

## Supporting Information

### **Organic Cations Directed One-Dimensional Cuprous Halide Compounds: Syntheses, Crystal Structures and Photoluminescent Properties**

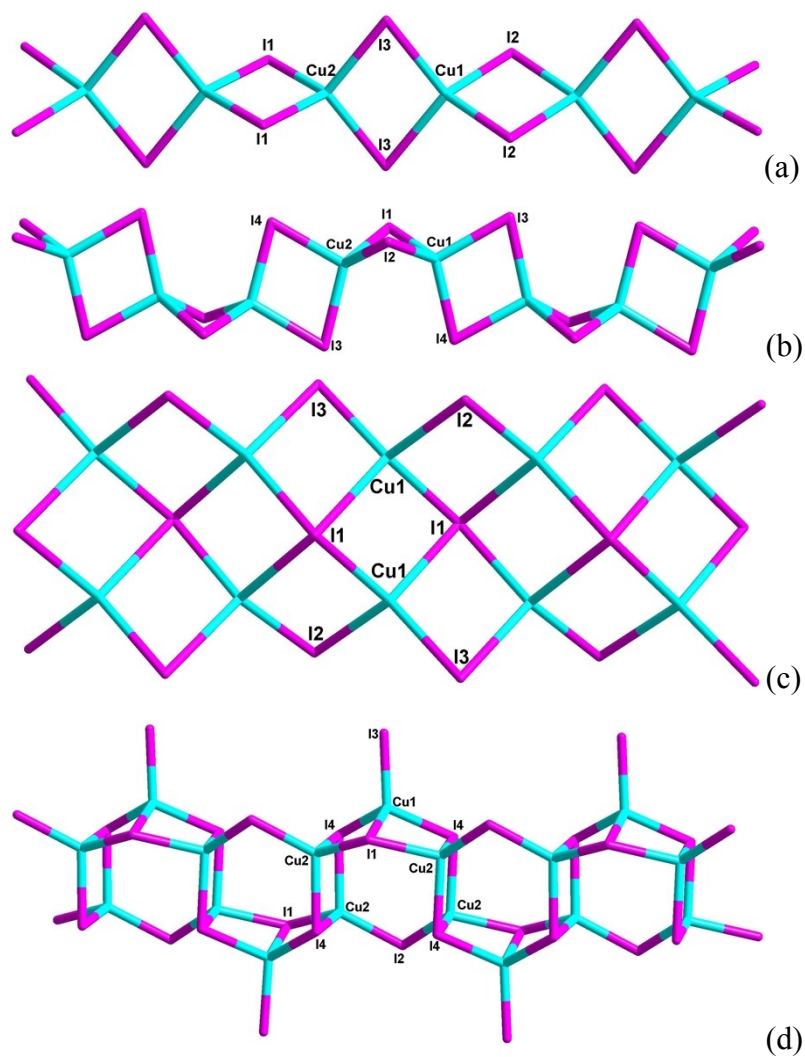
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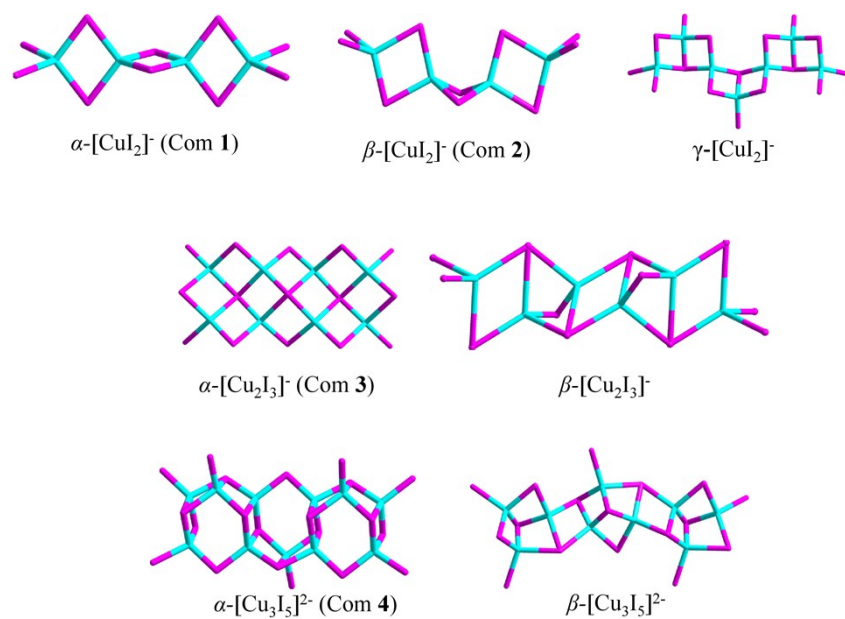
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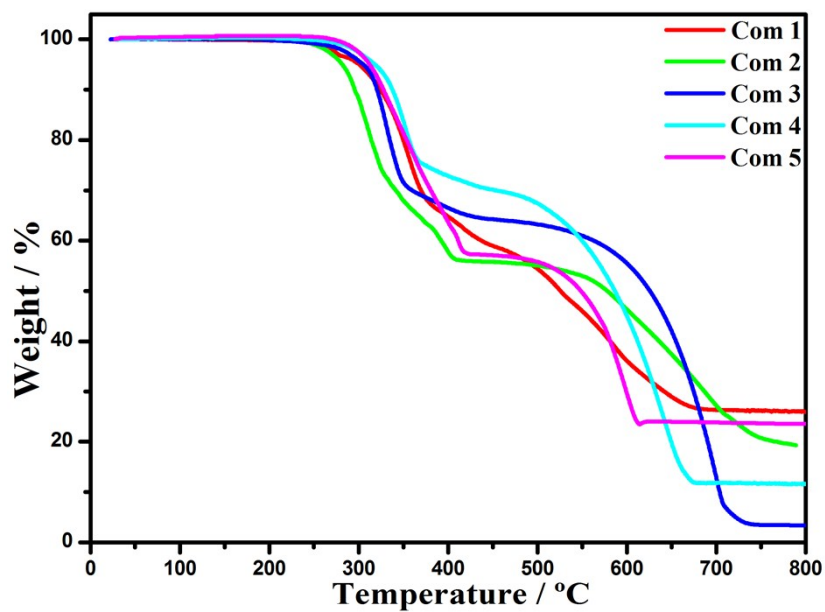
*E-mail address:* hubing@fjirsm.ac.cn; xwlei\_jnu@163.com



