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## **Supporting Information**

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T1	$T^{\mathrm{Al}}_1\text{-}O^{H}\text{-}T^{\mathrm{Si}}_2$	$T^{\mathrm{Al}}_{1}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{10}$	$T^{\mathrm{Al}}_1\text{-}O^{H}\text{-}T^{\mathrm{Si}}_5$	$T^{Al}_{1}\text{-}O^{H}\text{-}T^{Si}_{4}$
Ε	-2296.63	-2296.54	-2296.53	-2296.34
T2	$T^{\mathrm{Al}}_{2}$ - $O^{H}$ - $T^{\mathrm{Si}}_{6}$	$T^{Al}_2\text{-}O^{H}\text{-}T^{Si}_1$	$T^{Al}_2\text{-}O^{H}\text{-}T^{Si}_8$	$T^{\mathrm{Al}}_2\text{-}O^{H}\text{-}T^{\mathrm{Si}}_3$
Ε	-2296.68	-2296.61	-2296.54	-2296.49
Т3	$T^{Al}_{3}\text{-}O^{H}\text{-}T^{Si}_{4}$	$T^{Al}_{3}\text{-}O^{H}\text{-}T^{Si}_{6}$	$T_3^{\text{Al}}\text{-}O^{\text{H}}\text{-}T_{12}^{\text{Si}}$	$T^{Al}_{3}\text{-}O^{H}\text{-}T^{Si}_{2}$
Ε	-2296.67	-2296.65	-2296.63	-2296.61
T4	$T_4^{\mathrm{Al}} ext{-}O^{H} ext{-}T_7^{\mathrm{Si}}$	$T^{Al}_{4}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T_{4}^{Al}\text{-}O^{H}\text{-}T_{1}^{Si}$	$T_4^{Al}$ - $O^{H}$ - $T_3^{Si}$
Ε	-2296.68	-2296.64	-2296.57	-2296.51
Т5	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{11}$	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{6}$	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{1}$	$T^{\mathrm{Al}}_{5}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{4}$
Ε	-2296.58	-2296.57	-2296.57	-2296.45
Т6	$T_{6}^{Al}\text{-}O^{H}\text{-}T_{2}^{Si}$	$T^{Al}_{6}\text{-}O^{H}\text{-}T^{Si}_{3}$	$T^{Al}_{6}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T_6^{\mathrm{Al}}$ - $O^{H}$ - $T_9^{\mathrm{Si}}$
Ε	-2296.63	-2296.60	-2296.58	-2296.51
Т7	$T_7^{\mathrm{Al}}\text{-}O^{H}\text{-}T_{11}^{\mathrm{Si}}$	$T^{Al}_{7}\text{-}O^{H}\text{-}T^{Si}_{8}$	$T_7^{Al}\text{-}O^{H}\text{-}T_4^{Si}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_7^{\mathrm{Si}}$
Ε	-2296.64	-2296.62	-2296.62	-2296.58
Т8	$T^{Al}_8\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{8}\text{-}O^{H}\text{-}T^{Si}_{7}$	$T^{Al}_{8}\text{-}O^{H}\text{-}T^{Si}_{9}$	$T^{Al}_8\text{-}O^{H}\text{-}T^{Si}_2$
Ε	-2296.73	-2296.59	-2296.50	-2296.46
Т9	$T_{9}^{\text{Al}}\text{-}O^{H}\text{-}T_{10}^{\text{Si}}$	$T_9^{\mathrm{Al}}$ - $O^{H}$ - $T_9^{\mathrm{Si}}$	$T_{9}^{\mathrm{Al}}\text{-}O^{H}\text{-}T_{6}^{\mathrm{Si}}$	$T_{9}^{Al}$ - $O^{H}$ - $T_{8}^{Si}$
Ε	-2296.69	-2296.54	-2296.53	-2296.47
T10	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{1}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{10}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{11}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{9}$
Ε	-2296.62	-2296.53	-2296.39	-2296.35
T11	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{5}$	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{7}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{10}$
Ε	-2296.62	-2296.59	-2296.58	-2296.46
T12	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{8}$	$T^{\mathrm{Al}}_{12} \text{-} O^{H} \text{-} T^{\mathrm{Si}}_{12}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{3}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{11}$
Е	-2296.70	-2296.63	-2296.62	-2296.58

