

Supplementary Information for

Reactions of Cerium Complexes with Transition Metal Nitrides: Synthesis and Structure of Heterometallic Cerium Complexes Containing Bridging Catecholate Ligands

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Table S1. Crystallographic data and experimental details for **4**, **6** and **9**.

	<b>4</b>	<b>6</b>	<b>9</b>
Formula	C <sub>53</sub> H <sub>111</sub> CeCl <sub>2</sub> Co <sub>3</sub> NO <sub>29</sub> P <sub>9</sub> Ru	C <sub>46</sub> H <sub>78</sub> CeCo <sub>2</sub> NO <sub>22</sub> P <sub>6</sub> Ru	C <sub>46</sub> H <sub>82</sub> CeCo <sub>2</sub> O <sub>26</sub> P <sub>6</sub> Re
Formula weight	1994.03	1541.96	1681.11
Crystal system	Triclinic	monoclinic	Monoclinic
Space group	P-1	C2/c	P2 <sub>1</sub> /c
<i>a</i> , Å	15.0838(5)	22.9360(3)	11.86004(14)
<i>b</i> , Å	17.4193(7)	13.2047(2)	22.5385(3)
<i>c</i> , Å	17.9226(7)	40.2851(7)	24.0165(2)
<i>α</i> , deg	114.874(4)	90	90
<i>β</i> , deg	92.656(3)	93.4905(16)	99.3323(10)
<i>γ</i> , deg	103.838(3)	90	90
<i>V</i> , Å <sup>3</sup>	4089.1(3)	12178.2(3)	6334.81(12)
<i>Z</i>	2	8	4
<i>ρ</i> <sub>calc</sub> , g cm <sup>-3</sup>	1.620	1.682	1.763
<i>T</i> , K	100.00(10)	100.01(10)	100.15
<i>μ</i> , mm <sup>-1</sup>	13.236	13.948	15.222
<i>F</i> (000)	2040.0	6264.0	3372.0
Total reflections	23782	33582	35347
Independent reflections	14549	10936	11375
<i>R</i> <sub>int</sub>	0.0430	0.0611	0.0430
GoF <sup>a</sup>	1.021	1.017	1.0036
<i>R</i> <sub>1</sub> , <sup>b</sup> <i>wR</i> <sub>2</sub> <sup>c</sup> [ <i>I</i> > 2σ( <i>I</i> )]	0.0394, 0.0886	0.0582, 0.1340	0.0367, 0.0902
<i>R</i> <sub>1</sub> , <i>wR</i> <sub>2</sub> (all data)	0.0535, 0.0937	0.0718, 0.1421	0.0450, 0.0943