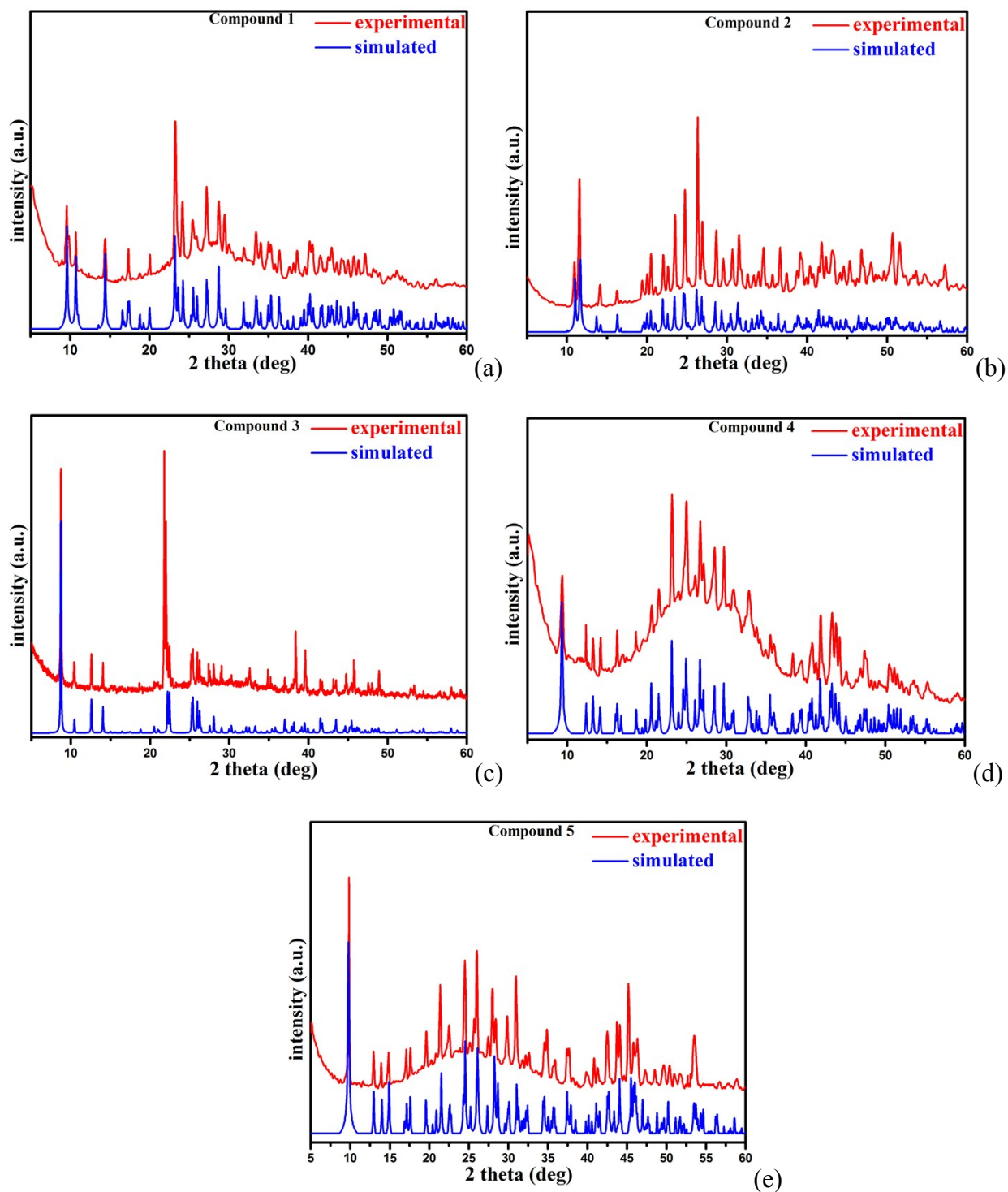
**Fig. S1.** Detailed view of the 1D  $[\text{CuI}_2]^-$  chain in compound **1** (a), 1D  $[\text{Cu}_2\text{I}_4]^{2-}$  chain in compound **2** (b), 1D  $[\text{Cu}_2\text{I}_3]^-$  chain in compound **3** (c) and 1D  $[\text{Cu}_3\text{I}_5]^{2-}$  chain in compound **4** (d).