	Table S1: Computed	l electronic energ	ies (eV	of all 48 acidic sites for one Al substitution
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Table S2: Bader charge	(BC	) for H atom in th	e bare HZSM5	structures
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T1	$T^{Al}_1\text{-}O^{H}\text{-}T^{Si}_2$	$T^{Al}_{1}\text{-}O^{H}\text{-}T^{Si}_{10}$	$T^{Al}_{1}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T^{Al}_{1}\text{-}O^{H}\text{-}T^{Si}_{4}$
BC/bare HZSM-5	0.67	0.68	0.67	0.65
T2	$T_2^{\mathrm{Al}}$ - $O^{H}$ - $T_6^{\mathrm{Si}}$	$T^{\mathrm{Al}}_2$ - $O^{H}$ - $T^{\mathrm{Si}}_1$	$T^{\mathrm{Al}}_{2}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{8}$	$T^{\mathrm{Al}}_2$ - $O^{H}$ - $T^{\mathrm{Si}}_3$
BC/bare HZSM-5	0.64	0.70	0.68	0.72
Т3	$T^{\mathrm{Al}}_{3}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{4}$	$T^{\mathrm{Al}}_{3}$ - $O^{H}$ - $T^{\mathrm{Si}}_{6}$	$T_3^{\mathrm{Al}}$ - $O^{H}$ - $T_{12}^{\mathrm{Si}}$	$T^{\mathrm{Al}}_{3}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{2}$
BC/bare HZSM-5	0.72	0.71	0.71	0.69
T4	$T^{\mathrm{Al}}_4 ext{-}O^{H} ext{-}T^{\mathrm{Si}}_7$	$T_4^{\mathrm{Al}}$ - $O^{H}$ - $T_5^{\mathrm{Si}}$	$T_4^{\mathrm{Al}}$ - $O^{H}$ - $T_1^{\mathrm{Si}}$	$T_{4}^{\mathrm{Al}}$ - $O^{H}$ - $T_{3}^{\mathrm{Si}}$
BC/bare HZSM-5	0.68	0.67	0.67	0.68
Т5	$T^{\mathrm{Al}}_{5}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{11}$	$T^{\mathrm{Al}}_{5}$ - $O^{H}$ - $T^{\mathrm{Si}}_{6}$	$T^{\mathrm{Al}}_{5}$ - $O^{H}$ - $T^{\mathrm{Si}}_{1}$	$T^{\mathrm{Al}}_{5}$ - $O^{H}$ - $T^{\mathrm{Si}}_{4}$
BC/bare HZSM-5	0.66	0.70	0.68	0.66
Т6	$T_{6}^{\mathrm{Al}}$ - $O^{H}$ - $T_{2}^{\mathrm{Si}}$	$T_{6}^{\mathrm{Al}}$ - $O^{H}$ - $T_{3}^{\mathrm{Si}}$	$T_{6}^{\mathrm{Al}}$ - $O^{H}$ - $T_{5}^{\mathrm{Si}}$	$T_{6}^{\mathrm{Al}}$ - $O^{H}$ - $T_{9}^{\mathrm{Si}}$
BC/bare HZSM-5	0.68	0.68	0.71	0.67
Τ7	$T^{\mathrm{Al}}_{7}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{11}$	$T^{Al}_{7}\text{-}O^{H}\text{-}T^{Si}_{8}$	$T_7^{\mathrm{Al}}\text{-}O^{H}\text{-}T_4^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_7^{\mathrm{Si}}$
BC/bare HZSM-5	0.67	0.67	0.67	0.67
Т8	$T^{\mathrm{Al}}_{8}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{12}$	$T^{\mathrm{Al}}_{8}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{7}$	$T^{\mathrm{Al}}_{8}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{9}$	$T^{\mathrm{Al}}_{8}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{2}$
BC/bare HZSM-5	0.69	0.70	0.68	0.67
Т9	$T_{9}^{\mathrm{Al}}$ - $O^{H}$ - $T_{10}^{\mathrm{Si}}$	$T_9^{\mathrm{Al}}$ - $O^{H}$ - $T_9^{\mathrm{Si}}$	$T_9^{\mathrm{Al}}\text{-}O^{H}\text{-}T_6^{\mathrm{Si}}$	$T_9^{\mathrm{Al}}\text{-}O^{H}\text{-}T_8^{\mathrm{Si}}$
BC/bare HZSM-5	0.70	0.69	0.67	0.69
T10	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{1}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{10}$	$T^{\mathrm{Al}}_{10}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{11}$	$T^{\mathrm{Al}}_{10}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{9}$
BC/bare HZSM-5	0.70	0.70	0.66	0.68
T11	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{7}$	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{12}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{10}$
BC/bare HZSM-5	0.66	0.64	0.67	0.65
T12	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{8}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{3}$	$T^{\mathrm{Al}}_{12}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{11}$
BC/bare HZSM-5	0.68	0.71	0.68	0.68

