

Supporting Information for the manuscript:

Amino Functionalized Zn/Cd- Metal–Organic Frameworks for Selective CO₂ Adsorption and Knoevenagel Condensation Reaction

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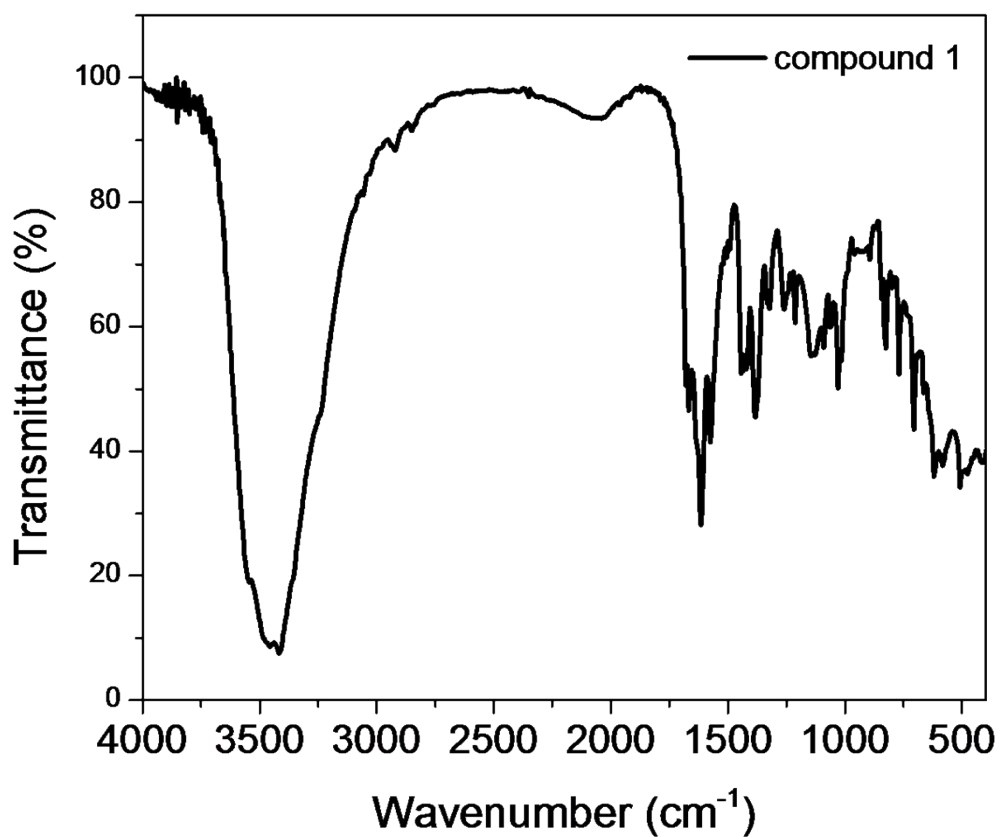


Figure S1. The IR spectra of 1.

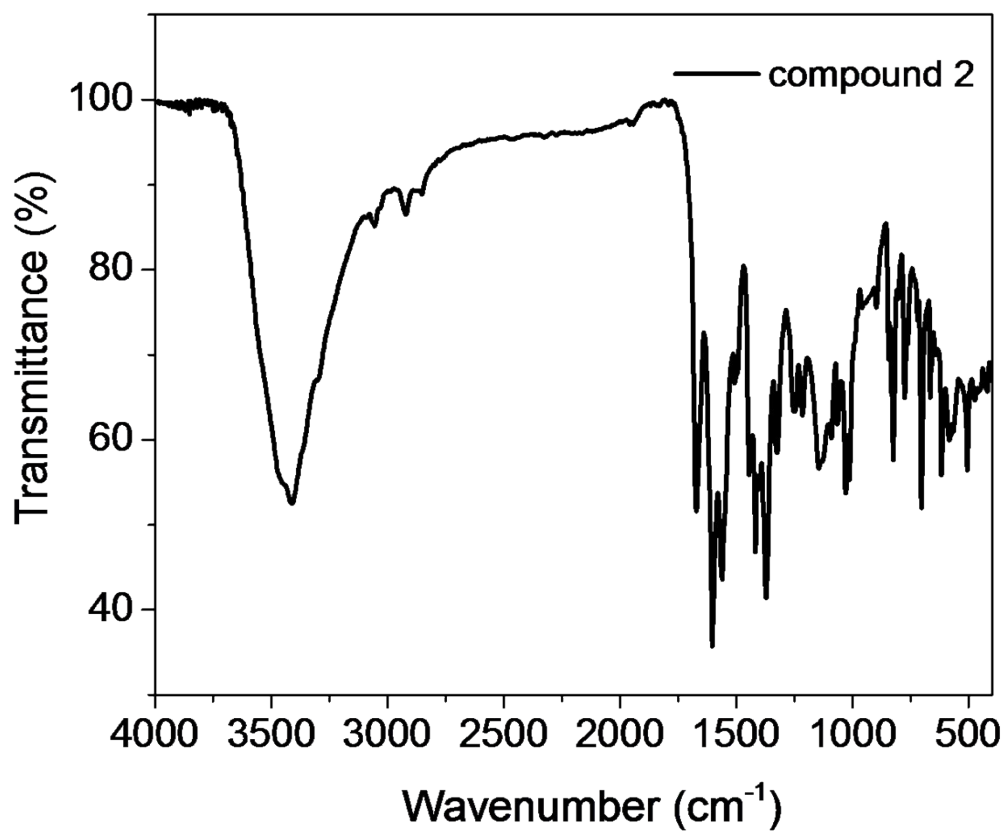


Figure S2. The IR spectra of 2.

