

### Supplementary Information

Comparing the host behaviour of *N,N'*-bis(9-phenyl-9-thioxanthenyl)ethylenediamine and *N,N'*-bis(9-phenyl-9-xanthenyl)ethylenediamine in the presence of various alkylated aromatic and aniline guests: crystal engineering considerations

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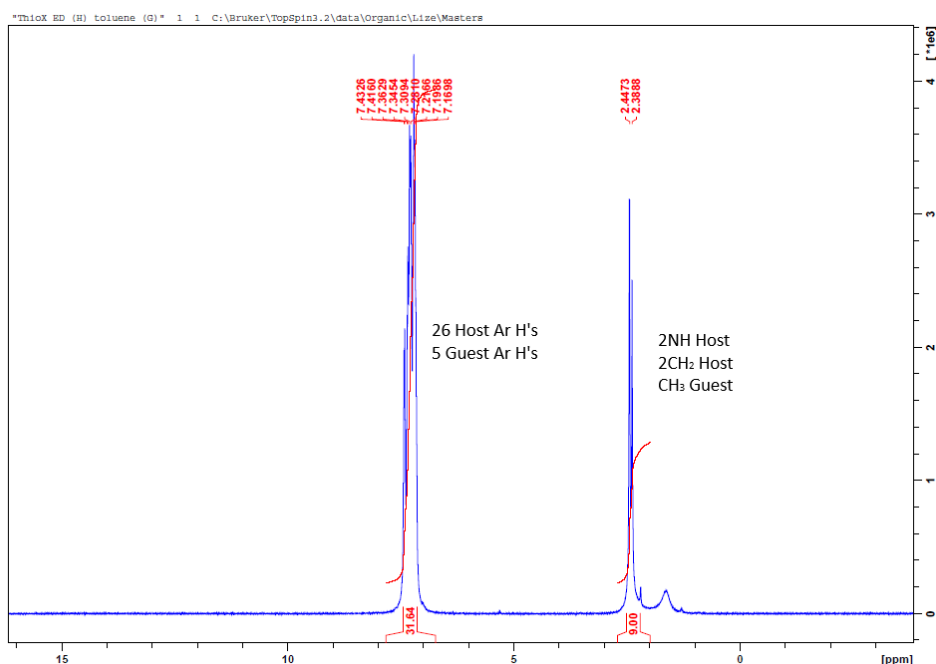


Figure S1a. <sup>1</sup>H-NMR spectrum for the complex between TOL and H<sub>1</sub>.

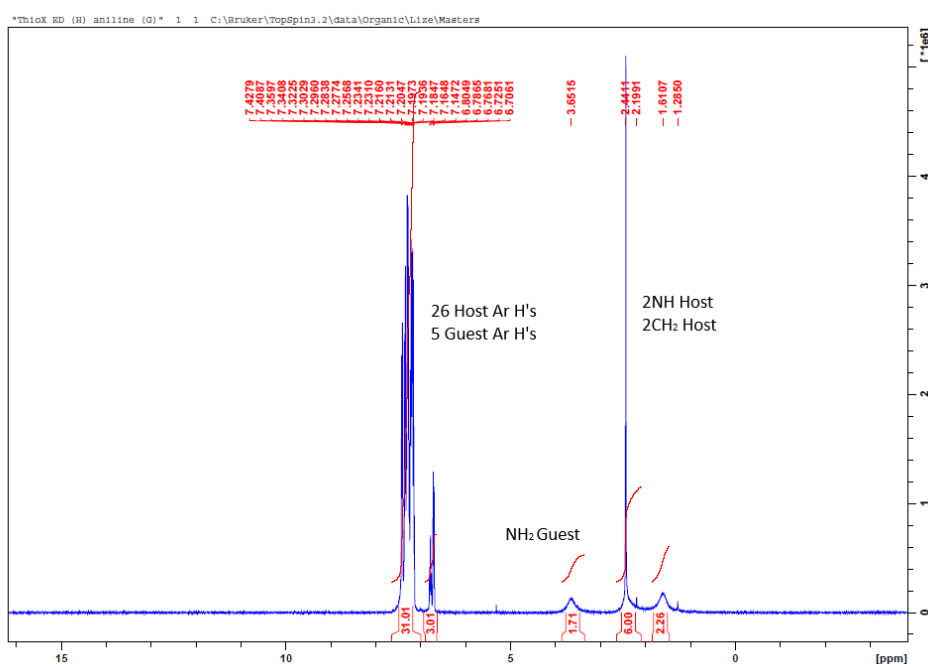


Figure S1b. <sup>1</sup>H-NMR spectrum for the complex between ANI and H<sub>1</sub>.

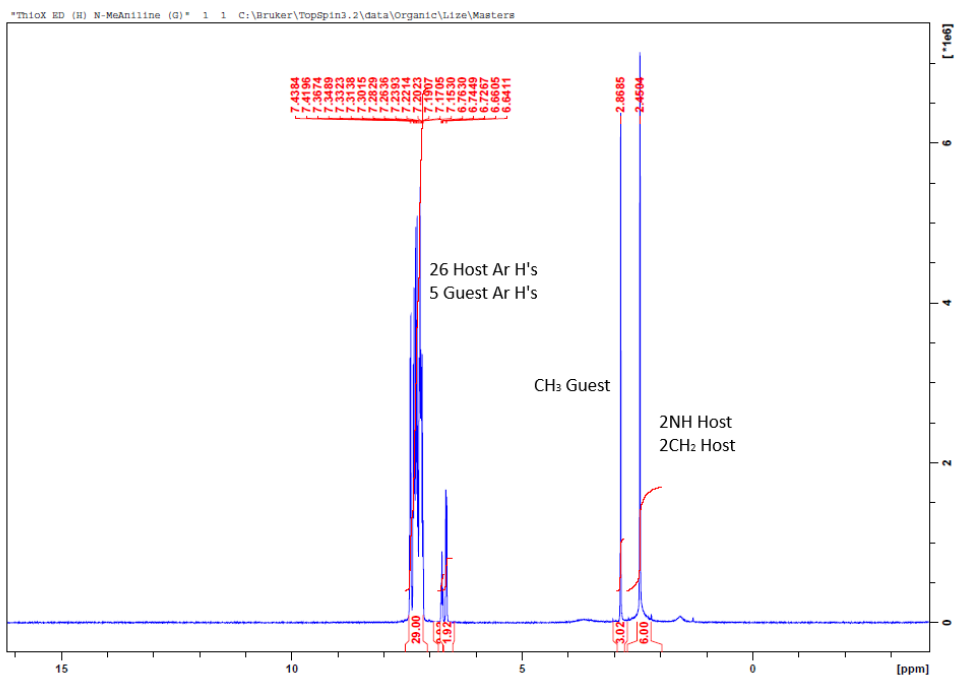


Figure S1c. <sup>1</sup>H-NMR spectrum for the complex between NMA and H<sub>1</sub>.