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T1	$T^{\mathrm{Al}}_1\text{-}O^{H}\text{-}T^{\mathrm{Si}}_2$	$T^{\mathrm{Al}}_{1}$ - $O^{H}$ - $T^{\mathrm{Si}}_{10}$	$T^{\mathrm{Al}}_{1}$ - $O^{H}$ - $T^{\mathrm{Si}}_{5}$	$T^{\mathrm{Al}}_1\text{-}O^{H}\text{-}T^{\mathrm{Si}}_4$
E/Pyridine	-2370.13	-2370.00	-2370.04	-2368.43
T2	$T_2^{\mathrm{Al}}$ - $O^{H}$ - $T_6^{\mathrm{Si}}$	$T_2^{\mathrm{Al}}$ - $O^{H}$ - $T_1^{\mathrm{Si}}$	$T^{\mathrm{Al}}_2$ -O <sup>H</sup> - $T^{\mathrm{Si}}_8$	$T_2^{Al}\text{-}O^{H}\text{-}T_3^{Si}$
E/Pyridine	-2369.82	-2370.03	-2369.42	-2369.96
ТЗ	$T^{\mathrm{Al}}_{3}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{4}$	$T_3^{\mathrm{Al}}$ - $O^{H}$ - $T_6^{\mathrm{Si}}$	$T^{\mathrm{Al}}_{3}$ - $O^{H}$ - $T^{\mathrm{Si}}_{12}$	$T^{\mathrm{Al}}_3$ -O <sup>H</sup> - $T^{\mathrm{Si}}_2$
E/Pyridine	-2369.88	-2370.02	-2370.07	-2370.05
Т4	$T_4^{\mathrm{Al}}$ - $O^{H}$ - $T_7^{\mathrm{Si}}$	$T_4^{\mathrm{Al}}$ - $O^{H}$ - $T_5^{\mathrm{Si}}$	$T_4^{\mathrm{Al}}$ - $O^{H}$ - $T_1^{\mathrm{Si}}$	$T_4^{\mathrm{Al}}$ - $O^{H}$ - $T_3^{\mathrm{Si}}$
E/Pyridine	-2369.68	-2369.87	-2368.47	-2369.90
Т5	$T^{\mathrm{Al}}_{5} ext{-}O^{H} ext{-}T^{\mathrm{Si}}_{11}$	$T^{\mathrm{Al}}_{5}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{6}$	$T^{\mathrm{Al}}_{5}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{1}$	$T_5^{\mathrm{Al}}$ -O <sup>H</sup> - $T_4^{\mathrm{Si}}$
E/Pyridine	-2368.37	-2370.03	-2369.68	-2369.67
Т6	$T_{6}^{\mathrm{Al}}$ -O <sup>H</sup> - $T_{2}^{\mathrm{Si}}$	$T_{6}^{\mathrm{Al}}$ -O <sup>H</sup> - $T_{3}^{\mathrm{Si}}$	$T_{6}^{\mathrm{Al}}$ -O <sup>H</sup> - $T_{5}^{\mathrm{Si}}$	$T_{6}^{\mathrm{Al}}$ -O <sup>H</sup> - $T_{9}^{\mathrm{Si}}$
E/Pyridine	-2369.56	-2369.17	-2369.86	-2369.84
Τ7	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_{11}^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_8^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_4^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_7^{\mathrm{Si}}$
E/Pyridine	-2370.13	-2370.07	-2370.11	-2370.09
Т8	$T^{\mathrm{Al}}_{8} ext{-}O^{H} ext{-}T^{\mathrm{Si}}_{12}$	$T^{\mathrm{Al}}_{8} ext{-}O^{H} ext{-}T^{\mathrm{Si}}_{7}$	$T^{\mathrm{Al}}_{8} ext{-}O^{H} ext{-}T^{\mathrm{Si}}_{9}$	$T^{\mathrm{Al}}_{8}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{2}$
E/Pyridine	-2368.49	-2370.00	-2369.97	-2369.48
Т9	$T^{\mathrm{Al}}_{9} ext{-}O^{H} ext{-}T^{\mathrm{Si}}_{10}$	$T_{9}^{\mathrm{Al}}$ - $O^{H}$ - $T_{9}^{\mathrm{Si}}$	$T_{9}^{\mathrm{Al}}$ -O <sup>H</sup> - $T_{6}^{\mathrm{Si}}$	$T_{9}^{Al}\text{-}O^{H}\text{-}T_{8}^{Si}$
E/Pyridine	-2369.62	-2369.80	-2369.83	-2369.98
T10	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{1}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{10}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{11}$	$T^{Al}_{10}\text{-}O^{H}\text{-}T^{Si}_{9}$
E/Pyridine	-2370.06	-2370.02	-2368.65	-2369.81
T11	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{7}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{10}$
E/Pyridine	-2369.35	-2370.02	-2370.09	-2369.85
T12	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{8}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{3}$	$T^{\mathrm{Al}}_{12}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{11}$
E/Pyridine	-2369.69	-2369.95	-2370.06	-2369.99

Table S3: Computed electronic energies (eV) of pyridine adsorption at all 48 acidic sites for one AI substitution

Table S4: Computed electronic e	energies (	eV)	of ammonia	adsorption	n at all 4	8 acidic site	es for one	Al substitution
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<b>T1</b>	$T_{1}^{Al}-O^{H}-T_{2}^{Si}$	$T_{1}^{Al}-O^{H}-T_{10}^{Si}$	$T_{1}^{Al}-O^{H}-T_{5}^{Si}$	$T_{1}^{Al}-O^{H}-T_{4}^{Si}$
τ2	$T_{2}^{Al}-O^{H}-T_{6}^{Si}$	$T_{2}^{Al}-O^{H}-T_{1}^{Si}$	$T_{2}^{Al}-O^{H}-T_{8}^{Si}$	$T_{2}^{Al}-O^{H}-T_{3}^{Si}$
е/мп <sub>3</sub>	$T_{3}^{Al}-O^{H}-T_{4}^{Si}$	$^{-2317.03}$	$^{-2317.45}$	$^{-2317.03}$
ТЗ		$T_{3}^{Al}-O^{H}-T_{6}^{Si}$	$T_{3}^{Al}-O^{H}-T_{12}^{Si}$	$T_{3}^{Al}-O^{H}-T_{2}^{Si}$
<b>T4</b> <i>F/NH</i> 2	$T_{4}^{Al}-O^{H}-T_{7}^{Si}$	$T_{4}^{Al}$ -O <sup>H</sup> -T_{5}^{Si}-2317.82	$T_{4}^{Al}-O^{H}-T_{1}^{Si}$ -2316.60	$T_{4}^{Al}$ -O <sup>H</sup> - $T_{3}^{Si}$ -2317.74
<b>5</b> <b>15</b> <i>Ε/ΝΗ</i> ₂	T <sup>Al</sup> <sub>5</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>11</sub> -2317.61	T <sup>Al</sup> <sub>5</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>6</sub> -2317.67	T <sup>Al</sup> <sub>5</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>1</sub> -2317.67	T <sup>Al</sup> <sub>5</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>4</sub> -2317.61
<b>Τ6</b>	T <sup>AI</sup> <sub>6</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>2</sub>	T <sup>Al</sup> <sub>6</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>3</sub>	T <sup>Al</sup> <sub>6</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>5</sub>	T <sup>Al</sup> <sub>6</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>9</sub>
<i>Ε/ΝΗ</i> ₃	-2317.59	-2317.55	-2317.64	-2317.56
<b>T7</b>	T <sup>Al</sup> <sub>7</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>11</sub>	T <sup>Al</sup> <sub>7</sub> -O <sup>H</sup> -T <sup>Si</sup>	T <sup>Al</sup> <sub>7</sub> -O <sup>H</sup> -T <sup>Si</sup>	T <sup>Al</sup> <sub>7</sub> -O <sup>H</sup> -T <sup>Si</sup>
E/NH <sub>3</sub>	-2317.72	-2317.72	-2317.72	-2317.72
<b>T8</b>	T <sup>Al</sup> <sub>8</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>12</sub>	T <sup>Al</sup> <sub>8</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>7</sub>	T <sup>Al</sup> <sub>8</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>9</sub>	$T_8^{Al}$ -0 <sup>H</sup> - $T_2^{Si}$
<i>E/NH</i> ₃	-2316.81	-2317.63	-2317.58	-2317.50
<b>T9</b>	T <sup>Al</sup> <sub>9</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>10</sub>	T <sup>Al</sup> <sub>9</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>9</sub>	T <sup>Al</sup> <sub>9</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>6</sub>	T <sup>Al</sup> <sub>9</sub> -O <sup>H</sup> -T <sup>Si</sup>
<i>E/NH</i> 3	-2316.20	-2317.64	-2317.59	-2317.64
<b>T10</b>	T <sup>Al</sup> <sub>10</sub> -O <sup>H</sup> -T <sup>Si</sup>	$T_{10}^{Al}$ -O <sup>H</sup> - $T_{10}^{Si}$	$T_{10}^{Al}$ -O <sup>H</sup> - $T_{11}^{Si}$	T <sup>Al</sup> <sub>10</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>9</sub>
<i>E/NH</i> <sub>3</sub>	-2317.65	-2317.68	-2317.68	-2317.47
T11	T <sup>Al</sup> <sub>11</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>5</sub>	T <sup>Al</sup> <sub>11</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>7</sub>	T <sup>Al</sup> <sub>11</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>12</sub>	T <sup>Al</sup> <sub>11</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>10</sub>
E/NH₃	-2317.65	-2317.64	-2317.71	-2317.54
-,, T12 E/NH₂	T <sup>Al</sup> <sub>12</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>8</sub> -2316.57	T <sup>Al</sup> <sub>12</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>12</sub> -2317.64	T <sup>Al</sup> <sub>12</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>3</sub> -2317.62	T <sup>Al</sup> <sub>12</sub> -O <sup>H</sup> -T <sup>Si</sup> <sub>11</sub> -2317.62

