Supporting Information for

Introducing catalytic gasification into chemical activation for the conversion of natural coal into hierarchically porous carbons with broadened pore size for enhanced supercapacitive utilization

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Proximate analysis (wt-%)				Ultimate analysis (wt-%, daf)				
M _{ad}	A _{ad}	FC_{ad}	V_{daf}	С	Н	O^*	N	S
11.79	3.68	56.64	32.70	73.52	6.55	18.51	0.91	0.51

Table S1. Proximate and ultimate analyses of Zhundong coal.

^{ad} Air-dried basis, ^d dry basis, ^{daf} dry and ash-free basis, * by difference.

Table S2. Ash composition analyses of Zhundong coal.

Ash composition analysis (wt-%)										
SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	TiO ₂	CaO	MgO	K ₂ O	Na ₂ O	MnO ₂	SO ₃	P_2O_5
13.33	10.71	6.19	0.47	37.75	9.98	0.62	9.78	0.16	6.52	0.19



Fig. S1. HRTEM images of Zhundong raw coal (a); MPC-500 (b) and HPC (c). All the HRTEM images demonstrate the mainly amorphous carbon nature.



Fig. S2 (a) XRD patterns of prepared porous carbons; (b) Raman spectra of prepared porous carbons. Both XRD and Raman spectra suggest that the obtained porous carbons are mainly amorphous which is in agreement with the HRTEM results (Fig. S1). However, the differences between prepared porous carbons reveal that MPC-500 prepared under lower temperature (500) shows a relatively higher graphitization degree (higher 002 peak in XRD and G-to-D band ratio in Raman) that those of other samples. This is because high temperature treatments destroy the graphite-like crystal structure of raw coal and lead to the resulting MPC-950, HPC and MF-PC with lower graphitization degree



Fig. S3 (a) CV curves of MPC-950 at various scan rates; (b) Galvanostatic chargedischarge curves of MPC-950 under different charge-discharge current densities.



Fig. S4 Comparison of MF-PC and MPC-950 in three electrode system using 6 M KOH as electrolyte. Without the mineral component in the coal structure, MF-PC mainly have micropores. Owing to the larger BET surface area, MF-PC perform much better performance compared with MPC-950. However, it provides both poorer gravimetric capacity and rate capacity compared with HPC.

		_	Rate perfo			
Sample	Electrolyte	System voltage	Capacitance	Current	Ref.	
			$(F g^{-1})$	$(A g^{-1})$		
Nitrogen-doped	2 M KOU	0~1.8 vs. SCE	260	1	S 1	
Carbon Nanosheets	2 WI KOII		177	20		
Nitrogen-doped	2 М КОН	0~1 vs. SCE	255	1	S2	
Porous Carbon			192	10	52	
Hierarchically		-1~0 vs.	281	0.2		
Biomass Carbons	1 M KOH	SCE	125	4	83	
Nitrogen-doped	<u>ALKON</u>	-1.2~0.2 vs.	293	1	S4	
Carbon	6М КОН	SCE	157	30		
Nitrogen-doped	6 M KOH	-1~0 vs. SCE	202	1	S5	
porous nanofibers			~170	30		
Nitrogen-doped	<u>AUKOU</u>	-0.9~0 vs.	227	0.2	S6	
Ordered Mesoporous	6M KOH	SCE	227	0.2		
Carbon	$I M H_2 SO_4$	0~0.8 vs. SCE	262	0.2		
Human hair-derived carbon flakes	6М КОН	-1~0 vs. SCE	128	80	S7	
Shape-controlled carbon nanosheets	1 M H ₂ SO ₄	0~1 vs. Ag/AgCl	145	30	S8	
Two-dimensional		-1~0 vs	300	0.5		
Porous Carbon Nanosheets	6М КОН	SCE	246	100	S9	
Yeast Cells Derived Carbon 6M KOH		-1.2~0.2 vs. Ag/AgCl	175	100	S10	
Hierarchical Porous	0.000	-1~0 vs.	290	1	S11	
Coal Tar Pitch		SCE	250	10		
НРС	6M KOH	-1~0 vs. SCE	308 202	1 100	Our work	

Table S3. Typical results of carbon materials for three-electrode test in literatures

 with aqueous electrolyte systems

Table S4. Typical results of carbon materials as cathode for lithium ion capacitors

 with organic electrolyte systems

Sample	Electrolyte	System voltage Rate performance		Cycling Stability	Ref.
Nitrogen- Doped Porous Carbon	1 M LiPF ₆	2.5-4.5 V vs. Li/Li+	$\begin{array}{c} 117 \ F \ g^{-1}(81.5 \ mAh \ g^{-1} \\ \) \ (0.1 \ A \ g^{-1}) \\ 60.8 \ F \ g^{-1}(42.3 \ mAh \\ g^{-1}) \ (30 \ A \ g^{-1}) \end{array}$	86% after 2000 cycles (5 A g ⁻¹)	S12
Activated Carbons	1 M LiPF ₆	3.0-4.6 V vs. Li/Li ⁺	$159 \text{ F g}^{-1}(110.6 \text{ mAh} \text{g}^{-1})$ (0.1 A g^{-1})	~82% after 1000 cycles (0.1 A g ⁻¹)	S13
3D Carbon Nanofibers	1 M LiPF ₆	2.0-4.5 V vs. Li/Li ⁺	162 F g ⁻¹ (113 mAh g ⁻¹) (0.1 A g ⁻¹) 90.6 F g ⁻¹ (63 mAh g ⁻¹) (10 A g ⁻¹)	87% after 5000 cycles (2 A g ⁻¹)	S14
LTO/ Graphene hybrid	1 M LiPF ₆	1-4 V vs. Li/Li ⁺	$\begin{array}{c} 69 \ \mathrm{F} \ \mathrm{g}^{-1}(178 \ \mathrm{mAh} \ \mathrm{g}^{-1}) \\ (0.25 \ \mathrm{A} \ \mathrm{g}^{-1}) \\ 58 \ \mathrm{F} \ \mathrm{g}^{-1}(120 \ \mathrm{mAh} \ \mathrm{g}^{-1}) \\ (10 \ \mathrm{A} \ \mathrm{g}^{-1}) \end{array}$	95% after 1000 cycles (5 A g ⁻¹)	S15
НРС	1 M LiPF ₆	2.0~4.5 V vs. Li/Li ⁺	$\begin{array}{c} 183 \ \mathrm{F} \ \mathrm{g}^{-1}(127 \ \mathrm{mAh} \ \mathrm{g}^{-1}) \\ (0.25 \ \mathrm{A} \ \mathrm{g}^{-1}) \\ 112 \ \mathrm{F} \ \mathrm{g}^{-1}(78 \ \mathrm{mAh} \ \mathrm{g}^{-1}) \\ (10 \ \mathrm{A} \ \mathrm{g}^{-1}) \end{array}$	88% after 5000 cycles (2 A g ⁻¹)	Our work

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