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Supplementary information for

Multiscale Investigation Elucidating the Structural Complexities and Electrochemical Properties of Layered-Layered Composite Cathode Materials Synthesized at Low Temperatures

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Supplementary information



Figure S1. Schematic diagram of the preparation procedures of the layered-layered composite materials.



Figure S2. XRD patterns of the HR5AC, HR2AC, and HR5SC samples.

As presented in Fig. S2, the final XRD patterns of the HR5AC, HR2AC, and HR5SC samples prepared using various firing schedules could be indexed to both of the Li₂MnO₃ (C2/m) and LiCoO₂ ($R\bar{3}m$) phases with quite broad XRD peaks. Moreover, a Co₃O₄ phase was also observed and their broad XRD patterns indicate that the samples exhibited low crystallinity. This results from a low annealing temperature (600 \mathbb{C} C). Figure 1 presents HRTEM results of the HR5AC, HR2AC and HR5SC samples. The results revealed that both the Li₂MnO₃-like domain with a C2/m space group and the LiCoO₂-like domain with an $R\bar{3}m$ space group were clearly observed coexisting within the same particles. The results indicate that the HR5AC, HR2AC, and HR5SC samples formed a LMRtype material with low crystallinity.



Figure S3. Firing schedule of each heat treatment condition.



Figure S4. XRD patterns of the HR5AC (a), HR2AC (b), and HR5SC (c) samples measured during heating processes with various heating and cooling rates.



Figure S5. Examples of Rietveld refinement results of the HR5AC (a), HR2AC (b), and HR5SC (c) samples.

The prepared samples contained Co and Mn, contributing to the high background signals of the diffraction patterns owing to the fluorescent effect. This led to rather low R factors. To improve these R factors, 80% of the background signals were removed without a significant effect on the refined lattice parameter values.

Table S1. The lattice parameters and agreement indices obtained from Reitveld refinement of XRDpatterns of HR5AC, HR2AC, and HR5SC samples.

Conditions	Lattice parameters										
		Li	₂MnO₃ (C2/n	LiCoO ₂ (R3m)							
			$\alpha = \gamma = 90^{\circ}$	α = β = $~90^\circ$ and γ = 120°							
	a (Å)	b (Å)	c (Å)	β(°)	Unit cell	a and b	C (Å)	Unit cell			
					volume	(Å)		volume			
					(A ³)	. ,		(A ³)			
HR5AC	4.9295	8.5569	4.9529	109.5359	197.1429	2.8259	14.0811	97.3829			
	± 0.0001	± 0.0002	± 0.0009	± 0.0050		± 0.0004	±0.0021				
HR2AC	4.9234	8.5383	4.9390	109.8551	197.2621	2.8281	14.0681	97.4375			
	± 0.0006	± 0.0003	± 0.0008	± 0.0011		± 0.0005	± 0.0050				
HR2SC	5.0261	8.4261	4.9453	109.4681	197.4593	2.8243	14.0741	97.2034			
	± 0.0005	± 0.0005	± 0.0006	± 0.0004		± 0.0004	± 0.0003				



Figure S6. XAS spectra of the HR5AC (a), HR2AC (b), and HR5SC (c) samples measured during heating processes with various heating and cooling rates.



Figure S7. EXAFS fitting of the HR5AC (a), HR2AC (b), and HR5SC (c) samples synthesized using various heating and cooling rates.

Table S2. The bond distances and agreement indices obtained from XAS spectra fitting of finalXAS spectra of the HR5AC, HR2AC, and HR5SC samples.

		Li ₂ M	nO₃ structure (M	n K- edge)		
Samples	Shell (atoms)		R (Å)	σ² (Ų)	R- factor	E ₀ (eV)
HR5AC	0	6	1.900	0.003		-4.052
			±0.015	± 0.001	0.013	
	Mn	3	2.866	0.004		
			±0.021	± 0.002		
HR2AC	0	6 3	1.909	0.003		-3.091
			±0.018	± 0.001	0.020	
	Mn		2.869	2.869	0.020	
			± 0.025	± 0.002		
HR5SC	0	6	1.912	0.004		-3.029
			±0.021	± 0.001	0.020	
	Mn	3	2.866	0.004	0.026	
			± 0.028	± 0.002		
		LiC	oO2 structure (Co	K-edge)		
Samples	Shell	N (atoms)	R (Å)	σ² (Ų)	R- factor	E ₀ (eV)
HR5AC	0	6	1.901	0.003		9.434
			± 0.022	± 0.002	0.012	
	Co	6	2.832	0.005	0.012	
			±0.02	± 0.002		
HR2AC	0	6	1.911	0.005		9.427
			± 0.027	± 0.003	0.016	
	Со	6	2.837	0.004	0.010	
			± 0.027	± 0.002		
HR5SC	0	6	1.909	0.004		-3.359
			±0.025	± 0.002	0.012	
	Со	6	2.851	0.006	0.013	
			10.020	± 0.002	I	



Figure S8. Cycling stabilities at C/10 (a) and rate performance (b) of the HR5AC, HR2AC, and HR5SC samples.