Table S1 Crystallographic data for **1** and **2**

Compounds	1	2
Formula	ZnC ₂₅ H ₂₀ N ₆ O ₅ S ₂	CdC ₂₅ H ₂₁ N ₆ O _{5.5} S ₂
Formula weight	613.96	670.01
Temperature (K)	150	150
Crystal system	orthorhombic	monoclinic
Space group	<i>Pccn</i>	<i>P2₁/n</i>
<i>a</i> (Å)	18.5564(2)	13.56520(10)
<i>b</i> (Å)	17.2237(2)	14.40230(10)
<i>c</i> (Å)	16.5743(2)	13.93520(10)
α (°)	90	90
β (°)	90	100.8040(10)
γ (°)	90	90
<i>V</i> (Å ³)	5297.31(11)	2674.26(3)
<i>Z</i>	8	4
<i>D</i> _{calc} (g cm ⁻³)	1.540	1.664
<i>F</i> (000)	2512.0	1168.0
Reflections Collected	18616	14867
Independent Reflections	5234	5251
observed data [<i>I</i> > 2σ(<i>I</i>)]	4601	4722
<i>R</i> _{int}	0.0311	0.0375
GOF on <i>F</i> ²	1.043	1.097
<i>R</i> ₁ ^a , <i>wR</i> ₂ ^b [<i>I</i> > 2σ(<i>I</i>)]	0.0332, 0.0842	0.0397, 0.1120
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0387, 0.0870	0.0428, 0.1140

$${}^a R_1 = \frac{\sum \|F_o\| - |F_c|}{\sum |F_o|}, {}^b wR_2 = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

Table S2. Selected bond lengths (Å) and angles (°) for **1** and **2**.

1			
Zn(1)–O(1)	2.0003(14)	Zn(1)–O(2)#2	2.0352(14)
Zn(1)–O(3)#1	2.0907(15)	Zn(1)–O(4)#1	2.3179(16)
Zn(1)–N(1)	2.1512(17)	Zn(1)–N(2)#3	2.1893(18)
O(2)–Zn(1)#2	2.0351(14)	O(3)–Zn(1)#4	2.0906(15)
O(4)–Zn(1)#4	2.3179(16)	N(2)–Zn(1)#6	2.1893(18)
O(1)–Zn(1)–O(2)#1	120.73(6)	O(1)–Zn(1)–O(3)#2	151.72(6)
O(1)–Zn(1)–O(4)#2	92.31(6)	O(1)–Zn(1)–N(1)	88.37(6)
O(1)–Zn(1)–N(2)#3	91.35(6)	O(2)#1–Zn(1)–O(3)#2	87.55(6)
O(2)#1–Zn(1)–O(4)#2	146.63(6)	O(2)#1–Zn(1)–N(1)	87.88(6)
O(2)#1–Zn(1)–N(2)#3	92.37(6)	O(3)#2–Zn(1)–O(4)#2	59.46(6)
O(3)#2–Zn(1)–N(1)	92.32(6)	O(3)#2–Zn(1)–N(2)#3	87.87(6)
N(1)–Zn(1)–O(4)#2	88.53(6)	N(1)–Zn(1)–N(2)#3	179.70(7)
N(2)#3–Zn(1)–O(4)#2	91.36(6)		
2			
Cd(1)–O(1)	2.242(2)	Cd(1)–O(2)#2	2.291(2)
Cd(1)–O(3)#1	2.339(2)	Cd(1)–O(4)#1	2.409(2)
Cd(1)–N(1)	2.321(3)	Cd(1)–N(2)#3	2.340(3)
O(2)–Cd(1)#2	2.291(2)	O(3)–Cd(1)#4	2.339(2)
O(4)–Cd(1)#4	2.409(2)	N(2)–Cd(1)#5	2.339(3)
O(1)–Cd(1)–O(2)#2	123.34(8)	O(1)–Cd(1)–O(3)#1	149.18(8)
O(1)–Cd(1)–O(4)#1	96.73(8)	O(1)–Cd(1)–N(1)	97.86(9)
O(1)–Cd(1)–N(2)#3	85.62(9)	O(2)#2–Cd(1)–O(3)#1	85.87(8)
O(2)#2–Cd(1)–O(4)#1	139.92(8)	O(2)#2–Cd(1)–N(1)	86.03(9)
O(2)#2–Cd(1)–N(2)#3	84.31(9)	O(3)#1–Cd(1)–O(4)#1	54.93(8)
O(3)#1–Cd(1)–N(2)#3	88.15(11)	N(1)–Cd(1)–O(3)#1	93.40(11)
N(1)–Cd(1)–O(4)#1	88.13(10)	N(1)–Cd(1)–N(2)#3	170.09(11)
N(2)#3–Cd(1)–O(4)#1	100.73(10)		