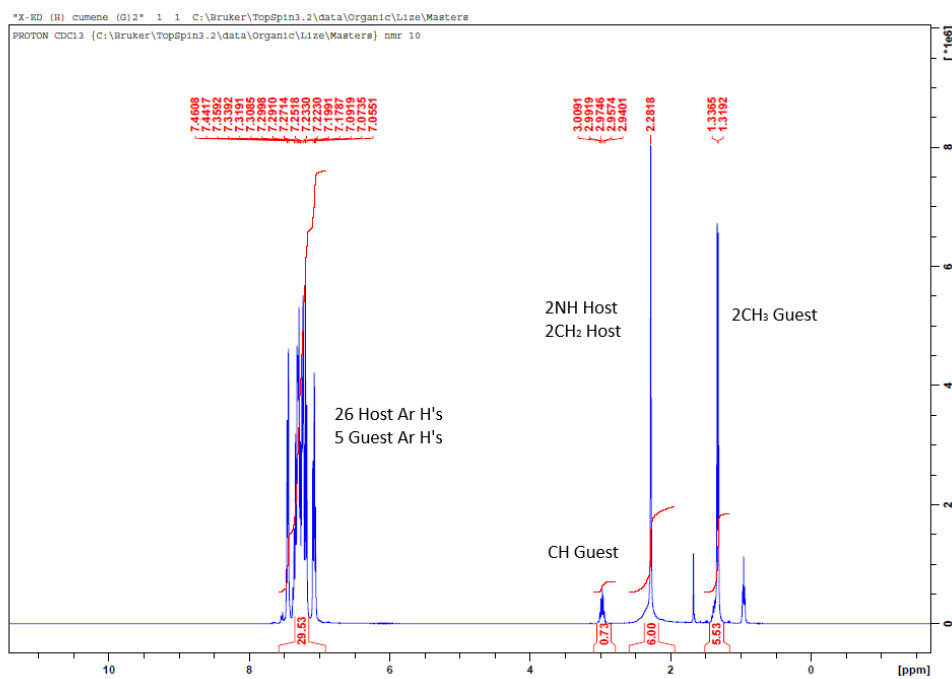


Figure S1d. <sup>1</sup>H-NMR spectrum for the complex between CU and H<sub>2</sub>.

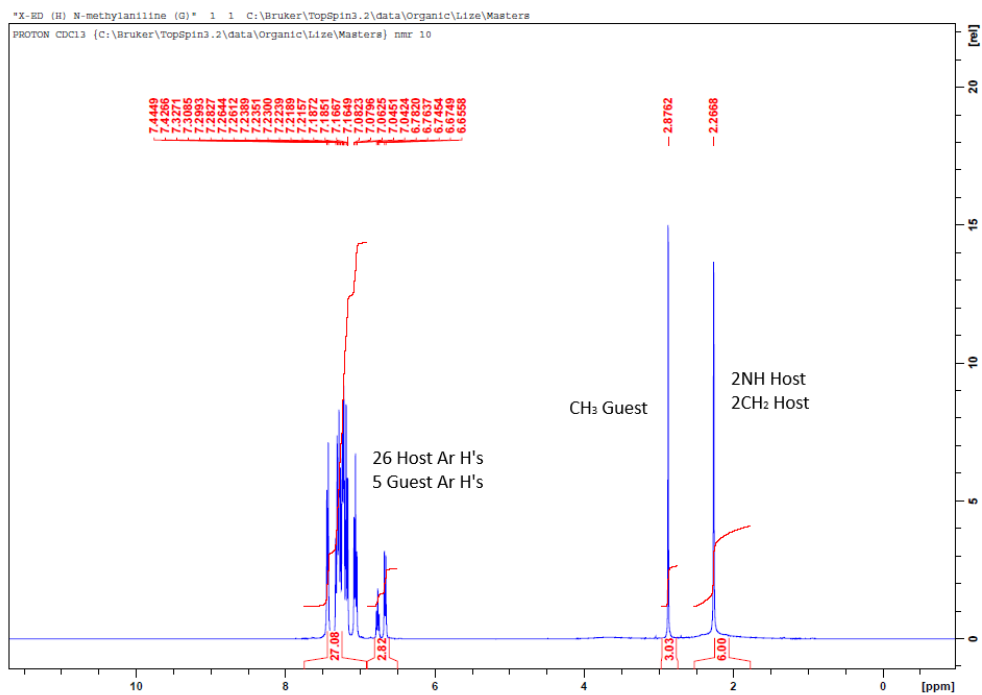


Figure S1e. <sup>1</sup>H-NMR spectrum for the complex between NMA and H<sub>2</sub>.

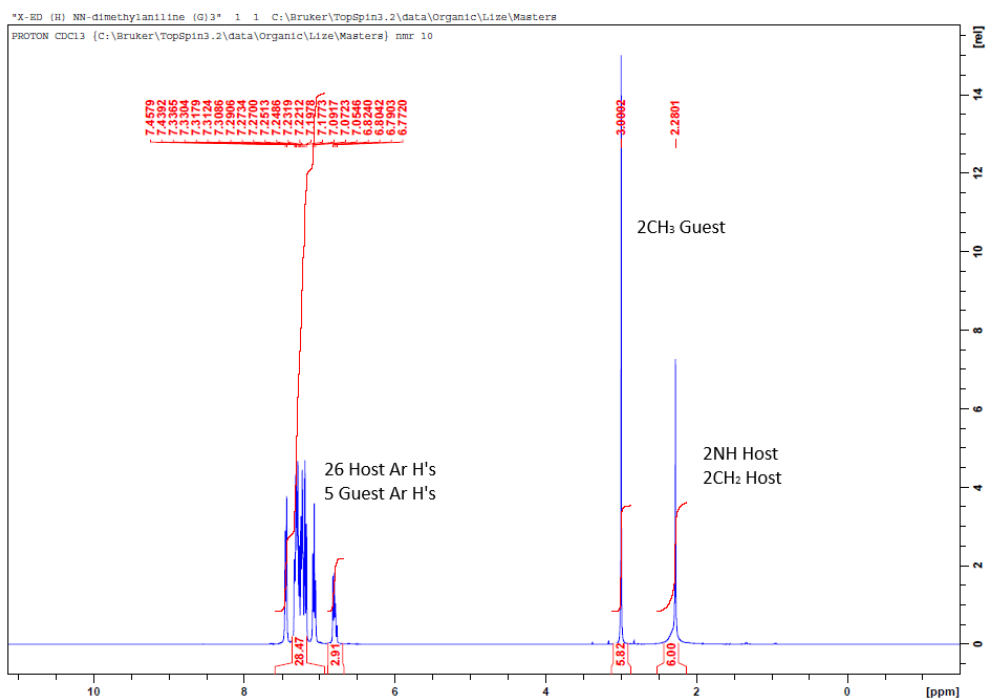


Figure S1f. <sup>1</sup>H-NMR spectrum for the complex between NNDMA and H<sub>2</sub>.

**Table S2.** Duplicate data for equimolar competition experiments of ANI, NMA and NNDMA with H<sub>1</sub>

Comp:	Batch 1	Batch 2	Average	% e.s.d.s
ANI + NMA	90.00:10.00	88.24:11.76	89.12:10.88	(0.88): (0.88)
ANI + NNDMA	93.77: 6.23	90.69:9.31	92.23:7.77	(1.54): (1.54)
NMA + NNDMA	<sup>a</sup>	<sup>a</sup>	-	-
ANI + NMA + NNDMA	89.50:5.73:4.78	88.90:5.91:5.20	89.20:5.82:4.98	(0.30): (0.09):(0.21)

<sup>a</sup>No inclusion occurred.**Table S3.** Duplicate data for equimolar competition experiments of TOL, EB and CU with H<sub>1</sub>

Comp:	Batch 1	Batch 2	Average	% e.s.d.s
TOL + EB	92.07:7.94	91.44:8.54	91.76:8.24	(0.31): (0.31)
TOL + CU	96.77:3.24	96.13:3.87	96.44:3.56	(0.32): (0.32)
EB + CU	62.09:37.91	63.70:36.30	62.90:37.10	(0.81):(0.81)
TOL + EB + CU	89.08:8.74:2.18	90.50:7.70:1.80	89.90:8.22:1.88	(0.71): (0.52):(0.19)

**Table S4.** Duplicate data for equimolar competition experiments of ANI, NMA and NNDMA with H<sub>2</sub>

Guests:	Batch 1	Batch 2	Average	% e.s.d.s
ANI + NMA	24.03:75.97	21.74:78.26	22.89:77.12	(1.15): (1.15)
ANI + NNDMA	1.35:98.65	1.28:98.72	1.32:98.69	(0.03): (0.03)
NMA + NNDMA	7.57:92.43	5.77:94.23	6.67:93.33	(0.90):(0.90)
ANI + NMA + NNDMA	1.99:5.99:92.01	5.38:6.17:88.46	3.69:6.08:90.24	(1.70): (0.09):(1.78)

