

Supplementary material for:

Antimony(I)→Pd(II) complexes with (μ -Sb)Pd₂ coordination framework

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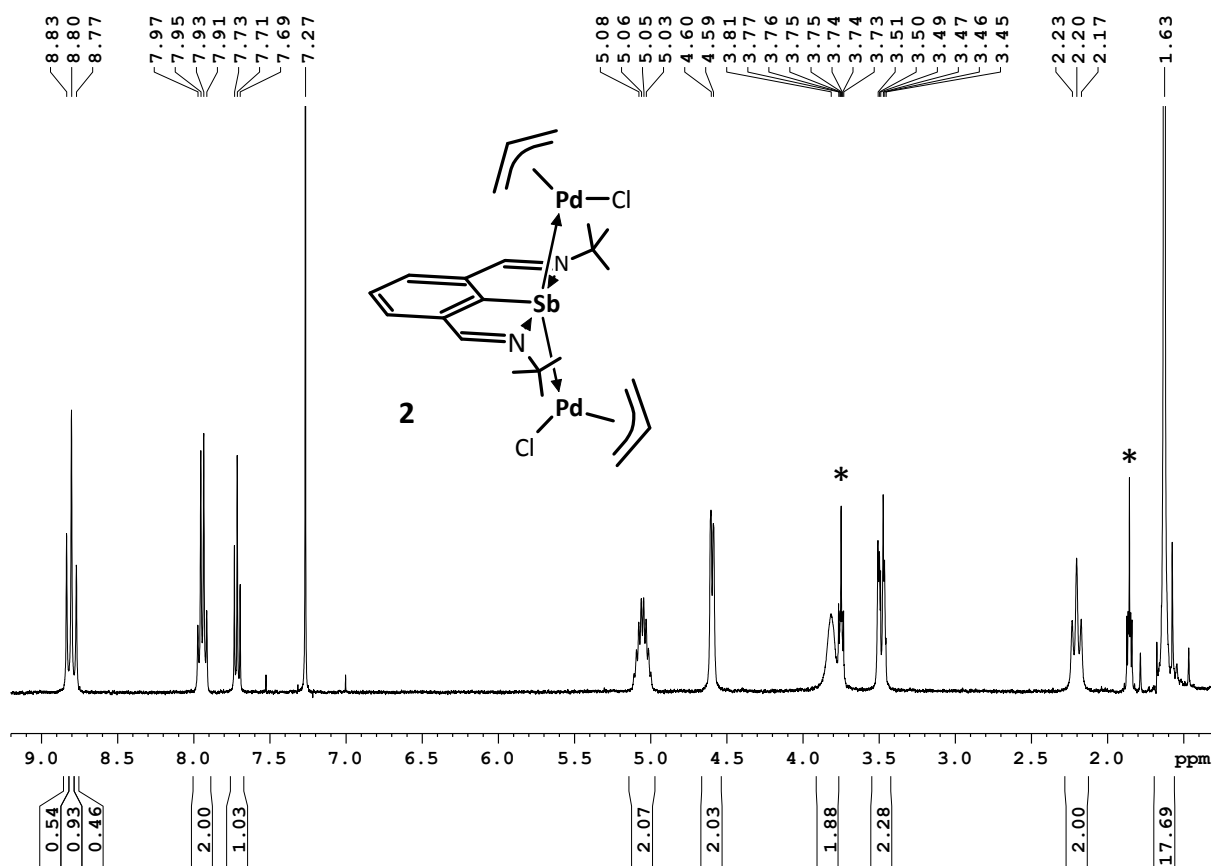


Figure S1: ¹H NMR of compound 2 (500.20 MHz, CDCl₃, 294 K). * - traces of THF.

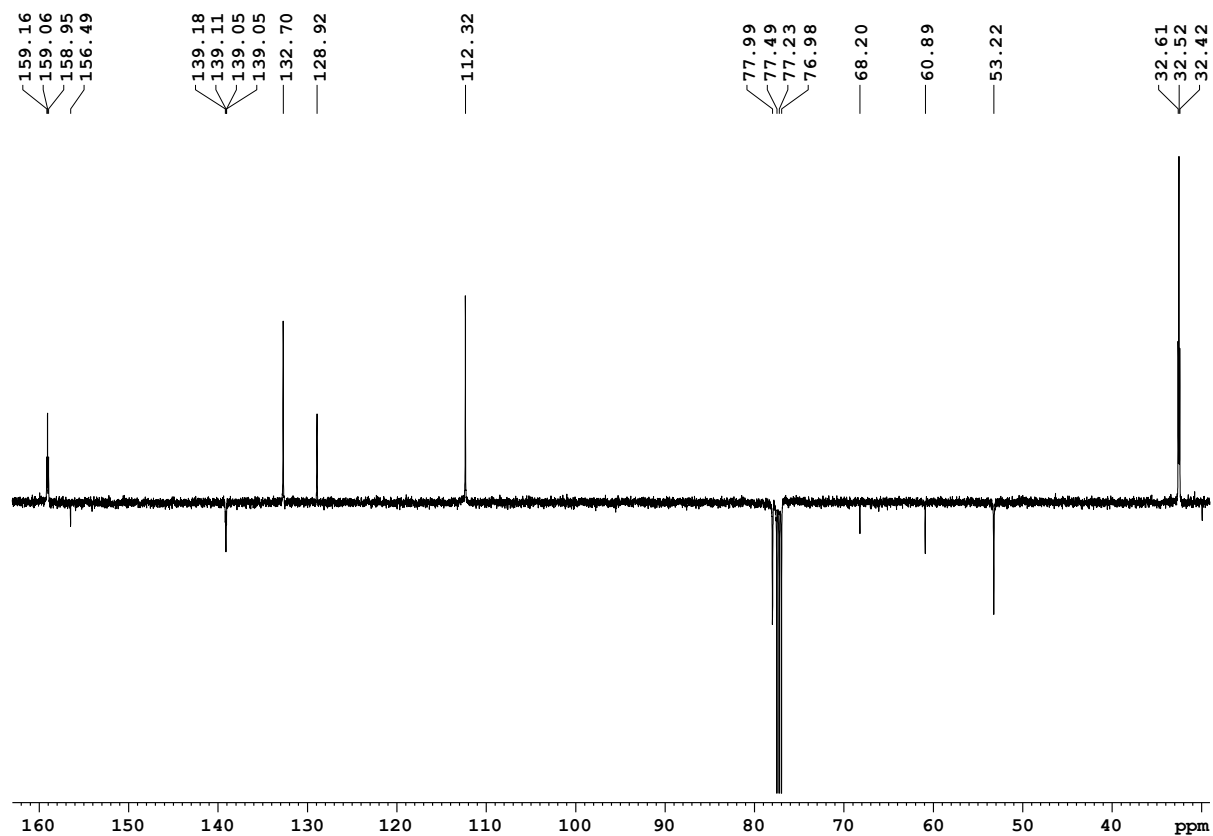
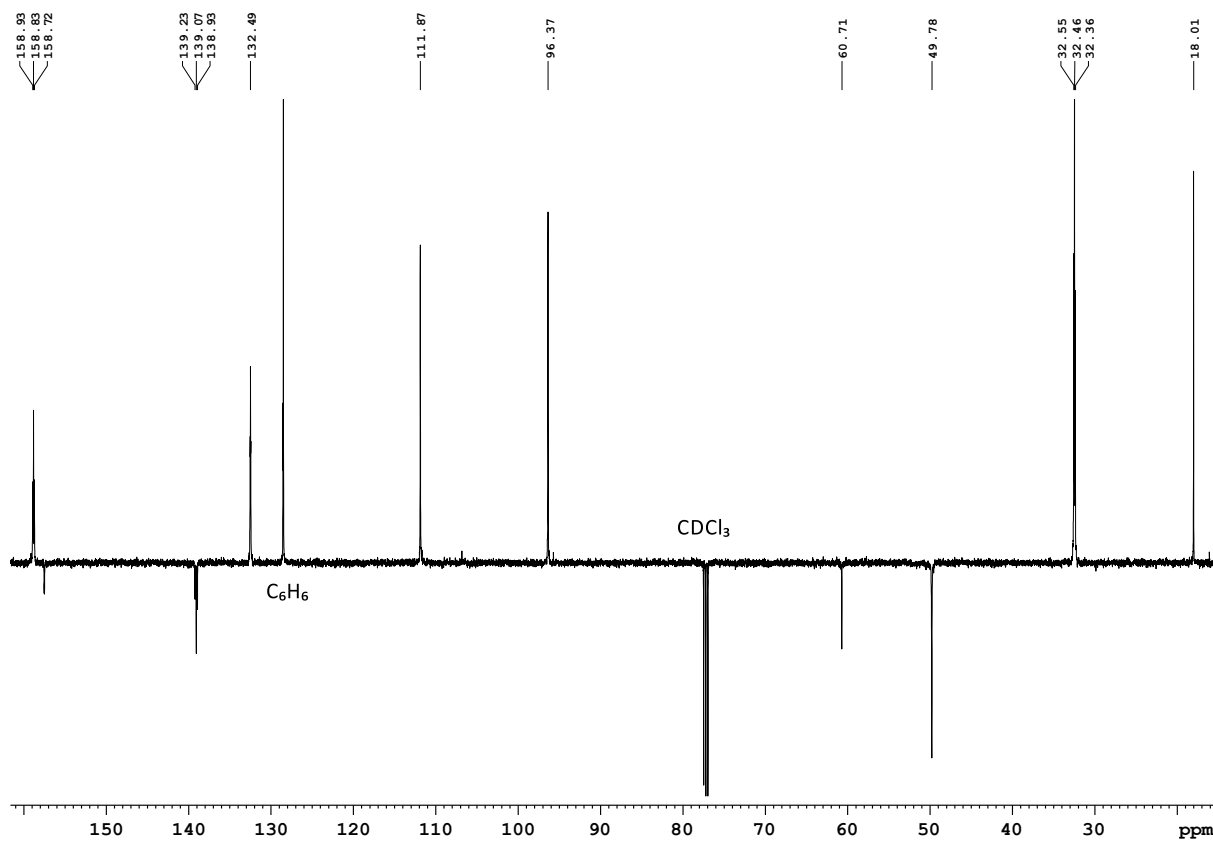
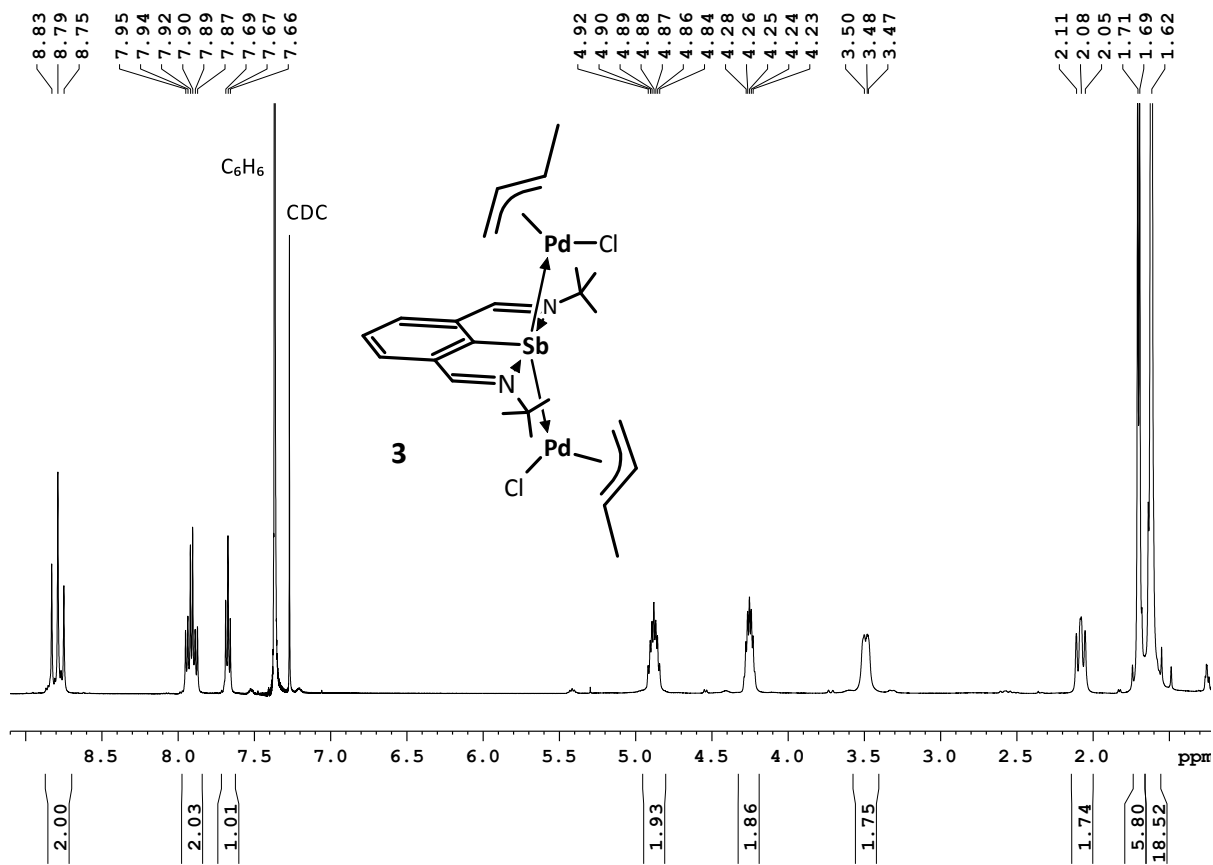


