

Supplementary information for

Energy efficiency of ionic transport through proton conducting ceramics electrolytes for energy conversion applications

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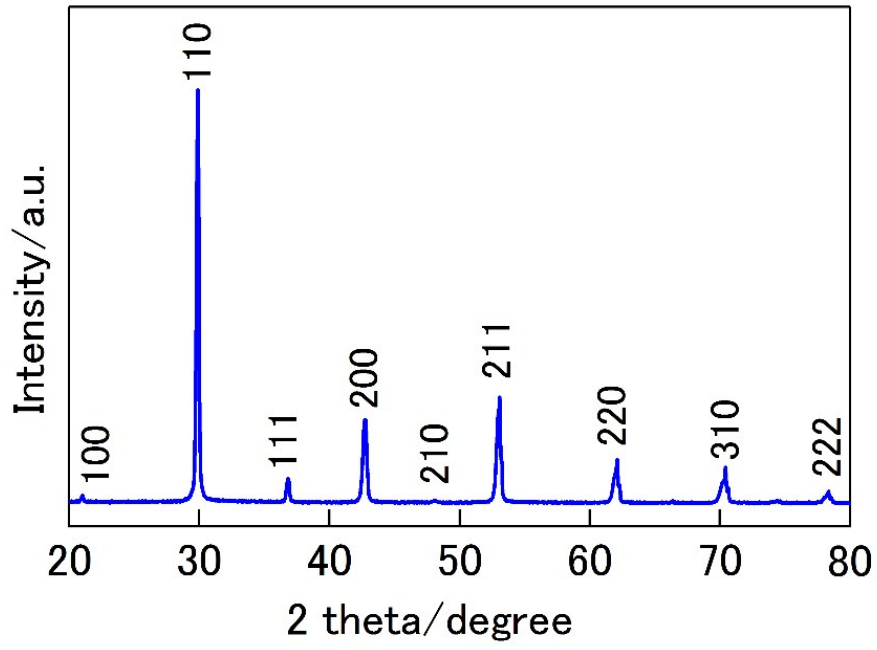


Figure S1. XRD patterns of BaZr_{0.8}Y_{0.2}O_{3-δ}.

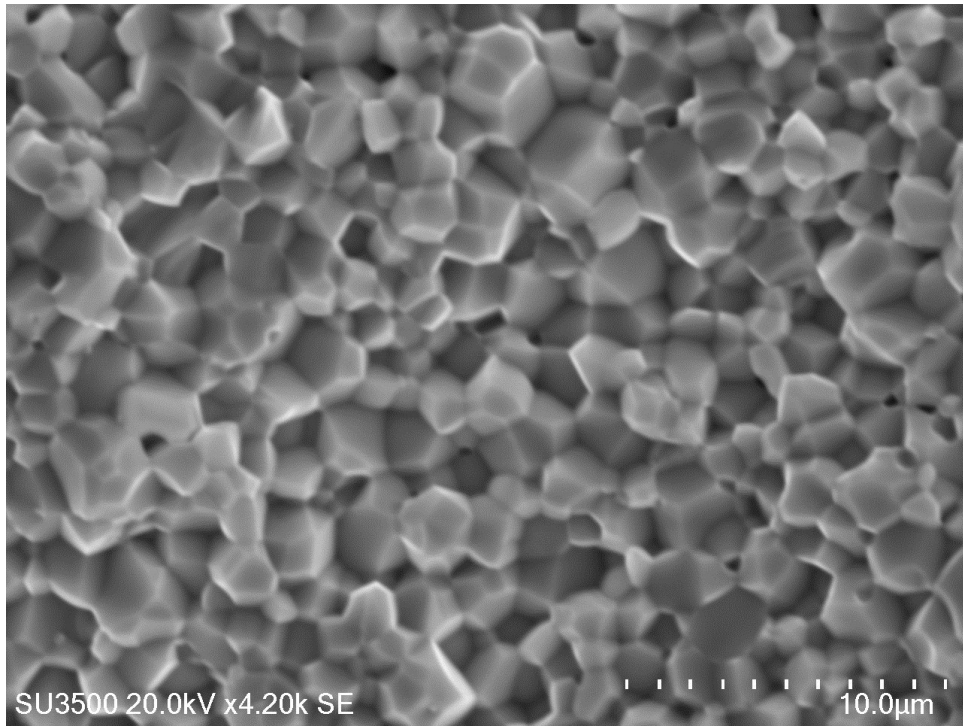


Figure S2. SEM image of the cross section of the dense $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ specimen.