**Table S5.** Duplicate data for equimolar competition experiments of TOL, EB and CU with H<sub>2</sub>

Guests:	Batch 1	Batch 2	Batch 3	Average	% e.s.d.s
TOL + EB	<sup>a</sup>	<sup>a</sup>	<sup>a</sup>	-	-
TOL + CU	<sup>a</sup>	<sup>a</sup>	<sup>a</sup>	-	-
EB + CU	49.19:50.81	46.51:53.50	49.98:50.02	48.56:51.44	(1.48):(1.48)
TOL + EB + CU	<sup>a</sup>	<sup>a</sup>	<sup>a</sup>	-	-

<sup>a</sup>No inclusion occurred

**Table S6.** K values for competition experiment of ANI/NMA with H<sub>1</sub>.

ANI ml	ANI c	NMA ml	NMA c	K values
1	1	0	0	
0,74531	0,92048	0,25469	0,07952	3,95560512
0,5402	0,94585	0,4598	0,05415	14,8675085
0,43038	0,91012	0,56962	0,08988	13,4019731
0,34044	0,90494	0,65956	0,09506	18,4431757
0,14723	0,02795	0,85277	0,97205	0,16654393
0	0	1	1	10,1669613

**Table S7.** K values for competition experiment of ANI/NNDMA with H<sub>1</sub>.

ANI ml	ANI c	NNDMA ml	NNDMA c	K values
1	1	0	0	
0,7598	0,93357	0,2402	0,06643	4,44279576
0,57566	0,90274	0,42434	0,09726	6,84189397
0,47608	0,91477	0,52392	0,08523	11,811484
0,34071	0,86149	0,65929	0,13851	12,0354053
0,22098	0,70008	0,77902	0,29992	8,22882605
0	0	1	1	8,67208101

**Table S8.** K values for competition experiment of ANI/NNDMA with H<sub>2</sub>.

NNDMA ml	NNDMA c	ANI ml	ANI c	K value
1	1	0	0	
0,84303	0,98557	0,15697	0,01443	12,7172958
0,63608	0,96269	0,36392	0,03731	14,7623465
0,56845	0,93178	0,43155	0,06822	10,369087
0,41709	0,94891	0,58291	0,05109	25,9573796
0,23338	0,60841	0,76662	0,39159	5,10365443
0	0	1	1	13,7819527

**Table S9.** K values for competition experiment of NMA/NNDMA with H<sub>2</sub>.

NNDMA ml	NNDMA c	NMA ml	NMA c	K value
1	1	0	0	
0,81391	0,96842	0,18609	0,03158	7,01129557
0,61555	0,93217	0,38445	0,06783	8,5832119
0,5	0,93524	0,5	0,06476	14,4416306
0,43407	0,43577	0,56593	0,56423	1,00694117
0,20745	0,04099	0,79255	0,95901	0,16329317
0	0	1	1	6,24127449

**Table S10.** K values for competition experiment of ANI/NMA with H<sub>2</sub>.

NMA ml	NMA c	ANI ml	ANI c	K value
1	1	0	0	
0,84382	0,94534	0,15618	0,05466	3,20106145
0,63729	0,80534	0,36271	0,19466	2,35464249
0,5147	0,79952	0,4853	0,20048	3,76022993
0,46543	0,75464	0,53457	0,24536	3,53253333
0,22835	0,18754	0,77165	0,81246	0,78002993
0	0	1	1	2,72569943

**Table 11.** K values for competition experiment of TOL/EB with H<sub>1</sub>.

TOL ml	TOL c	EB ml	EB c	K value
1	1	0	0	
0,74434	0,96978	0,25566	0,03022	11,0222483
0,55354	0,92547	0,44646	0,07453	10,0153186
0,43868	0,91253	0,56132	0,08747	13,3490607
0,34253	0,87077	0,65747	0,12923	12,9335409
0,18537	0,21562	0,81463	0,78438	1,20804602
0	0	1	1	9,70564289

**Table 12.** K values for competition experiment of TOL/CU with H<sub>1</sub>.

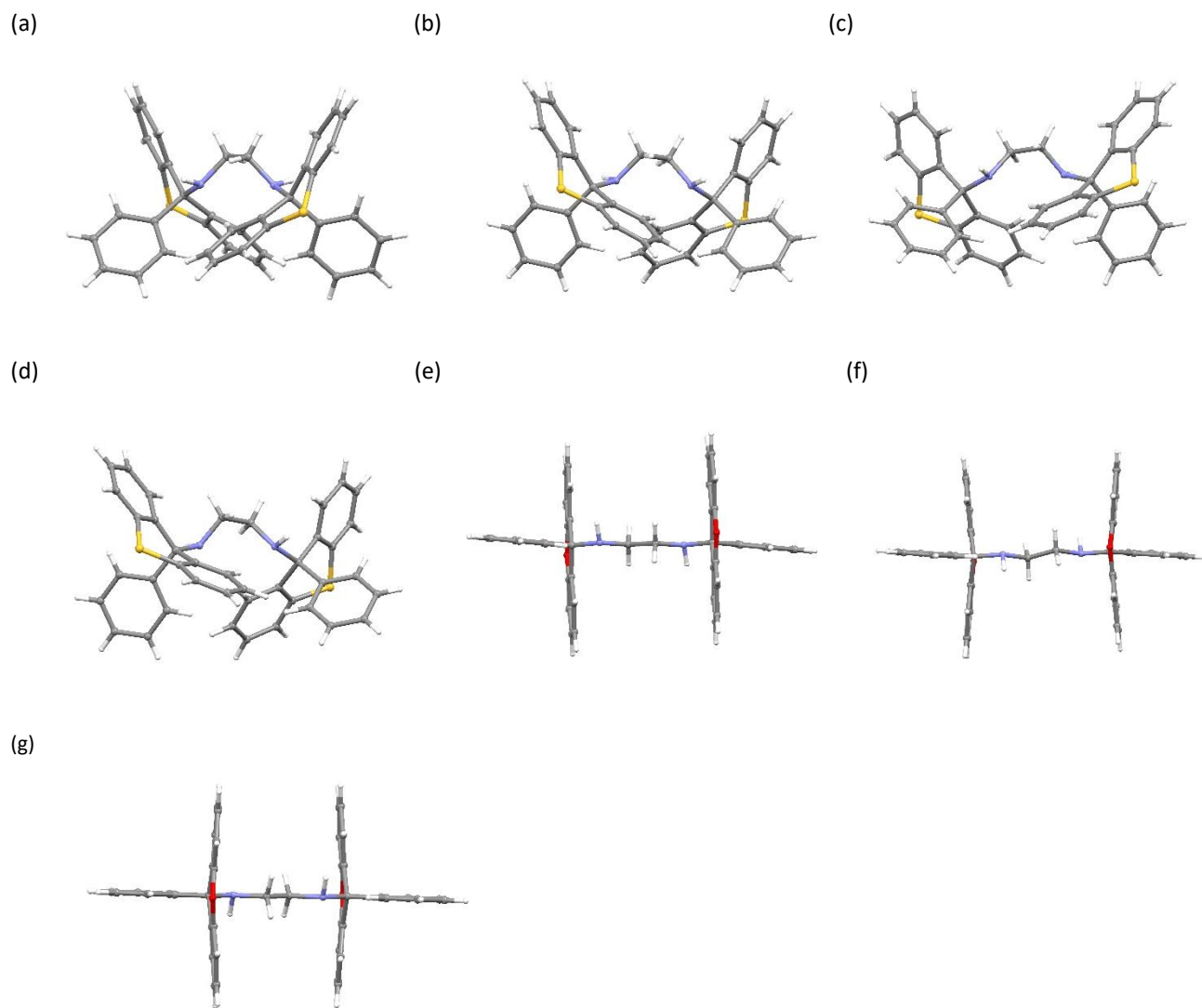
TOL ml	TOL c	CU ml	CU c	K value
1	1	0	0	
0,73171	0,98423	0,26829	0,01577	22,8839189
0,48083	0,9712	0,51917	0,0288	36,4111351
0,40474	0,96633	0,59526	0,03367	42,2097635
0,3137	0,94882	0,6863	0,05118	40,5586069
0,16365	0,51494	0,83635	0,48506	5,42541817
0	0	1	1	29,4977685