Symmetry code for **1**: #1. $1/2 + x, -y, 1/2 - z$; #2. $1 - x, -y, 1 - z$; #3. $+x, -1 + y, +z$; #4. $-1/2 + x, -y, 1/2 - z$; #5. $3/2 - x, 1/2 - y, +z$; #6. $+x, 1 + y, +z$; #7. $1/2 - x, 1/2 - y, +z$. Symmetry code for **2**: #2. $2 - x, 1 - y, 2 - z$; #1. $1/2 + x, 3/2 - y, -1/2 + z$; #3. $1 + x, +y, 1 + z$; #4. $-1/2 + x, 3/2 - y, 1/2 + z$; #5. $-1 + x, +y, -1 + z$.

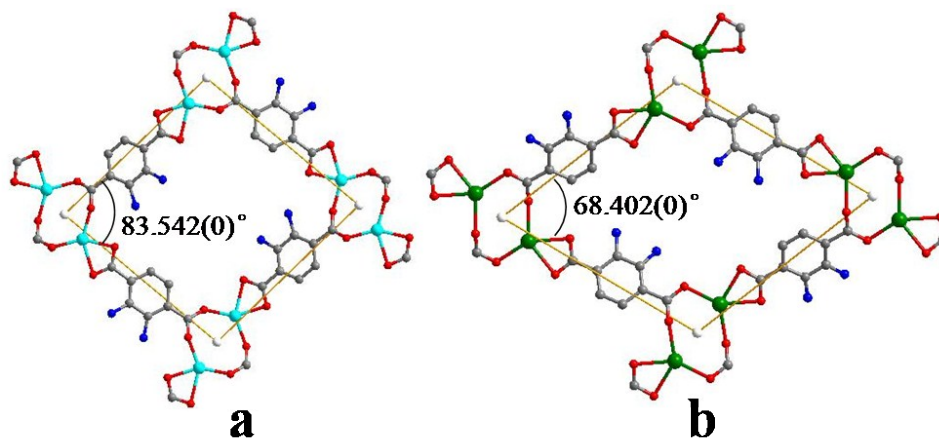


Figure S3. The rhombus-shaped grids in 2D sheet layers for **1** and **2**.

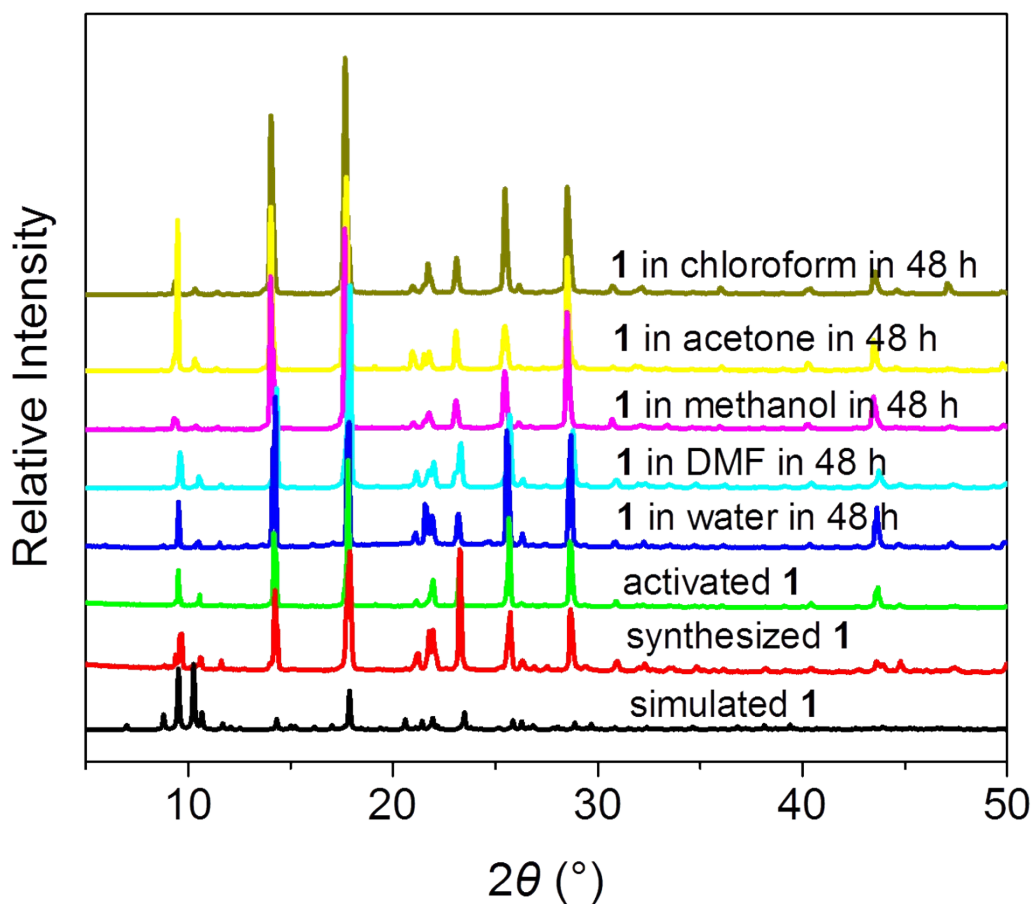


Figure S4. PXRD patterns of **1**.

