Supplementary Information for

Bisubstituted-Biquinoline Cu(I) complexes: synthesis, mesomorphism and photophysical studies in solution and condensed states

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Figure S1. First and second cycle DSC traces of complex **CuL2_ClO**₄ obtained with a heating-cooling rate of 10°C/min.



Figure S2. First and second cycle DSC traces of complex $CuL2_BF_4$ obtained with a heating-cooling rate of 10°C/min.



Figure S3. First and second cycle DSC traces of complex **CuL3_ClO**₄ obtained with a heating-cooling rate of 10°C/min.



Figure S4. First and second cycle DSC traces of complex $CuL3_BF_4$ obtained with a heating-cooling rate of 10°C/min.



Figure S5. X-ray powder diffraction patterns of complex **CuL2_BF**₄ recorded on heating in the II cycle at 75 °C (red trace) and at 135 °C (black trace)



Figure S6: X-ray powder diffraction patterns of complex **CuL2_ClO**₄ recorded on heating in the II cycle at r.t. (black trace), 57° (red trace), 70 °C (blue trace) and 130 °C (sky-blue trace)



Figure S7. Absorption spectrum of L2 in dichloromethane solution



Figure S8. Absorption spectrum of L3 in dichloromethane solution



Figure S9. Emission spectra of the ligands L2 and L3 in dichloromethane solution



Figure S10. Absorption spectrum of CuL2_ClO4 in dichloromethane solution



Figure S11. Absorption spectrum of CuL3_CIO4 in dichloromethane solution



Figure S12. Absorption spectrum of CuL3_BF4 in dichloromethane solution



Figure S13. Emission spectrum of CuL2_X from a solid sample at room temperature (λ_{ex} = 570 nm).



Figure S14. Emission spectrum of CuL3_X from a solid sample at room temperature (λ_{ex} = 570 nm).