**Table 13.** K values for competition experiment of EB/CU with H<sub>1</sub>.

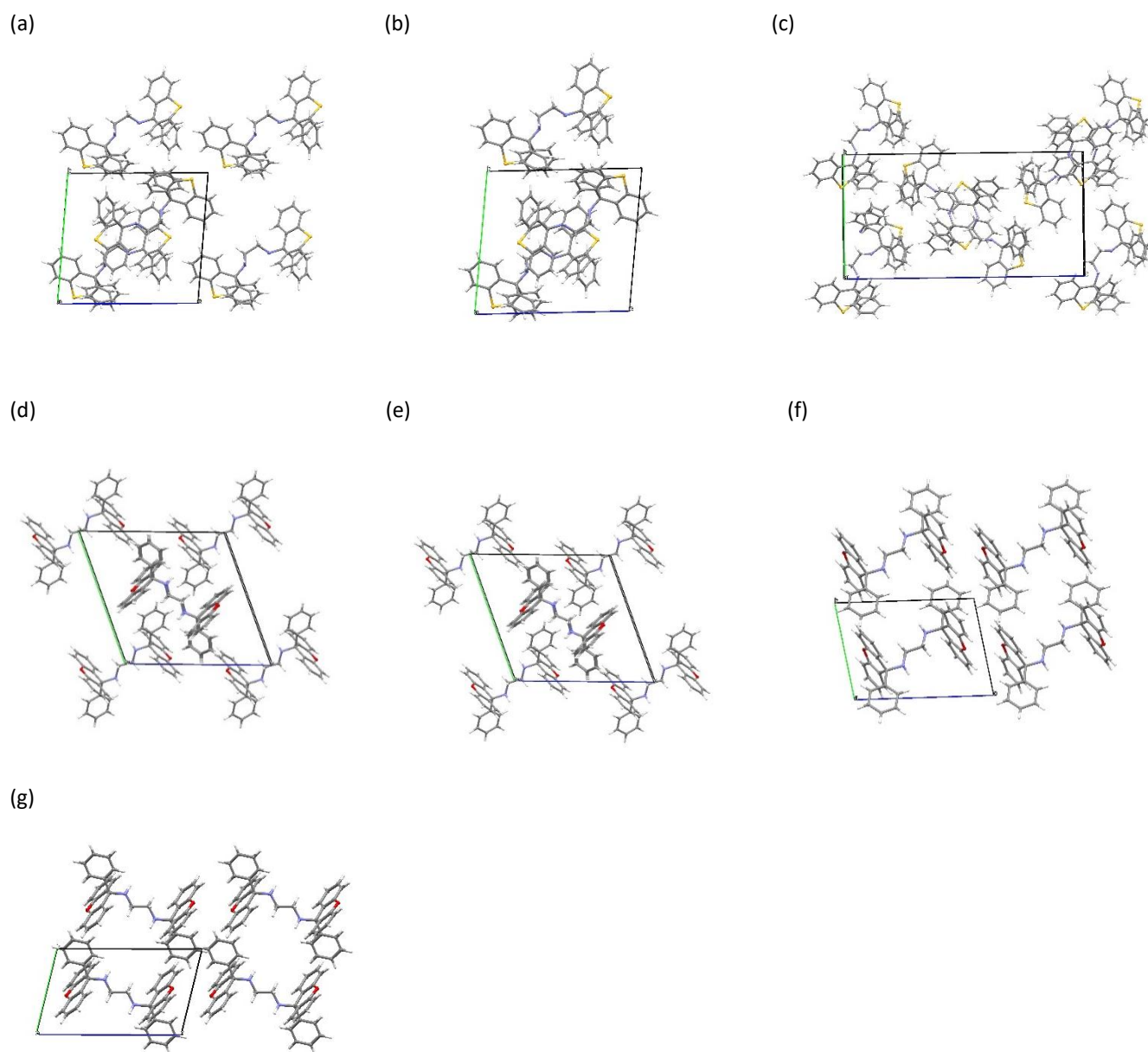
EB ml	EB c	CU ml	CU c	K value
1	1	0	0	
0,77593	0,87745	0,22407	0,12255	2,06761766
0,54517	0,70261	0,45483	0,29739	1,97108394
0,4639	0,63175	0,5361	0,36825	1,982549
0,35935	0,5158	0,64065	0,4842	1,89915204
0,18003	0,2538	0,81997	0,7462	1,54913567
0	0	1	1	1,89390766

**Table S14.** K values for competition experiment of CU/EB with H<sub>2</sub>.

CU ml	CU c	EB ml	EB c	K value
1	1	0	0	
0,78946	0,8187	0,21054	0,1813	1,20429109
0,55886	0,62882	0,44114	0,37118	1,33725796
0,47916	0,50022	0,52084	0,49978	1,08794253
0,36782	0,35244	0,63218	0,64756	0,93542848
0,19814	0,15228	0,80186	0,84772	0,72697057
0	0	1	1	1,05837813



**Figure S15.** Host geometry in complexes (a) H<sub>1</sub>·TOL, (b) H<sub>1</sub>·EB, (c) H<sub>1</sub>·ANI, (d) H<sub>1</sub>·NMA, (e) H<sub>2</sub>·CU, (f) H<sub>2</sub>·NMA and (g) H<sub>2</sub>·NNDMA; guests were removed for ease of examination.



**Figure S16.** Host packing in complexes (a) H<sub>1</sub>-TOL, (b) H<sub>1</sub>-EB, (c) H<sub>1</sub>-ANI, (d) H<sub>1</sub>-NMA, (e) H<sub>2</sub>-CU, (f) H<sub>2</sub>-NMA and (g) H<sub>2</sub>-NNDMA; guests were removed for ease of examination.



**Table S17.** Crystallographic data for H<sub>1</sub>·TOL and H<sub>1</sub>·EB.<sup>a</sup>

Non-covalent interaction	H <sub>1</sub> ·TOL	H <sub>1</sub> ·EB	Symmetry
$\pi \cdots \pi$ (H $\cdots$ H and H $\cdots$ G)	4.361(1) – 5.919(1) Å 5.045(1) – 5.919(1) Å [8]	4.521(1) – 5.997(1) Å 4.999(1) – 5.997(1) Å [7]	
CH $\cdots$ $\pi$ (host-host)			
C <sub>(host)</sub> –H <sub>32</sub> $\cdots$ C <sub>(host)</sub>	2.80 Å, 134°		x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub>	2.71 Å, 147°		3/2–x, –1/2+y, 1/2–z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub>	2.84 Å, 138°		x, 1+y, z
C <sub>(guest)</sub> –H <sub>(guest)</sub> $\cdots$ C <sub>(host)</sub>	2.66 Å, 162°		3/2–x, –1/2+y, 1/2–z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub>		2.96 Å, 139°	x, y, 1+z
C <sub>(guest)</sub> –H <sub>(guest 1)</sub> $\cdots$ C <sub>(host)</sub>		2.99 Å, 155°	1+x, y, –1+z
C <sub>(guest)</sub> –H <sub>(guest 2)</sub> $\cdots$ C <sub>(host)</sub>		2.88 Å, 163°	1+x, y, –1+z
H-bonding (intramolecular)	Non-classical	Non-classical	
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	2.64(2) Å, 103°	2.761(2) Å, 103°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	3.443(2) Å, 152°	3.432(2) Å, 154°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	2.906(2) Å, 102°	2.904(2) Å, 102°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>		3.456(2) Å, 151°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	2.766(2) Å, 103°	2.764(2) Å, 103°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	2.897(2) Å, 102°	2.899(2) Å, 102°	x, y, z
Other short contacts (host/ guest and guest/guest)			
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(guest)</sub> –C <sub>(guest)</sub>	2.35 Å, 138°(<)		3/2–x, 1/2+y, 1/2–z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(guest)</sub> –C <sub>(guest)</sub>	2.37 Å, 129°(<)		x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> –C <sub>(host)</sub>		2.86 Å, 145° (<)	–x, 1–y, 1–z
C <sub>(guest 2)</sub> –C <sub>(guest 2)</sub> $\cdots$ H <sub>(guest 2)</sub> –C <sub>(guest 2)</sub>		1.89 Å, 131° (<<)	2–x, 1–y, –z

<sup>a</sup>Distances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved while those denoted by << is this sum minus 0.2 Å.