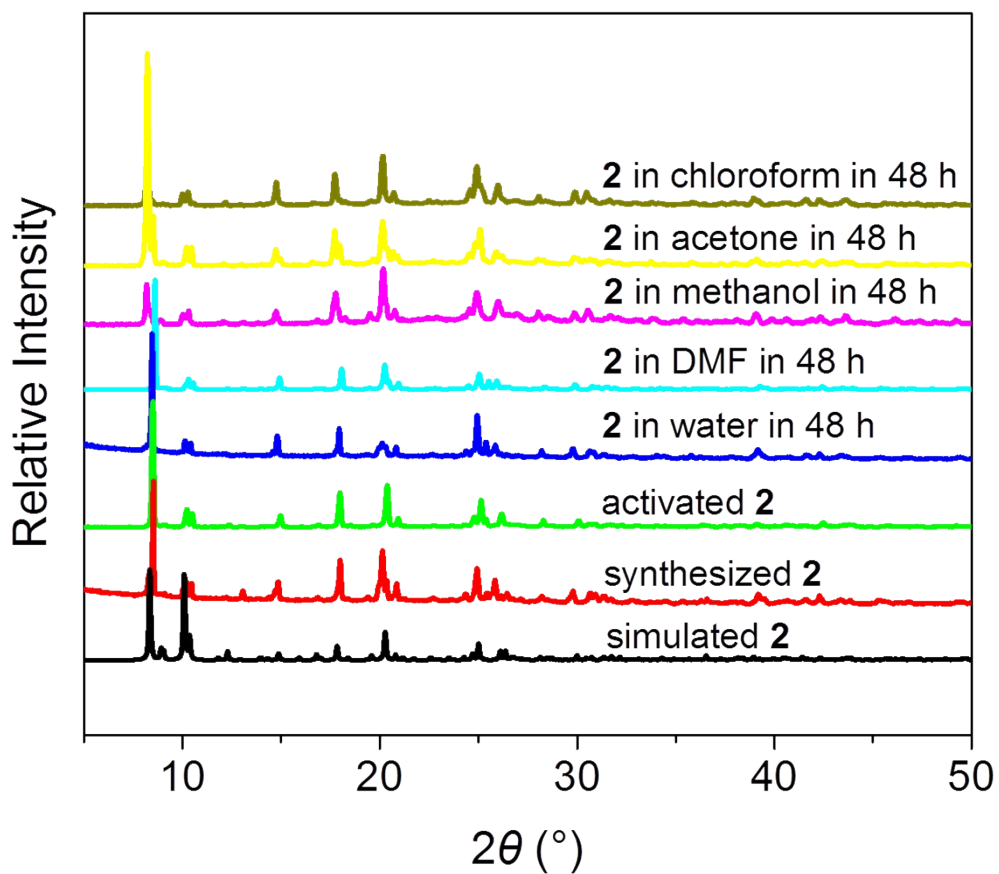


Figure S5. PXRD patterns of 2.

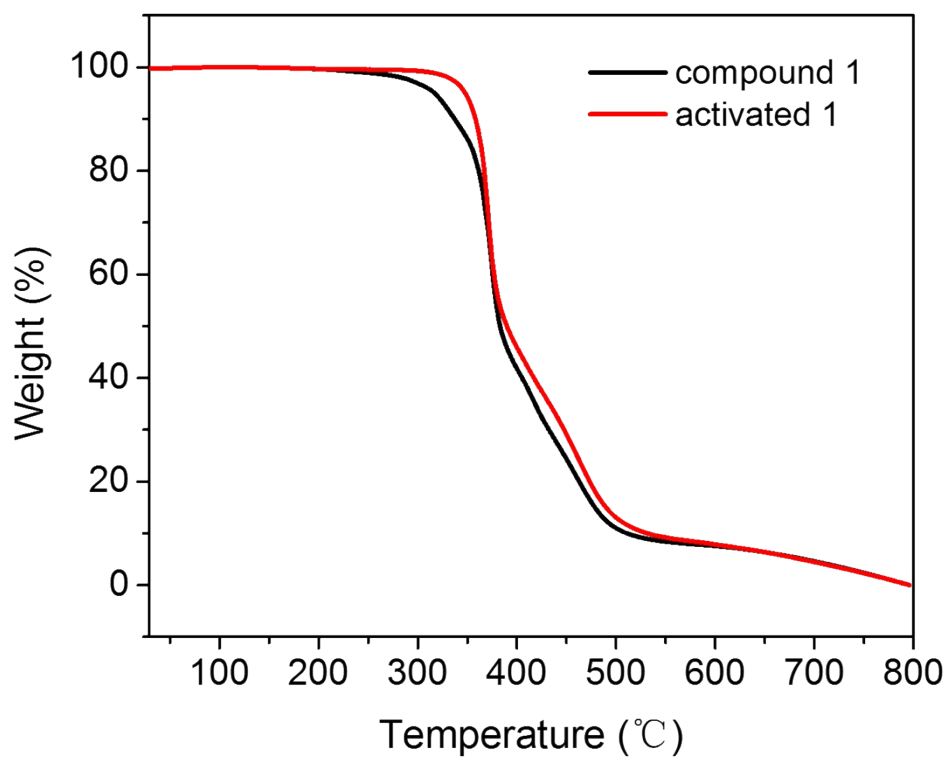


Figure S6. Thermogravimetric analyses plots of synthesized 1 and activated 1.