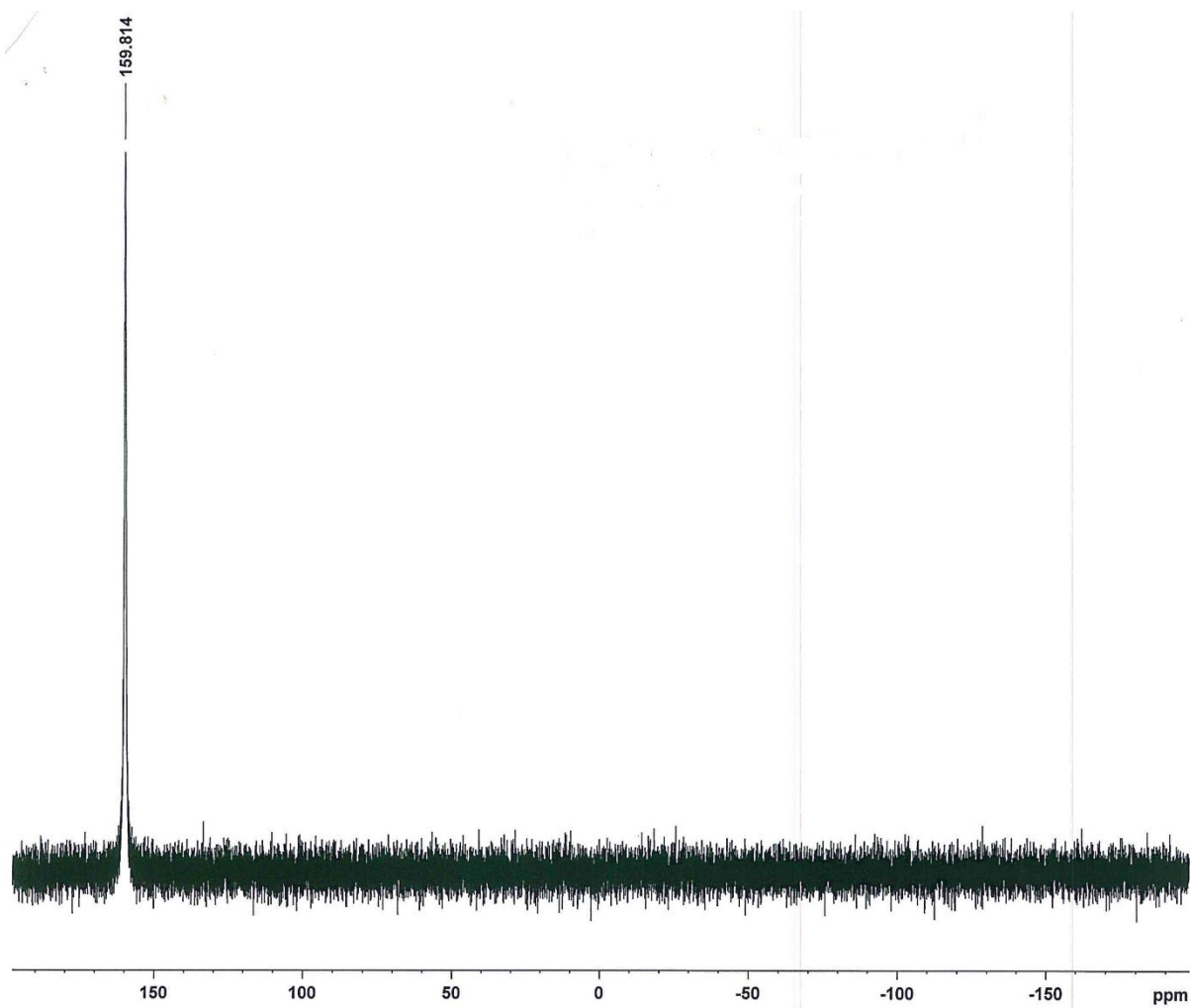


Figure S1.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz, 298 K,  $\text{CD}_3\text{CN}$ ) spectrum of  $[(\text{L}_{\text{OEt}})_2(\text{H}_2\text{O})\text{Ce}^{\text{III}}\{\mu\text{-O,N-MeC(O)NH}\}\text{Ru}^{\text{III}}(\text{L}_{\text{OEt}})\text{Cl}_2]$  (**4**).

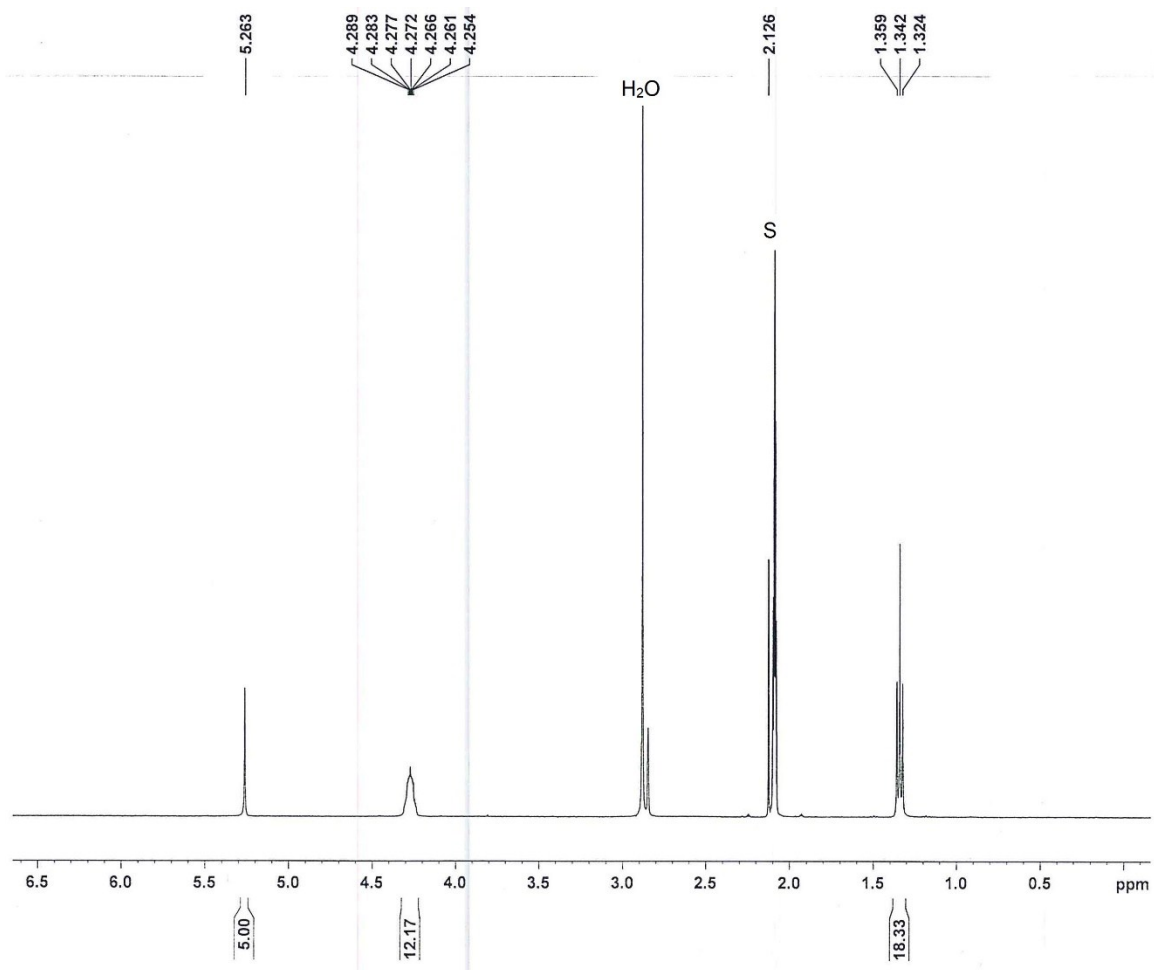


Figure S2. <sup>1</sup>H NMR (400 MHz, 298 K, acetone-*d*<sub>6</sub>) spectrum of [Ce(LOEt)<sub>2</sub>(H<sub>2</sub>O){Mn(N)(CN)<sub>4</sub>}] (**5**) (S = residual solvent).

