

## Supplementary information

### **A Novel Textile-like Carbon Wrapping for High-performance Silicon Anodes in Lithium-ion Batteries**

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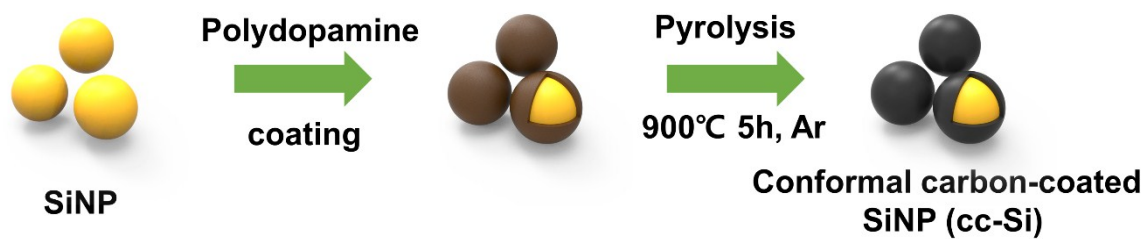
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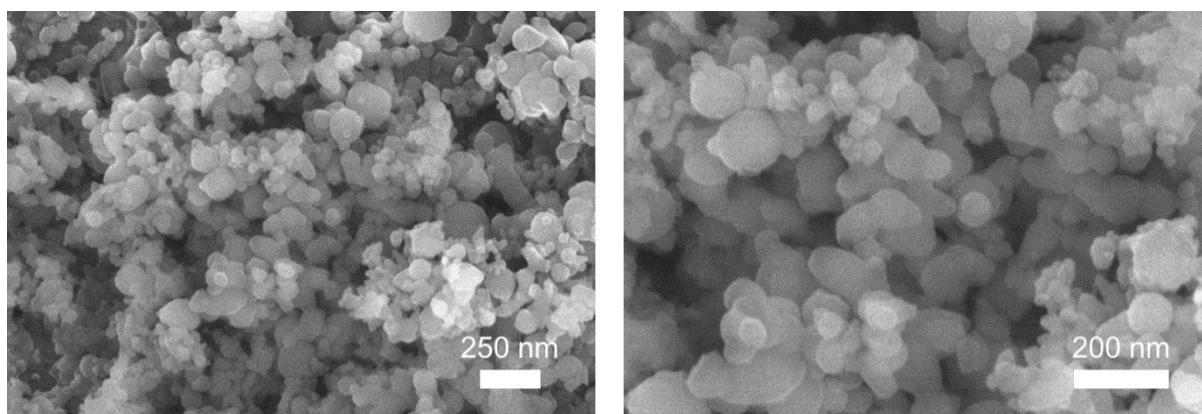
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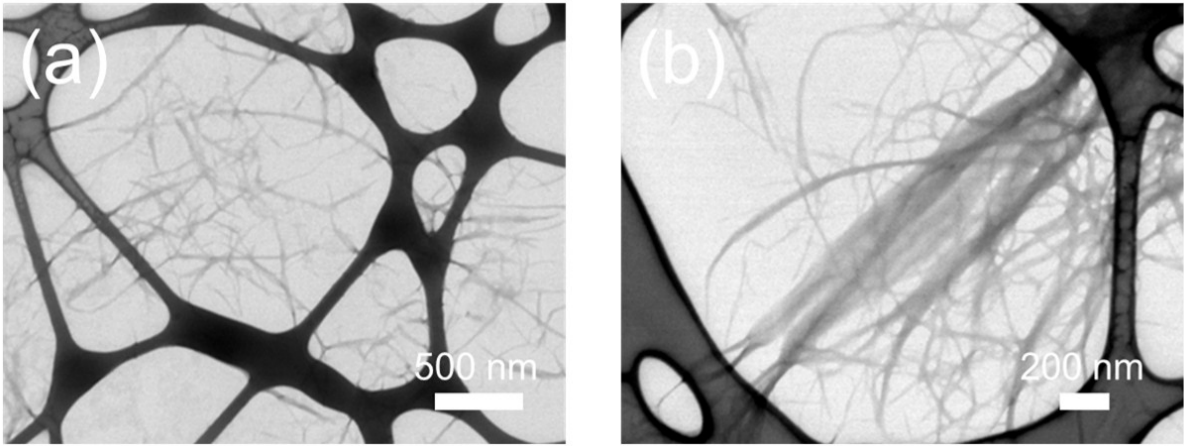
E-mail: german@mmk.su.se (G. Salazar-Alvarez) Tel: +46-816-3942