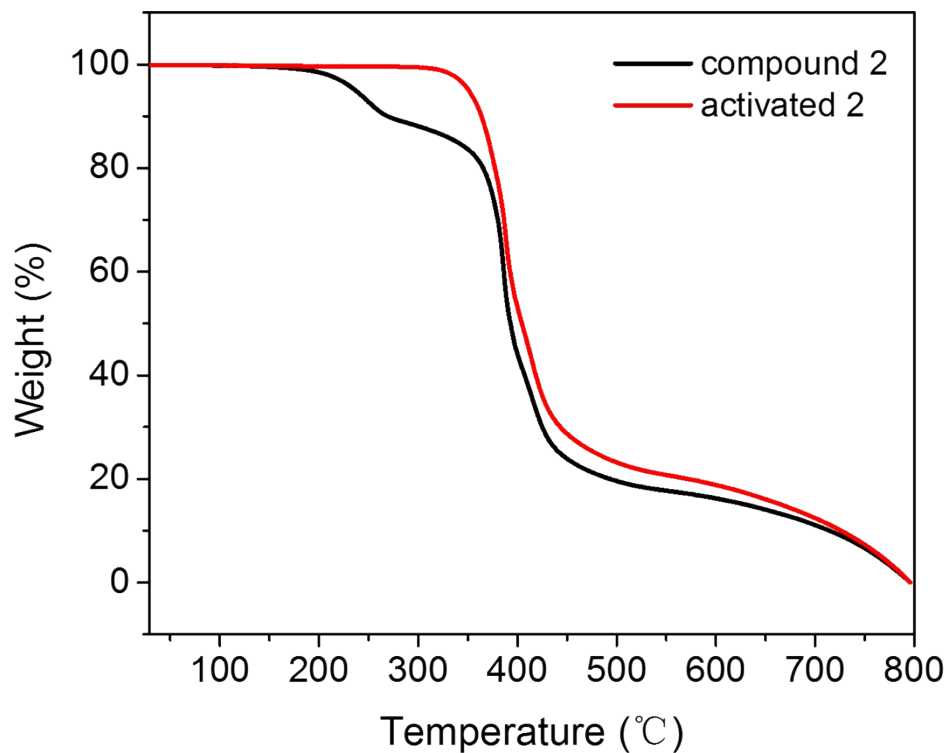


Figure S7. Thermogravimetric analyses plots of synthesized **2** and activated **2**.

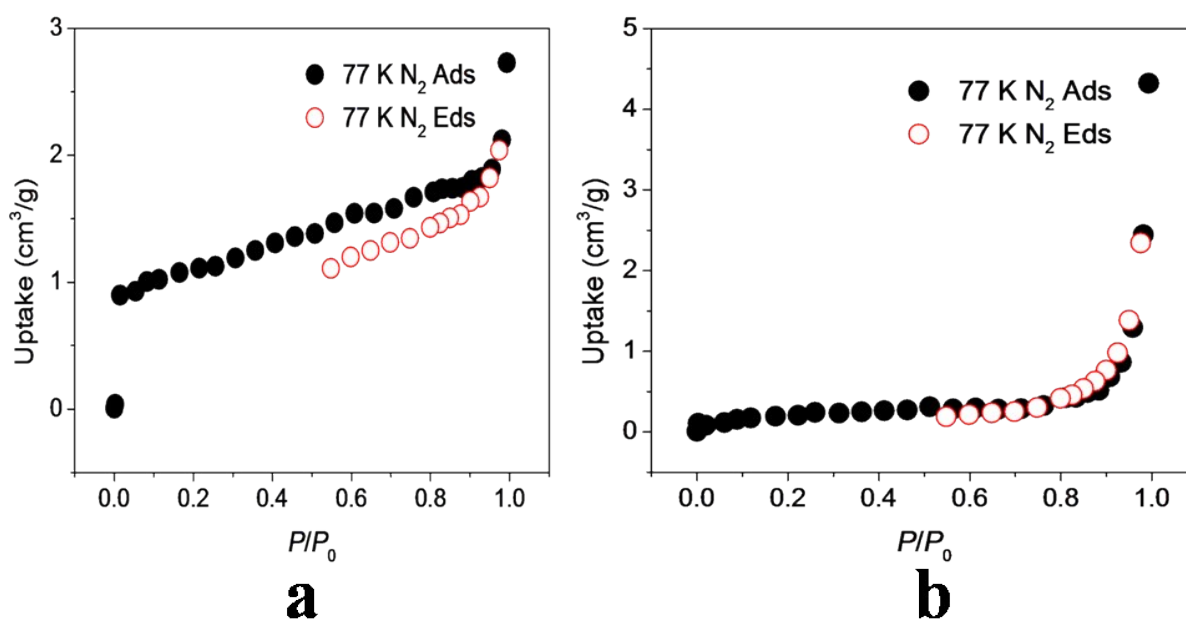


Figure S8. (a) N_2 adsorption isotherms at 77 K for **1**. (b) N_2 adsorption isotherms at 77 K for **2**. Filled and open symbols represent adsorption and desorption, respectively.