T1	$T^{\mathrm{Al}}_{1}$ - $O^{H}$ - $T^{\mathrm{Si}}_{2}$	$T^{\mathrm{Al}}_{1}$ - $O^{H}$ - $T^{\mathrm{Si}}_{10}$	$T^{\mathrm{Al}}_{1}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{5}$	$T^{\mathrm{Al}}_{1}$ -O <sup>H</sup> - $T^{\mathrm{Si}}_{4}$
E/NH <sub>2</sub> CH <sub>3</sub>	-2334.23	-2334.16	-2334.23	-2332.68
E/NH(CH <sub>3</sub> ) <sub>2</sub>	-2350.82	-2350.79	-2350.82	-2348.95
<i>E/</i> N(CH <sub>3</sub> ) <sub>3</sub>	-2367.28	-2367.21	-2367.10	-2365.55
Т3	$T^{\mathrm{Al}}_{3}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{4}$	$T^{Al}_{3}\text{-}O^{H}\text{-}T^{Si}_{6}$	$T_3^{Al}\text{-}O^{H}\text{-}T_{12}^{Si}$	$T^{Al}_{3}\text{-}O^{H}\text{-}T^{Si}_{2}$
E/NH <sub>2</sub> CH <sub>3</sub>	-2334.24	-2334.16	-2334.16	-2334.16
E/NH(CH <sub>3</sub> ) <sub>2</sub>	-2350.79	-2350.79	-2350.76	-2350.82
<i>E/</i> N(CH <sub>3</sub> ) <sub>3</sub>	-2367.21	-2367.21	-2366.91	-2367.18
Т5	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{11}$	$T^{\mathrm{Al}}_{5}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{6}$	$T_5^{\mathrm{Al}}\text{-}O^{H}\text{-}T_1^{\mathrm{Si}}$	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{4}$
E/NH <sub>2</sub> CH <sub>3</sub>	-2334.15	-2334.21	-2334.16	-2334.18
E/NH(CH <sub>3</sub> ) <sub>2</sub>	-2350.76	-2350.82	-2350.78	-2350.80
<i>E/</i> N(CH <sub>3</sub> ) <sub>3</sub>	-2366.86	-2367.08	-2367.25	-2367.22
Τ7	$T_7^{\text{Al}}\text{-}O^{\text{H}}\text{-}T_{11}^{\text{Si}}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_8^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}\text{-}O^{H}\text{-}T_4^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}$ - $O^{H}$ - $T_7^{\mathrm{Si}}$
<i>E/</i> NH <sub>2</sub> CH <sub>3</sub>	-2334.17	-2334.25	-2334.36	-2334.27
E/NH(CH <sub>3</sub> ) <sub>2</sub>	-2350.84	-2350.78	-2350.94	-2350.89
<i>E</i> /N(CH <sub>3</sub> ) <sub>3</sub>	-2367.22	-2367.22	-2367.29	-2367.11
T11	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{5}$	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{7}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{10}$
E/NH <sub>2</sub> CH <sub>3</sub>	-2334.04	-2334.06	-2334.23	-2334.05
E/NH(CH <sub>3</sub> ) <sub>2</sub>	-2350.78	-2350.83	-2350.85	-2350.20
<i>E/</i> N(CH <sub>3</sub> ) <sub>3</sub>	-2366.98	-2367.23	-2367.20	-2365.49
T12	$T^{\mathrm{Al}}_{12}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{8}$	$T^{\mathrm{Al}}_{12} \text{-} O^{H} \text{-} T^{\mathrm{Si}}_{12}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{3}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{11}$
E/NH <sub>2</sub> CH <sub>3</sub>	-2332.06	-2334.17	-2334.04	-2334.13
E/NH(CH <sub>3</sub> ) <sub>2</sub>	-2348.75	-2350.73	-2350.75	-2350.71
$E/N(CH_3)_3$	-2365.63	-2367.27	-2367.24	-2367.18