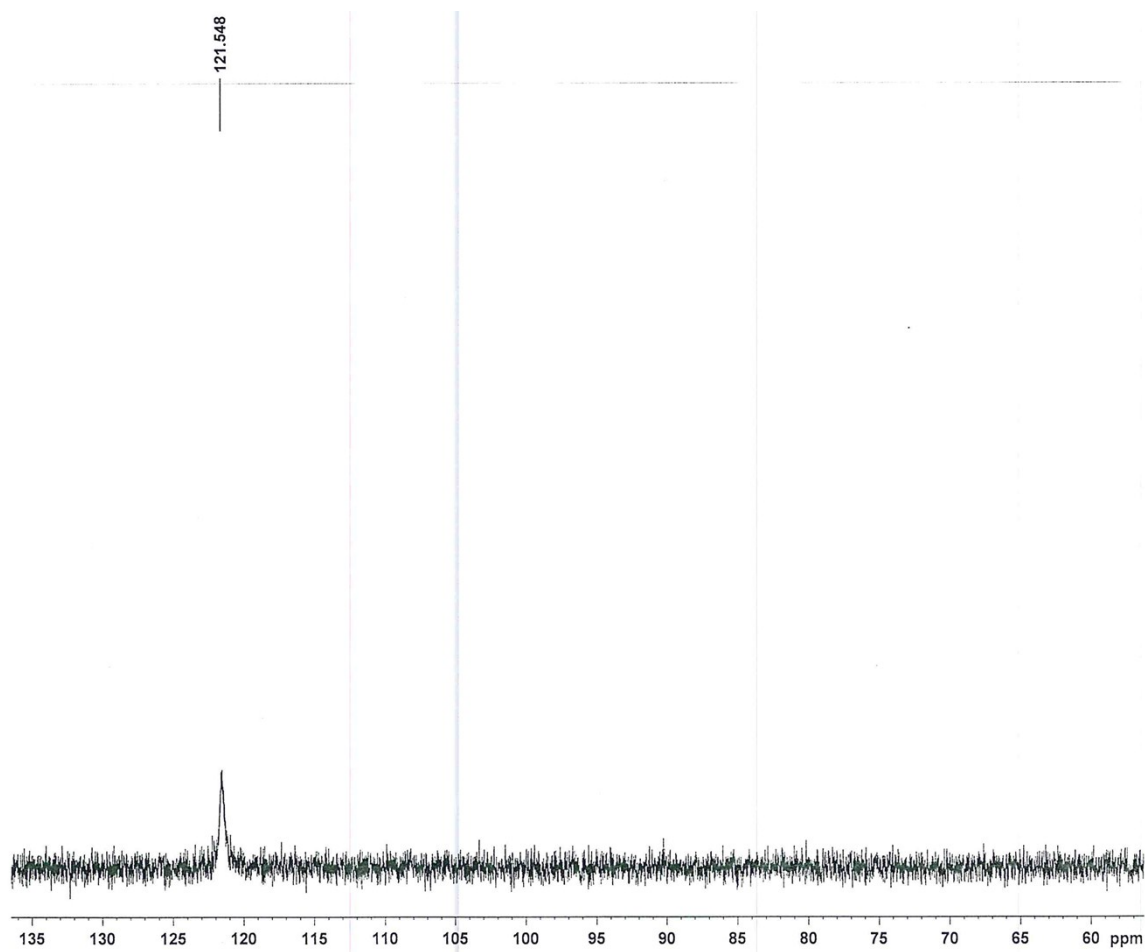


Figure S3.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz, 298 K, acetone- $d_6$ ) spectrum of  $[\text{Ce}(\text{L}_{\text{OEt}})_2(\text{H}_2\text{O})\{\text{Mn}(\text{N})(\text{CN})_4\}]$  (**5**).