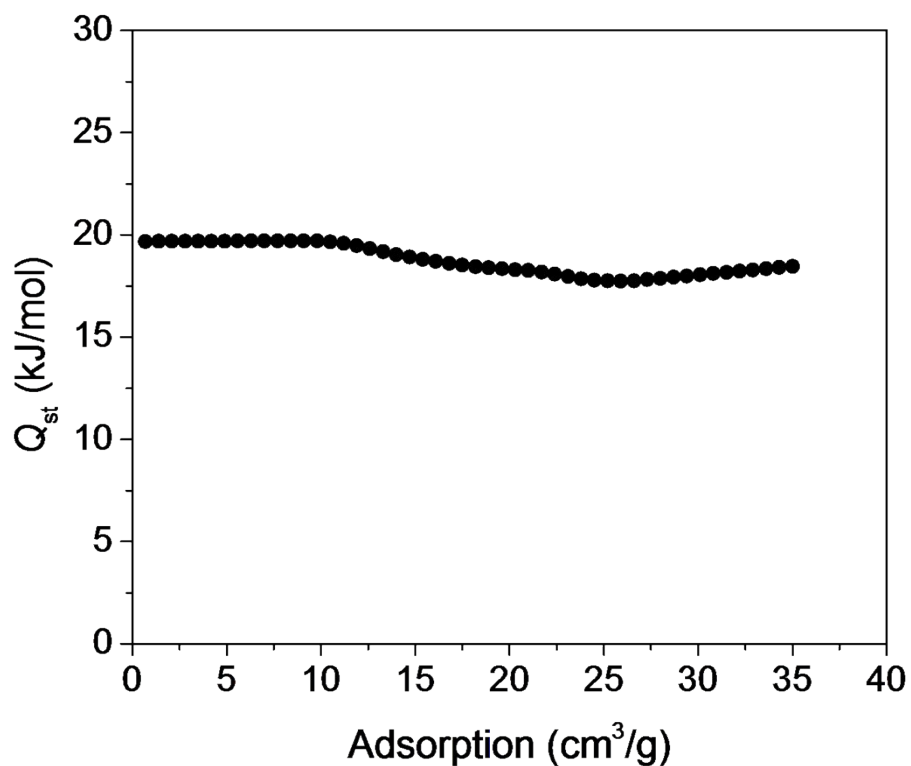


Figure S9. Isosteric heat of adsorption versus CO₂ loading for 2.

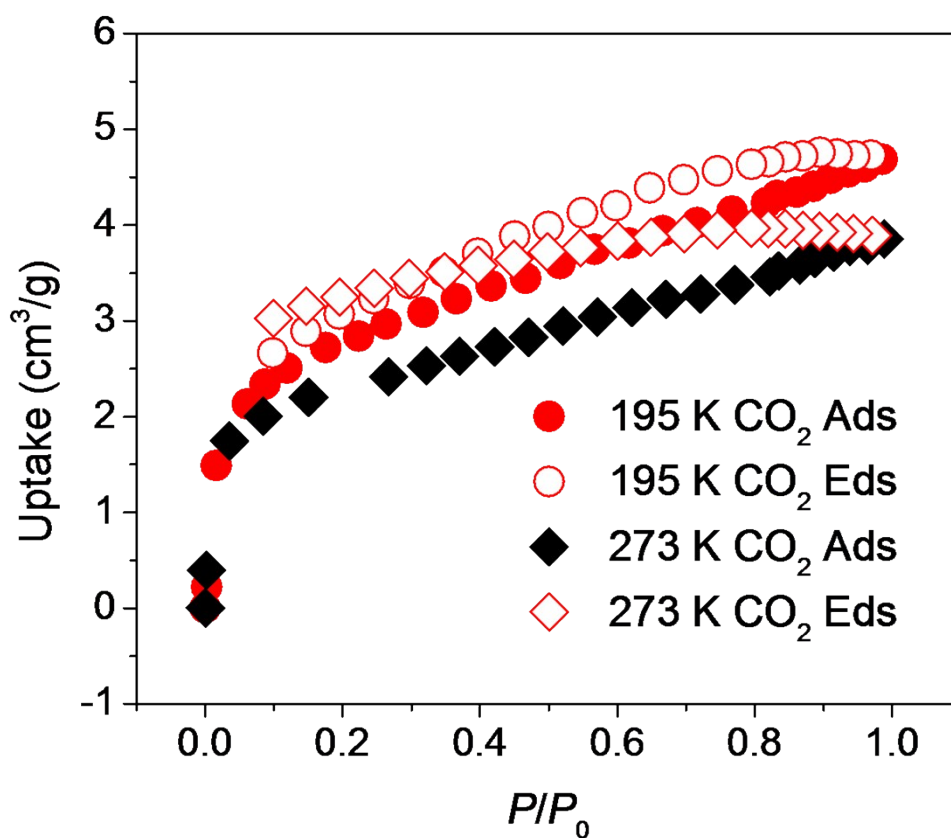


Figure S10. CO₂ adsorption isotherms at 195 K and 273 K for 1. Filled and open symbols represent adsorption and desorption, respectively.