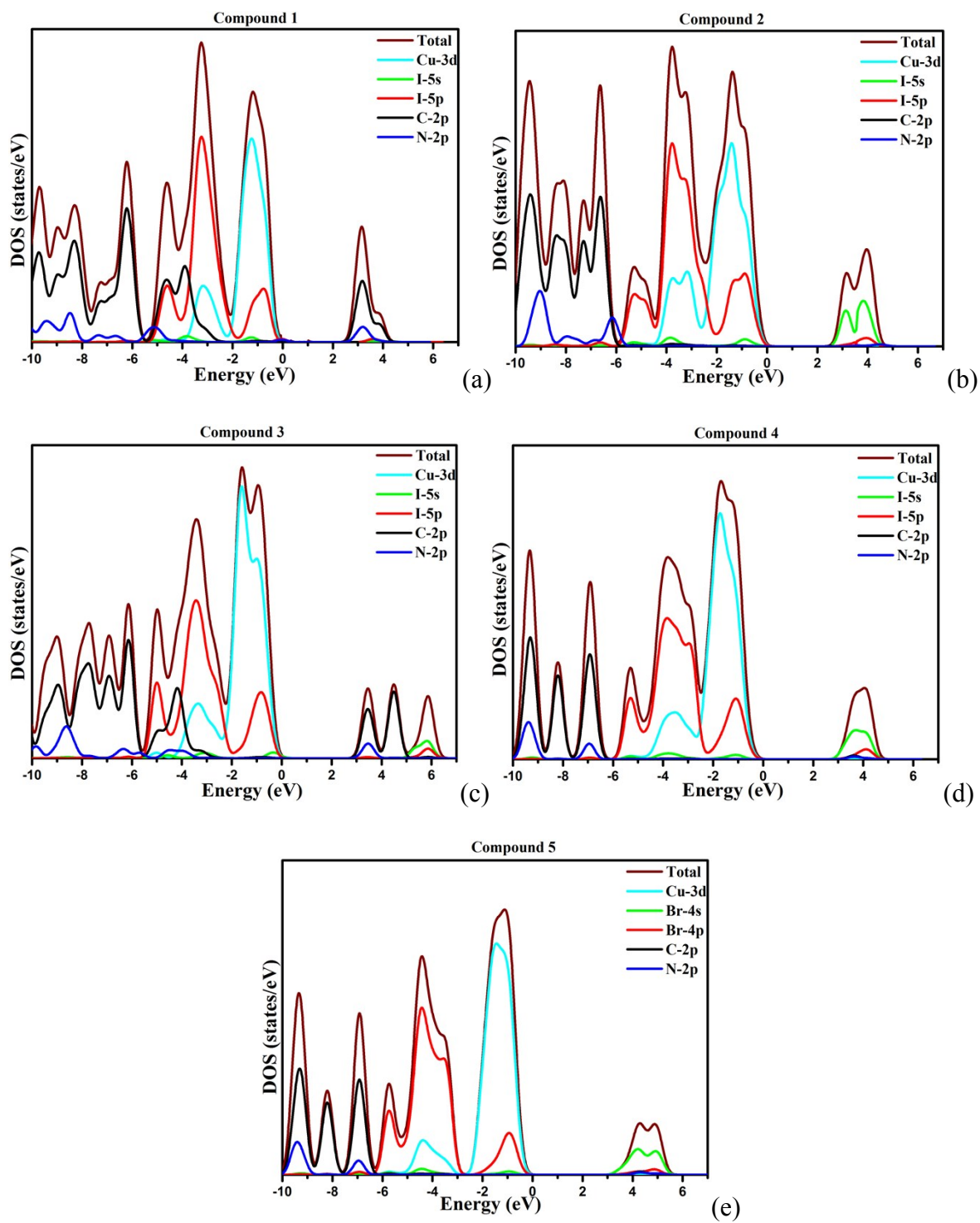
**Fig. S2.** Variation of the partial structural motifs of 1D [CuI<sub>2</sub>]<sup>-</sup>, [Cu<sub>2</sub>I<sub>3</sub>]<sup>-</sup> and [Cu<sub>3</sub>I<sub>5</sub>]<sup>2-</sup> chains based on [CuI<sub>4</sub>] tetrahedrons.