Figure S2: ¹³C{¹H} APT NMR of compound 2 (125.78 MHz, CDCl₃, 294 K).



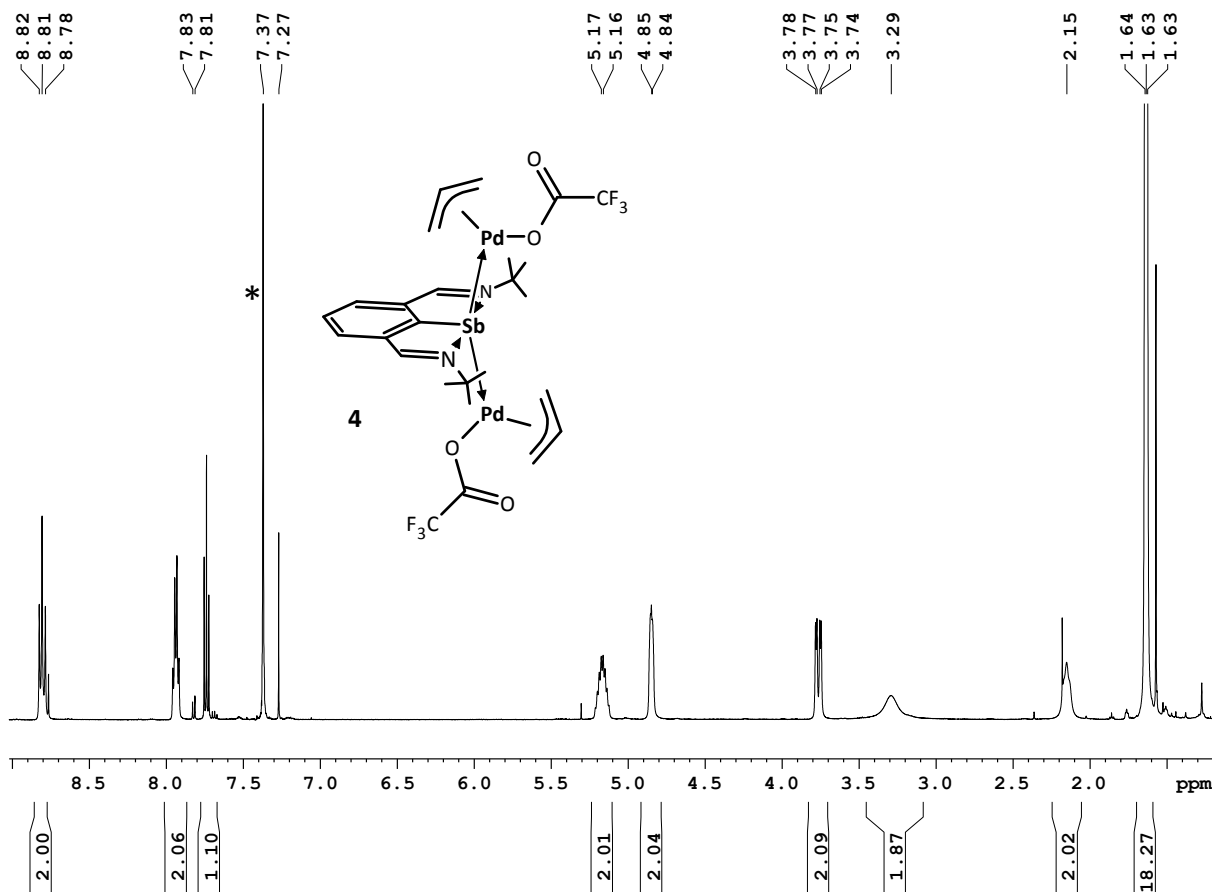


Figure S5: ¹H NMR of compound **4** (500.20 MHz, CDCl₃, 294 K). * - C₆H₆ (since benzene solvate single crystals of **4** were dissolved in CDCl₃).