**Table S18.** Crystallographic data for  $H_1 \cdot ANI$  and  $H_1 \cdot NMA$ .<sup>a,b</sup>

Non-covalent interaction	$H_1 \cdot ANI$	$H_1 \cdot NMA$	Symmetry
$\pi \cdots \pi$ (H $\cdots$ H and H $\cdots$ G) H $\cdots$ G major H $\cdots$ G minor	4.379(1) – 5.935(3) Å 4.885(1) – 5.927(1) Å [7] 4.714(5) – 5.935(3) Å [8]	4.658(1) – 5.932(1) Å 4.935(1) – 5.893 (1) Å [7]	
CH $\cdots$ $\pi$ (host-host)  C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>G(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>G(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>G(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>G(host)</sub>  C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>G(host)</sub> C <sub>(guest)</sub> –H <sub>(guest)</sub> $\cdots$ C <sub>G(host)</sub>	2.78 Å, 134° 2.71 Å, 149° 2.84 Å, 137° 2.93 Å, 133°  2.94 Å, 139° 2.93 Å, 140°		x, y, z 3/2–x, –1/2+y, 1/2–z x, 1+y, z x, y, z  x, y, 1+z –x, 1–y, 1–z
X–Y $\cdots$ $\pi$ (guest-host)  C <sub>(guest)</sub> –N <sub>(guest)</sub> $\cdots$ C <sub>G(host)</sub>	3.518 (4) Å, 109°		2–x, 1–y, 1–z
H-bonding (Intramolecular)  C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	Non-classical  2.766(2) Å, 103° 3.436(2) Å, 152° 2.913(2) Å, 102°  2.775(2) Å, 103° 2.896(2) Å, 102°	Non-classical  2.767(2) Å, 103° 3.466(2) Å, 158° 2.910(2) Å, 102° 3.454(2) Å, 148° 2.768(2) Å, 103° 2.900(2) Å, 102°	x, y, z x, y, z x, y, z x, y, z x, y, z x, y, z
Other short contacts (host/ guest and guest/guest)  C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> –C <sub>(host)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(guest 1)</sub> –C <sub>(guest 1)</sub> C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(guest 2)</sub> –C <sub>(guest 2)</sub> C <sub>(guest 2)</sub> –N <sub>(guest 2)</sub> $\cdots$ N <sub>(guest 2)</sub> –C <sub>(Guest 2)</sub> C <sub>(guest 2)</sub> –H <sub>(guest 2)</sub> $\cdots$ H <sub>(host)</sub> –C <sub>(host)</sub> C <sub>(guest 2)</sub> –N <sub>(guest 2)</sub> $\cdots$ C <sub>(host)</sub> –C <sub>(host)</sub>  C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(guest 1)</sub> –C <sub>(guest 1)</sub>	2.89Å, 153°(<) 2.34Å, 134°(<) 2.30Å, 149°(<) 2.3890Å, 125.1(6)°(<<) 2.35Å, 138°(<) 3.244Å, 105.5(2)°(<)  2.37 Å, 143° (<)		x, 1+y, z –1/2+x, 1/2–y, –1/2+z –1/2+x, 1/2–y, –1/2+z 2–x, –y, 1–z 1–x, 1–y, 1–z x, 1+y, z  x, 1+y, 1+z

<sup>a</sup>Number of H-G interactions indicated in parentheses.

<sup>b</sup>Distances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved while those denoted by << is this sum minus 0.2 Å.

**Table S19.** Crystallographic data for H<sub>2</sub>·CU.<sup>a</sup>

Non-covalent interaction	H <sub>2</sub> ·CU	Symmetry
$\pi \cdots \pi$ (H $\cdots$ H and H $\cdots$ G)	3.981(1)–5.977(1) Å	
CH $\cdots$ $\pi$		
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.85 Å, 90°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.96 Å, 84°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.99 Å, 80°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.74 Å, 94°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.94 Å, 143°	x, –1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(guest)</sub> G <sub>(guest)</sub>	3.00 Å, 117°	–1+x, –1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.68 Å, 105°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.97 Å, 151°	1+x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.77 Å, 104°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(host)</sub> G <sub>(host)</sub>	2.69 Å, 156°	1–x, 1–y, 1–z
H-bonding (intramolecular)	Non-classical	
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	2.790(3) Å, 102°	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ N <sub>(host)</sub>	2.758 (3) Å, 102°	x, y, z
Other short contacts (host/guest and guest/guest)		
C <sub>(host)</sub> –C <sub>(host)</sub> $\cdots$ H <sub>(host)</sub> –C <sub>(host)</sub>	2.87 Å, 112° (<)	–x, 1–y, 1–z
C <sub>(host)</sub> –C <sub>(host)</sub> $\cdots$ C <sub>(guest 1)</sub> –C <sub>(guest 1)</sub>	3.33 Å, 112° (<)	–1+x, –1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(host)</sub> –C <sub>(host)</sub>	2.35 Å, 131° (<)	x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ C <sub>(guest 1)</sub> –C <sub>(guest 1)</sub>	2.88 Å, 131° (<)	1–x, 1–y, –z
C <sub>(host)</sub> –H <sub>(host)</sub> $\cdots$ H <sub>(host)</sub> –C <sub>(host)</sub>	2.78 Å, 101° (<)	–1+x, y, z
C <sub>(guest 1)</sub> –H <sub>(guest 1)</sub> $\cdots$ C <sub>(host)</sub> –O <sub>(host)</sub>	2.83 Å, 169° (<)	1+x, y, z
C <sub>(guest 2)</sub> –H <sub>(guest 2)</sub> $\cdots$ C <sub>(host)</sub> –O <sub>(host)</sub>	2.72 Å, 150° (<)	1–x, 1–y, 1–z

<sup>a</sup>Distances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved.

