, J = 8.0, 4.7 Hz, 1H).

^{13}C NMR (151 MHz, dmsO) δ 152.17, 143.99, 133.38, 132.20, 131.99, 131.73, 127.79, 121.63, 118.13.

6-bromo-2-(2-iodophenyl)-1H-benzo[d]imidazole



^1H NMR (600 MHz, dmsO) δ 12.94 (s, 1H), 8.06 (d, J = 7.3 Hz, 1H), 7.89 (s, 0.5H), 7.71 (s, 0.5H), 7.68 – 7.63 (m, 0.5H), 7.62 (dd, J = 7.6, 1.6 Hz, 1H), 7.56 (td, J = 7.5, 0.8 Hz, 1H), 7.51 (d, J = 7.7 Hz, 0.5H), 7.37 (dd, J = 24.4, 16.2 Hz, 1H), 7.29 (td, J = 7.7, 1.6 Hz, 1H).

^{13}C NMR (151 MHz, dmsO) δ 139.65, 136.02, 131.42, 131.32, 128.15, 125.27, 124.55, 121.50, 120.86, 114.11, 113.32, 97.24.

Preparation of CuI NPs

The nanocatalyst was produced by ultrasonic irradiation using copper sulfate (CuSO_4) as the Cu source. First, CuSO_4 (1 mmol) was cleaned for 20 s in acetone under ultrasonic irradiation followed. The dried substrate was then slowly dipped into a solution of potassium iodide (1 mmol) in 40 mL of distilled water, and the mixture was sonicated for 30 min. When the reaction was complete, the obtained gray precipitate was isolated by filtration, washed with distilled water and dried to afford pure CuI NPs.

According to the diffraction data card (JCPDS, 75-0832); all the peaks can be perfectly indexed to the peak positions characteristic of CuI (**Fig. S2**).

Scanning electron microscopy (SEM) was used to confirm the size of the CuI nanoparticles, and the SEM micrograph is shown in **Fig. S3**. This figure illustrates that, the CuI NPs obtained under ultrasonic irradiation had diameters on the order of nanometers.

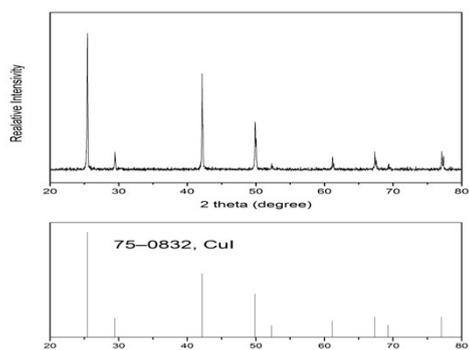


Fig S2. XRD spectra of CuI NPs

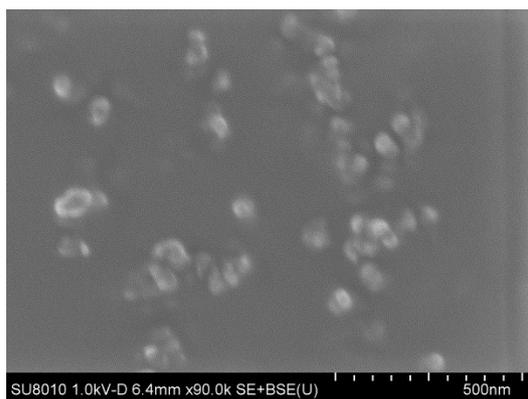
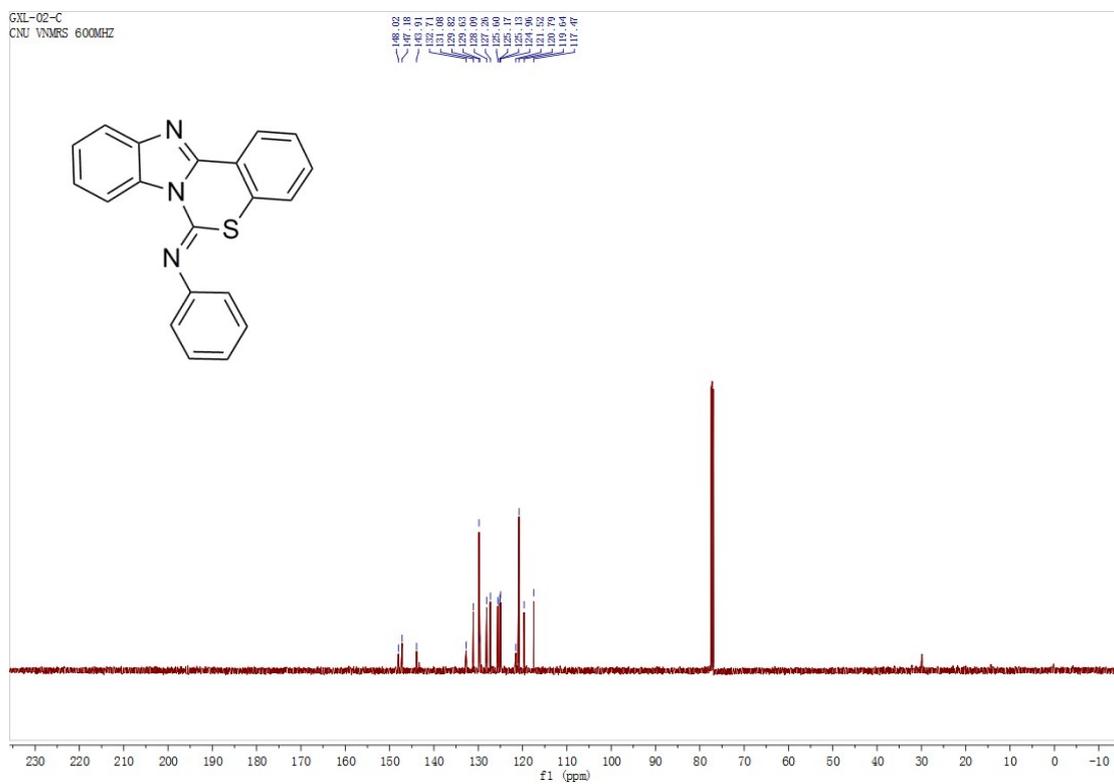
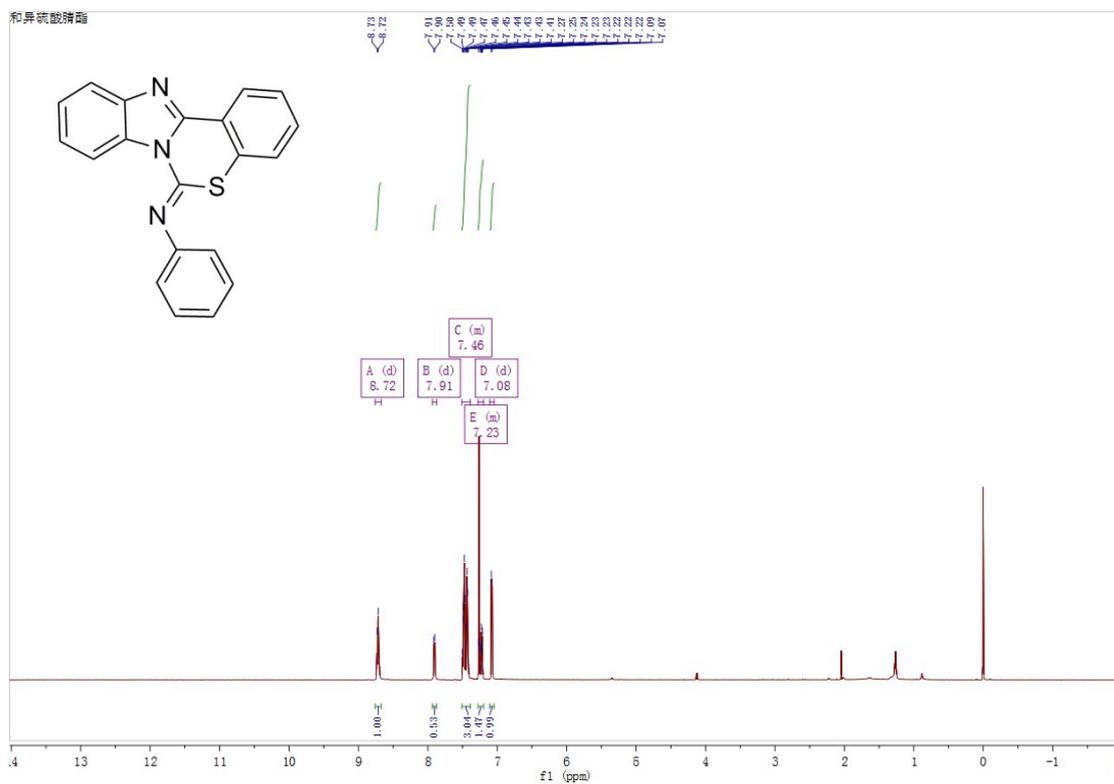