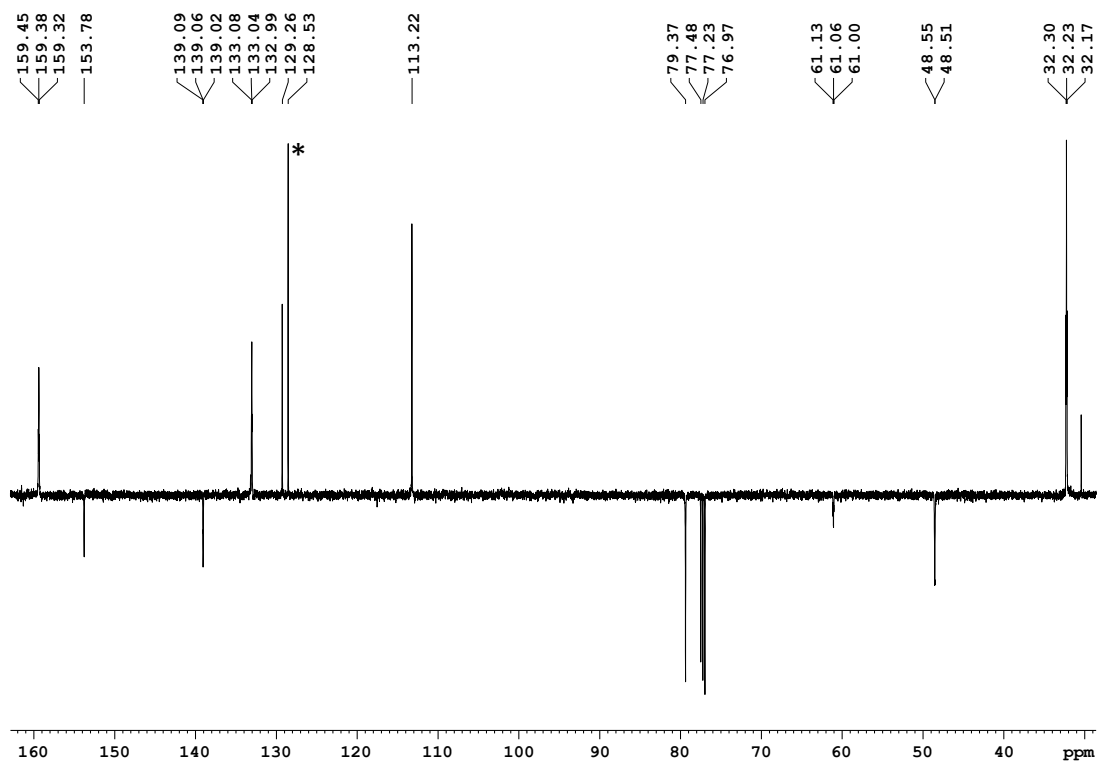


Figure S6: ¹³C{¹H} APT NMR of compound **4** (125.78 MHz, CDCl₃, 294 K). * - C₆H₆ (since benzene solvate single crystals of **4** were dissolved in CDCl₃).

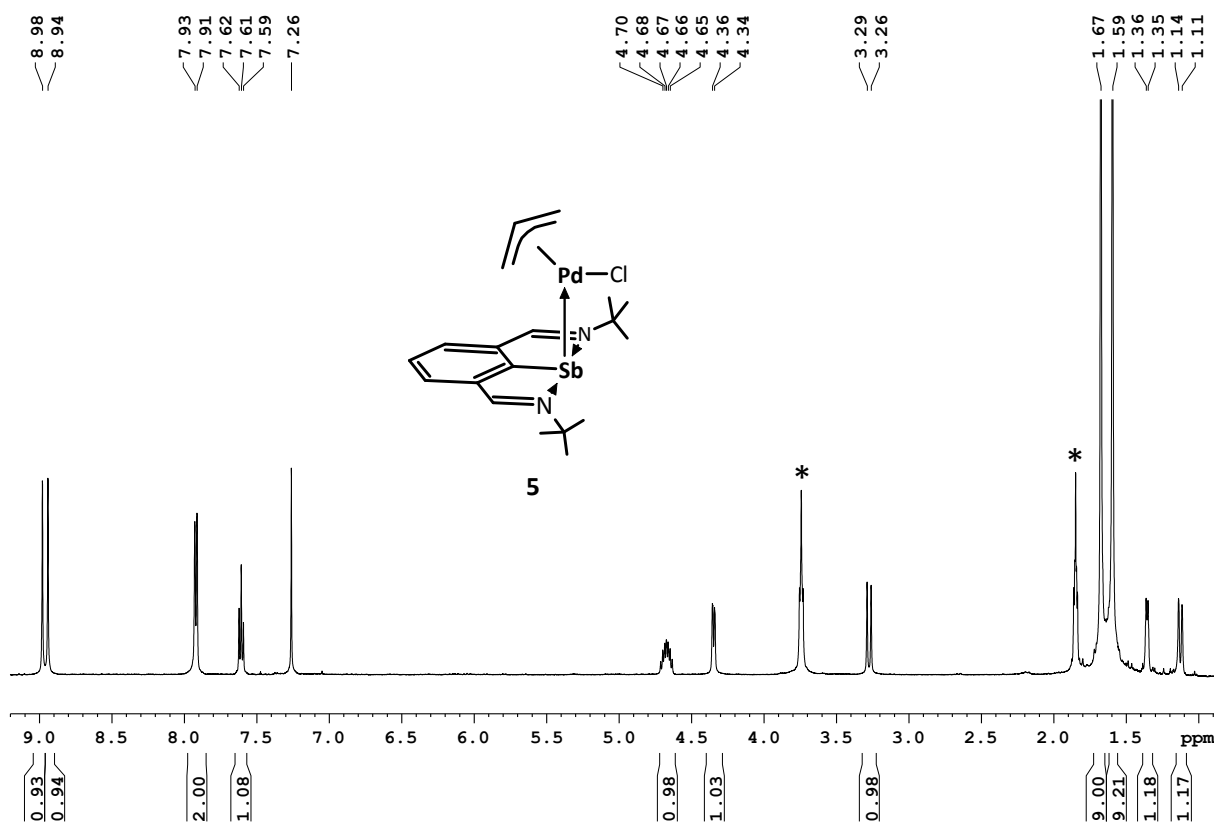


Figure S7: $^1\text{H NMR}$ of compound 5 (500.20 MHz, CDCl_3 , 294 K). * - traces of THF.