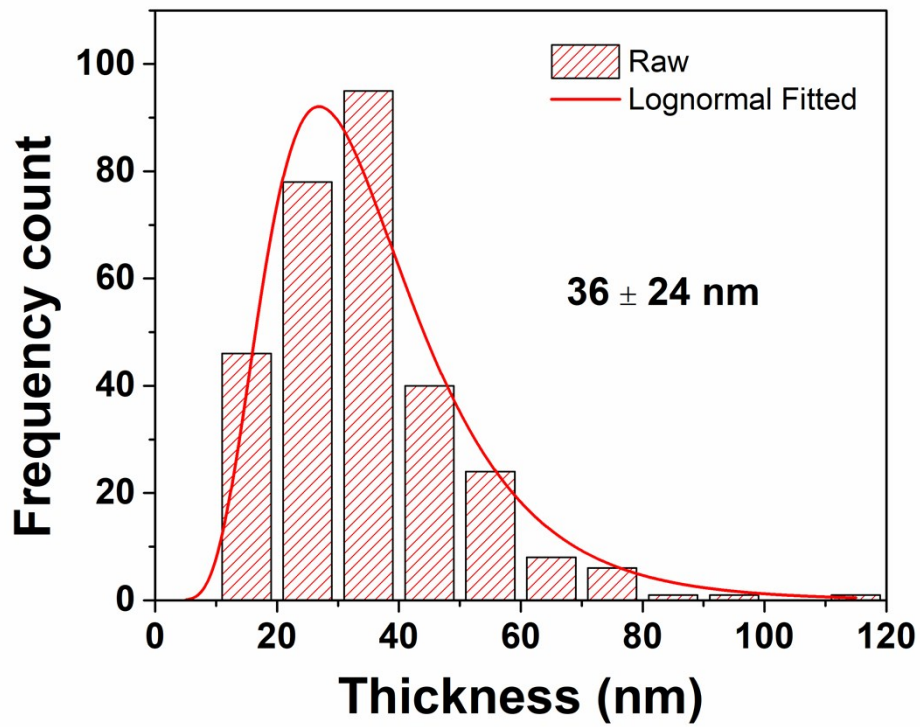
**Fig. S1** A schematic illustration of cc-Si synthesis process.



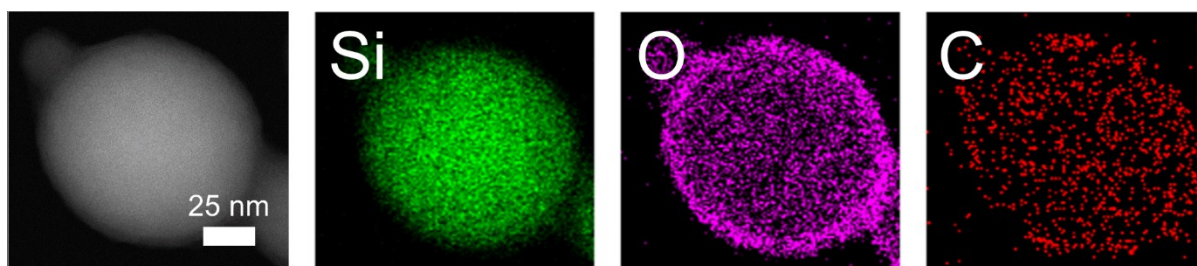
**Fig. S2** SEM images of Bare-Si powder.



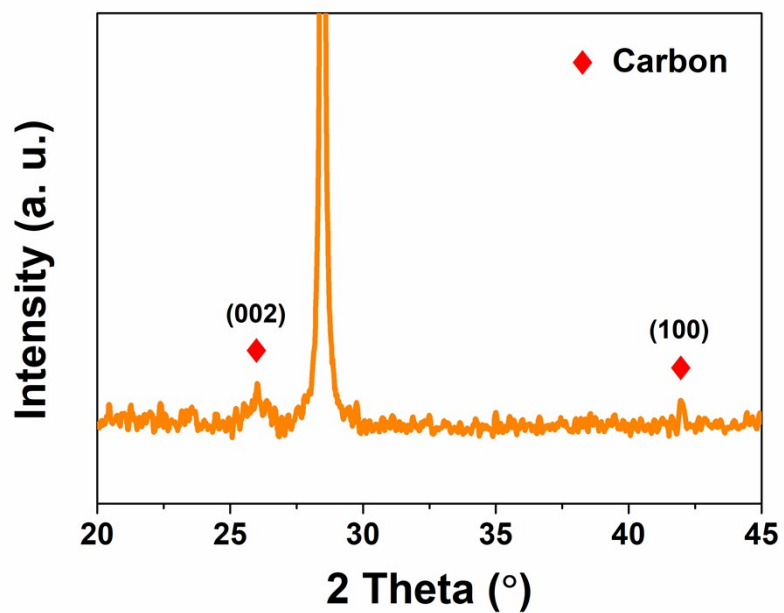
**Fig. S3** STEM images of (a) dispersed and (b) agglomerated enzymatically hydrolyzed cellulose nanofibers.



