

Electronic Supplementary Information  
For  
**Micro-Supercapacitors Based on Oriented Coordination Polymer  
Thin Films for AC Line-Filtering**

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### Experimental section

*Synthesis of Co-BTA*: A solution of 80 mg (0.28 mmol) of benzene-1,2,4,5-tetrayltetraamine tetrahydrochloride purchased from Beijing Huaweiruike Chemical Co.,Ltd in 20 mL of water and 1120  $\mu$ L of triethylamine (1.4 mol L<sup>-1</sup>) was added to a solution of 26 mg (0.11 mmol) of cobalt acetate tetrahydrate (Co(CH<sub>3</sub>COO)<sub>2</sub>•4H<sub>2</sub>O) in 20 mL of water. The mixture was stirred in a bottle and kept at 60 °C for 3 hours, leading to the formation of a brown film at the gas-

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liquid interface and deep brown powders at the bottom of the reaction bottle. The resulting film product was rinsed with ethanol and deionized (DI) water, finally transferred onto certain substrates and dried under vacuum at 60 °C for further characterizations and applications. Meanwhile, the powder product was collected, centrifuged, filtered, and then washed with DI water, acetone for 15 min by ultrasonic bath (the water and acetone repeated for three times during washing), respectively. The solid was then dried under vacuum at 60 °C.

*Fabrication of Micro-supercapacitors Based on Co-BTA:* First, 150 nm Au interdigital electrode was evaporated on the silicon wafer. Then the brown film obtained at the air-liquid interface was transferred onto the Au patterned substrate by immersing the substrate into the solution. Then, the H<sub>2</sub>SO<sub>4</sub>-polyvinyl alcohol (H<sub>2</sub>SO<sub>4</sub>-PVA) gel electrolyte was cast on the Co-BTA-Au:SiO<sub>2</sub> substrate. Finally, the device was left to solidify overnight to form a solid-state Co-BTA-Au MSC.

*Material characterization :* Scanning electron microscopy images were taken by using a Jsm-7800F scanning electron microscope. X-ray diffraction patterns of out-of-plane and in-plane were obtained by using Smartlab (3KW) and Smartlab III, respectively. Reflex module implemented in Material studio 8.0 (Accelrys Inc.) was employed to simulate the Co-BTA XRD pattern. UV-visible absorption spectra was acquired from Shimadzu UV-1750 at room temperature. Phi 5000 VersaProbe was used to record the X-ray photoelectron spectroscopy images. Tensor 27 was used to obtain IR spectra. Cyclic voltammograms of product and ferrocene were gained by using CHI 660D electrochemical workstation with a scanning rate of 100 mV/s.

*Electrochemical measurements:* Cyclic voltammetry (CV) and electrochemical impedance spectroscopy (EIS) were performed by CHI 660D electrochemical workstation (Chenhua). The CV was tested with the scan rate ranging from 0.05 to 1000 V s<sup>-1</sup>. EIS was recorded in the

frequency range of 1 Hz to 100 kHz with a 5 mV ac amplitude. The H<sub>2</sub>SO<sub>4</sub>- PVA gel electrolyte was prepared by mixing 1 g H<sub>2</sub>SO<sub>4</sub> and 1 g PVA in 10 mL deionized water and heated up to 60 °C for 6 h with a stirring rate of 800 rpm. The capacitance values of the device were calculated from the CV data by equation (1):

$$C_{device} = \frac{1}{v \times (V_f - V_i)} \int_{V_i}^{V_f} I(V) dV$$

(1)

the  $C_{device}$  is used as the capacitance contribution from Co-BTA film electrodes;  $v$  is the scan rate (in V s<sup>-1</sup>);  $V_f$  and  $V_i$  are the intergration potential limits of the voltammetric curve and  $I(V)$

$$\int_{V_i}^{V_f} I(V) dV$$

is the voltammetric discharge current (in amperes); is the integrated area from the CV curves. The total surface area of the device including the spacing between the electrodes was 0.6 cm<sup>2</sup>, and the thickness of the active materials are 60 nm. This is used to calculate the power and energy density.

