Size and Shape Dependences of Adsorption Kinetics of Malachite Green on Nano-MgO: a Theoretical and

Experimental Study

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(Supplementary Information)

The bar charts of size distribution are shown in Figure S1, and it can be observed that the size distribution is homogeneous.



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Figure S1 The bar of equivalent particle size distribution of nano-MgO of spherical (a)-(e) and cubic (f)-(j) shapes.



The plots of q_t to t are shown in Figure S2, and the results are shown in Table S1.

Figure S2 The fit plots of pseudo-first-order kinetic equation for the adsorption of malachite green on nano-MgO with different particle sizes at 298 K.

Sphere nano-MgO		Cube nano-MgO	
d/nm	pseudo-first-	d/nm	pseudo-first-
	order (R ²)		order (R ²)
40.4	0.9973	59.1	0.9928
54.8	0.9940	66.8	0.9953
64.4	0.9925	71.0	0.9935
80.0	0.9894	97.0	0.9881
90.2	0.9882	114.7	0.9844

 Table S1. The correlation coefficient parameters of pseudo-first-order kinetic equation for the adsorption of malachite green on nano-MgO at 298 K.

As illustrated in Table S1, we found the nonlinear fitting for the pseudo-first-order kinetic equation with the correlation coefficients $R^2 < 0.9998$.