**Fig. S4** The thickness of enzymatically hydrolyzed cellulose nanofibers.



**Fig. S5** EDS analysis images of cc-Si with the element mapping of silicon, oxygen and carbon.



**Fig. S6** Local XRD pattern of ctex-Si.

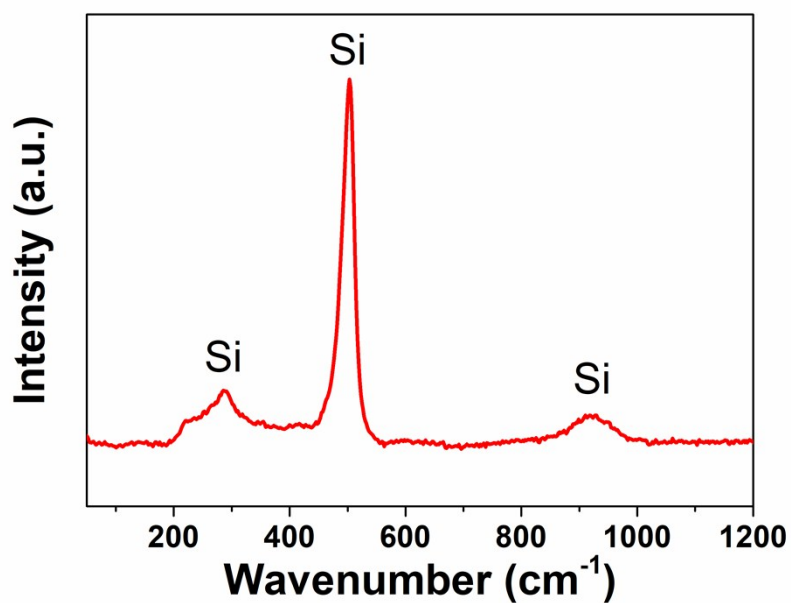
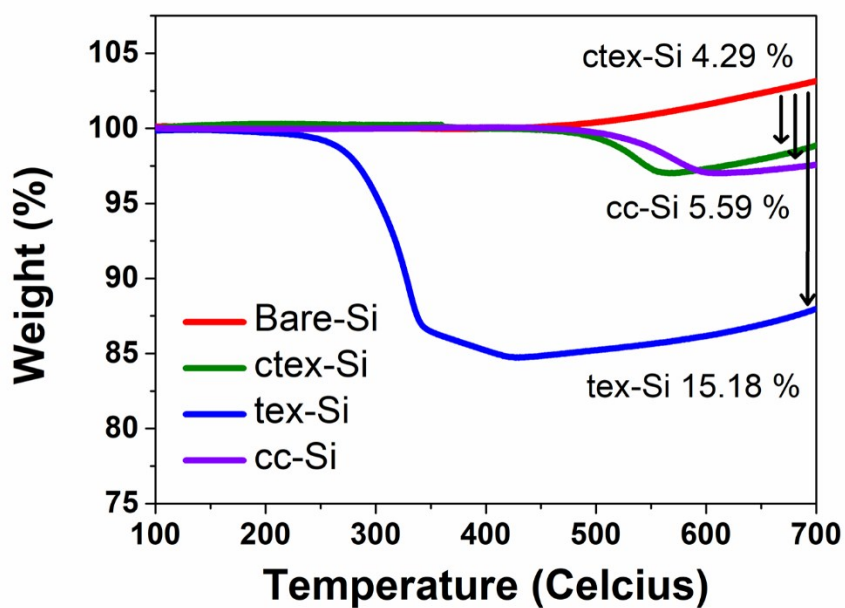
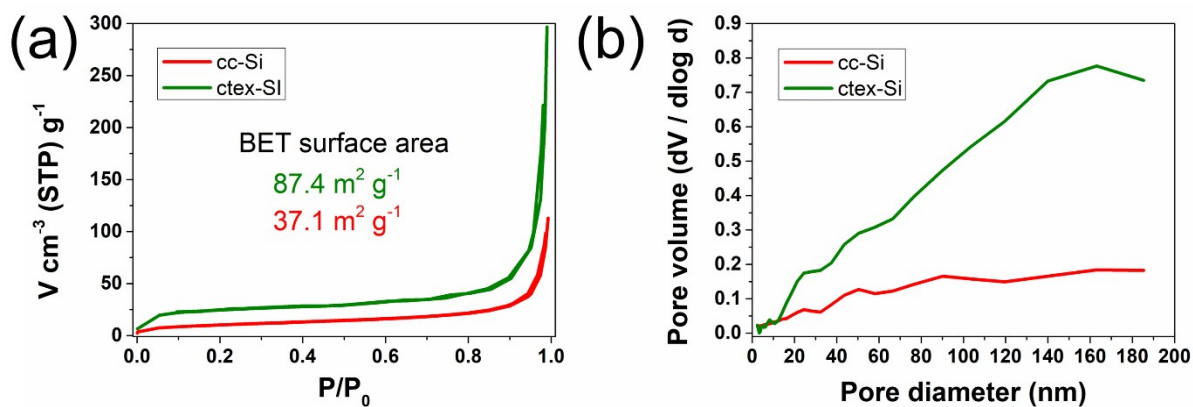