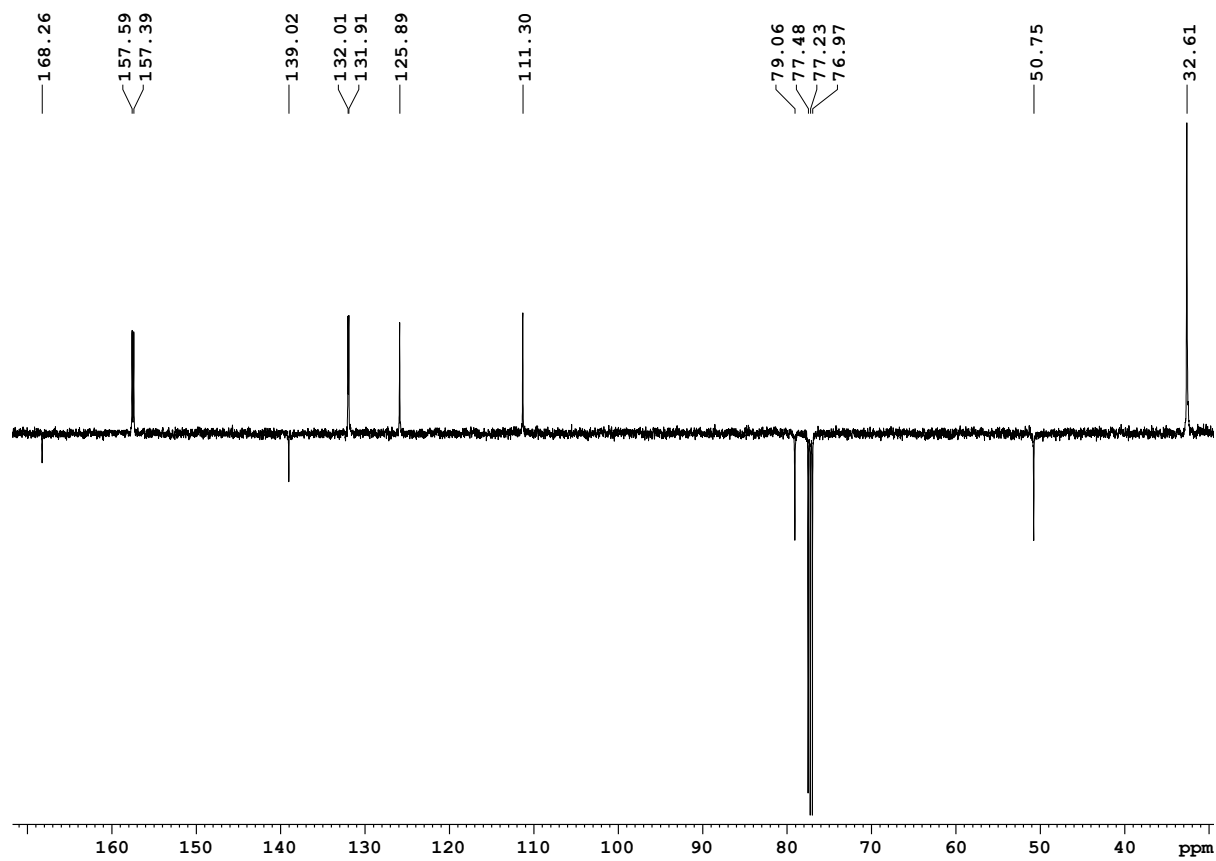


Figure S8: $^{13}\text{C}\{^1\text{H}\}$ APT NMR of compound 5 (125.78 MHz, CDCl_3 , 294 K).

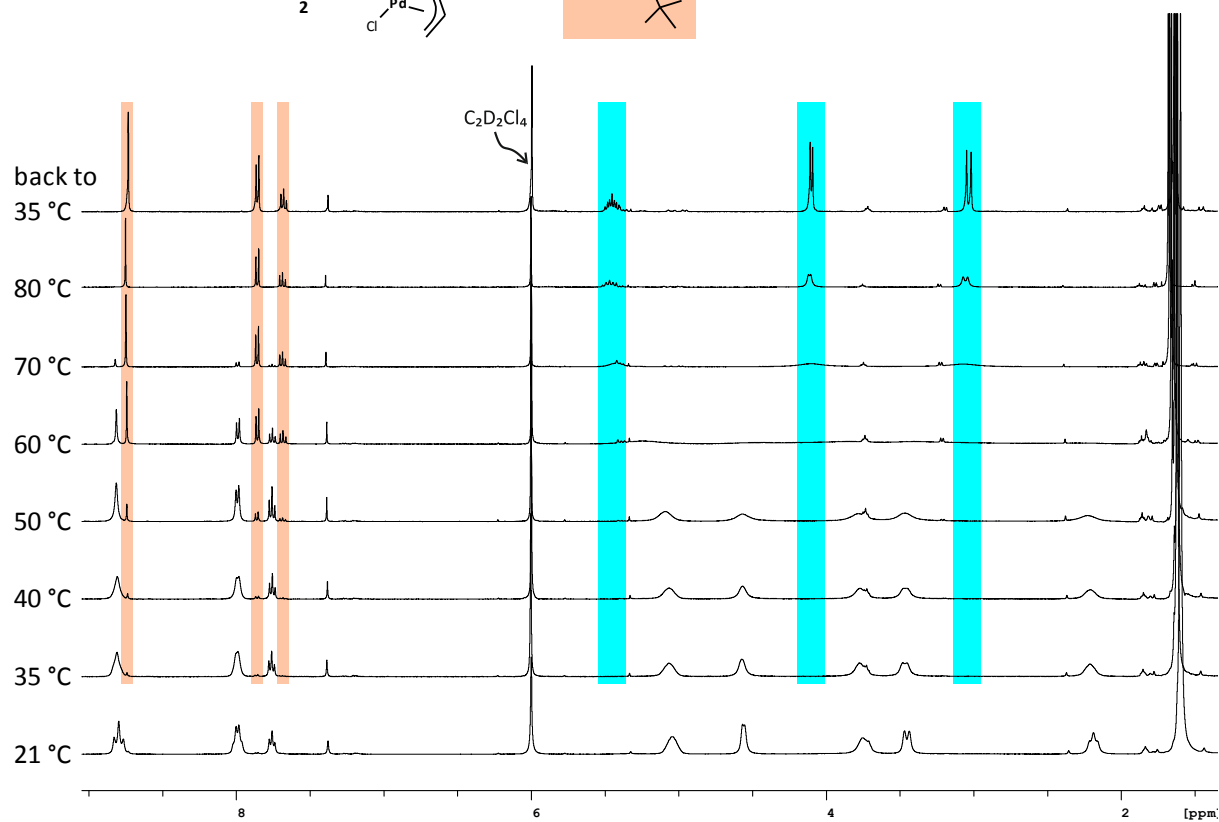
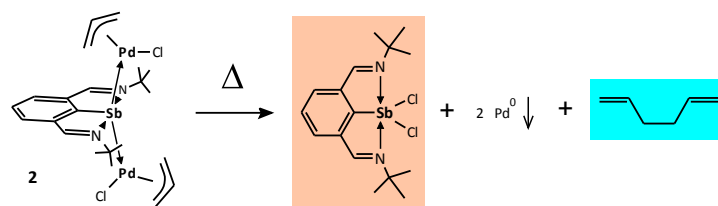


Figure S9: Stacked plot of result of VT- ^1H NMR (500.20 MHz) measurement of freshly prepared NMR sample of compound **2** in $\text{C}_2\text{D}_2\text{Cl}_4$.

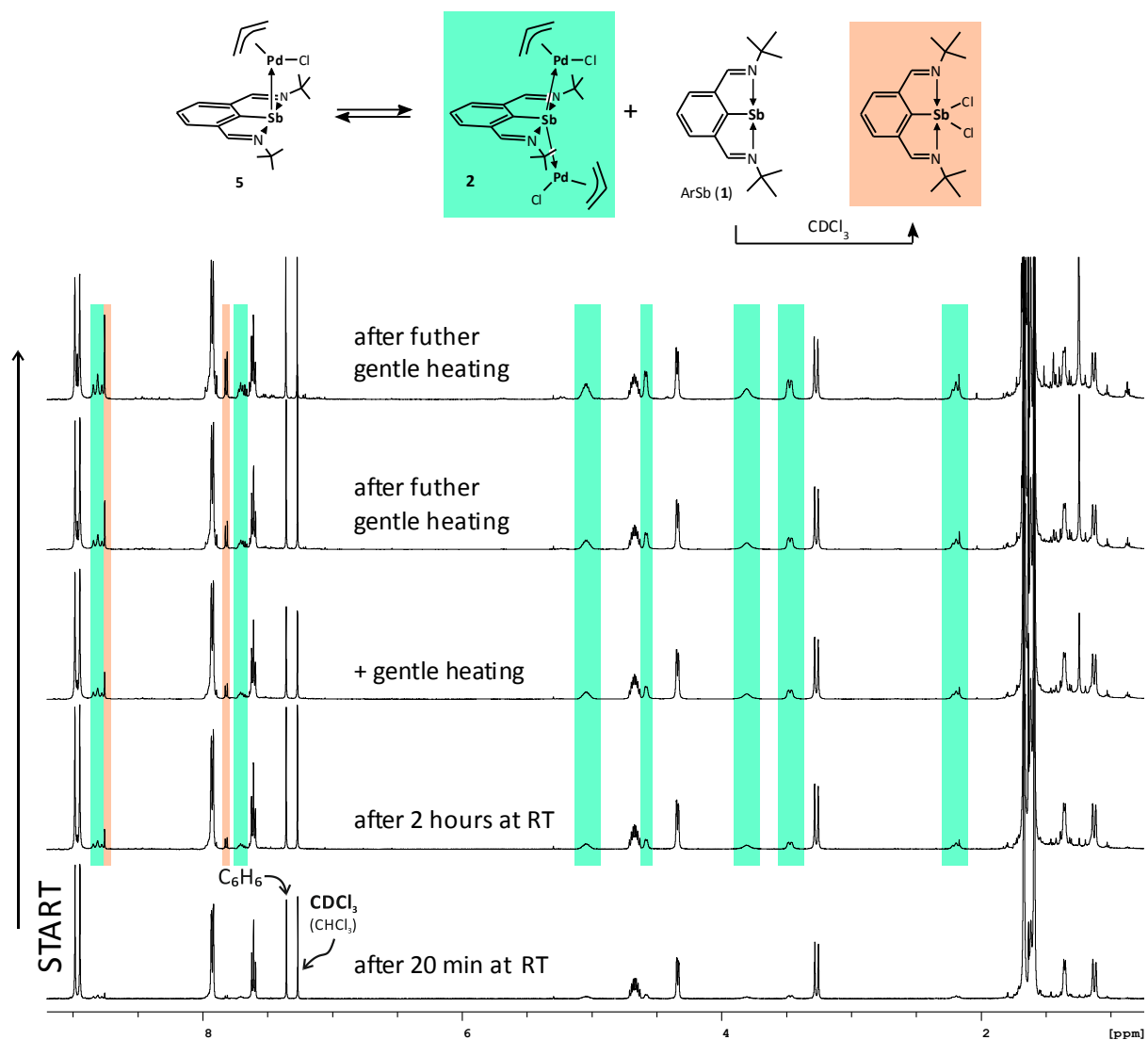


Figure S10: Stacked plot of ^1H NMR (500.20 MHz) spectra showing decomposition of compound **5** in CDCl_3 (gentle heating means heating the NMR tube up for couple of seconds outside the NMR spectrometer with heat gun set to 60°C).