The volumetric ( $C_V$  in F cm<sup>-3</sup>) capacitance was calculated by equation (2):

$$C_V = \frac{C_{device}}{V}$$

(2)

$V$  is the volume of the device, respectively.

The volumetric ( $E_V$  in Wh cm<sup>-3</sup>) energy densities was calculated from the equation (3):

$$E_V = \frac{1}{2} \times C_V \times \frac{(\Delta V)^2}{3600}$$

(3)

the  $\Delta V$  is the potential range (in volts).

The volumetric ( $P_V$  in W cm<sup>-3</sup>) power densities was calculated by equation (4):

$$P_V = \frac{E_V}{\Delta t} \times 3600 \quad (4)$$

$\Delta t$  is the discharge time in seconds.

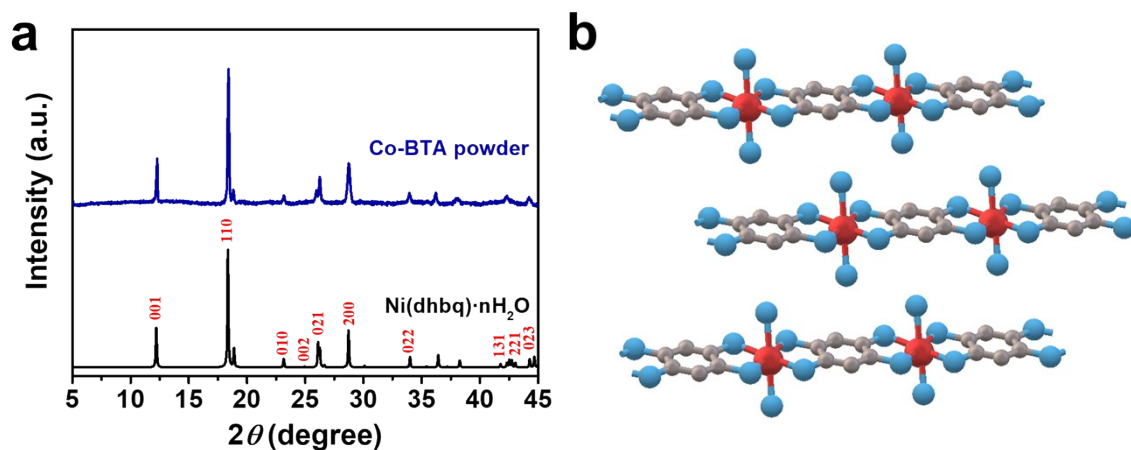
The EIS was measured to investigate the AC-line filtering performance of the micro-supercapacitors based on Co-BTA film. The specific capacitance of the microdevice can be described by  $C'(f)$  and  $C''(f)$  on the basis of the equation (5) and (6):

$$C'(f) = \frac{-Z''(f)}{2\pi fS|Z(f)|^2}$$

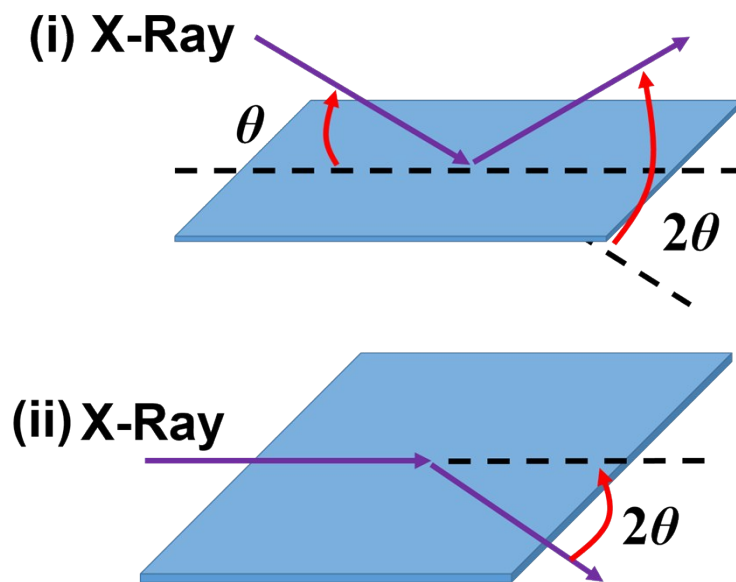
(5)