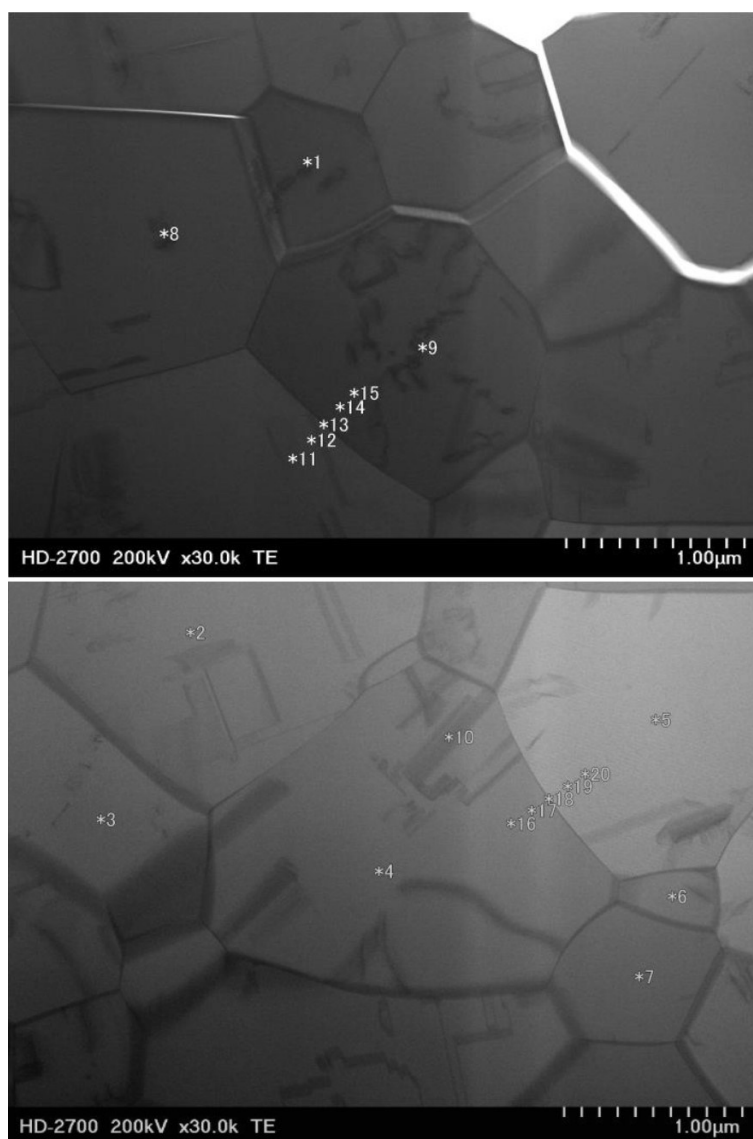


Figure S3. TEM images of the dense $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ specimen and the points of EDX analysis.