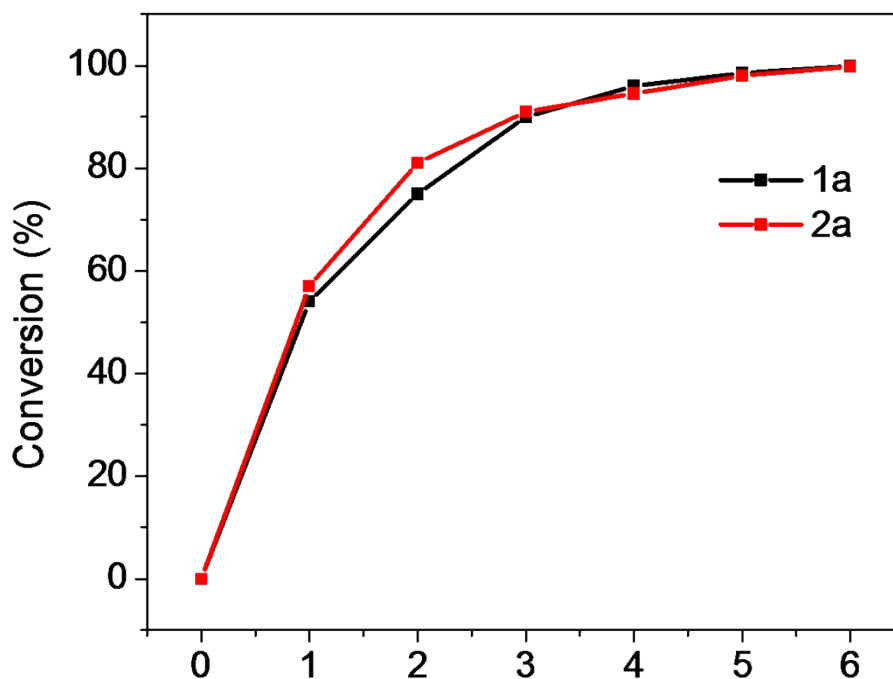


Figure S11. Reaction conversion versus reaction time for Knoevenagel condensation reaction of benzaldehyde and malononitrile with **1a** and **2a** as catalyst.

Table S4. Knoevenagel condensation reactions of benzaldehyde catalyzed by different catalysts.

Entry	Catalyst	Yield (%)
1	Catalyst-free	0
2	2-NH ₂ -H ₂ BDC	1
3	Py ₂ TTz	72
4	2-NH ₂ -H ₂ BDC + Py ₂ TTz	54
5	Ph ₂ TTz	48
6	1a	99.9
7	2a	99.8

Table S5. Recyclable experiments of **1a** and **2a** as catalyst for Knoevenagel reaction.

Entry	Yield (% 1a as catalyst)	Yield (% 2a as catalyst)
Round 1	99.9	99.8
Round 2	98.5	98.1
Round 3	97.8	98.3
Round 4	97.3	97.6
Round 5	96.7	97.3

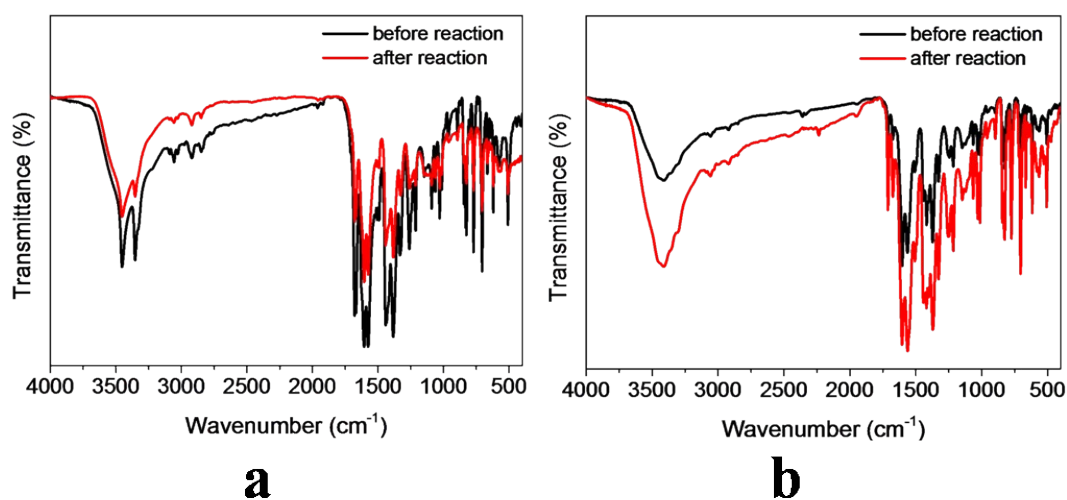


Figure S12. (a) IR spectra of **1a** before and after five round catalytic experiments. (b) IR spectra of **2a** before and after five round catalytic experiments.