Table S5: Computed electronic energies (eV) of the adsorption of  $NH_2CH_3$ ,  $NH(CH_3)_2$  and  $N(CH_3)_3$  at the T1, T3, T5, T7, T11 and T12 centers for one AI substitution

T1		A=E11/K	A=E12/K	A=E13/K	A=E14/K
$\Delta H_{ads}/NH_3$	-128.2/-1.33	482	452	425	401
$\Delta H_{ads}$ /Pyridine	-204.7/-2.12	760	712	670	631
T2					
$\Delta H_{ads}/NH_3$	-125.6/-1.30	473	442	416	392
$\Delta H_{ads}$ /Pyridine	-191.0/-1.98	711	666	626	591
Т3					
$\Delta H_{ads}/NH_3$	-128.3/-1.33	484	451	424	400
$\Delta H_{ads}$ /Pyridine	-204.5/-2.12	758	712	670	631
T4					
$\Delta H_{ads}/NH_3$	-141.6/-1.47	531	498	468	441
$\Delta H_{ads}$ /Pyridine	-175.7/-1.83	654	614	577	545
T5					
$\Delta H_{ads}/NH_3$	-139.0/-1.44	521	488	458	433
$\Delta H_{ads}$ /Pyridine	-200.6/-2.08	744	699	656	620
Т6					
$\Delta H_{ads}/NH_3$	-128.6/-1.33	483	453	426	402
$\Delta H_{ads}$ /Pyridine	-180.0/-1.87	671	629	592	558
Τ7					
$\Delta H_{ads}/NH_3$	-140.7/-1.46	528	495	464	438
$\Delta H_{ads}$ /Pyridine	-203.4/-2.12	756	707	665	629
Т8					
$\Delta H_{ads}/NH_3$	-120.5/-1.25	454	426	400	376
$\Delta H_{ads}$ /Pyridine	-182.1/-1.89	678	637	598	563
Т9					
$\Delta H_{ads}/NH_3$	-122.5/-1.27	462	432	407	382
$\Delta H_{ads}$ /Pyridine	-182.0/-1.89	678	634	597	564
T10					
$\Delta H_{ads}/NH_3$	-133.9/-1.39	502	472	442	418
$\Delta H_{ads}$ /Pyridine	-178.0/-1.84	663	621	584	552
T11					
$\Delta H_{ads}/NH_3$	-136.1/-1.41	511	479	450	424
$\Delta H_{ads}$ /Pyridine	-201.3/-2.09	747	700	659	623
T12					
$\Delta H_{\rm ads}/\rm NH_3$	-126.1/-1.31	475	444	417	394
$\Delta H_{ads}$ /Pyridine	-189.2/-1.96	706	662	621	586

Table S6: All simulated desorption temperature of pyridine and  $\ensuremath{\mathsf{NH}}_3$ 

T1	$T_1^{\text{Al}}\text{-}O^{\text{H}}\text{-}T_2^{\text{Si}}$	$T_1^{Al}\text{-}O^{H}\text{-}T_{10}^{Si}$	$T^{Al}_{1}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T^{Al}_{1}\text{-}O^{H}\text{-}T^{Si}_{4}$
<i>E/</i> benzene	-2373.73	-2373.45	-2373.57	-2372.94
<i>E/</i> toluene	-2390.43	-2390.38	-2390.35	-2390.07
<i>E</i> /p-xylene	-2407.26	-2406.98	-2407.17	-2406.66
ТЗ	$T_3^{Al}\text{-}O^{H}\text{-}T_4^{Si}$	$T_3^{Al}\text{-}O^{H}\text{-}T_6^{Si}$	$T_3^{Al}\text{-}O^{H}\text{-}T_{12}^{Si}$	$T_3^{Al}\text{-}O^{H}\text{-}T_2^{Si}$
<i>E/</i> benzene	-2373.43	-2373.43	-2373.70	-2373.55
<i>E/</i> toluene	-2390.30	-2390.32	-2390.51	-2390.51
E/p-xylene	-2406.87	-2406.40	-2407.17	-2407.19
Т5	$T_5^{Al}\text{-}O^{H}\text{-}T_{11}^{Si}$	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{6}$	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{1}$	$T^{Al}_{5}\text{-}O^{H}\text{-}T^{Si}_{4}$
<i>E/</i> benzene	-2373.27	-2373.54	-2373.55	-2373.38
<i>E/</i> toluene	-2389.98	-2390.15	-2390.31	-2389.83
<i>E</i> /p-xylene	-2406.62	-2406.96	-2407.21	-2406.87
77	$T_{7}^{\text{Al}}\text{-}O^{\text{H}}\text{-}T_{11}^{\text{Si}}$	$T_7^{\text{Al}}\text{-}O^{\text{H}}\text{-}T_8^{\text{Si}}$	$T_7^{\mathrm{Al}}\text{-}O^{\mathrm{H}}\text{-}T_4^{\mathrm{Si}}$	$T_7^{\mathrm{Al}}\text{-}O^{\mathrm{H}}\text{-}T_7^{\mathrm{Si}}$
<i>E/</i> benzene	-2373.43	-2373.71	-2373.62	-2373.17
<i>E/</i> toluene	-2390.40	-2390.55	-2390.36	-2390.03
<i>E</i> /p-xylene	-2407.23	-2407.16	-2407.18	-2406.67
T11	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{5}$	$T^{\mathrm{Al}}_{11}\text{-}O^{H}\text{-}T^{\mathrm{Si}}_{7}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{11}\text{-}O^{H}\text{-}T^{Si}_{10}$
<i>E/</i> benzene	-2373.22	-2373.47	-2373.62	-2373.19
<i>E/</i> toluene	-2389.71	-2390.33	-2390.40	-2389.99
<i>E</i> /p-xylene	-2406.78	-2407.07	-2406.92	-2406.77
T12	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{8}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{12}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{3}$	$T^{Al}_{12}\text{-}O^{H}\text{-}T^{Si}_{11}$
<i>E/</i> benzene	-2372.56	-2373.63	-2373.68	-2373.52
<i>E/t</i> oluene	-2390.20	-2390.55	-2390.44	-2390.32
E/p-xylene	-2406.78	-2406.89	-2407.16	-2406.94