**Table S20.** Crystallographic data for H<sub>2</sub>·NMA and H<sub>2</sub>·NNDMA.<sup>a</sup>

Non-covalent interaction	H <sub>2</sub> ·NMA	H <sub>2</sub> ·NNDMA	Symmetry
π···π (H···H and H···G) H···G interactions	4.090(1)–5.930(1) Å 4.735(1)–5.930(1) Å [6]	3.919(2)–5.839(3) Å 4.977(3)–5.839(3) Å [8]	
CH···π (host-host)			
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>	2.91 Å, 77°		x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>	2.57 Å, 97°		x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(guest)</sub> G <sub>(guest)</sub>	2.95 Å, 127°		x, 1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>	2.61 Å, 163°		2–x, 2–y, 2–z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>	2.83 Å, 104°		x, y, z
C <sub>(guest)</sub> –H <sub>(guest)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>	2.89 Å, 139°		x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>		3.00 Å, 142°	x, –1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>		2.93 Å, 142°	1+x, y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(guest)</sub> G <sub>(guest)</sub>		2.95 Å, 141°	1–x, 1–y, 1–z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>		2.90 Å, 141°	1–x, 2–y, –z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> G <sub>(host)</sub>		2.57 Å, 106°	x, y, z
H-bonding (intramolecular)	Non-classical	Non-classical	
C <sub>(host)</sub> –H <sub>(host)</sub> ···N <sub>(host)</sub>	2.804(1) Å, 102°	2.770(2) Å, 103°	x, y, z
Other short contacts (host/ guest and guest/guest)			
C <sub>(host)</sub> –H <sub>(host)</sub> ···H <sub>(guest)</sub> –C <sub>(guest)</sub>	2.32 Å, 144° (<)		2–x, 1–y, 1–z
C <sub>(host)</sub> –H <sub>(host)</sub> ···O <sub>(host)</sub> –C <sub>(host)</sub>	2.64 Å, 165° (<)		1+x, 1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(host)</sub> –C <sub>(host)</sub>		2.81 Å, 136° (<)	1–x, 2–y, –z
N <sub>(host)</sub> –H <sub>(host)</sub> ···H <sub>(guest)</sub> –C <sub>(guest)</sub>		2.26 Å, 168° (<)	–1+x, 1+y, z
C <sub>(host)</sub> –H <sub>(host)</sub> ···C <sub>(guest)</sub> –N <sub>(guest)</sub>		2.75 Å, 146° (<)	1–x, 1–y, 1–z

<sup>a</sup>Distances denoted by < are contacts that measure less than the sum of the van der Waals radii of the atoms involved.

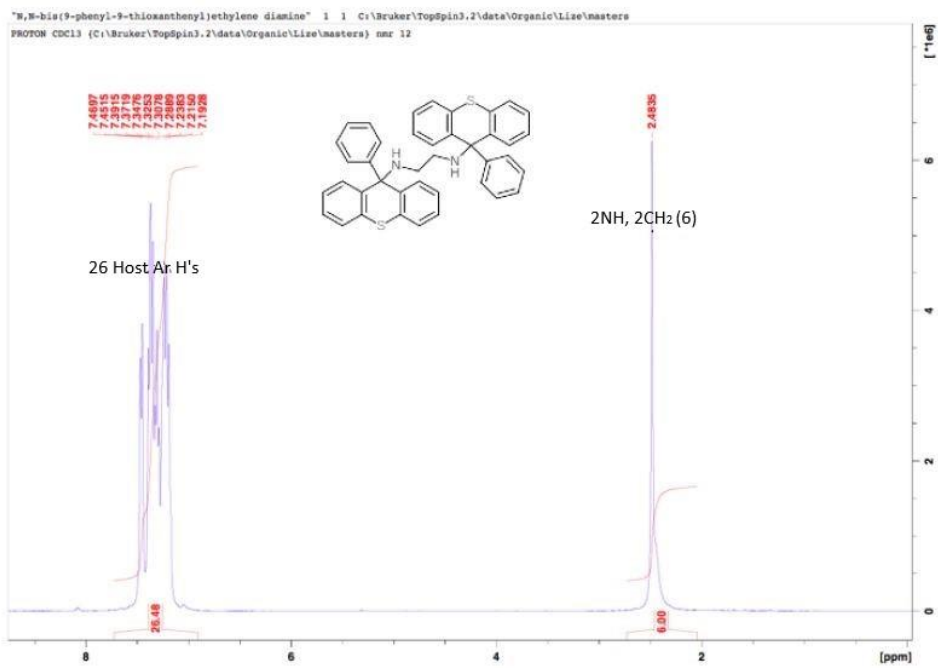


Figure S21a. <sup>1</sup>H-NMR spectrum for H<sub>1</sub>.

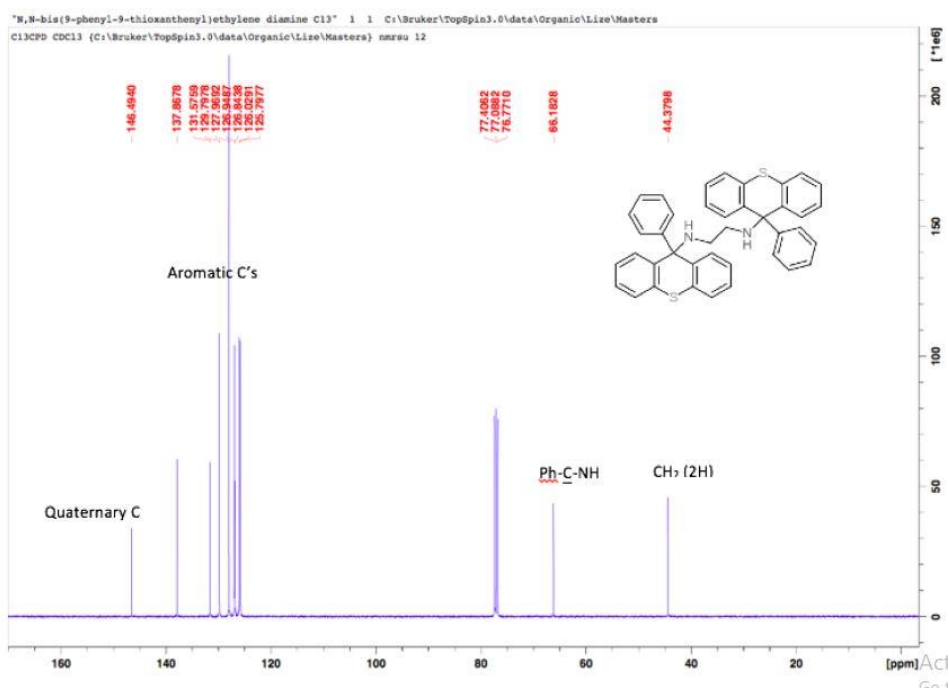


Figure S21b. <sup>13</sup>C-NMR spectrum for H<sub>1</sub>.

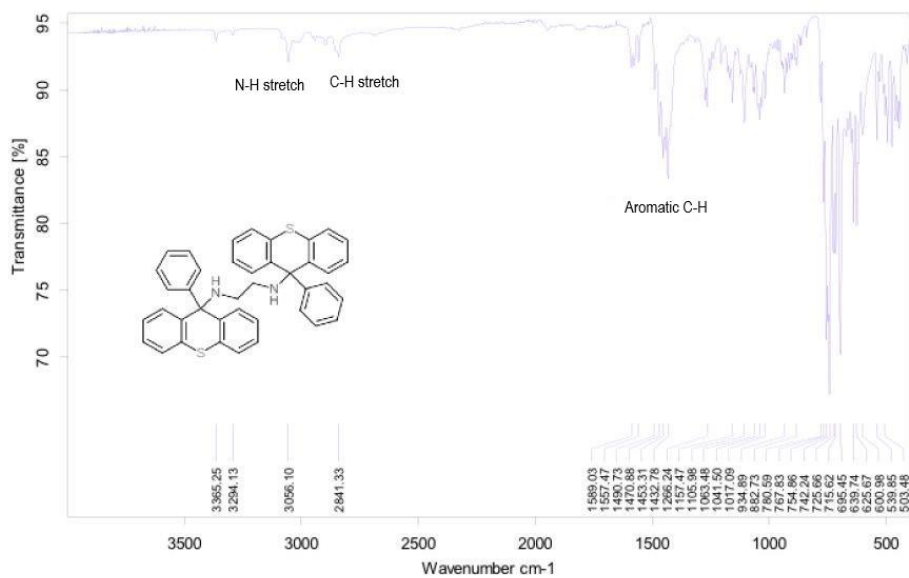


Figure S21c. IR spectrum for H<sub>1</sub>.