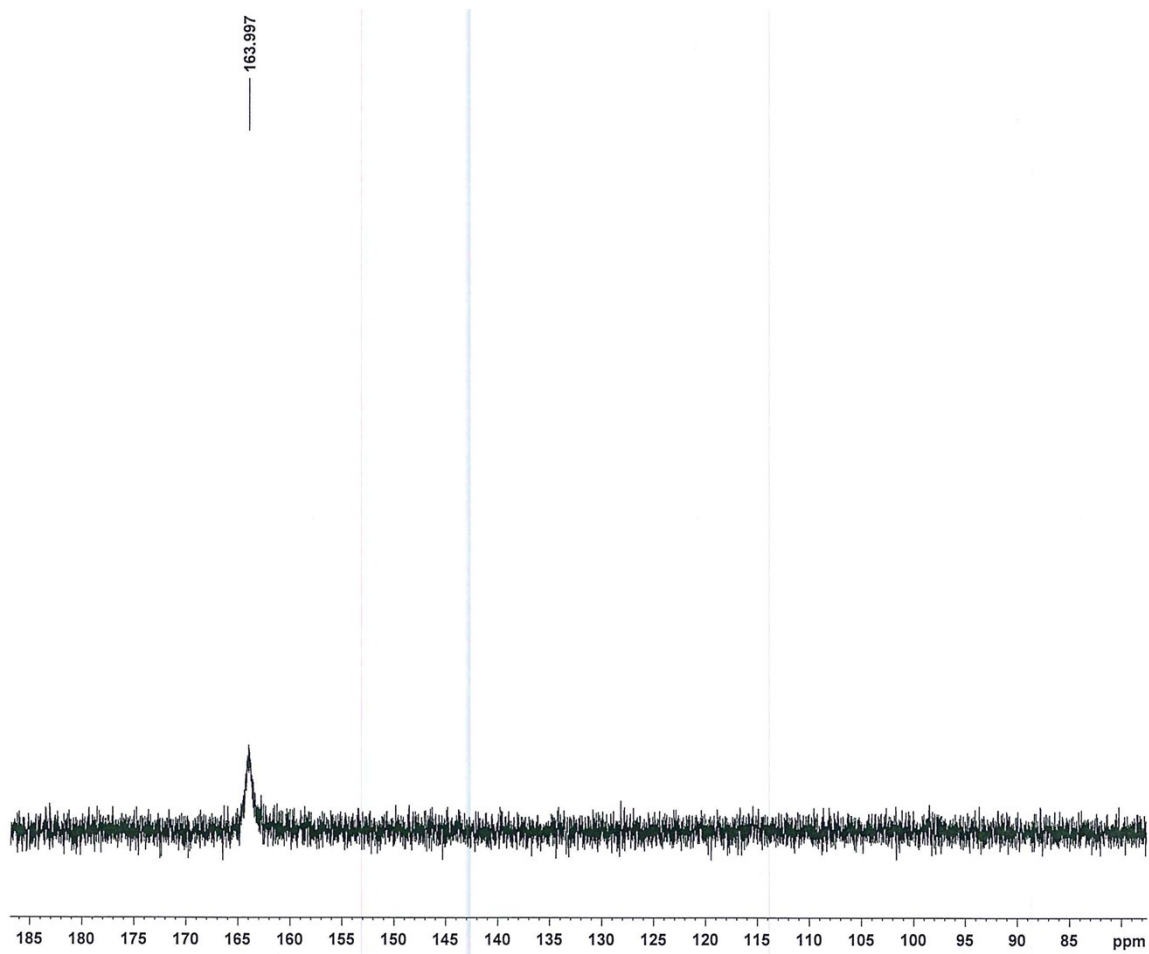


Figure S4.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz, 298 K, acetone- $d_6$ ) spectrum of  $[(\text{L}_{\text{OEt}})_2\text{Ce}^{\text{III}}\{(\mu\text{-cat})_2\text{Ru}^{\text{VI}}(\text{N})\}]$  (6).