Table S7: Computed electronic energies (eV) of the adsorption of benzene, toluene and p-xylene at the T1, T3, T5, T7, T11 and T12 centers for one Al substitution





 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 





 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{5}$ 





 $\mathsf{T}_1^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 



























































 $\mathsf{T}_4^{Al}\text{-}\mathsf{O}^{H}\text{-}\mathsf{T}_7^{Si}$ 

















 $\mathsf{T}^{Al}_{5}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 





























 $T_7^{Al} - O^H - T_4^{Si}$ 



 $\mathsf{T}_7^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_7^{\text{Si}}$ 





 $\mathsf{T}^{Al}_{\phantom{Al}7}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{\phantom{B}8}$ 





 $\mathsf{T}^{Al}_{7}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 





 $\mathsf{T}^{Al}_8\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_2$ 

 $\mathsf{T}^{Al}_8\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_7$ 



b ↓→ a











 $\mathsf{T}^{Al}_8\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 









 $\mathsf{T}^{Al}_{9}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{8}$ 



a

b c→a



b ↓→ a



 $\mathsf{T}_9^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_9^{\text{Si}}$ 





 $\mathsf{T}_9^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_{10}^{\text{Si}}$ 





 $\mathsf{T}^{Al}_{10}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{1}$ 



 $\mathsf{T}^{Al}_{10}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{9}$ 

 ${\sf T}_{10}^{Al} {\text{--}} {\sf O}^{\sf H} {\text{--}} {\sf T}_{10}^{Si}$ 





a





 $\mathsf{T}^{Al}_{10}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 







¢ b



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{5}$ 



b ↓→ a

b ↓→a





















 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{8}$ 





 ${\sf T}_{12}^{\rm Al} {\rm -} {\sf O}^{\rm H} {\rm -} {\sf T}_{11}^{\rm Si}$ 







Figure S1 The optimized configurations of all 48 acidic sites for one Al substitution in the framework



Figure S2 Calculated Bader charge as a function of (a) relative energy  $\Delta E$  of all 48 acid sites and (b) pyridine adsorption enthalpy  $\Delta H_{ads}$ 

















 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 







 $\mathsf{T}^{Al}_{2}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{8}$ 





 $\mathsf{T}^{Al}_{2}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{6}$ 





T<sup>Al</sup> T<sup>Al</sup> D





 $\mathsf{T}^{Al}_{\phantom{3}3}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{\phantom{2}2}$ 



Transformed a second seco





 $\mathsf{T}^{Al}_{3}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{6}$ 





 $\mathsf{T}_3^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_{12}^{Si}$ 









 $\mathsf{T}_4^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_3^{Si}$ 





 $\mathsf{T}_4^{Al}\text{-}\mathsf{O}^{H}\text{-}\mathsf{T}_5^{Si}$ 



 $\mathsf{T}_4^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_7^{Si}$ 

b ↓→a







b ↓→a





 $\mathsf{T}_5^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_6^{Si}$ 











 $\mathsf{T}^{Al}_{5}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{4}$ 





c a b

b  $\leftarrow^{c}$  a

















b ↓→a

b L



























c a b

 $\mathsf{T}^{Al}_8\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 







b c → a







 $\mathsf{T}^{Al}_8\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_3$ 















 $\mathsf{T}_9^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_9^{\text{Si}}$ 





 $\mathsf{T}_9^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_{10}^{\text{Si}}$ 













¢ c



 $\mathsf{T}^{Al}_{10}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 







 $\mathsf{T}^{Al}_{10}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 







 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 





≙





c a b

c a b







c a b











¢ c







b c → a





 $\mathsf{T}^{Al}_{12} \text{-} \mathsf{O}^{\mathsf{H}} \text{-} \mathsf{T}^{Si}_{12}$ 



Figure S3 The optimized configurations of pyridine adsorption at all 48 acidic sites of one AI substitution in the framework

b

























b ↓→a ¢ c

• c b

¢ c













T<sup>Al</sup><sub>3</sub>-O<sup>H</sup>-T<sup>Si</sup><sub>4</sub>
























 $\mathsf{T}_4^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_5^{Si}$ 

































 $\mathsf{T}^{\mathrm{Al}}_{6}$ -O<sup>H</sup>- $\mathsf{T}^{\mathrm{Si}}_{5}$ 



 $\mathsf{T}_6^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_3^{Si}$ 





 $\mathsf{T}^{Al}_{\phantom{6}6} \text{-} \mathsf{O}^{\mathsf{H}} \text{-} \mathsf{T}^{Si}_{\phantom{2}2}$ 







 $\mathsf{T}^{\mathrm{Al}}_{7}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{\mathrm{Si}}_{11}$ 









 $\mathsf{T}_7^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_7^{\text{Si}}$ 





 $\mathsf{T}_7^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 





T<sup>Al</sup> <sup>78</sup> <sup>75</sup> <sup>75</sup> <sup>75</sup> <sup>6</sup>





