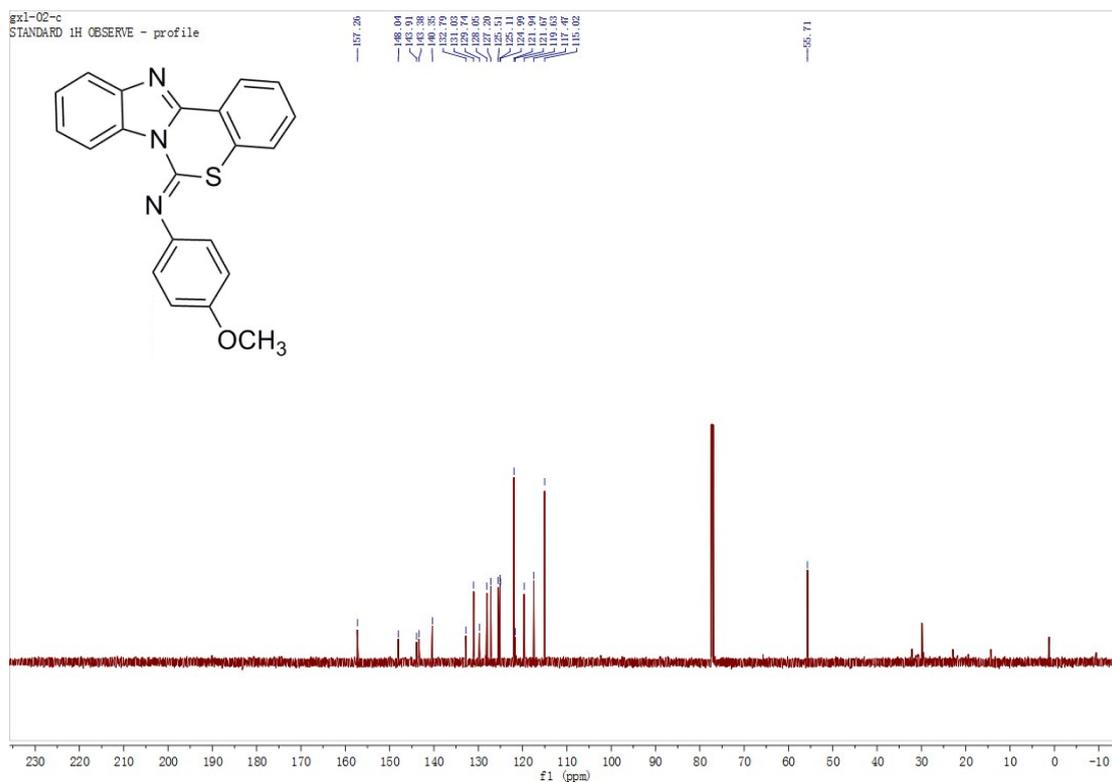
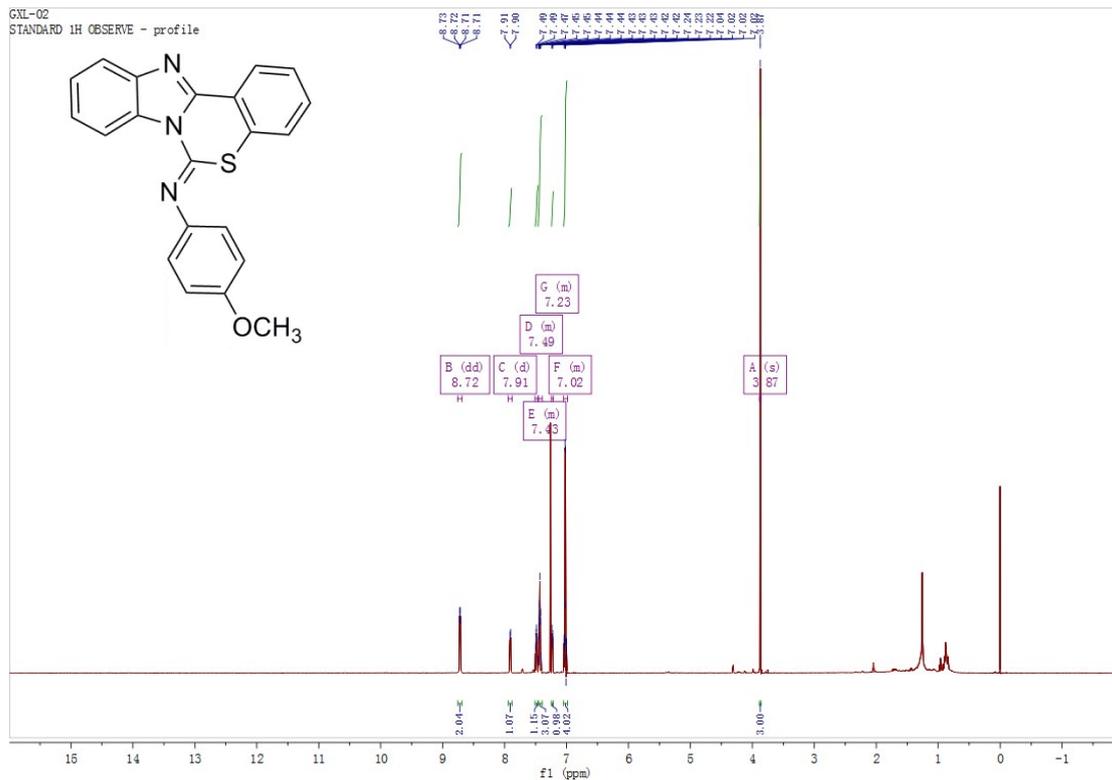
Fig S3. SEM image of CuI NPs