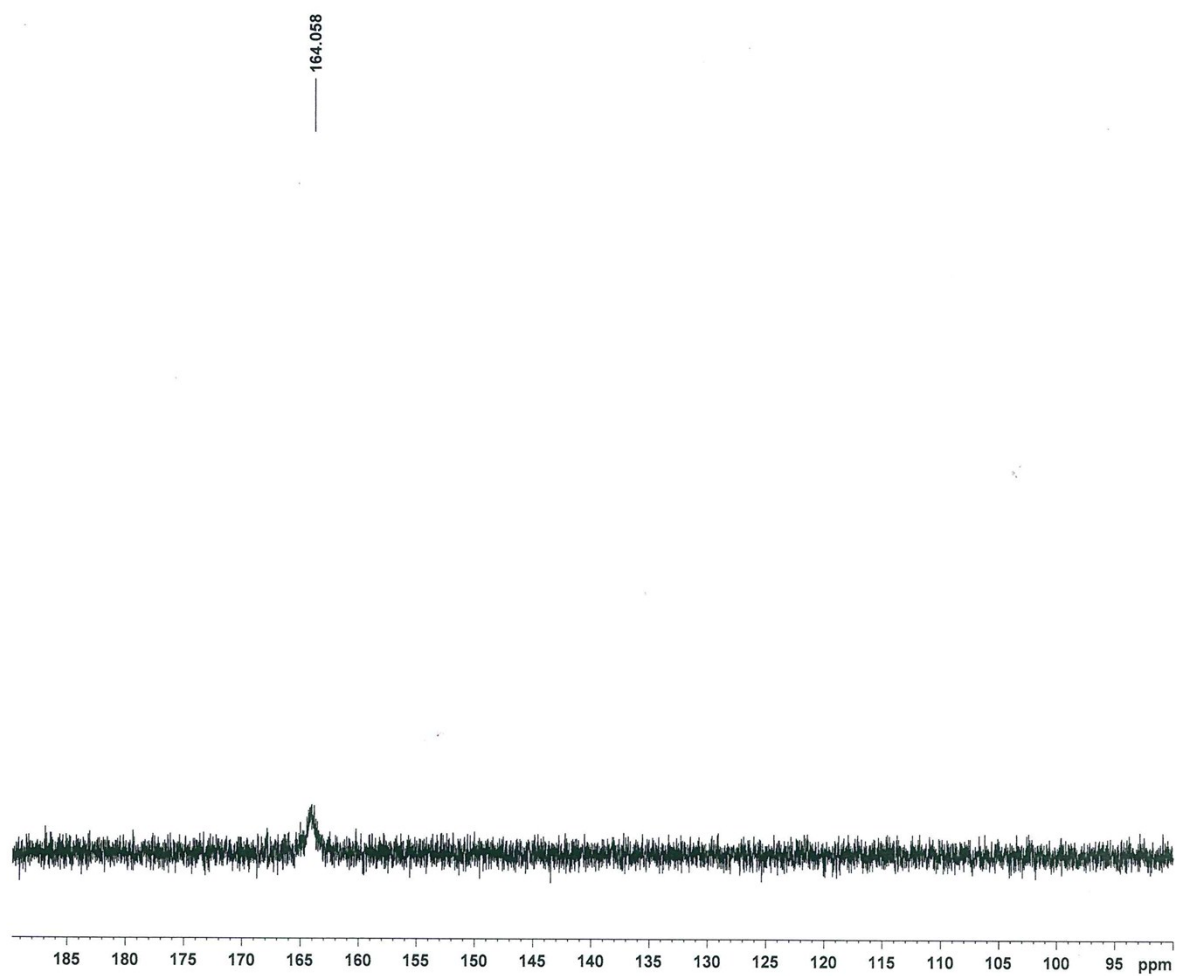


Figure S5.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz, 298 K, acetone- $d_6$ ) spectrum of  $[(\text{L}_{\text{OEt}})_2\text{Ce}^{\text{III}}\{(\mu\text{-cat})_2\text{Os}^{\text{VI}}(\text{N})\}]$  (7).