$$C''(f) = \frac{Z'(f)}{2\pi fS|Z(f)|^2}$$

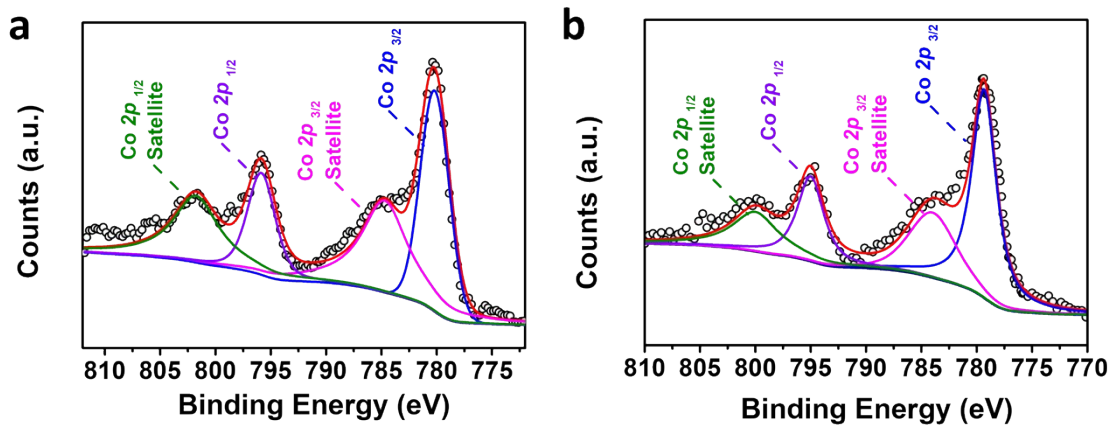
(6)



**Figure S1.** (a) PXRD profile of Co-BTA powder and simulated PXRD pattern of Ni(dhbq)·nH<sub>2</sub>O.<sup>1</sup> (b) Simulated crystal structure of Ni(dhbq)·nH<sub>2</sub>O.<sup>1</sup> Red, blue and grey balls correspond to nickel, oxygen and carbon atoms, respectively. Hydrogen atoms are deleted in the crystal structure for a simplified view.



**Figure S2.** The schemes of the orientation of the films for (i) out-of-plane XRD scan and (ii) in-plane XRD scan.



**Figure S3.** Co 2p core level spectra of (a) Co-BTA film and (b) Co-BTA powder. The Co 2p spectra of XPS could be fitted into two sets of doublets, among which the peak at ~780 eV together with the satellite peak at ~785 eV corresponds to Co 2p<sub>3/2</sub> and the peak at ~795 eV together with the satellite peak at ~801 eV belongs to the Co 2p<sub>1/2</sub>, respectively.