Fig. S7 Raman spectrum of ctex-Si.

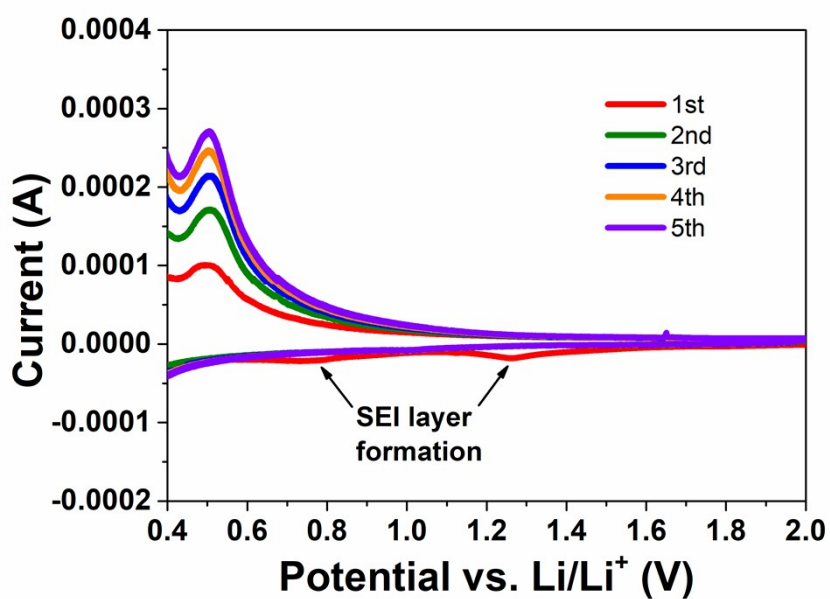


$$\text{C content (\%)} = \text{The weight of Bare-Si at } 700\text{ }^{\circ}\text{C} - \text{The weight of a sample at } 700\text{ }^{\circ}\text{C}$$

Fig. S8 TGA graph of Bare-Si, cc-S, tex-Si and ctex-Si. The equation for C content calculation.