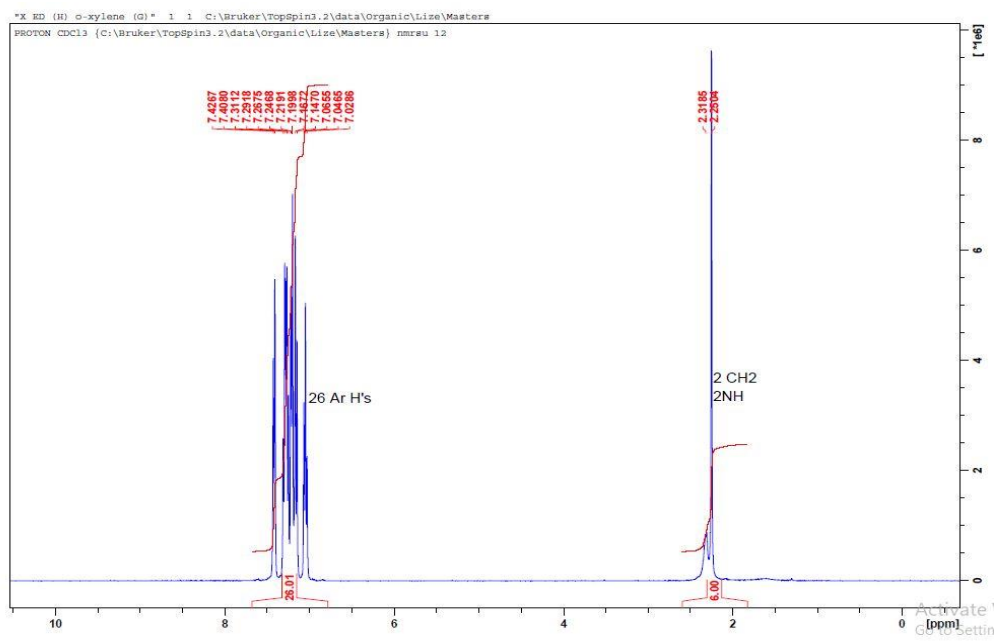


Figure S21d. <sup>1</sup>H-NMR spectrum for H<sub>2</sub>.

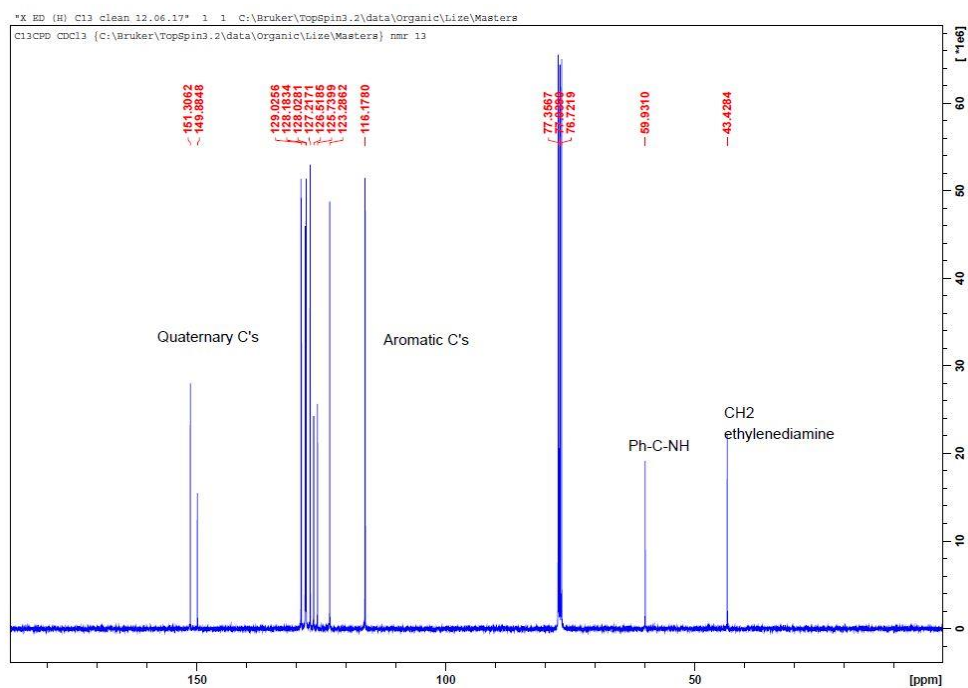


Figure S21e.  $^{13}\text{C}$ -NMR spectrum for  $\text{H}_2$ .

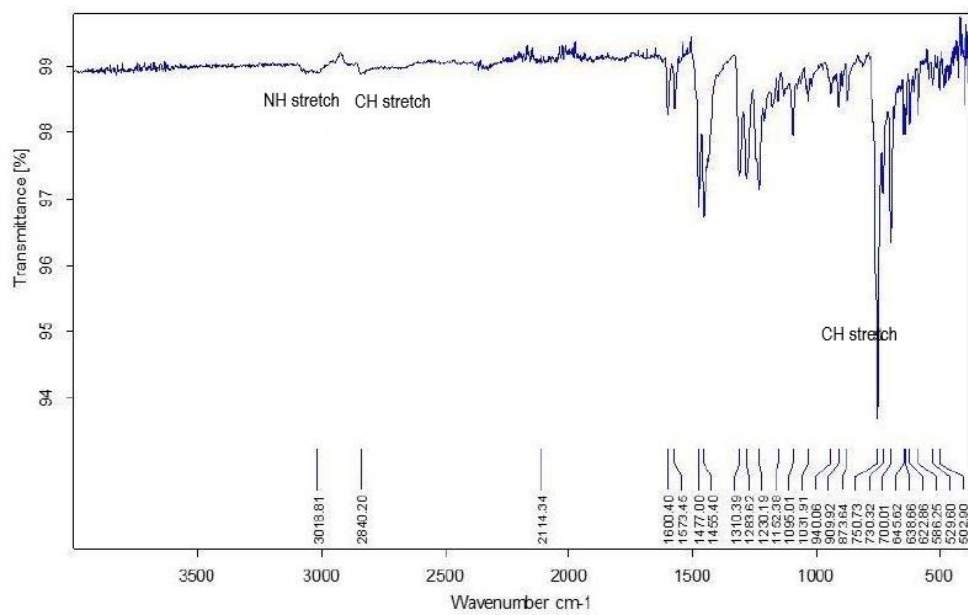


Figure S21f. IR spectrum for  $\text{H}_2$ .

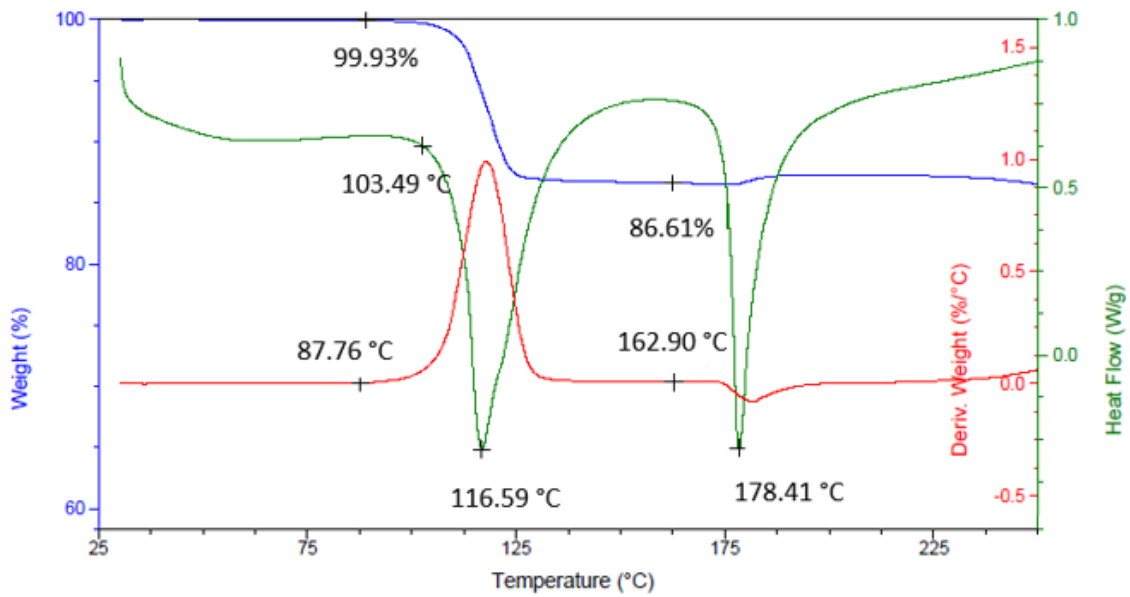


Figure S22a. Overlaid thermal traces (DSC, TG and DTG) for the  $H_1$ -ANI complex