 $\mathsf{T}_9^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_6^{\text{Si}}$ 





















 $\mathsf{T}^{\mathrm{Al}}_{11} \text{-} \mathsf{O}^{\mathsf{H}} \text{-} \mathsf{T}^{\mathrm{Si}}_{12}$ 































Figure S4 The optimized configurations of ammonia adsorption at all 48 acidic sites of one Al substitution in the framework



 $T_{1}^{Al} - 0^{H} - T_{10}^{Si}$ 





 $\mathsf{T}_1^{Al}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}_2^{Si}$ 



 $\begin{array}{c} & & \\$ 







 $\mathsf{T}_1^{Al}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}_5^{Si}$ 





 $\mathsf{T}^{Al}_{\mathbf{3}}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}^{Si}_{\mathbf{2}}$ 





 $\mathsf{T}^{Al}_{\mathbf{3}}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}^{Si}_{\mathbf{4}}$ 





 $\mathsf{T}^{Al}_{3}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}^{Si}_{6}$ 



 $\int_{c}^{a} b$ 



 $\downarrow^{b}_{c}$  a















e a b





 ${\rm T}_{5}^{Al}\!\!-\!\!0^{\rm H}\!\!-\!\!{\rm T}_{4}^{Si}$ 



↓ c b

 $\mathtt{T}_5^{Al}\!\!-\!\!0^{\scriptscriptstyle H}\!\!-\!\!\mathtt{T}_1^{Si}$ 





 $T_{7}^{Al} = 0^{H} = T_{11}^{Si}$ 









 $\mathsf{T}_7^{Al}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}_7^{Si}$ 













↓ c b

 ${\rm T}_{\bf 11}^{\bf Al}\!\!-\!\!{\rm O}^{\rm H}\!\!-\!\!{\rm T}_{\bf 10}^{\bf Si}$ 

 $T_{11}^{Al} = 0^{H} = T_{12}^{Si}$ 





 $T_{11}^{Al} = 0^{H} = T_{7}^{Si}$ 





 ${\rm T}_{11}^{Al}\!\!-\!\!0^{\rm H}\!\!-\!\!{\rm T}_{5}^{Si}$ 



Figure S5 The optimized configurations of the adsorption of  $NH_2CH_3$  at the acidic sites of the Al substituted T1, T3, T5, T7, T11 and T12 sites in the framework



 $\mathsf{T}^{\mathbf{Al}}_{\mathbf{1}}\!\!-\!\!\mathsf{O}^{\!\scriptscriptstyle H}\!\!-\!\!\mathsf{T}^{\mathbf{Si}}_{\mathbf{10}}$ 





 $\mathsf{T}_1^{Al}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}_2^{Si}$ 





















 $\stackrel{b}{\underset{c}{\longrightarrow}}$ a

b K a

 $\mathsf{T}^{Al}_{1}\!\!-\!\!\mathsf{O}^{\text{H}}\!\!-\!\!\mathsf{T}^{Si}_{5}$ 

 $\begin{array}{c} b \\ f \\ c \\ c \end{array} a$ 





 $\mathsf{T}^{Al}_{\mathbf{3}}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}^{Si}_{\mathbf{12}}$ 



↓ c b





























 $\mathsf{T}_7^{\mathbf{Al}}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}_4^{\mathbf{Si}}$ 









 $T_{7}^{Al} - 0^{H} - T_{7}^{Si}$ 





 $T_{7}^{Al} = 0^{H} = T_{11}^{Si}$ 





 ${\rm T}_{11}^{Al}\!\!-\!\!0^{\rm H}\!\!-\!\!{\rm T}_{5}^{Si}$ 







 $T_{11}^{Al} = 0^{H} = T_{10}^{Si}$ 



 $T_{11}^{Al} = 0^{H} = T_{12}^{Si}$ 

 $\downarrow^{b}_{c} \stackrel{a}{\longrightarrow} a$ 











 $T_{12}^{Al} - 0^{H} - T_{8}^{Si}$ 





 $\begin{array}{c} b \\ c \\ c \end{array} a$ 







Figure S6 The optimized configurations of the adsorption of  $NH(CH_3)_2$  at the acidic sites of the Al substituted T1, T3, T5, T7, T11 and T12 sites in the framework





 $\mathsf{T}^{Al}_{1}\!\!-\!\!\mathsf{O}^{\text{H}}\!\!-\!\!\mathsf{T}^{Si}_{5}$ 





,⊭rrrrrrrrra a

 $\mathtt{T_1^{Al}}\mathtt{-}\mathtt{O}^{\mathtt{H}}\mathtt{-}\mathtt{T_4^{Si}}$ 

















 $\mathsf{T}^{Al}_{3}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}^{Si}_{4}$ 



 $\mathsf{T}^{Al}_{3}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}^{Si}_{6}$ 



 $\mathsf{T}^{Al}_{3}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}^{Si}_{12}$ 

 $\downarrow^{b}_{c} \stackrel{a}{\longrightarrow} a$ 





 $T_{5}^{Al} - 0^{H} - T_{1}^{Si}$ 



 $\mathsf{T}_5^{Al}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}_4^{Si}$ 





























 $\stackrel{b}{\underset{c}{\longrightarrow}}^{a}$ 









 $\mathsf{T}^{Al}_{5}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}^{Si}_{11}$ 







 $\mathsf{T}^{\mathbf{Al}}_{\mathbf{7}}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!\mathsf{H}}\!\!-\!\!\mathsf{T}^{\mathbf{Si}}_{\mathbf{7}}$ 



 $T_7^{Al} - 0^H - T_8^{Si}$ 





 $\mathsf{T}^{Al}_{7}\!\!-\!\!0^{\scriptscriptstyle H}\!\!-\!\!\mathsf{T}^{Si}_{11}$ 