**Fig. S3.** The thermogravimetric analysis curves for compounds 1–5 under nitrogen atmosphere.



**Fig. S4.** Experimental and simulated XRD patterns of compounds 1-5.



**Fig. S5.** The calculated total and partial density of states for compounds 1-5.

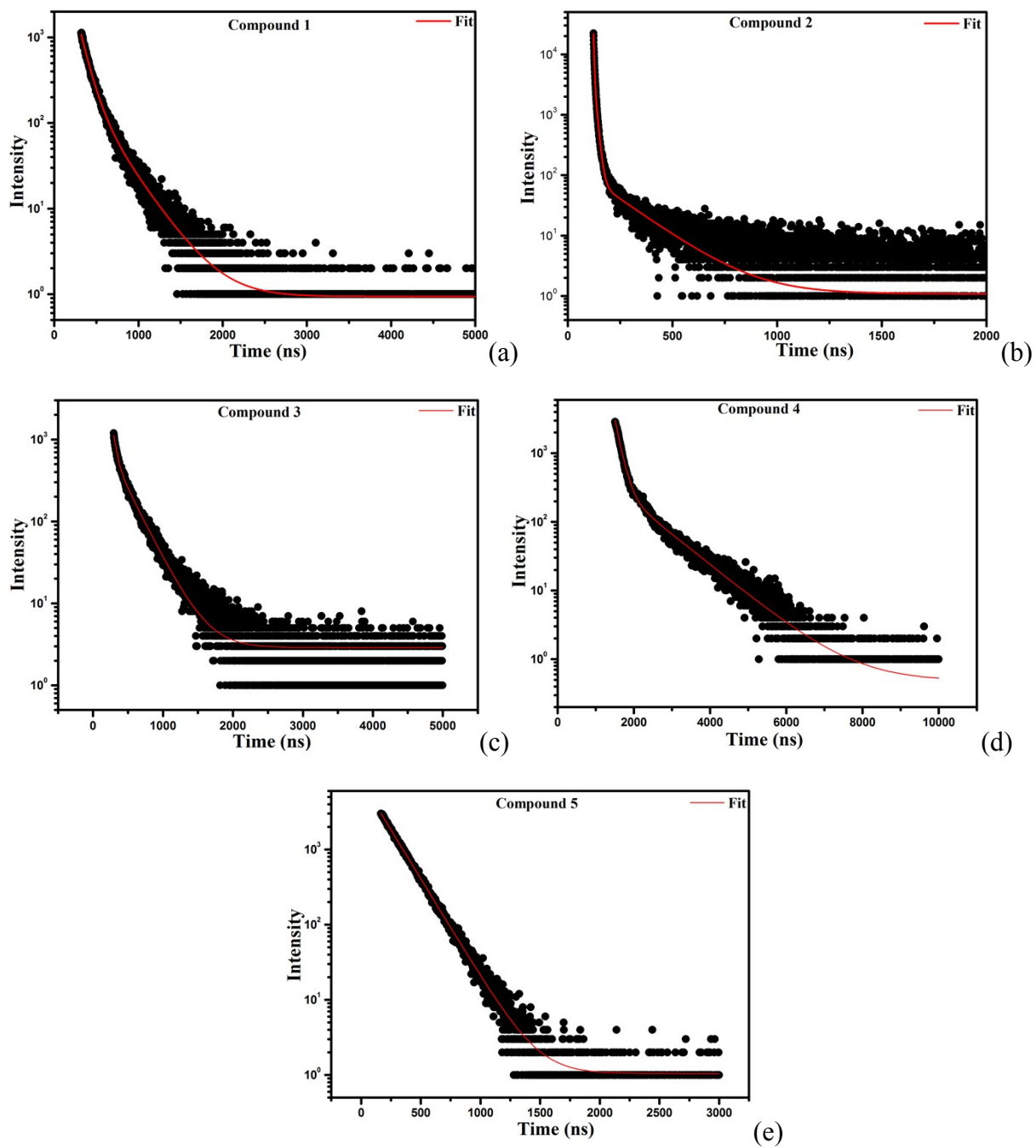
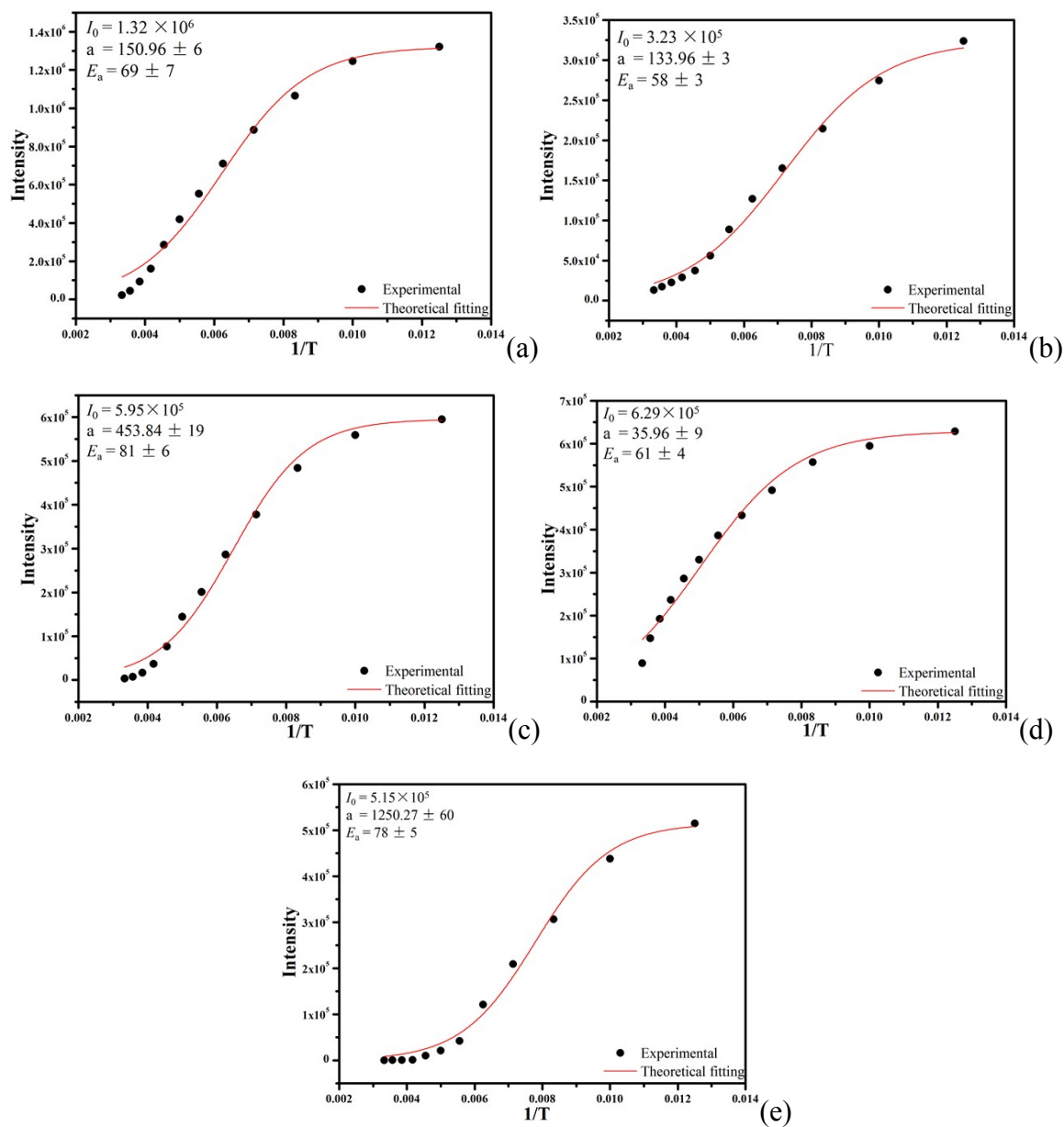