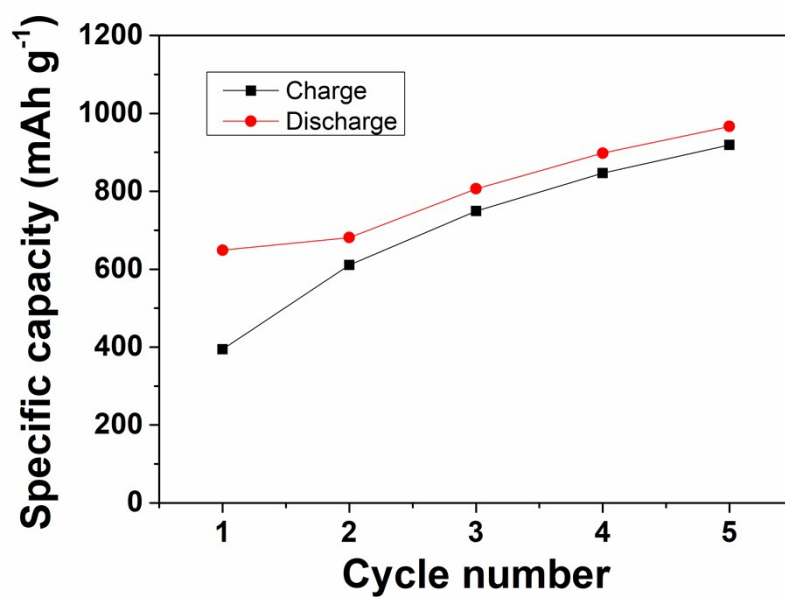


**Fig. S9** (a)  $\text{N}_2$  adsorption/desorption isotherms of cc-Si and ctex-Si. (b) BJH pore distribution diagrams of cc-Si and ctex-Si.