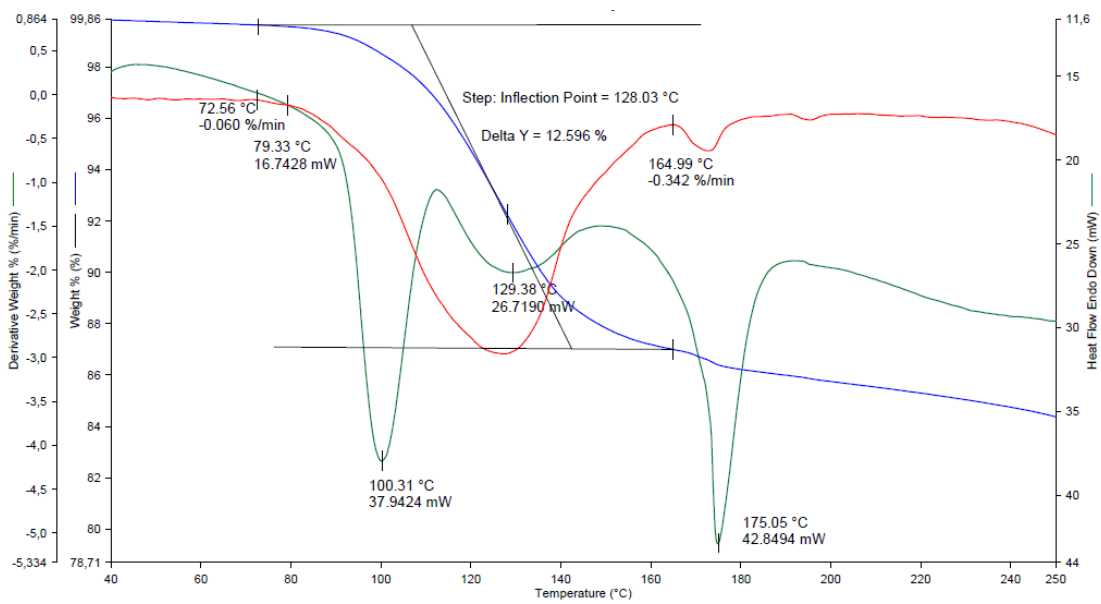


Figure S22b. Overlaid thermal traces (DSC, TG and DTG) for the  $H_1$ -NMA complex



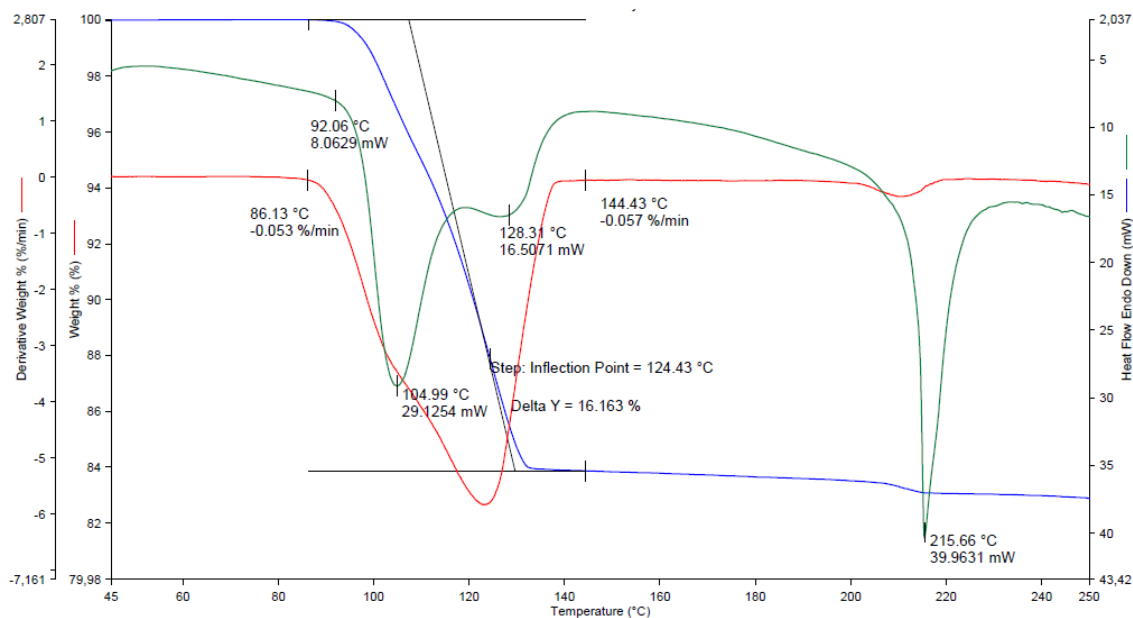


Figure S22c. Overlaid thermal traces (DSC, TG and DTG) for the  $H_2 \cdot NMA \cdot 0.406(H_2O)$  complex

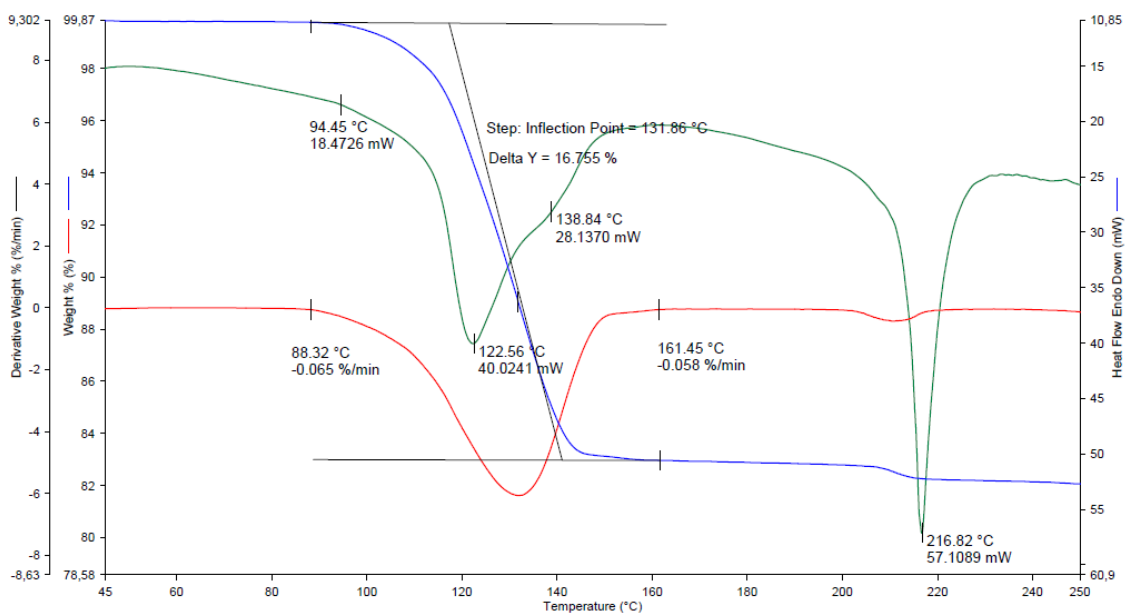


Figure S22d. Overlaid thermal traces (DSC, TG and DTG) for the  $H_2 \cdot NNDMA$  complex