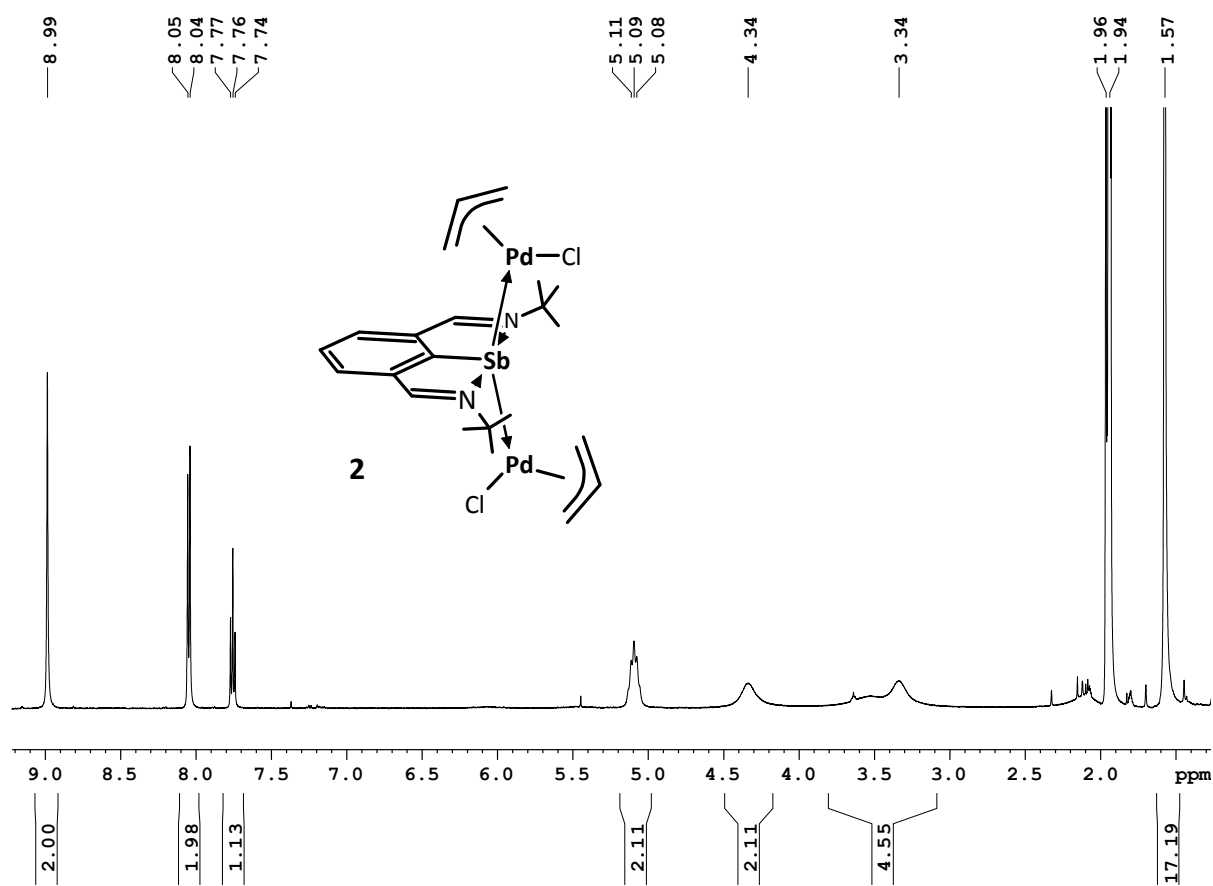


Figure S11: ¹H NMR of compound **2** (500.20 MHz, CD₃CN, 294 K).

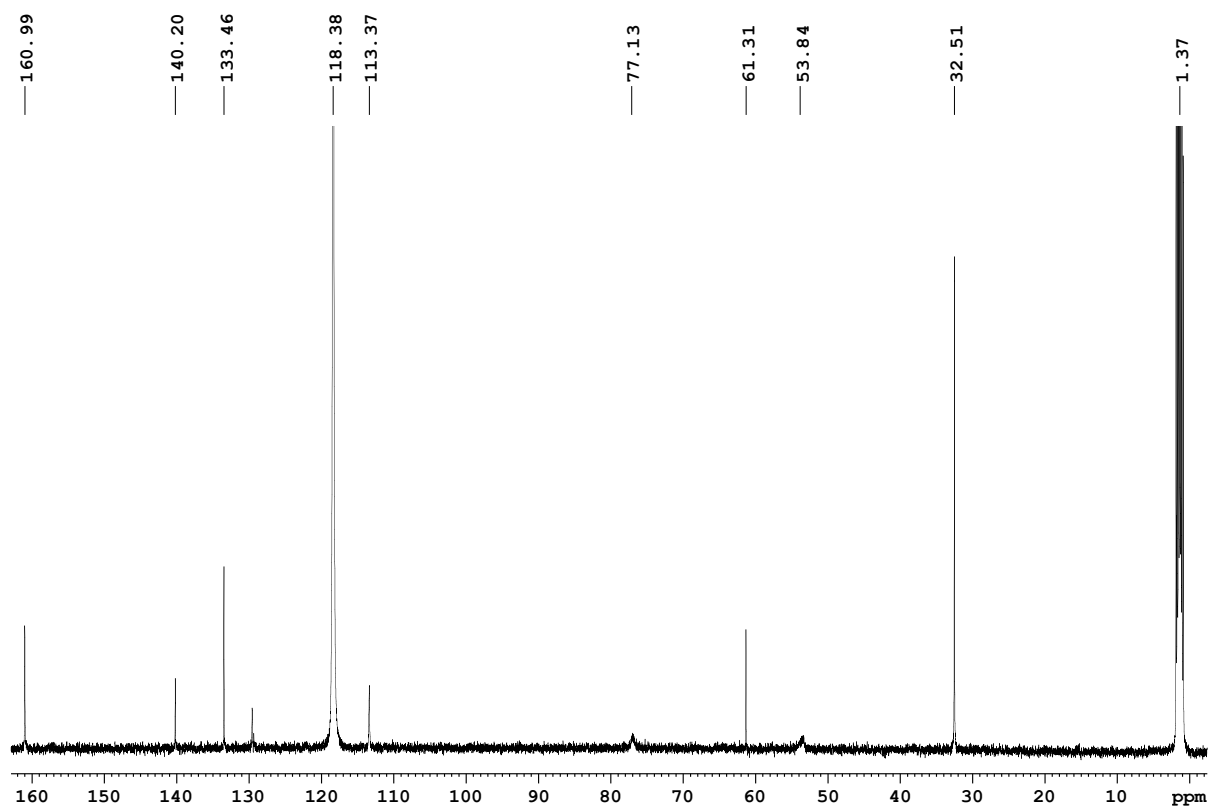


Figure S12: ¹³C{¹H} NMR of compound **2** (125.78 MHz, CD₃CN, 294 K).