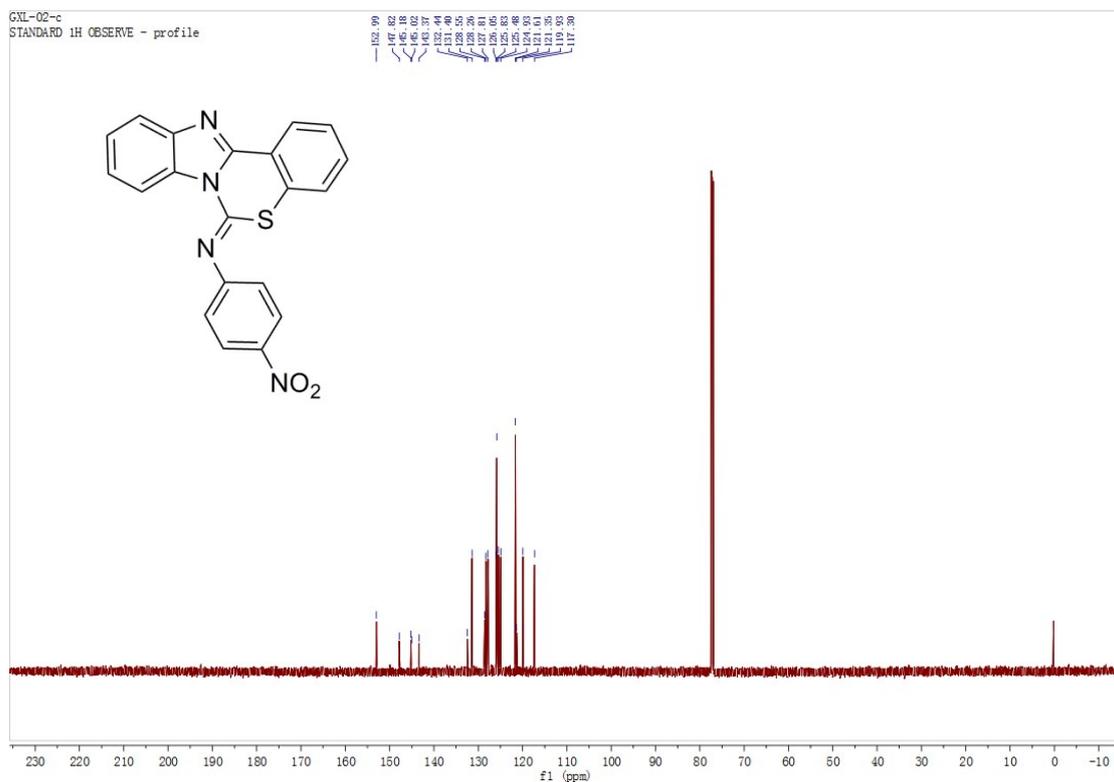
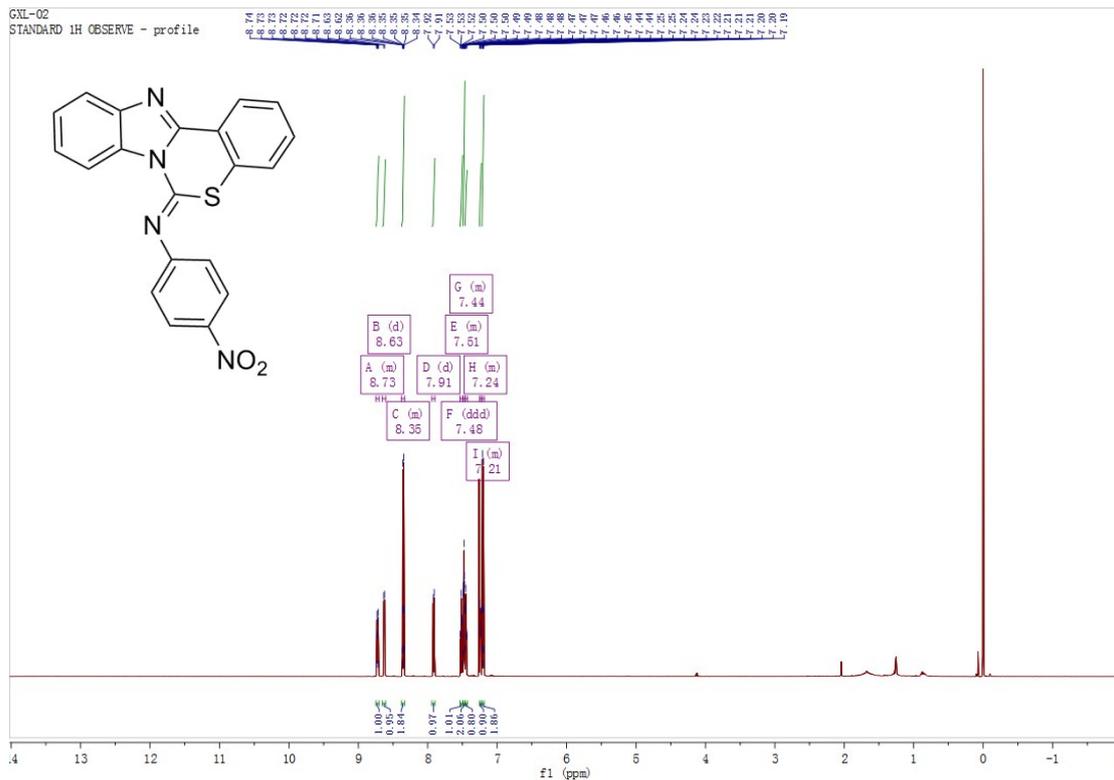
3. References

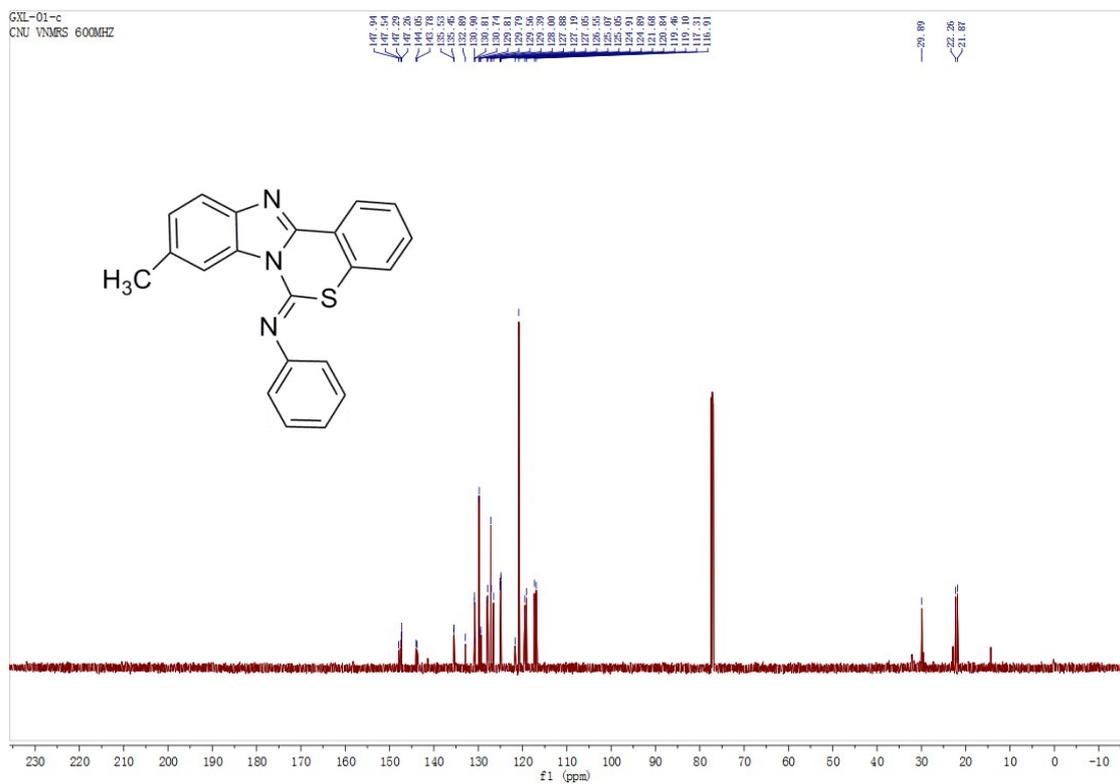
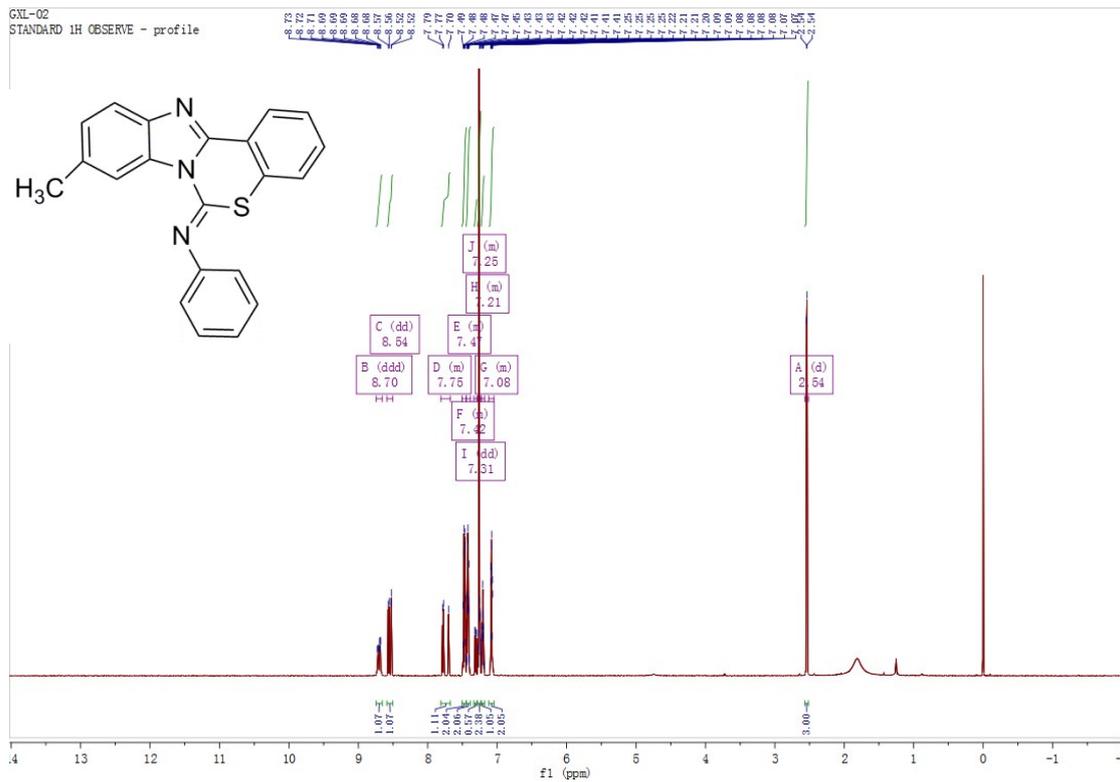
- (1) Xu, S.; Lu, J.; Fu, H. Chem. Commun. 2011, 47, 5596.
- (2) Javad, S-G.; Zeinab A.; Abolfazl Z. RSC Adv. 2014, 4, 16385.