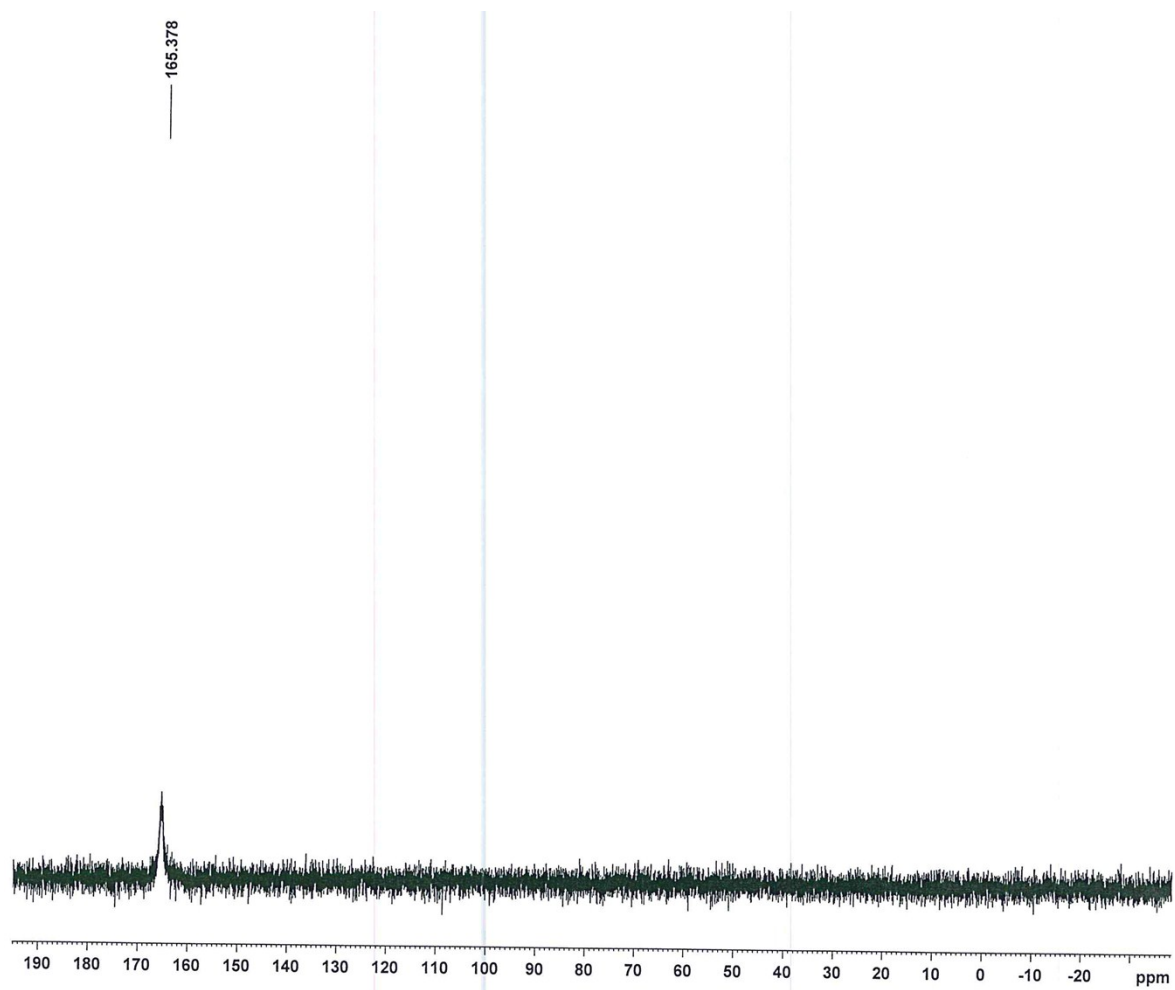


Figure S6.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz, 298 K,  $\text{CDCl}_3$ ) spectrum of  $[(\text{L}_{\text{OEt}})_2\text{Ce}^{\text{III}}\{(\mu\text{-cat})_2\text{Re}^{\text{V}}(\text{O})\}]$  (**8**).



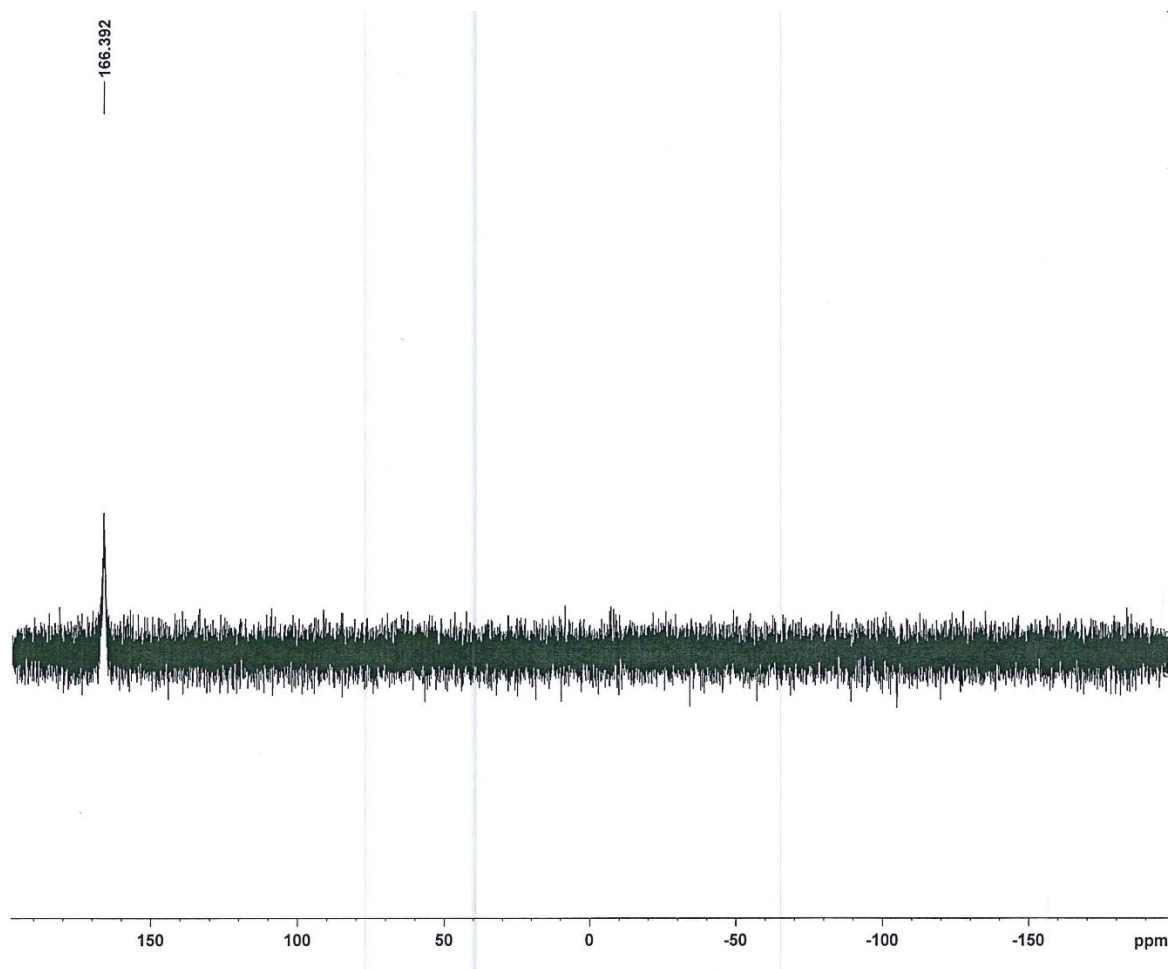


Figure S7.  $^{31}\text{P}\{^1\text{H}\}$  NMR (162 MHz, 298 K,  $\text{CDCl}_3$ ) spectrum of  $[\text{Ce}^{\text{III}}(\text{LOEt})_2(\text{H}_2\text{O})_2][\text{cis}\text{-}\{\text{Re}^{\text{VII}}(\text{O})_2(\text{cat})_2\}]$  (**9**).