 $\overset{b}{\underset{c}{\overset{b}{\longrightarrow}}}^{a}$ 



 $\mathsf{T}_7^{Al}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!H}\!\!-\!\!\mathsf{T}_4^{Si}$ 













 $T_{11}^{Al} = 0^{H} = T_{10}^{Si}$ 

 $\mathsf{T}_{\mathtt{1}\mathtt{1}}^{\mathtt{Al}}\!\!-\!\!\mathsf{O}^{\mathtt{H}}\!\!-\!\!\mathsf{T}_{\mathtt{7}}^{\mathtt{Si}}$ 





 $\mathsf{T}^{\mathbf{Al}}_{\mathbf{11}}\!\!-\!\!\mathsf{O}^{\scriptscriptstyle\!\mathsf{H}}\!\!-\!\!\mathsf{T}^{\mathbf{Si}}_{\mathbf{12}}$ 





Figure S7 The optimized configurations of the adsorption of  $N(CH_3)_3$  at the acidic sites of the Al substituted T1, T3, T5, T7, T11 and T12 sites in the framework



 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 





 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{2}$ 





 $\mathsf{T}_1^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_4^{\text{Si}}$ 





 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{5}$ 





















 $\mathsf{T}_3^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 



 $\mathsf{T}^{Al}_{3}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{2}$ 









 $\mathsf{T}_5^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_1^{Si}$ 



 $\mathsf{T}_5^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 



 $\mathsf{T}_5^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_6^{\text{Si}}$ 



 $\mathsf{T}^{Al}_{5}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 









 $\Gamma_7^{Al}$ 











c a b

¢ c





 $\mathsf{T}_7^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_4^{\text{Si}}$ 















 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{7}$ 



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{5}$ 









 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{8}$ 



 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 

 $a \rightarrow a$ 





Figure S8 The optimized configurations of benzene adsorption at the acidic sites of the AI substituted T1, T3, T5, T7, T11 and T12 sites in the framework



a c c



 $\mathsf{T}^{Al}_1\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_2$ 

 $\mathsf{T}_1^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 

 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 











 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{5}$ 







b ↓↓→a

 $\mathsf{T}_3^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 









 $\mathsf{T}^{Al}_{3}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 



c a b

¢ c



¢ c





K<sub>T</sub>Al S<sup>3</sup> K<sub>T</sub>Al S<sup>3</sup>











 $\mathsf{T}_5^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_{11}^{\text{Si}}$ 






 $F_{T_{7}^{\text{Si}}}$ 

 $\mathsf{T}_7^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_7^{\text{Si}}$ 





 $\mathsf{T}^{Al}_{\phantom{Al}7}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{\phantom{B}8}$ 



¢ c























T<sub>7</sub> T<sub>11</sub>

c a b

c a b

c a b



 $\mathsf{T}^{\mathrm{Al}}_{11} \text{-} \mathsf{O}^{\mathsf{H}} \text{-} \mathsf{T}^{\mathrm{Si}}_{7}$ 





 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{3}$ 

 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{8}$ 









 $T_{12}^{Al}$ 





b a√c

b a√c

T<sup>A</sup><sub>12</sub> T<sup>A</sup><sub>12</sub>

 $T_{12}^{Al}$ -0<sup>H</sup>- $T_{12}^{Si}$ 



Figure S9 The optimized configurations of toluene adsorption at the acidic sites of the Al substituted T1, T3, T5, T7, T11 and T12 sites in the framework



 $\mathsf{T}^{\mathrm{Al}}_1 \text{-} \mathsf{O}^{\mathsf{H}} \cdot \mathsf{T}^{\mathrm{Si}}_{10}$ 



 $\mathsf{T}^{Al}_1\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_2$ 







 $\mathsf{T}^{Al}_{1}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{4}$ 





 $\mathsf{T}^{Al}_1\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_5$ 







 $\mathsf{T}^{Al}_{3}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 





 $\mathsf{T}_3^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_6^{\text{Si}}$ 







¢ c

¢<sup>a</sup>b

¢<sup>a</sup>b

 $\mathsf{T}_3^{Al}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}_4^{Si}$ 









 $\mathsf{T}_5^{\text{Al}}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}_4^{\text{Si}}$ 





 $\mathsf{T}_5^{\mathrm{Al}}$ - $\mathsf{O}^{\mathsf{H}}$ - $\mathsf{T}_6^{\mathrm{Si}}$ 





¢ c

¢ c



 $\mathsf{T}^{Al}_{5}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 









 $\int_{c}^{a} b$ 

¢ c

¢<sup>a</sup>b

¢<sup>a</sup>b



 $\mathsf{T}_7^{\mathrm{Al}}$ - $\mathsf{O}^{\mathsf{H}}$ - $\mathsf{T}_7^{\mathrm{Si}}$ 





 $\mathsf{T}^{\mathrm{Al}}_{7}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{\mathrm{Si}}_{8}$ 





 $\mathsf{T}^{\text{Al}}_{7}\text{-}\mathsf{O}^{\text{H}}\text{-}\mathsf{T}^{\text{Si}}_{11}$ 















 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{10}$ 



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{7}$ 

b ↓→a



 $\mathsf{T}^{Al}_{11}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{5}$ 





 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{3}$ 



 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{8}$ 



 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{11}$ 





 $\mathsf{T}^{Al}_{12}\text{-}\mathsf{O}^{\mathsf{H}}\text{-}\mathsf{T}^{Si}_{12}$ 

*↓*→a



Figure S10 The optimized configurations of p-xylene adsorption at the acidic sites of the Al substituted T1, T3, T5, T7, T11 and T12 sites in the framework