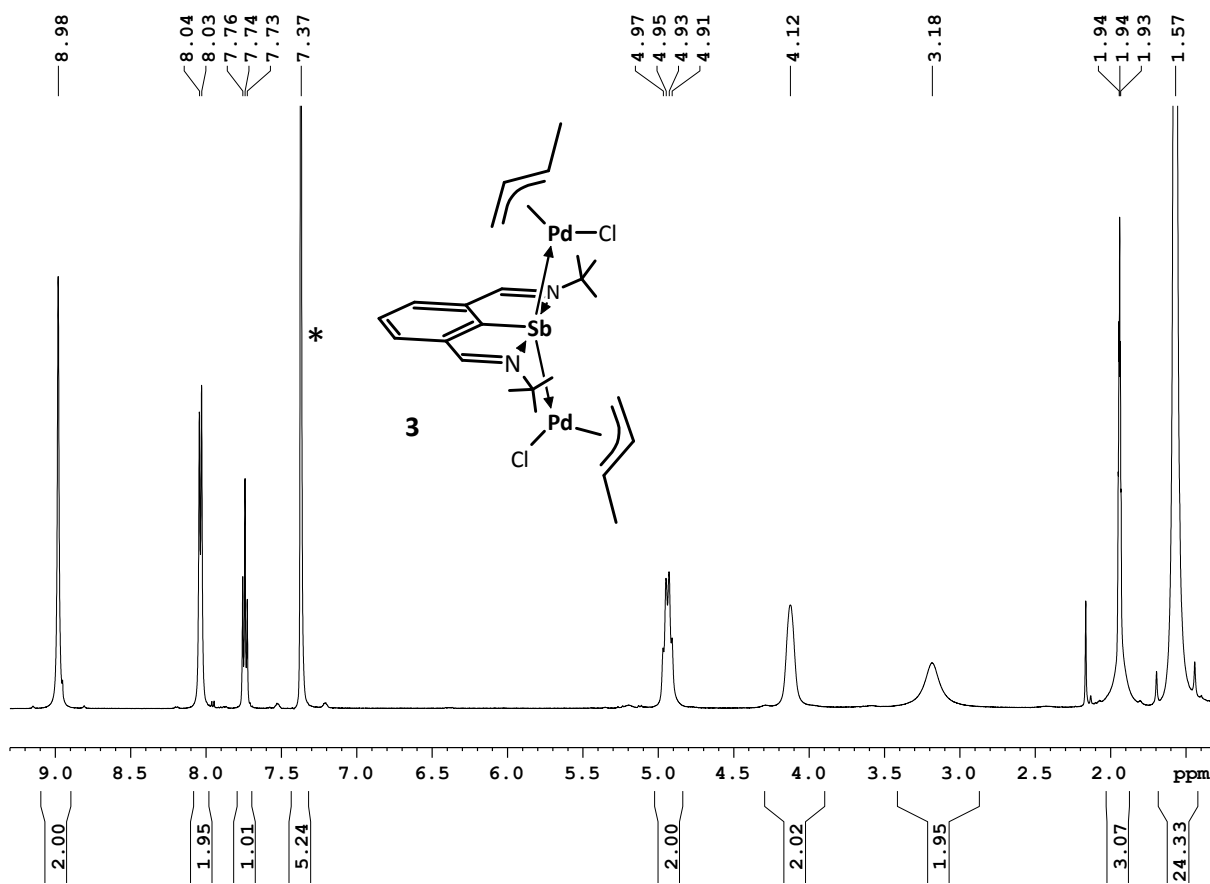


Figure S13: ^1H NMR of compound **3** (500.20 MHz, CD_3CN , 294 K). * - benzene (since benzene solvate single crystals of **3** were dissolved in CD_3CN).

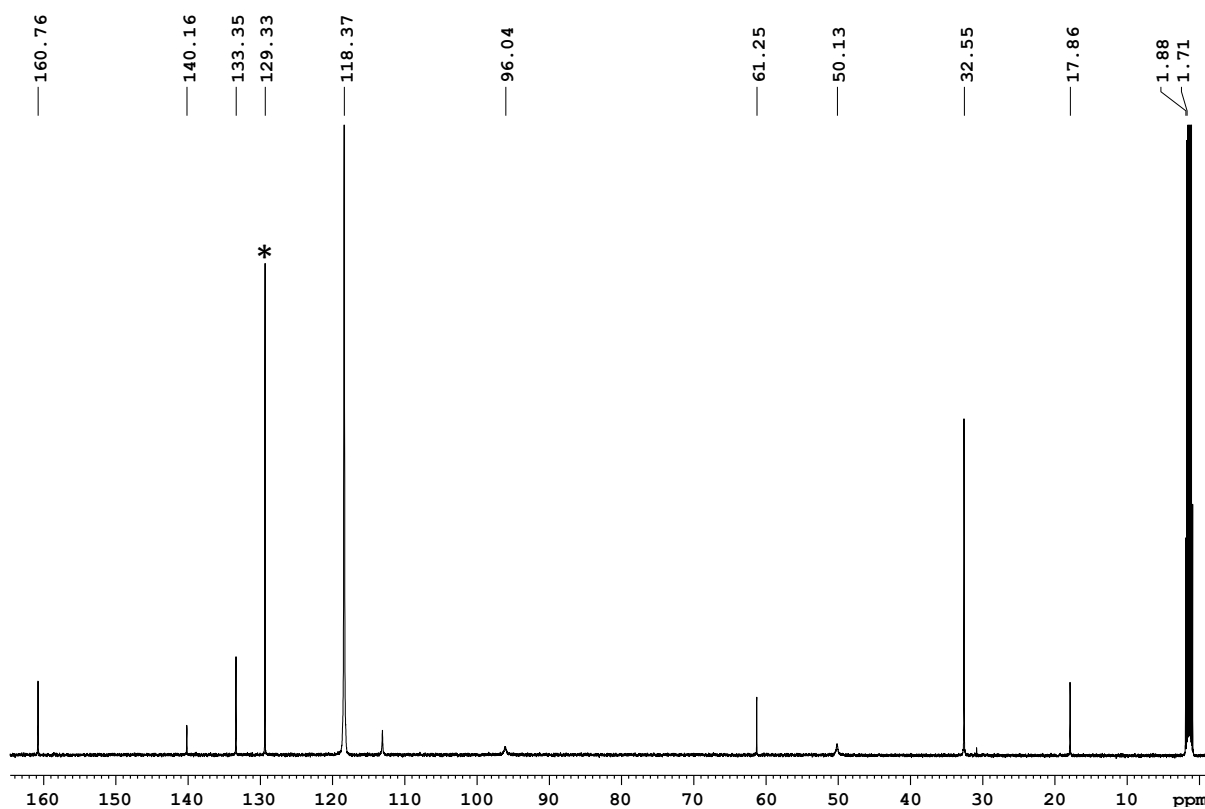


Figure S14: $^{13}\text{C}\{^1\text{H}\}$ NMR of compound **3** (125.78 MHz, CD_3CN , 294 K).

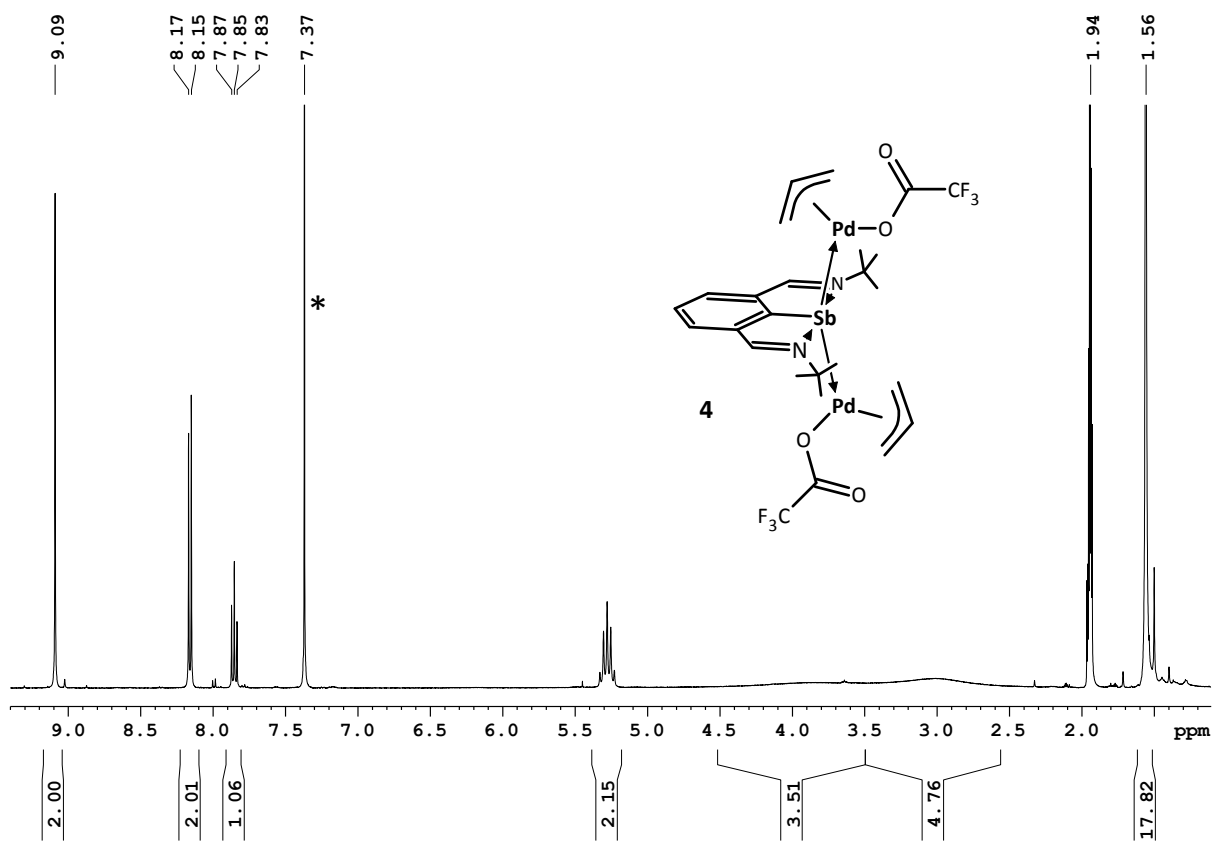


Figure S15: ^1H NMR of compound **4** (500.20 MHz, CD_3CN , 294 K). * - benzene (since benzene solvate single crystals of **4** were dissolved in CD_3CN).