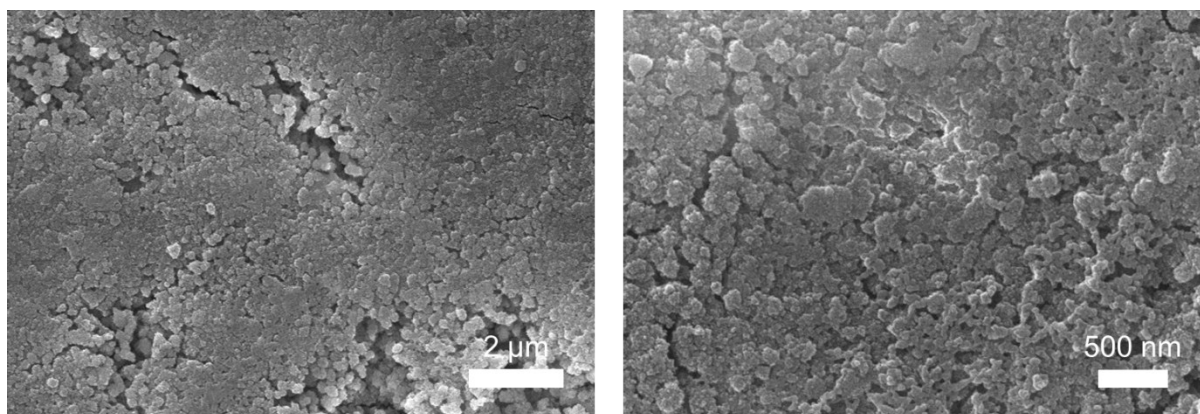


**Fig. S10** A magnified CV curve of ctex-Si.

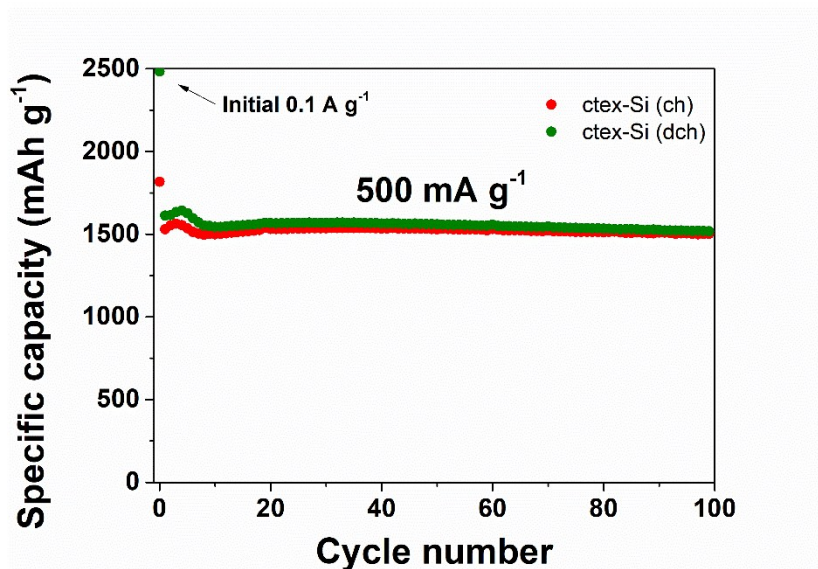




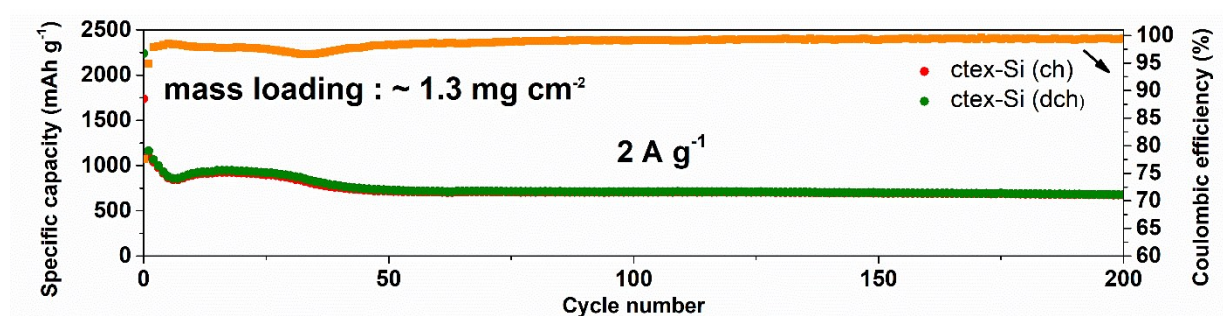
**Fig. S11** Specific capacity of ctex-Si during CV with a scan rate of  $0.1 \text{ mV s}^{-1}$ .



**Fig. S12** SEM images of a ctex-Si electrode after 500 cycles at  $2.0 \text{ A g}^{-1}$ .



**Fig. S13** Cycling performance of ctex-Si for 100 cycles at 500 mA g<sup>-1</sup>.



**Fig. S14** Cycling performance of ctex-Si with higher active material mass loading ( $\sim 1.3$  mg cm<sup>-2</sup>) for 200 cycles at 2 A g<sup>-1</sup>.

**Table S1.** EIS parameters for the fresh and 100 cycled cell of cc-Si and ctex-Si.

Sample	$R_S$ ( $\Omega$ )	$R_{SEI}$ ( $\Omega$ )	$R_{ct}$ ( $\Omega$ )	Reduced Chi-square
Fresh cc-Si	$4.26 \pm 0.06$	-	$6.37 \pm 0.57$	$1.01 \times 10^{-4}$
Fresh ctex-Si	$4.53 \pm 0.05$	-	$3.84 \pm 0.14$	$1.17 \times 10^{-4}$
Cycled cc-Si	$3.09 \pm 0.04$	$19.34 \pm 0.14$	$17.83 \pm 0.36$	$4.79 \times 10^{-5}$
Cycled ctex-Si	$3.89 \pm 0.05$	$11.54 \pm 0.14$	$4.85 \pm 0.31$	$1.23 \times 10^{-4}$



**Table S2.** Comparison table with other carbon-coated silicon, yolk-shell structured silicon and silicon-carbon composite.

Sample	Synthesis Method	Carbon source	Si (wt%) In active	Active: conductive additive: binder	Total carbon content (%)	Retention (%) (Cycle)	A g <sup>-1</sup>	Ref
<b>Textile-like carbon wrapped silicon</b>	<b>Pyrolysis</b>	<b>Cellulose nanofiber</b>	<b>95.71</b>	<b>70:15:15</b>	<b>18.0</b>	<b>94.5 (500)</b>	<b>2</b>	<b>Our work</b>
Yolk-shell Si@void@C	Sol-gel / HF etching	Polydopamine	71	65:20:15	38.8	74 (1000)	4.2	Ref S1
Yolk-shell silicon-mesoporous carbon	Sol-gel / HF etching	Phenolic resin	73.3 42.2	60:20:20	36.02 54.6	~ 29 (400) 63 (400)	0.42	Ref S2
Nonfilling carbon-coated porous silicon	Pyrolysis / HF etching	Resorcinol-formaldehyde resin	89	80:10:10	18.8	100 (1000)	1.05	Ref S3
a phenolic resin-based carbon interfacial coated silicon	Sol-gel / Pyrolysis	Resorcinol-formaldehyde resin	81.7	60:20:20	30.9	69 (200)	0.5	Ref S4
Nanosized core/shell silicon@carbon	Pyrolysis	Polyvinylidene fluoride	95.7	60:16:24	18.6	97 (30)	0.05	Ref S5
Cocoon-like silicon based composite (porous Si@void@N-doped C)	Magnesium-thermic reduction / CVD	Pyridine vapor	79.3	85:0:15	18.1	84.9 (300) 100 (300)	2 16	Ref S6
Silicon Nanoparticles Embedded in Micro-Carbon	Hydro-thermal / Pyrolysis	Sucrose	30.5	60:20:20	61.4	93.5 (500)	0.8	Ref S7
double-shelled-yolkstructured silicon (carbon@void@silicon)	CVD / HCl etching	Acetylene	60 (Carbon =24)	60:20:20	34.4	84.6 (1000)	5	Ref S8

## References

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