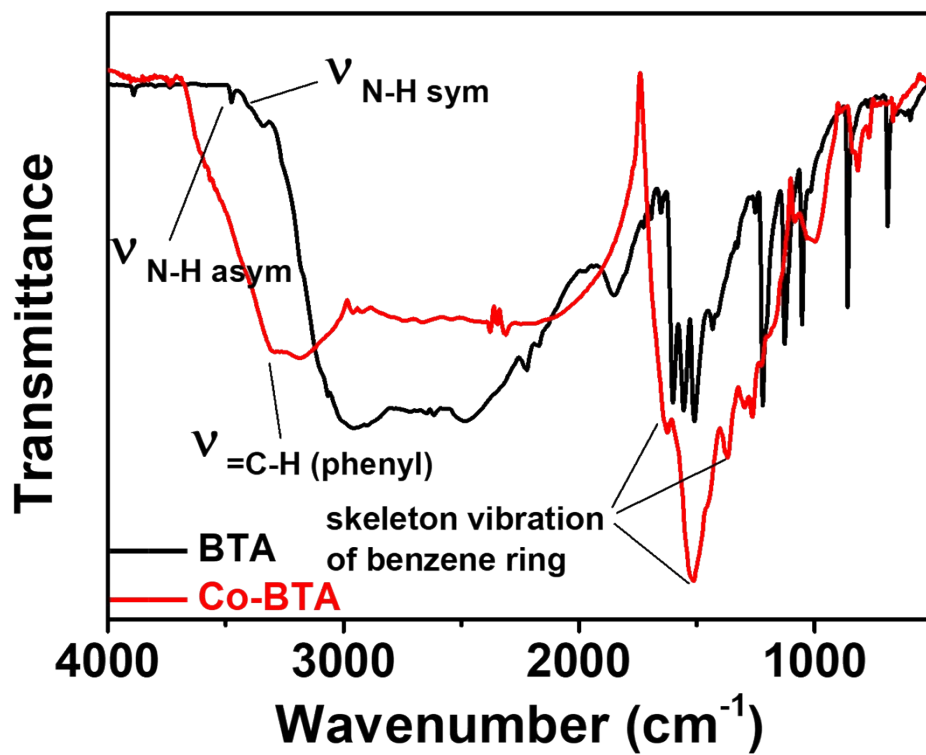


Figure S4. IR spectra of the Co-BTA film and BTA ligand.