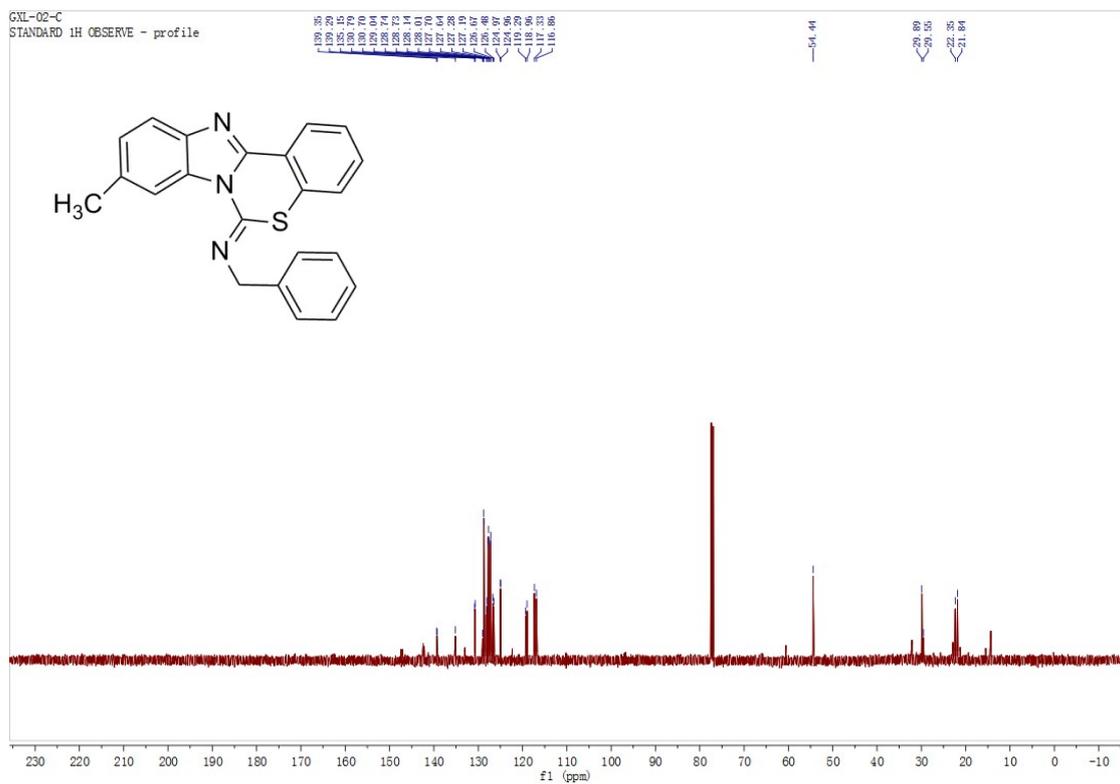
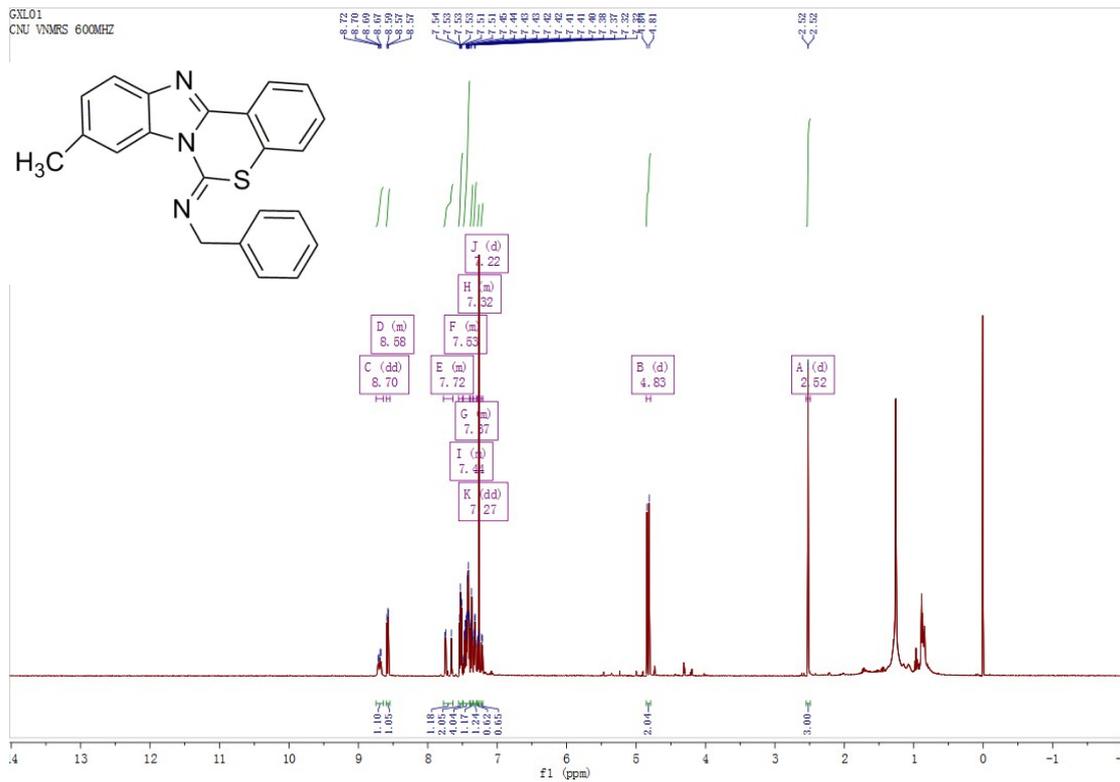
4. Copies of ^1H and ^{13}C NMR Spectra