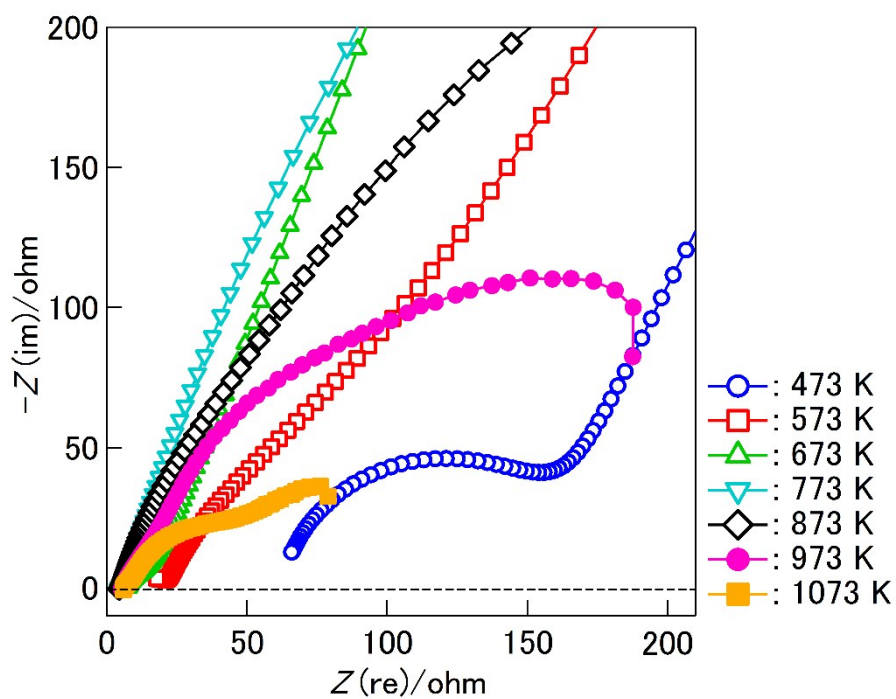


Figure S4. Typical impedance spectra of BaZr_{0.8}Y_{0.2}O_{3-δ} electrolyte measured in 1.9%H₂O-1%H₂-Ar at 473 to 1073 K.

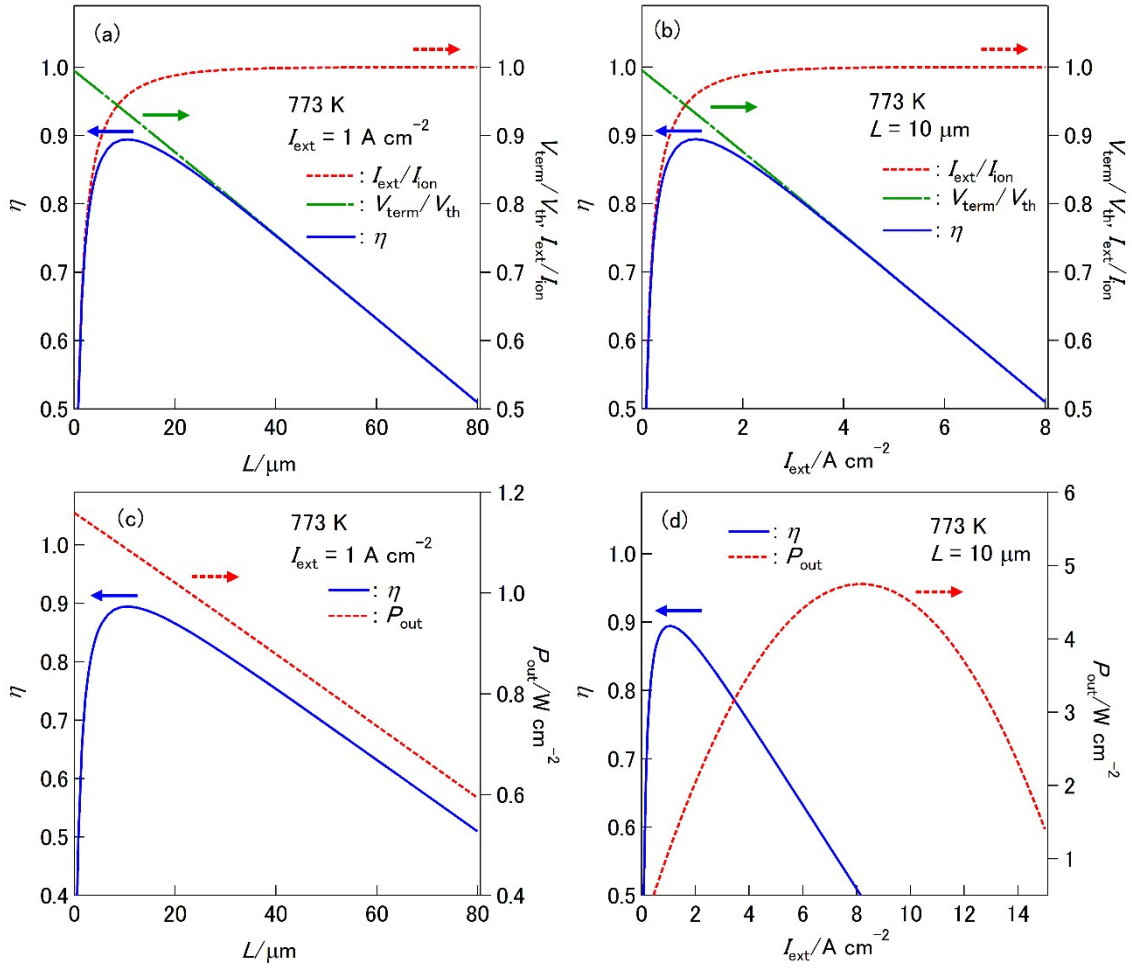


Figure S5. Energy efficiency, current ratio, and voltage ratio of the proton transport through a $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ electrolyte, (a) the electrolyte thickness dependence and (b) the external current dependence. Energy efficiency and power output of the proton transport through a $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ electrolyte, (c) the electrolyte thickness dependence and (d) the external current dependence. Anode: 4.2% H_2O - H_2 , cathode: 4.2% H_2O - O_2 , temperature: 773 K.