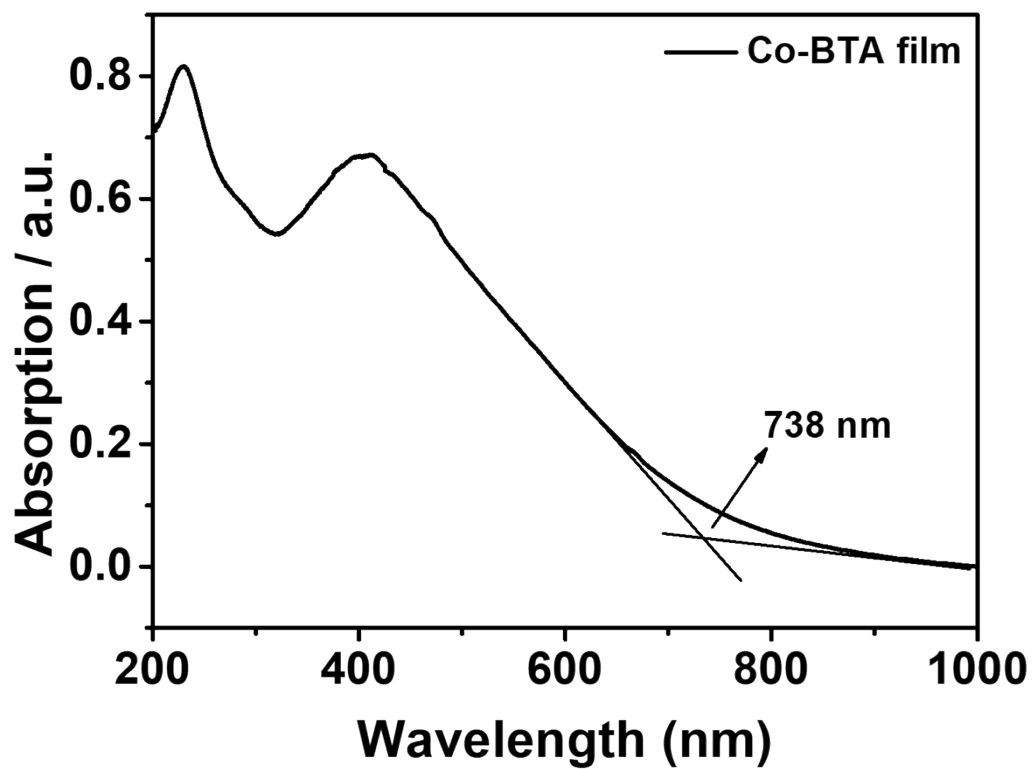
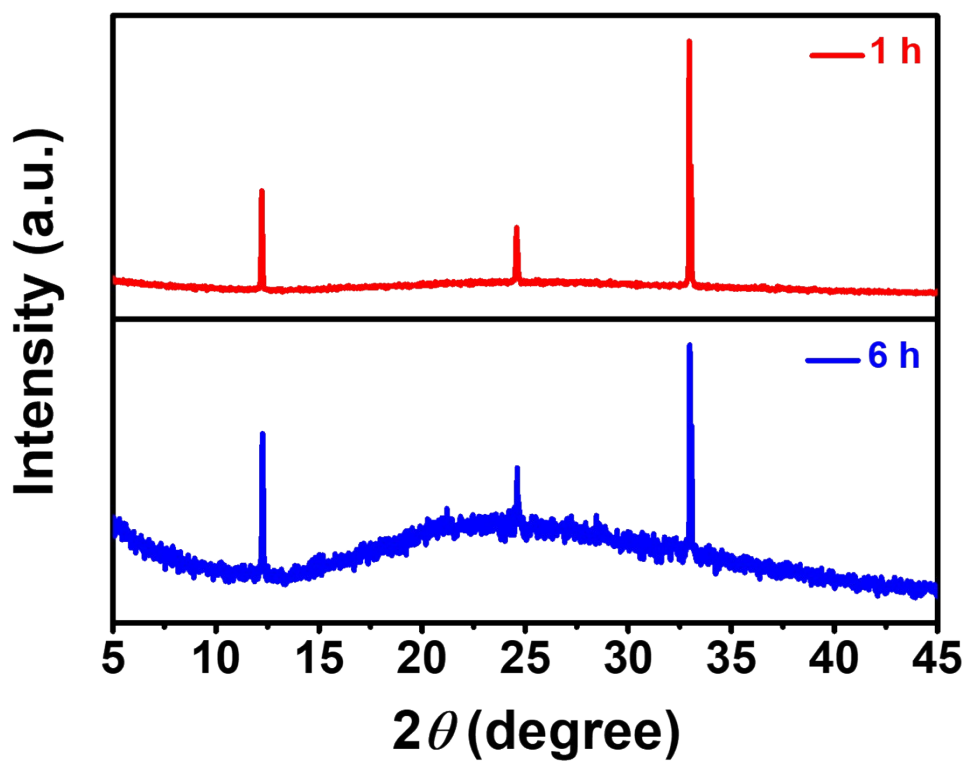
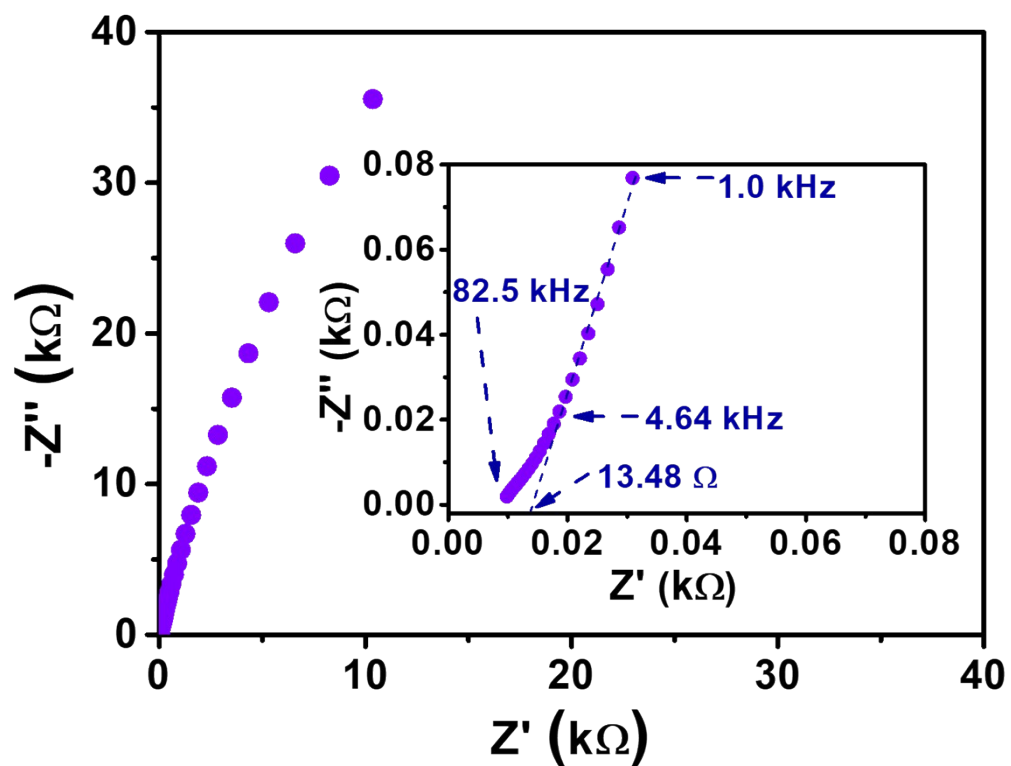