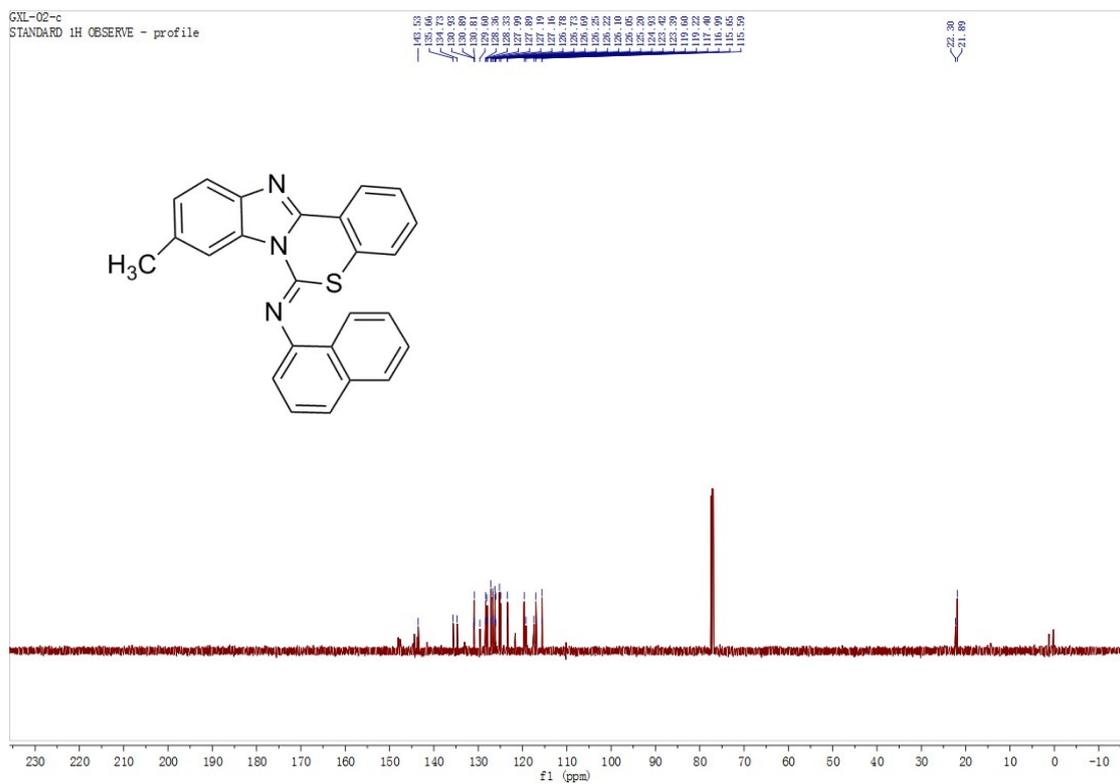
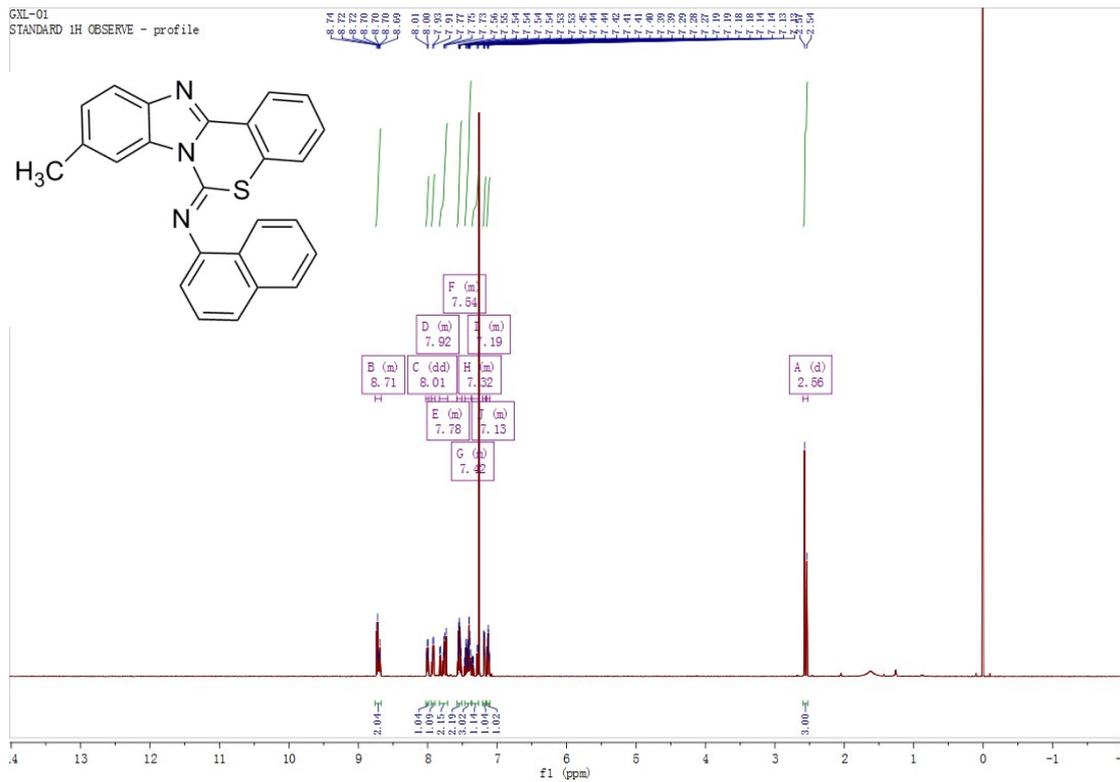


