Electronic Supplementary Material (ESI) for RSC Advances. This journal is © The Royal Society of Chemistry 2018

Supporting Information

Remarkable Colorimetric Sensing Behavior of Pyrazole-based Chemosensor Towards Cu(II) Ion Detection: Synthesis, Characterization and Theoretical Investigations

Nagaraj Nayak,^a Kollur Shiva Prasad,^{*, a} Renjith Raveendran Pillai,^b Stevan Armaković,^c and Sanja J.

Armaković^d

^a Chemistry Group, Manipal Centre for Natural Sciences, Manipal Academy of Higher Education, Manipal, Karnataka – 576 104, India.

^b Department of Physics, TKM College of Arts and Science, Karicode, Kollam, Kerala, India.

^c University of Novi Sad, Faculty of Sciences, Department of Physics, Trg D. Obradovića 4, 21000 Novi Sad, Serbia.

^d University of Novi Sad, Faculty of Sciences, Department of Chemistry, Biochemistry and Environmental Protection, Trg D. Obradovića 3, 21000 Novi Sad, Serbia.

Contents

Figure S1: The FT-IR spectra of the HL and Cu(II) complexes, 1 and 2.

Figure S2: The ¹³C NMR spectrum of compound HL.

Figure S3: ESI-HRMS of complex 1.

Figure S4: ESI-HRMS of complex 2.

Table S1: Detection limit towards Cu²⁺ ion sensing by various chemosensors.

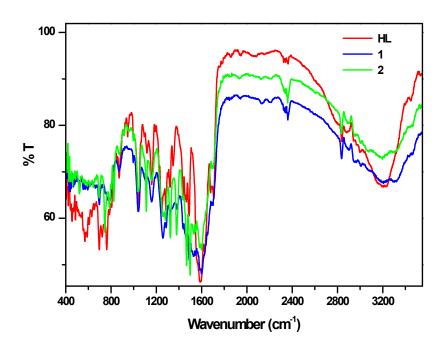


Figure S1: The FT-IR spectra of the HL and Cu(II) complexes, 1 and 2.

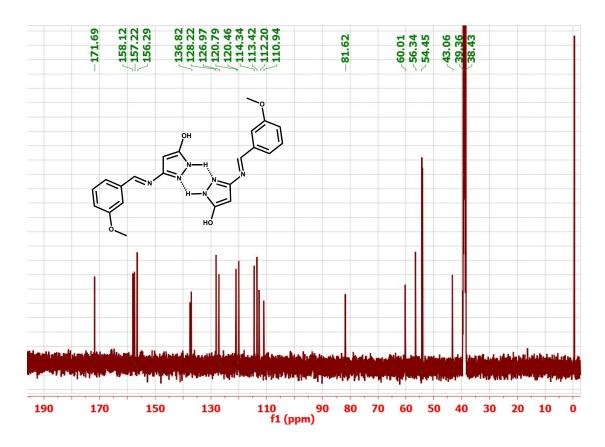


Figure S2: The ¹³C NMR spectrum of compound HL.

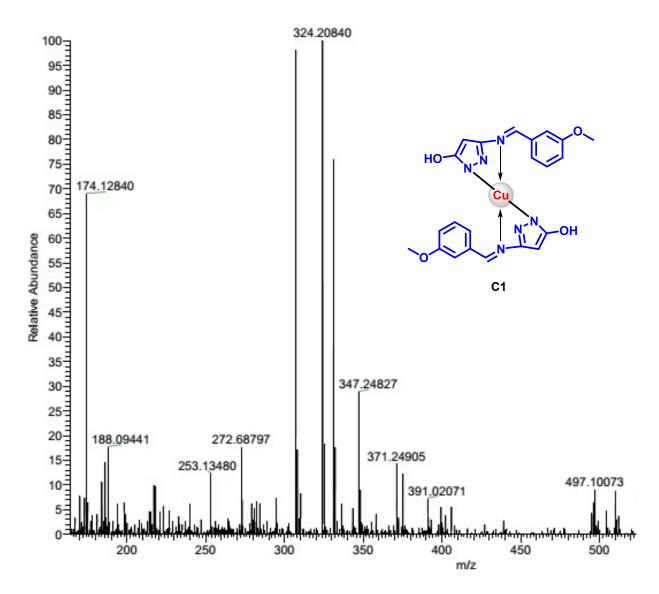


Figure **S3**: ESI-HRMS of complex **1**.

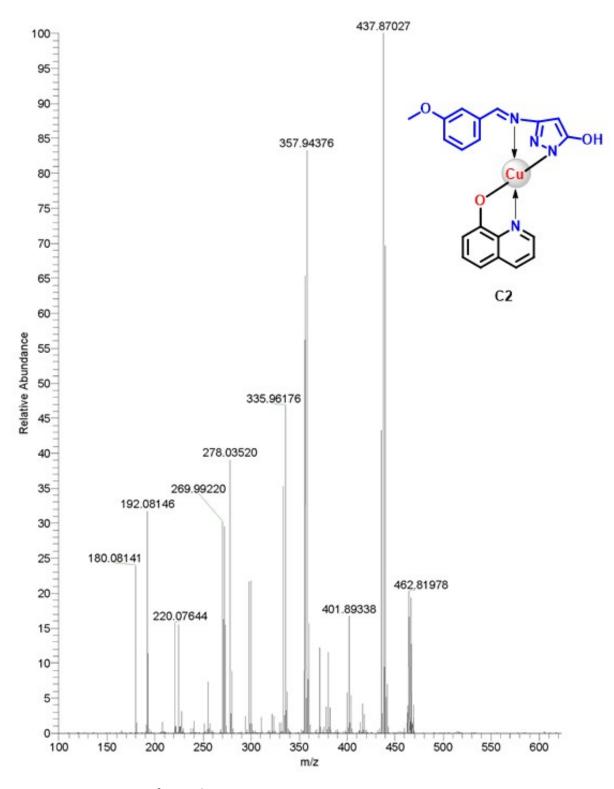


Figure **\$4**: ESI-HRMS of complex **2**.

Table S1: Detection limit towards Cu²⁺ ion sensing by various chemosensors.

| Sensor reported in the literature | Lowest detecting ability of Cu ²⁺ ions from solution |
|--|---|
| Colorimetric sensor doi: 10.1080/00958972.2017.1420787 | 1.66 μΜ |
| Colorimetric sensor doi: 10.1016/j.snb.2016.01.133 | 2.9 μΜ |
| Colorimetric sensor doi: 10.1016/j.tetlet.2013.09.126 | 5 μΜ |

| | 2.73 μΜ |
|---|---------|
| Fluorescence sensor doi: 10.1016/j.snb.2015.06.124 | |
| Fluorescence sensor doi:10.1039/C3DT51047F | 5.4 μΜ |
| Our system | 1.60 μΜ |