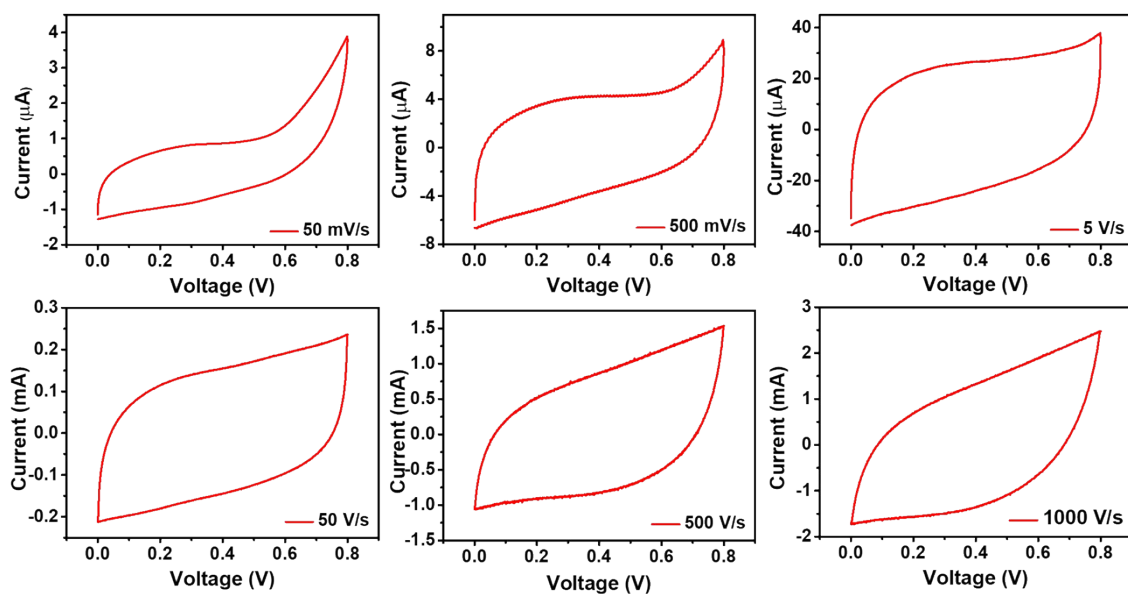
Figure S5. The normalized UV-Vis spectra of the Co-BTA film.



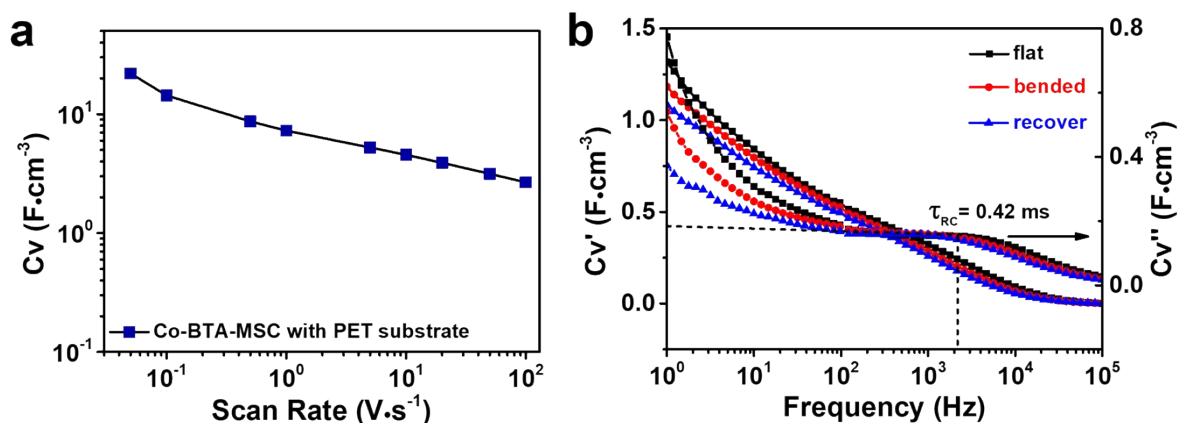
**Figure S6.** Out-of-plane PXRD profiles of Co-BTA film after immersed into the acid solution for 1 hour and 6 hour, respectively.