Fig. S6. The time-resolved PL emissions of compounds 1-5.



**Fig. S7.** Temperature dependence of the PL intensity and theoretical fitting data from Arrhenius-type model for compounds 1 (a), 2 (b), 3 (c), 4 (d) and 5 (e).

**Table S1.** Crystal Data and Structure Refinements for compounds **1-5**.

Compound	<b>1</b>	<b>2</b>	<b>3</b>	<b>4</b>	<b>5</b>
chemical	C <sub>6</sub> NH <sub>8</sub> CuI <sub>2</sub>	C <sub>8</sub> N <sub>2</sub> H <sub>18</sub> Cu <sub>2</sub> I <sub>4</sub>	C <sub>7</sub> NH <sub>10</sub> Cu <sub>2</sub> I <sub>3</sub>	C <sub>6</sub> N <sub>2</sub> H <sub>14</sub> Cu <sub>3</sub> I <sub>5</sub>	C <sub>6</sub> N <sub>2</sub> H <sub>14</sub> Cu <sub>3</sub> Br <sub>5</sub>
fw	411.47	776.92	615.94	939.31	704.36
Space group	<i>Cmca</i>	<i>P2<sub>1</sub>/c</i>	<i>Pnma</i>	<i>Pnma</i>	<i>Pnma</i>
<i>a</i> /Å	13.102(10)	9.2225(9)	16.9103(11)	18.748(3)	17.951(2)
<i>b</i> /Å	18.403(14)	10.9983(11)	6.3543(4)	8.3956(12)	7.9123(10)
<i>c</i> /Å	16.480(13)	15.5063(16)	12.5868(8)	11.0005(16)	10.4907(14)
$\beta$ /°	90	93.8620(10)	90	90	90
<i>V</i> (Å <sup>3</sup> )	3974(5)	1569.3(3)	1352.49(15)	1731.5(4)	1490.0(3)
<i>Z</i>	16	4	4	4	4
<i>D</i> <sub>calcd</sub> (g·cm <sup>-3</sup> )	2.751	3.289	3.025	3.603	3.140
Temp (K)	298(2)	298(2)	298(2)	298(2)	298(2)
$\mu$ (mm <sup>-1</sup> )	8.348	10.558	9.954	12.545	17.626
<i>F</i> (000)	2976	1400	1104	1664	1304
Reflections	16376	17756	15117	19071	16396
Unique	2383	3595	1701	2155	1828
Reflections	2223	2825	1519	2042	1538
GOF on <i>F</i> <sup>2</sup>	1.067	1.058	1.067	1.018	1.033
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> ( <i>I</i> > 2 $\sigma$ ( <i>I</i> )) <sup>a</sup>	0.0333/0.0958	0.0648/0.2193	0.0217/0.0495	0.0208/0.0495	0.0443/0.1105
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (all)	0.0382/0.0998	0.0806/0.2323	0.0258/0.0513	0.0226/0.0504	0.0537/0.1176
$\Delta\rho$ <sub>max</sub> (e/Å <sup>3</sup> )	0.693	4.148	1.161	1.164	1.719
$\Delta\rho$ <sub>min</sub> (e/Å <sup>3</sup> )	-0.847	-3.495	-0.618	-0.777	-1.721