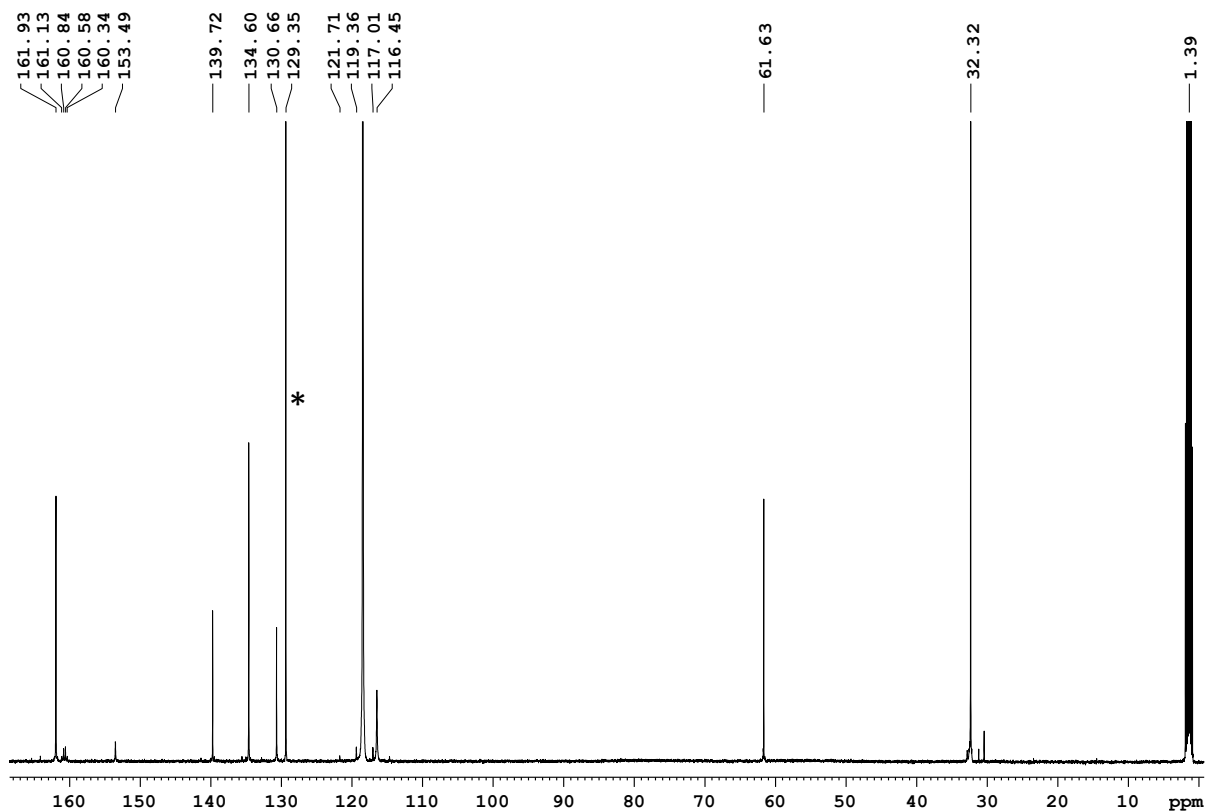


Figure S16: $^{13}\text{C}\{^1\text{H}\}$ NMR of compound **4** (125.78 MHz, CD_3CN , 294 K). * - benzene (since benzene solvate single crystals of **4** were dissolved in CD_3CN).

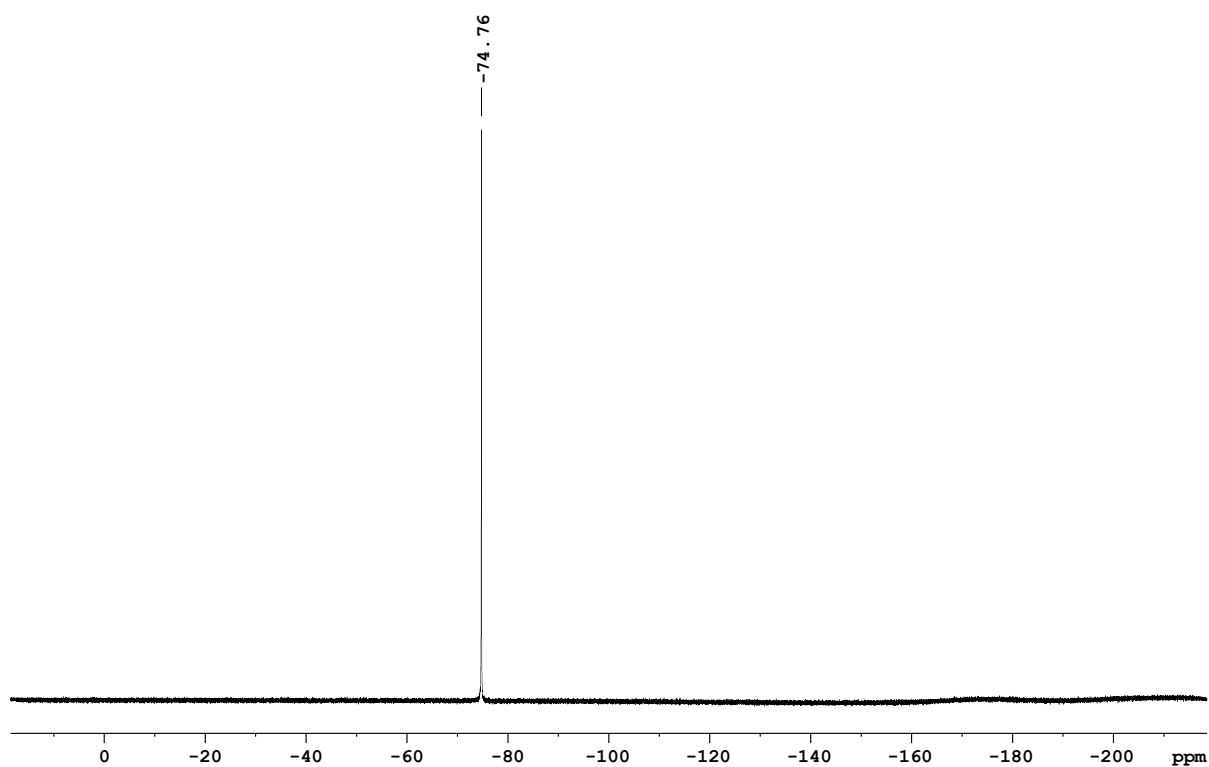


Figure S17: ^{19}F NMR of compound **4** (376.50 MHz, CD_3CN , 294 K).