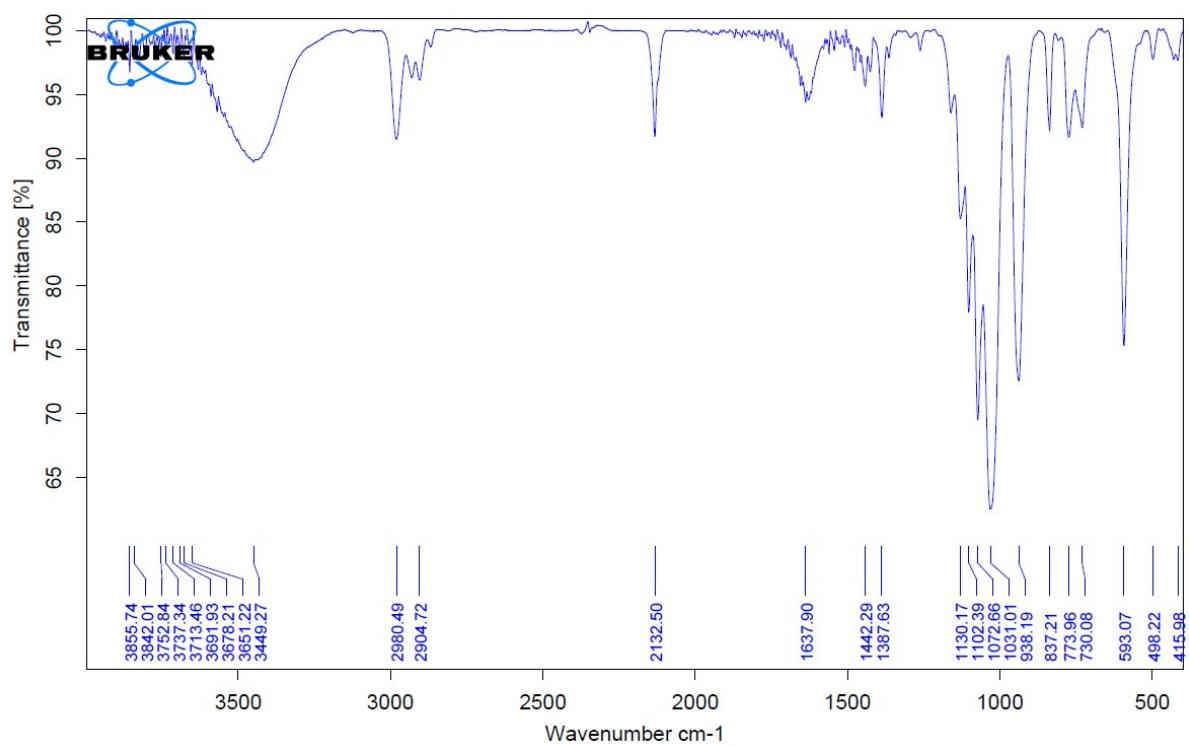


Figure S8. FT-IR spectrum of  $[\text{Ce}(\text{L}_{\text{OEt}})_2(\text{H}_2\text{O})\{\text{Mn}(\text{N})(\text{CN})_4\}]$  (**5**).