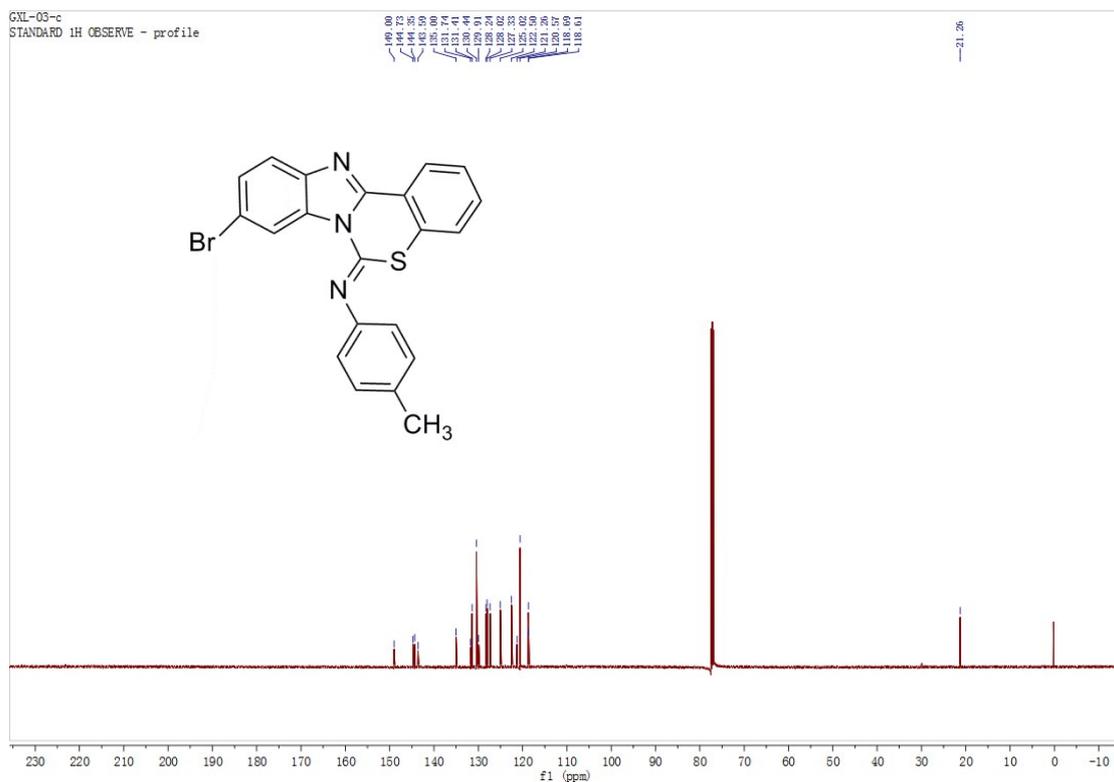
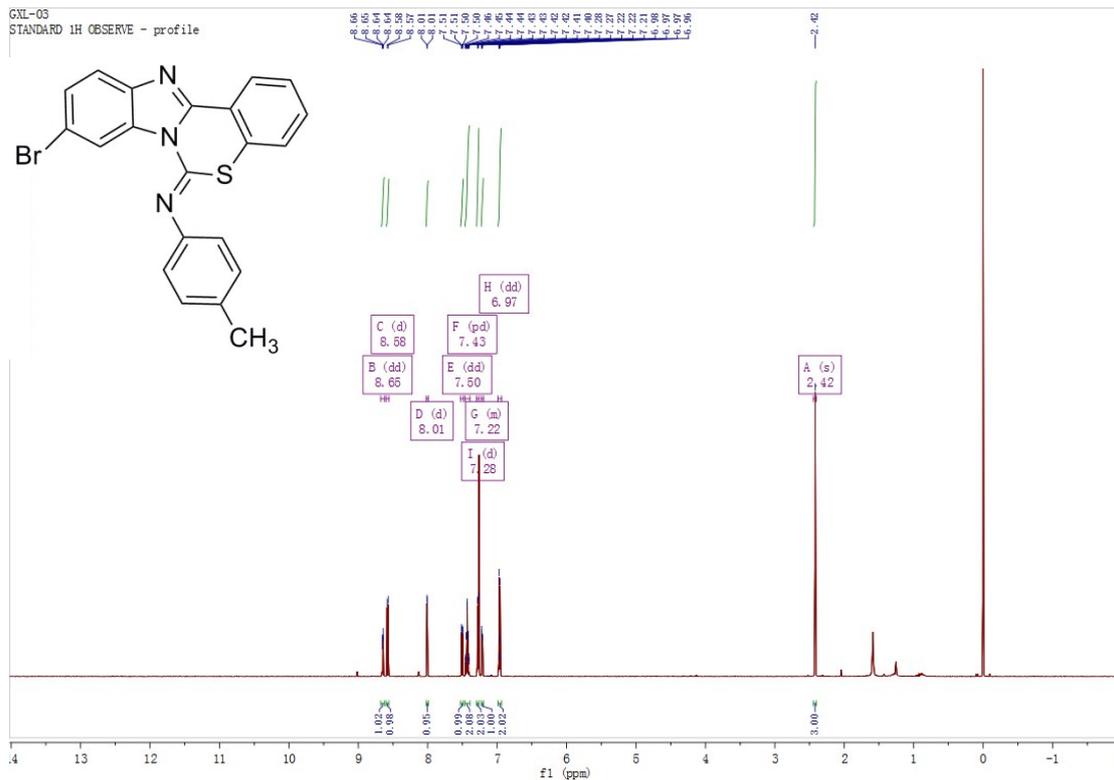


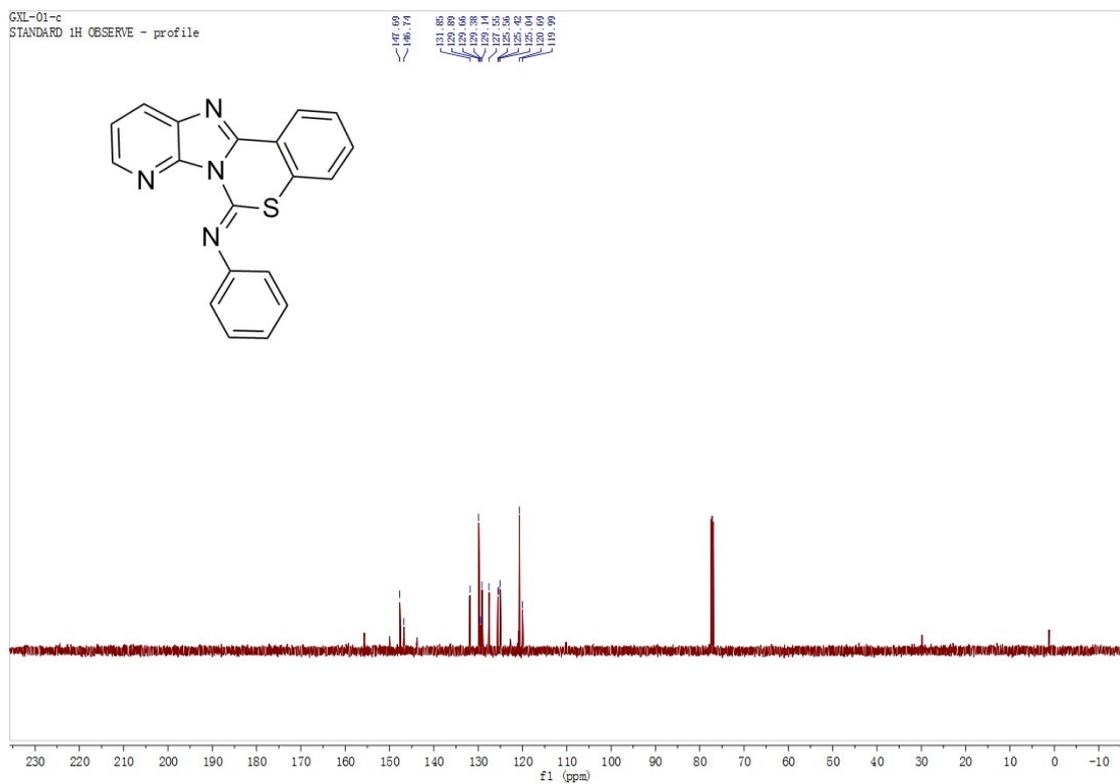
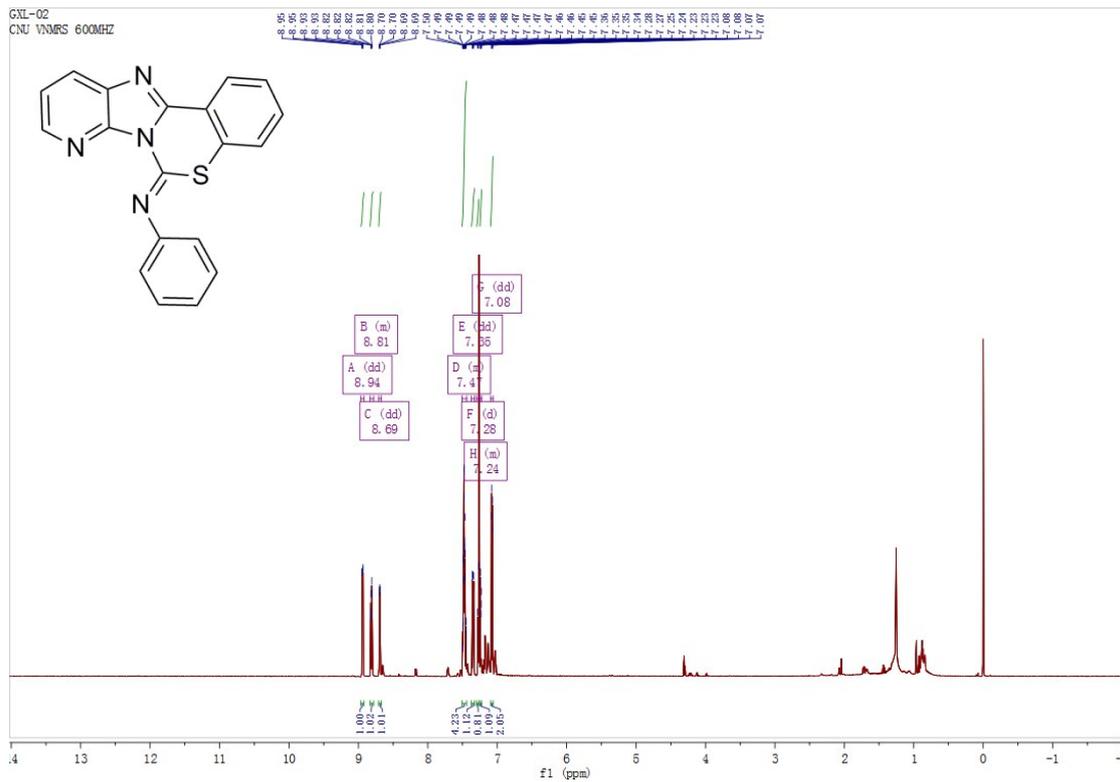