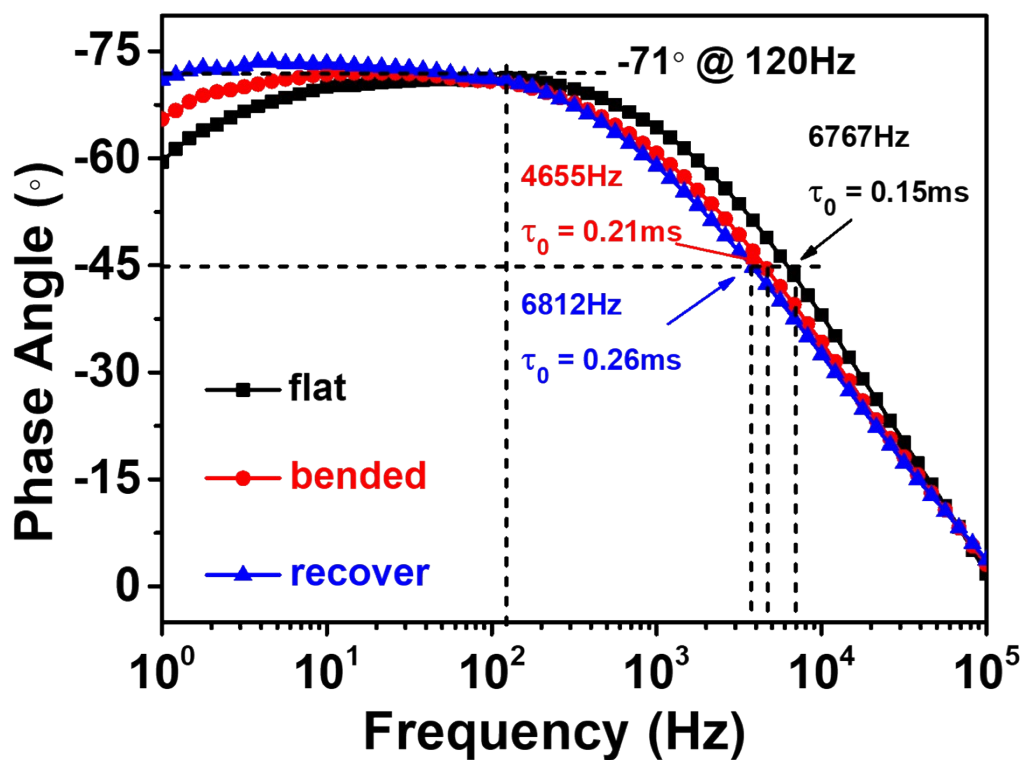
**Figure S7.** Complex plane plot of the impedance of the Co-BTA-based microdevices. Inset displays a magnification of the high-frequency region.



**Figure S8.** CV curves of Co-BTA-based MSC with a PET substrate in the  $\text{H}_2\text{SO}_4$ -PVA gel electrolyte at different scan rates.



**Figure S9.** Electrochemical performance of Co-BTA-based-MSC with flexible PET substrate. (a) CV evolution of the as-fabricated flexible MSC before bending test. (b) Plot of capacitance ( $Cv'$ =volumetric real capacitance and  $Cv''$ =imaginary capacitance) versus the frequency of flexible MSC in different states (flat, bended, and recover), bending radius = 1 cm.



**Figure S10.** Impedance phase angle on the frequency of Co-BTA-based flexible microdevices in different states (flat, bended and recover), bending radius = 1 cm.

