Electronic Supplementary Material (ESI) for Catalysis Science & Technology. This journal is © The Royal Society of Chemistry 2019

1	Supporting Information
2	The synergetic mechanism of NO _x and chlorobenzene
3	degradation in municipal solid waste incinerators
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16 SI figure captions

- 17 Figure S1. The effect of CB on N₂ selectivity and the effect of SCR gas on CO₂
- 18 selectivity for 10-h test at 300 °C. Reaction conditions: NO 500 ppm (when used), CB
- 19 50 ppm (when used), NH_3 500 ppm (when used), O_2 10 vol.%, N_2 as the balance gas,
- 20 GHSV 60,000 mL/(g·h).
- 21 Figure S2. The NO_x conversion of CB-SCR of MnO_x-CeO₂ catalyst at 300 °C.
- 22 Reaction conditions: NO 500 ppm, CB 50 ppm (when used), NH₃ 500 ppm (when used),
- 23 O₂ 10 vol.%, N₂ as the balance gas, GHSV 60,000 mL/(g·h).
- 24 Figure S3. The concentrations of CB, CO, CO₂, and N₂O, NO, NO₂ during the SCR-
- 25 transient reaction with CB at 300 °C. Reaction conditions: NO 500 ppm, CB 50 ppm,
- 26 NH₃ 500 ppm, O₂ 10 vol.%, N₂ as the balance gas, GHSV 60,000 mL/(g·h).
- 27 Figure S4. The concentrations of N₂O, NO, NO₂ and CB, CO₂ during the SCR reactions
- 28 without/with CB and CB oxidation reactions with/without SCR flue gas at 300 °C.
- 29 Reaction conditions: NO 500 ppm (when used), CB 50 ppm (when used), NH₃ 500 ppm
- 30 (when used), O₂ 10 vol.%, N₂ as the balance gas, GHSV 60,000 mL/(g·h).
- 31 Figure S5. XPS spectra: (a) Mn 2p, and (b) Ce 3d of the fresh and used catalysts.





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Figure S3. The concentrations of CB, CO, CO₂, and N₂O, NO, NO₂ during the SCRtransient reaction with CB at 300 °C. Reaction conditions: NO 500 ppm, CB 50 ppm,
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51 reactions without/with CB and CB oxidation reactions with/without SCR flue gas at

52 300 °C. Reaction conditions: NO 500 ppm (when used), CB 50 ppm (when used),

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57 Figure S5. XPS spectra: (a) Mn 2p, and (b) Ce 3d of the fresh and used catalysts.

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