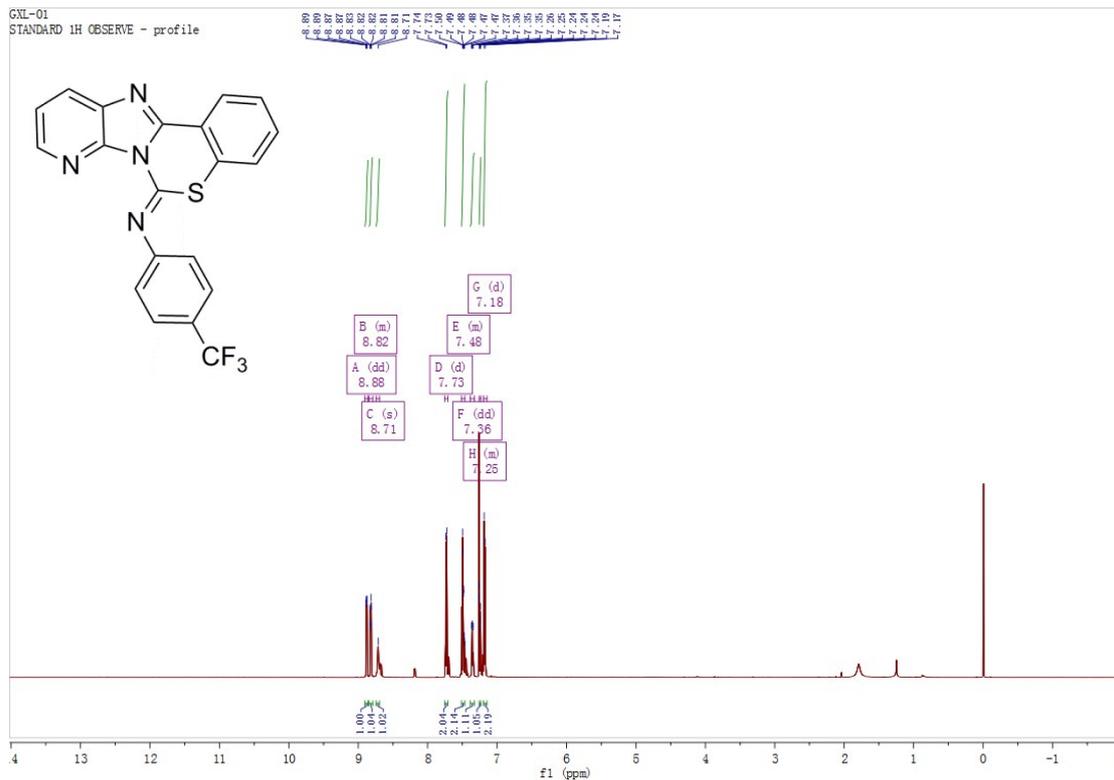




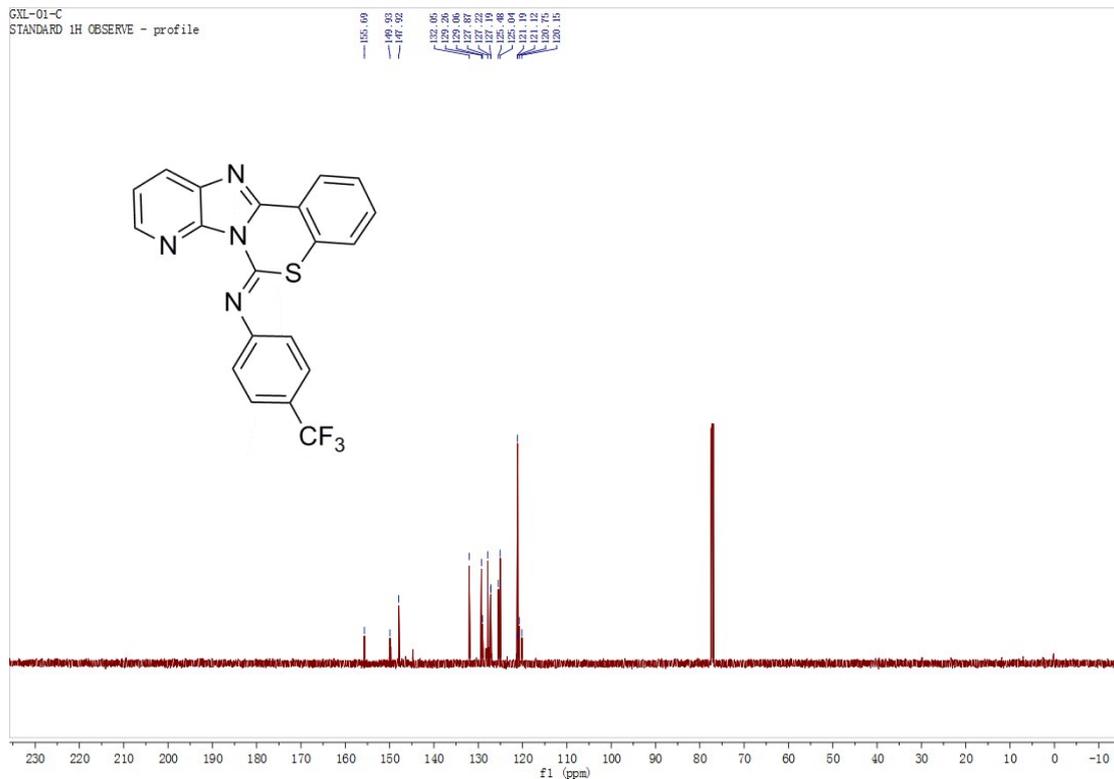


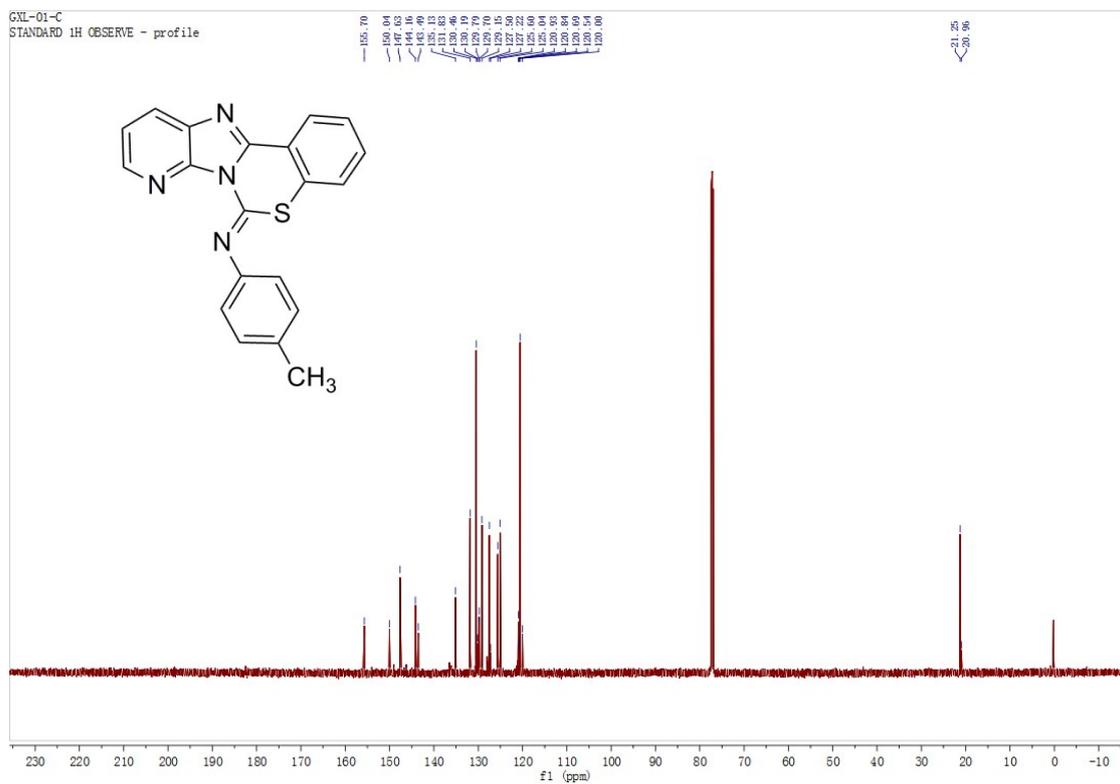
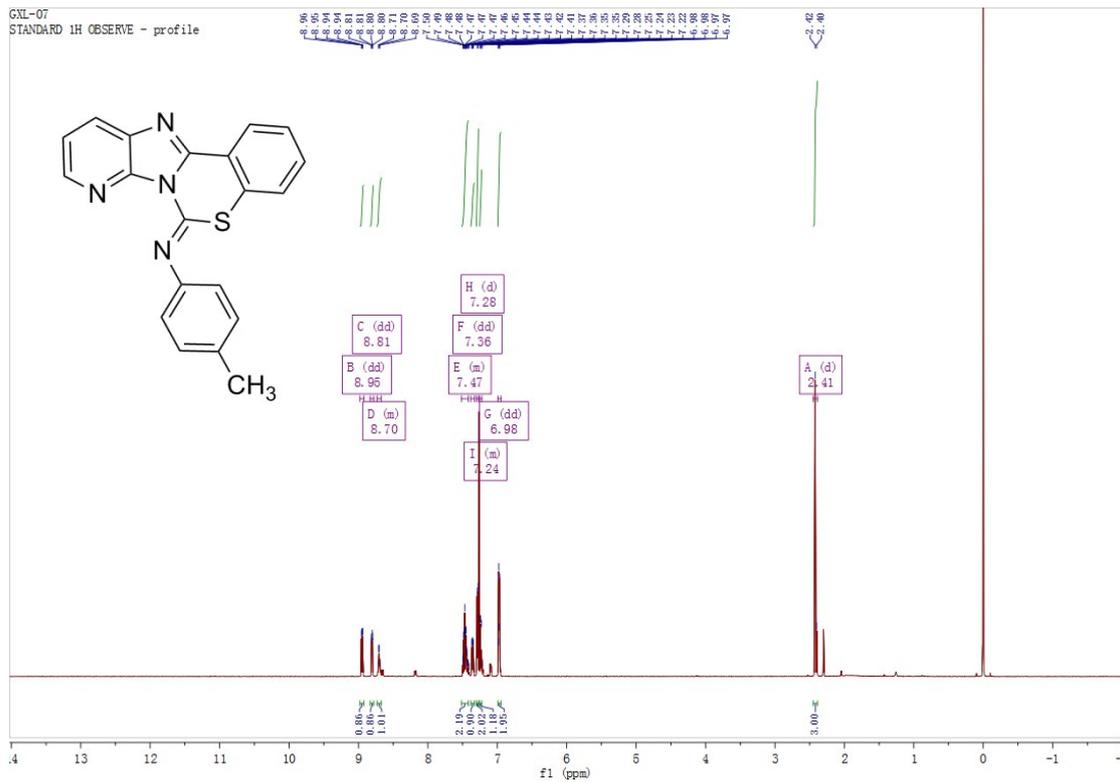


GXL-01