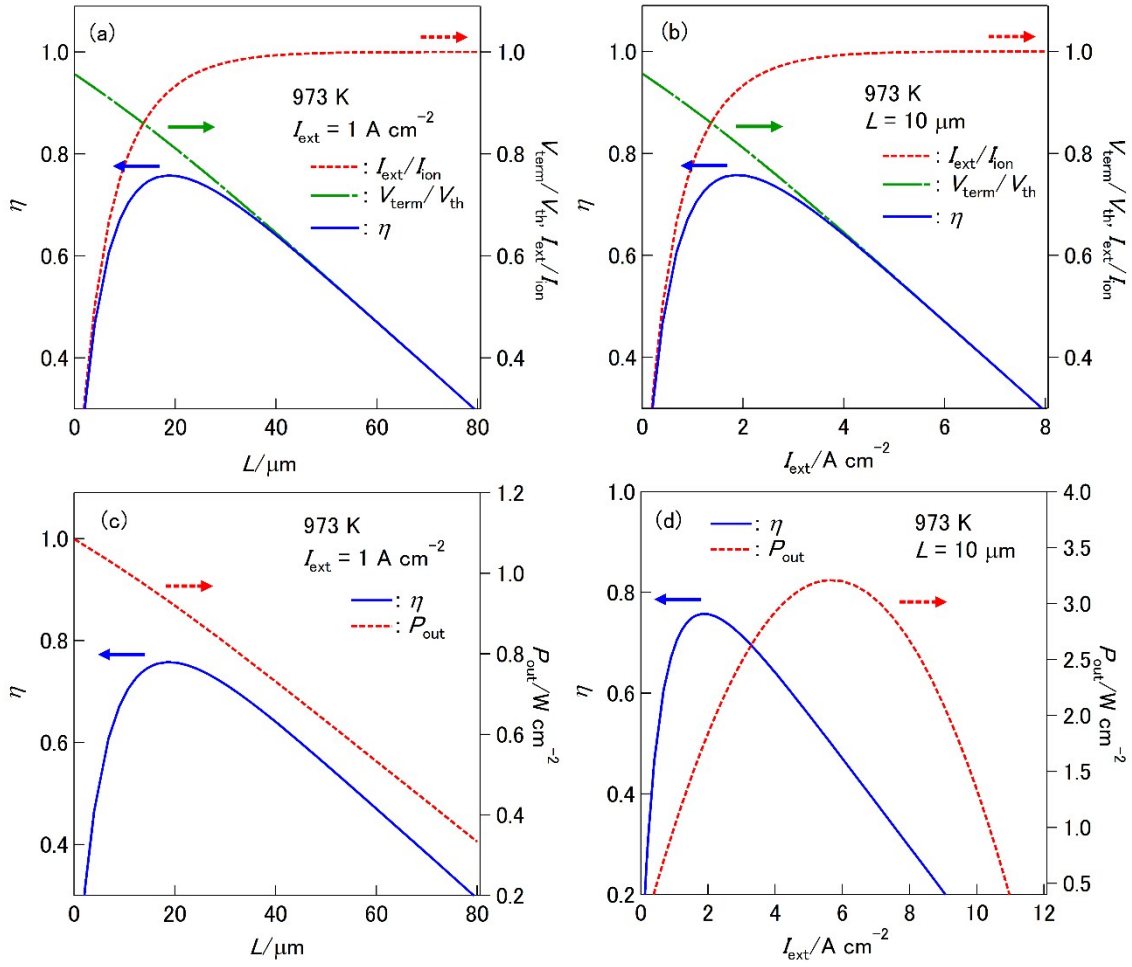


Figure S6. Energy efficiency, current ratio, and voltage ratio of the proton transport through a $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ electrolyte, (a) the electrolyte thickness dependence and (b) the external current dependence. Energy efficiency and power output of the proton transport through a $\text{BaZr}_{0.8}\text{Y}_{0.2}\text{O}_{3-\delta}$ electrolyte, (c) the electrolyte thickness dependence and (d) the external current dependence. Anode: $4.2\% \text{H}_2\text{O} - \text{H}_2$, cathode: $4.2\% \text{H}_2\text{O} - \text{O}_2$, temperature: 973 K.