**Table S1.** Element quantities analysis of C, O, N, Co of Co-BTA film and Co-BTA powder based on the XPS analysis.

Co-BTA	Element	C	O	N	Co
Film	At%	56.06	23.64	15.80	4.50
Powder	At%	60.75	21.23	14.20	3.82

**Table S2.** The performance of the reported micro-supercapacitors based on different materials.

<b>Material</b>	<b>E<sup>[a]</sup></b>	<b>VW<sup>[b]</sup></b>	<b>C<sub>v</sub><sup>[c]</sup></b>	<b>E<sub>v</sub><sup>[d]</sup></b>	<b>P<sub>v</sub><sup>[e]</sup></b>	<b>angle<sup>[f]</sup></b>	<b>f<sub>0</sub><sup>[g]</sup></b>	<b>τ<sub>0</sub><sup>[h]</sup></b>	<b>τ<sub>RC</sub><sup>[i]</sup></b>	<b>Ref</b>
Co-BTA	PVA-H <sub>2</sub> SO <sub>4</sub>	0-0.8	23 at 50 mV s <sup>-1</sup>	1.6 at 50 mV s <sup>-1</sup>	1056 at 1000 V s <sup>-1</sup>	-78.6°	6812	0.15	0.32	<b>This Work</b>
Onion-like Carbon	TEABF <sub>4</sub> -PC	0-3.0	1.35 at 1 V s <sup>-1</sup>	1.6 at 1 V s <sup>-1</sup>	250 at 200 V s <sup>-1</sup>	-	<100	<10	26	<b>2</b>
Carbon Nanotubes/ Graphene	1M Na <sub>2</sub> SO <sub>4</sub>	0-1.0	1.1 at 1.0 A cm <sup>-3</sup>	0.16	115	-81.5°	1343	0.82	0.195	<b>3</b>
PiCBA	PVA-H <sub>2</sub> SO <sub>4</sub>	0-1.0	34.1 at 50 mV s <sup>-1</sup>	4.7 at 50 mV s <sup>-1</sup>	1323 at 1000 V s <sup>-1</sup>	-73°	3620	0.27	0.83	<b>4</b>
inkjet-printed EC	1M Et <sub>4</sub> NBF <sub>4</sub>	0-2.5	-	-	-	-	< 5	< 200	-	<b>5</b>
UPSCs-25	PVA-H <sub>2</sub> SO <sub>4</sub>	0-1.0	348	12 at 10 mV s <sup>-1</sup>	4386 at 2000 V s <sup>-1</sup>	-72°	1000	1	0.47	<b>6</b>
vertically oriented graphene	25% KOH	-	3	-	-	-82°	15000	0.067	0.2	<b>7</b>
Sulfur-Doped Graphene	PVA-H <sub>2</sub> SO <sub>4</sub>	0-1.0	582	3.1 at 10 mV s <sup>-1</sup>	1191 at 6000 V s <sup>-1</sup>	-73°	3836	0.26	-	<b>8</b>
Carbon Nanotubes/ Reduced Graphene Oxide	3M KCl	0-1.0	3.1 at 10 mV s <sup>-1</sup>	0.68 at 1 V s <sup>-1</sup>	77 at 50 V s <sup>-1</sup>	-	208.6	4.8	-	<b>9</b>
EG/PH1000	PVA-H <sub>2</sub> SO <sub>4</sub>	0-1.0	27			-84°	4200	0.24	1.35	<b>10</b>
MPG-MSCs	PVA-H <sub>2</sub> SO <sub>4</sub>	0-1.0	17.9	2.5 at 0.01 V s <sup>-1</sup>	495 at 1000 V s <sup>-1</sup>	-	3579	0.28	-	<b>11</b>

a) electrolyte; b) voltage window (V); c) specific volumetric capacitance (F cm<sup>-3</sup>); d) energy density (mWh cm<sup>-3</sup>); e) power density (W·cm<sup>-3</sup>); f) phase angle @120 Hz; g) characteristic



frequency (Hz); <sup>h</sup>) relaxation time (ms); <sup>i</sup>) resistance-capacitance time (ms).

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