<sup>a</sup>  $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ ,  $wR_2 = \{\sum w[(F_o)^2 - (F_c)^2]^2 / \sum w[(F_o)^2]^2\}^{1/2}$



**Table S2.** Selected bond lengths (Å) and angle (°) for compound 1.

Cu(1)-I(3)	2.6750(15)	Cu(2)-I(1)	2.6583(15)
Cu(1)-I(3)#1	2.6751(15)	Cu(2)-I(1)#3	2.6583(15)
Cu(1)-I(2)	2.7015(15)	Cu(2)-I(3)	2.6740(15)
Cu(1)-I(2)#2	2.7015(15)	Cu(2)-I(3)#1	2.6740(15)
I(3)-Cu(1)-I(3)#1	106.68(7)	I(1)-Cu(2)-I(1)#3	104.78(7)
I(3)-Cu(1)-I(2)	118.58(5)	I(1)-Cu(2)-I(3)	111.34(5)
I(3)#1-Cu(1)-I(2)	106.80(5)	I(1)#3-Cu(2)-I(3)	111.37(4)
I(3)-Cu(1)-I(2)#2	106.80(5)	I(1)-Cu(2)-I(3)#1	111.37(4)
I(3)#1-Cu(1)-I(2)#2	118.58(5)	I(1)#3-Cu(2)-I(3)#1	111.34(5)
I(2)-Cu(1)-I(2)#2	100.02(6)	I(3)-Cu(2)-I(3)#1	106.74(7)

Symmetry transformations used to generate equivalent atoms: #1  $x, -y-1, -z+1$ , #2  $-x-1, -y-1, -z+1$ , #3  $-x, -y-1, z+1$ .

**Table S3.** Selected bond lengths (Å) and angle (°) for compound 2.

Cu(1)-I(2)	2.573(2)	Cu(2)-I(1)	2.642(2)
Cu(1)-I(3)#1	2.557(2)	Cu(2)-I(2)	2.579(2)
Cu(1)-I(4)#1	2.495(2)	Cu(2)-I(3)	2.556(2)
Cu(1)-I(1)	2.619(2)	Cu(2)-I(4)	2.501(2)
Cu(2)-Cu(1)	2.737(2)	Cu(2)-Cu(1)#1	3.015(2)
I(4)#2-Cu(1)-I(3)#2	106.81(7)	I(4)-Cu(2)-I(3)	106.64(8)
I(4)#2-Cu(1)-I(2)	111.51(8)	I(4)-Cu(2)-I(2)	106.31(8)
I(3)#2-Cu(1)-I(2)	108.66(8)	I(3)-Cu(2)-I(2)	116.51(8)
I(4)#2-Cu(1)-I(1)	115.50(8)	I(4)-Cu(2)-I(1)	105.27(8)
I(3)#2-Cu(1)-I(1)	109.24(7)	I(3)-Cu(2)-I(1)	117.01(8)
I(2)-Cu(1)-I(1)	104.98(7)	I(2)-Cu(2)-I(1)	104.13(7)

Symmetry transformations used to generate equivalent atoms: #1  $-x+3/2, y-1/2, -z+1/2$ , #2  $-x+3/2, y+1/2, -z+1/2$ .