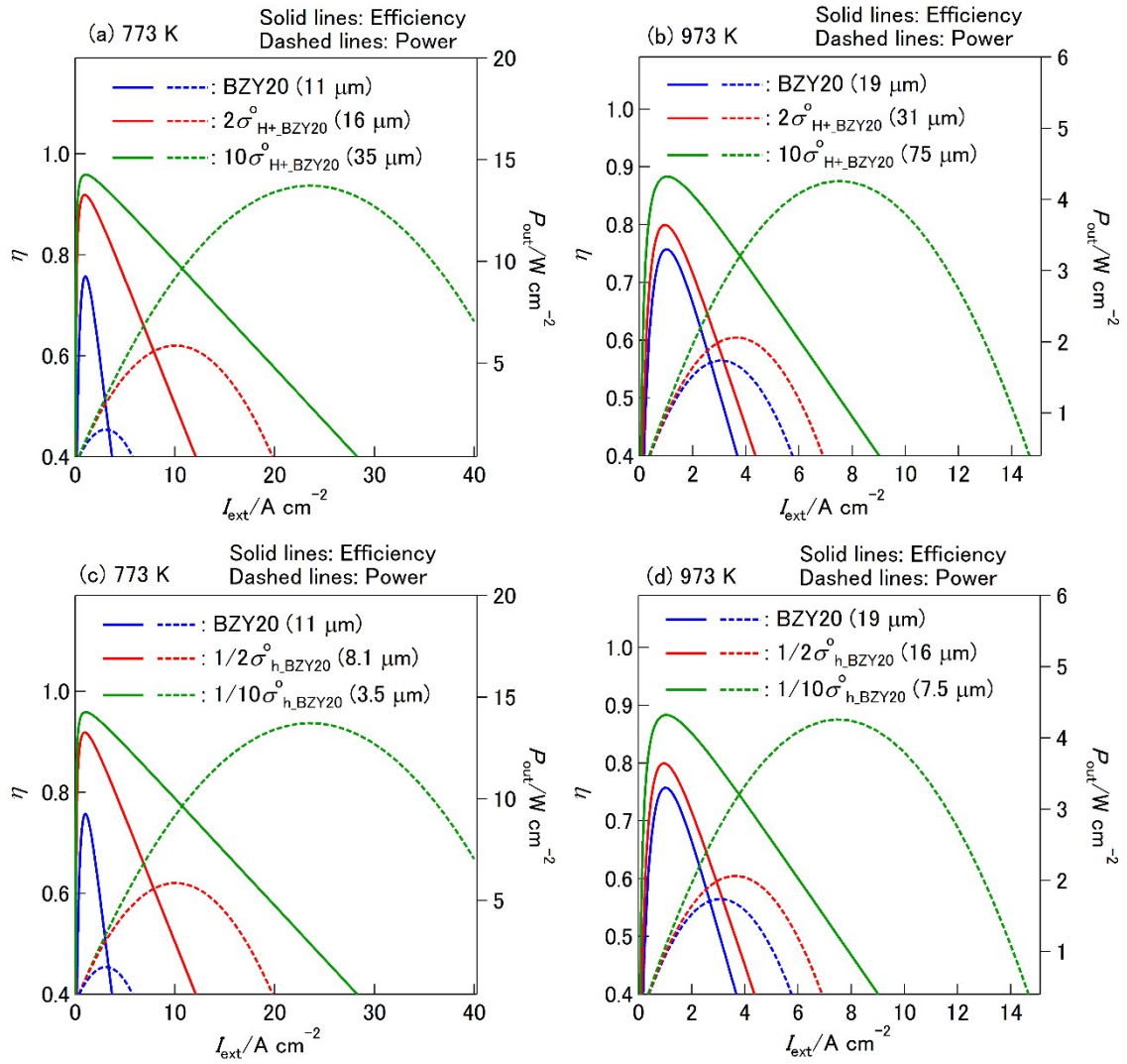


Figure S7. Influence of the partial conductivity variation on the energy efficiencies, the increase of $\sigma_{H^+}^o$ at (a) 773 K and (b) 973 K, and the decrease of σ_h^o at (c) 773 K and (d) 973 K.