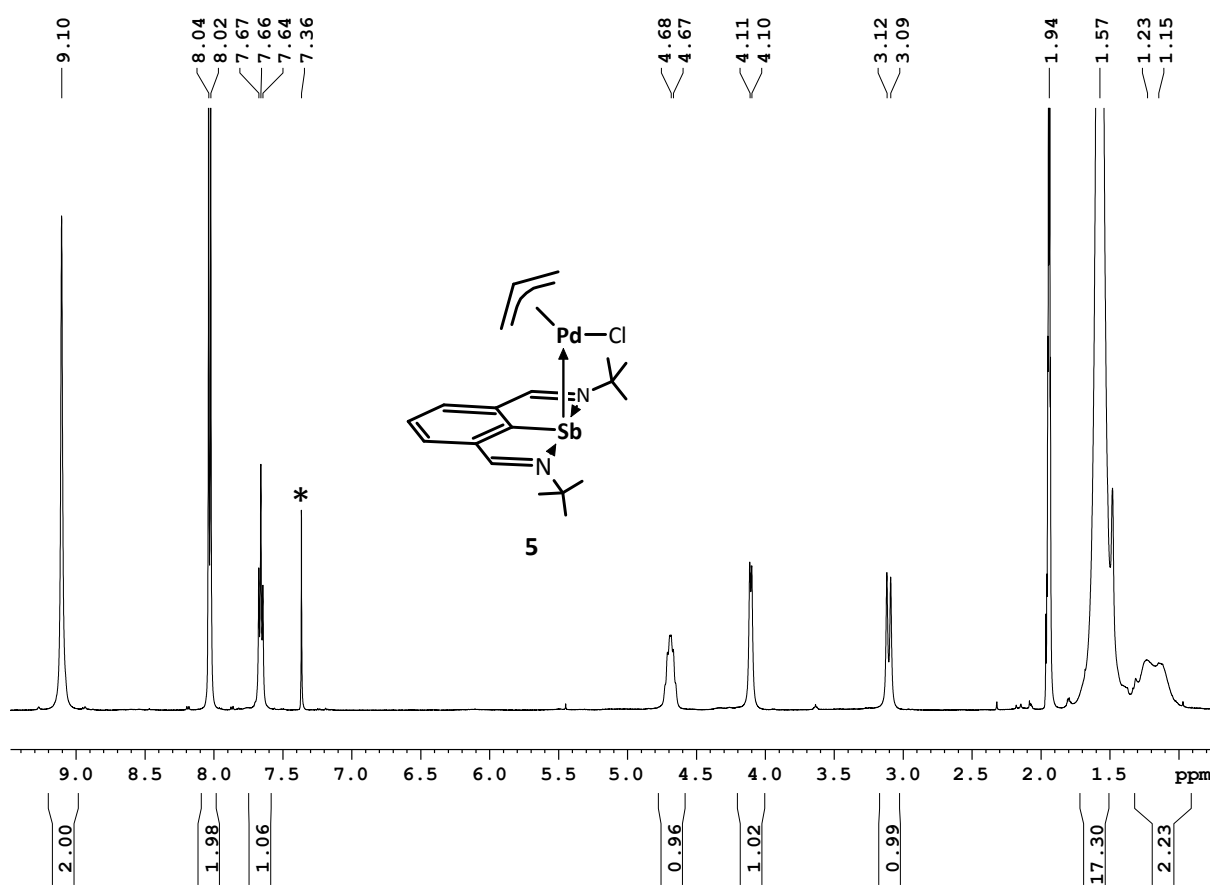


Figure S18: ^1H NMR of compound **5** (500.20 MHz, CD_3CN , 294 K). * - traces of benzene.

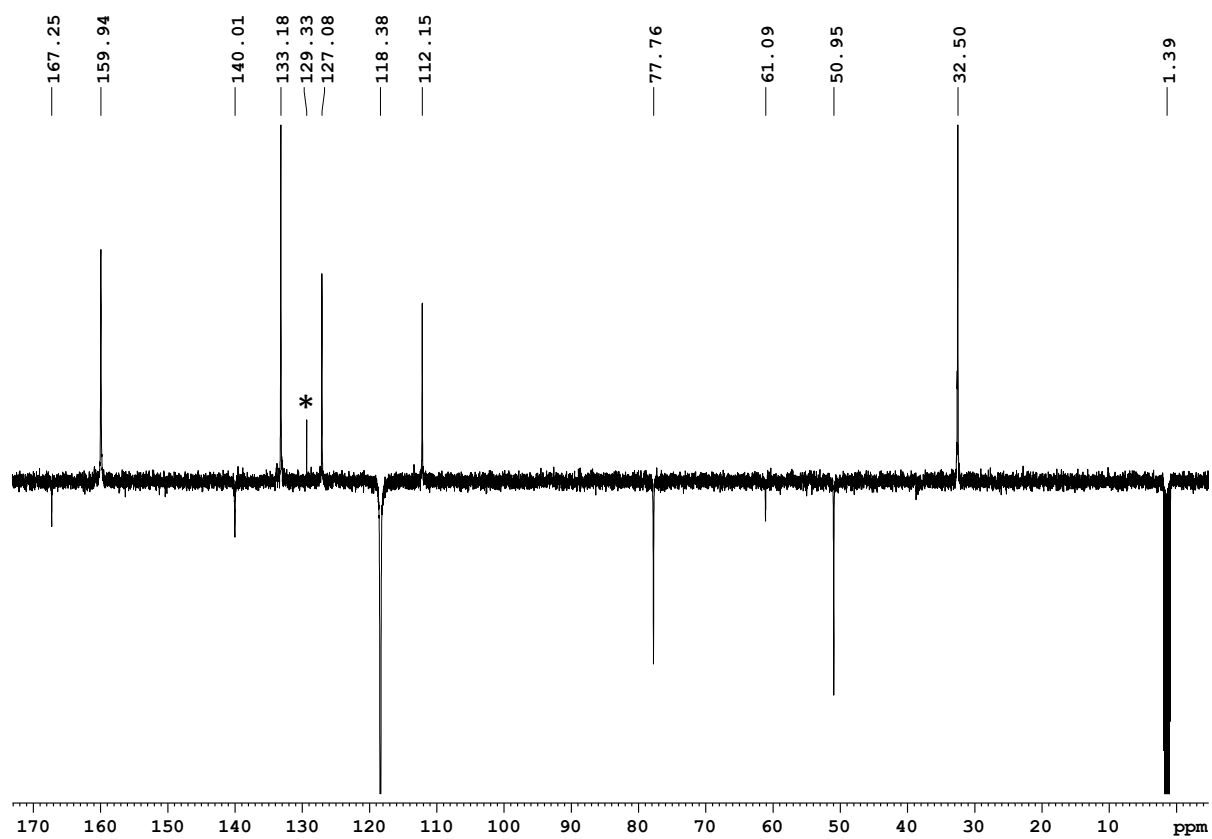


Figure S19: $^{13}\text{C}\{^1\text{H}\}$ APT NMR of compound **5** (125.78 MHz, CD_3CN , 294 K).

Table S1. Crystallographic data for studied compounds **2-5**.

	2	3	4	5
chemical formula	C ₂₂ H ₃₃ Cl ₂ N ₂ Pd ₂ Sb	C ₂₄ H ₃₇ Cl ₂ N ₂ Pd ₂ Sb	C ₂₆ H ₃₃ F ₆ N ₂ O ₄ Pd ₂ Sb	C ₁₉ H ₂₈ ClN ₂ PdSb
	.2(C ₆ H ₆)	.2(C ₆ H ₆)	.(C ₆ H ₆)	
Cryst syst	monoclinic	orthorhombic	orthorhombic	monoclinic
Space group	<i>P2₁/c</i>	<i>Pca2₁</i>	<i>Pca2₁</i>	<i>P2₁/c</i>
<i>a</i> [Å]	9.9651(10)	26.6878(9)	18.3001(13)	16.2454(7)
<i>b</i> [Å]	22.9000(14)	12.1213(4)	10.5668(6)	16.0559(7)
<i>c</i> [Å]	18.0049(13)	23.2984(8)	18.5561(14)	17.8662(8)
β[°]	118.278(7)	90	90	113.1910(10)
<i>Z</i>	4	8	4	8
μ[mm ⁻¹]	1.9	1.8	1.8	2.2
<i>D_x</i> [g cm ⁻³]	1.63	1.61	1.78	1.70
Crystal size [mm]	0.59×0.31×0.26	0.50×0.30×0.10	0.20×0.08×0.07	0.57×0.27×0.17
θ range, [deg]	1-27.5	1-32	1-27.9	1-32
<i>T_{min}</i> , <i>T_{max}</i>	0.579, 0.746	0.543, 0.746	0.466, 0.746	0.584, 0.746
no. of reflns measd	94 388	71 935	30 074	123 966
no. of unique reflns, <i>R_{int}</i>	53544, 0.022	21184, 0.107	8073, 0.134	13698, 0.083
no. of obsd reflns	6890	13371	5563	8821
no. of params	370	817	473	475
<i>S</i> all data	1.222	1.031	0.998	1.093
final <i>R</i> indices [<i>I</i> >2σ(<i>I</i>)]	0.024	0.067	0.058	0.058
w <i>R</i> 2 indices (all data)	0.051	0.084	0.075	0.069
Δρ, max., min. [e Å ⁻³]	0.95, -0.72	0.92, -1.38	0.71, -0.82	1.28, -0.95
CCDC	1908187	1908186	1908185	1908188

$R_{\text{int}} = \sum |F_o^2 - F_{o,\text{mean}}^2| / \sum F_o^2$, $\text{GOF} = [\sum (w(F_o^2 - F_c^2)^2) / (N_{\text{diffrs}} - N_{\text{params}})]^{1/2}$ for all data, $R(F) = \sum ||F_o| - |F_c|| / \sum |F_o|$ for observed data, $wR(F^2) = [\sum (w(F_o^2 - F_c^2)^2) / (\sum w(F_o^2)^2)]^{1/2}$ for all data.