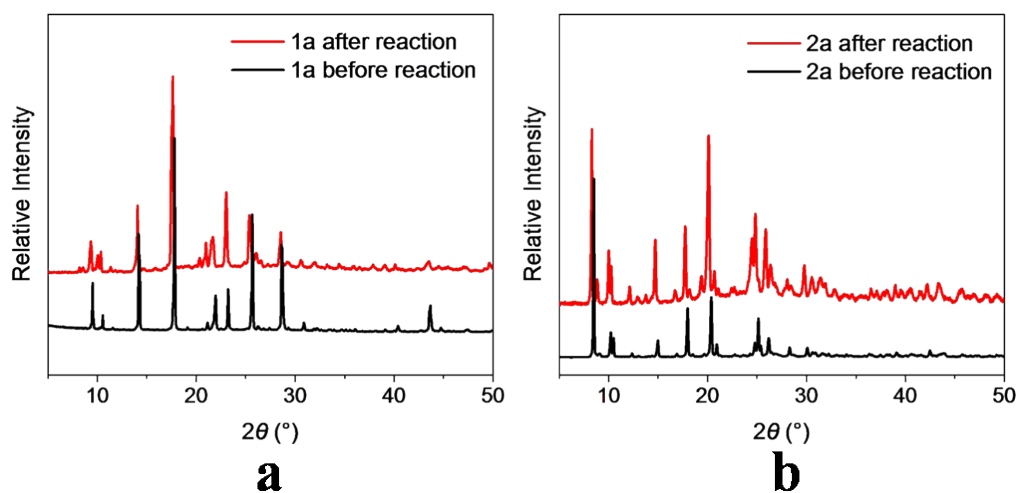


Figure S13. (a) PXRD patterns of **1a** before and after five round catalytic experiments. (b) PXRD patterns of **2a** before and after five round catalytic experiments.

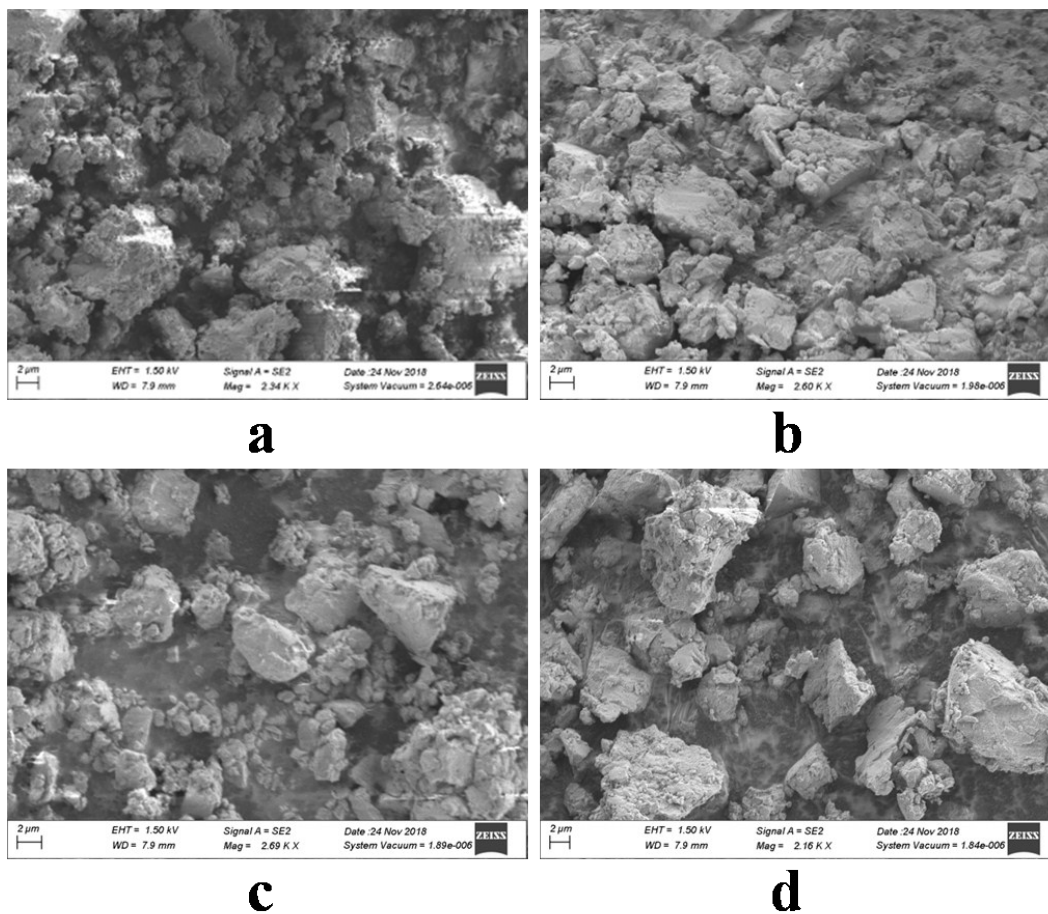


Figure S14. (a) SEM images of **1a** before catalytic experiment. (b) SEM images of **1a** after five round catalytic experiments. (c) SEM images of **2a** before catalytic experiment. (d) SEM images of **2a** after five round catalytic experiments.