STANDARD 1H OBSERVE - profile

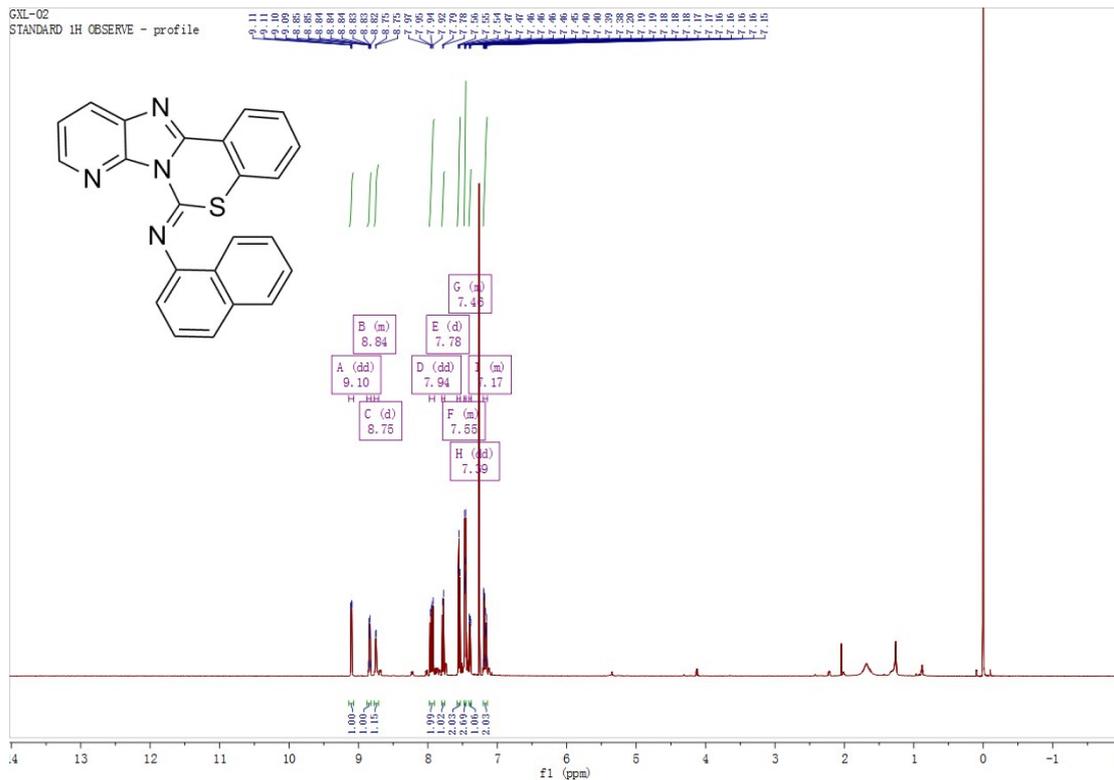


GXL-01-C
STANDARD 1H OBSERVE - profile





GXL-02
STANDARD 1H OBSERVE - profile



GXL-02-C
STANDARD 1H OBSERVE - profile