**Table S4.** Selected bond lengths (Å) and angle (°) for compound **3**.

Cu(1)-I(3)	2.6084(6)	Cu(1)-I(1)#1	2.7142(6)
Cu(1)-I(2)	2.6272(6)	Cu(1)-I(1)	2.7305(6)
I(3)-Cu(1)-I(2)	114.02(2)	I(3)-Cu(1)-I(1)	111.03(2)
I(3)-Cu(1)-I(1)#1	109.36(2)	I(2)-Cu(1)-I(1)	104.41(2)
I(2)-Cu(1)-I(1)#1	108.23(2)	I(1)#1-Cu(1)-I(1)	109.626(19)

Symmetry transformations used to generate equivalent atoms: #1 -x, -y, -z+1.

**Table S5.** Selected bond lengths (Å) and angle (°) for compound **4**.

Cu(1)-I(3)	2.6210(9)	Cu(2)-I(2)	2.6186(7)
Cu(1)-I(4)	2.6915(6)	Cu(2)-I(1)	2.6481(6)
Cu(1)-I(4)#1	2.6916(6)	Cu(2)-I(4)	2.6488(7)
Cu(1)-I(1)	2.7161(9)	Cu(2)-I(4)#2	2.7067(7)
Cu(1)-Cu(2)#1	2.8049(8)	Cu(1)-Cu(2)	2.8049(8)
I(3)-Cu(1)-I(4)	108.04(2)	I(2)-Cu(2)-I(1)	107.73(2)
I(3)-Cu(1)-I(4)#1	108.04(2)	I(2)-Cu(2)-I(4)	112.45(2)
I(4)-Cu(1)-I(4)#1	100.31(3)	I(1)-Cu(2)-I(4)	118.09(2)
I(3)-Cu(1)-I(1)	111.17(3)	I(2)-Cu(2)-I(4)#2	116.42(2)
I(4)-Cu(1)-I(1)	114.28(2)	I(1)-Cu(2)-I(4)#2	100.84(2)
I(4)#1-Cu(1)-I(1)	114.28(2)	I(4)-Cu(2)-I(4)#2	101.03(2)

Symmetry transformations used to generate equivalent atoms: #1 x, -y-1/2, z, #2 -x+1, -y, -z+1.

**Table S6.** Selected bond lengths (Å) and angle (°) for compound **5**.

Cu(1)-Br(2)	2.4112(14)	Cu(2)-Br(1)	2.4360(10)
Cu(1)-Br(3)	2.5841(15)	Cu(2)-Br(3)	2.5102(11)
Cu(1)-Br(4)	2.5264(10)	Cu(2)-Br(4)	2.4993(10)
Cu(1)-Br(4)#2	2.5264(10)	Cu(2)-Br(4)#3	2.5259(11)
Cu(1)-Cu(2)	2.7994(13)	Cu(1)-Cu(2)#2	2.7995(13)
Br(2)-Cu(1)-Br(4)#2	111.77(4)	Br(1)-Cu(2)-Br(4)	114.17(4)
Br(2)-Cu(1)-Br(4)	111.78(4)	Br(1)-Cu(2)-Br(3)	107.85(4)
Br(4)#2-Cu(1)-Br(4)	100.02(5)	Br(4)-Cu(2)-Br(3)	113.56(4)
Br(2)-Cu(1)-Br(3)	112.28(5)	Br(1)-Cu(2)-Br(4)#3	117.07(4)
Br(4)#2-Cu(1)-Br(3)	110.18(4)	Br(4)-Cu(2)-Br(4)#3	103.59(4)
Br(4)-Cu(1)-Br(3)	110.18(4)	Br(3)-Cu(2)-Br(4)#3	99.89(4)

Symmetry transformations used to generate equivalent atoms: #1  $x, -y+1/2, z$ , #2  $x, -y-1/2, z$ , #3  $-x+1, -y, -z+2$ .