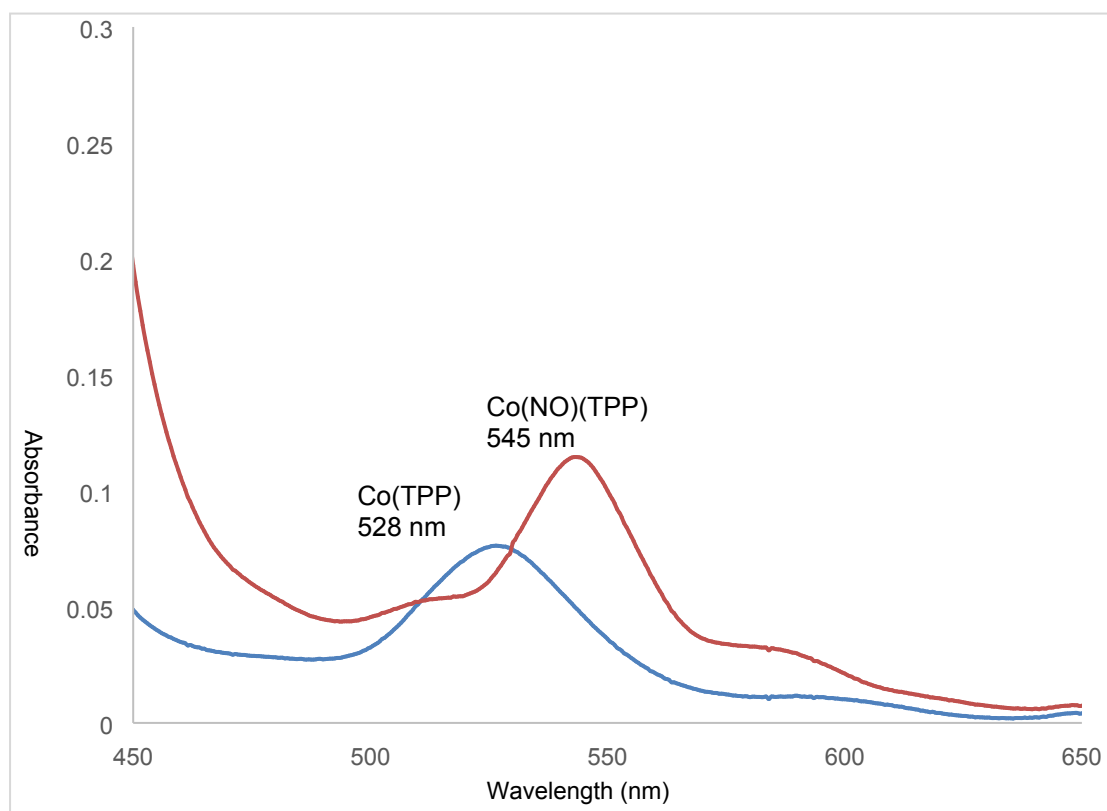


Fig. S9. UV-visible spectra (450-700 nm region) of [Co(TPP)] (0.059 mM in THF, blue) and [Co(NO)(TPP)] species that was generated upon reacting [Co(TPP)] with NO released from the reaction of **1** (10 mg, 0.076 mmol) with **2** (5.5 mg, 0.076 mmol) in tetrahydrofuran at room temperature.

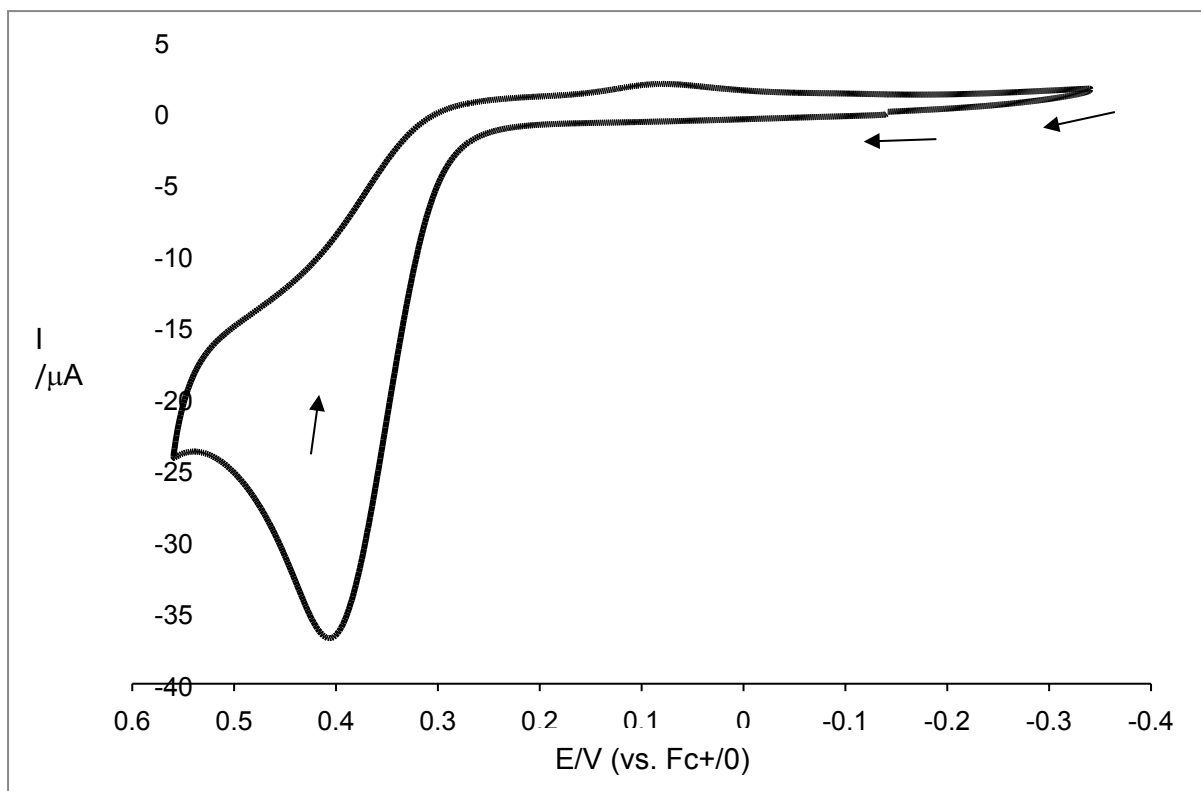


Figure S10. CV of  $[\text{nBu}_4\text{N}][\text{Ru}(\text{N})(\text{cat})_2]$ ; measured at a glassy carbon electrode in  $\text{CH}_2\text{Cl}_2$ , supporting electrolyte: 0.2 M of  $[\text{nBu}_4\text{N}][\text{PF}_6]$ , scan rate =  $100 \text{ mVs}^{-1}$ .

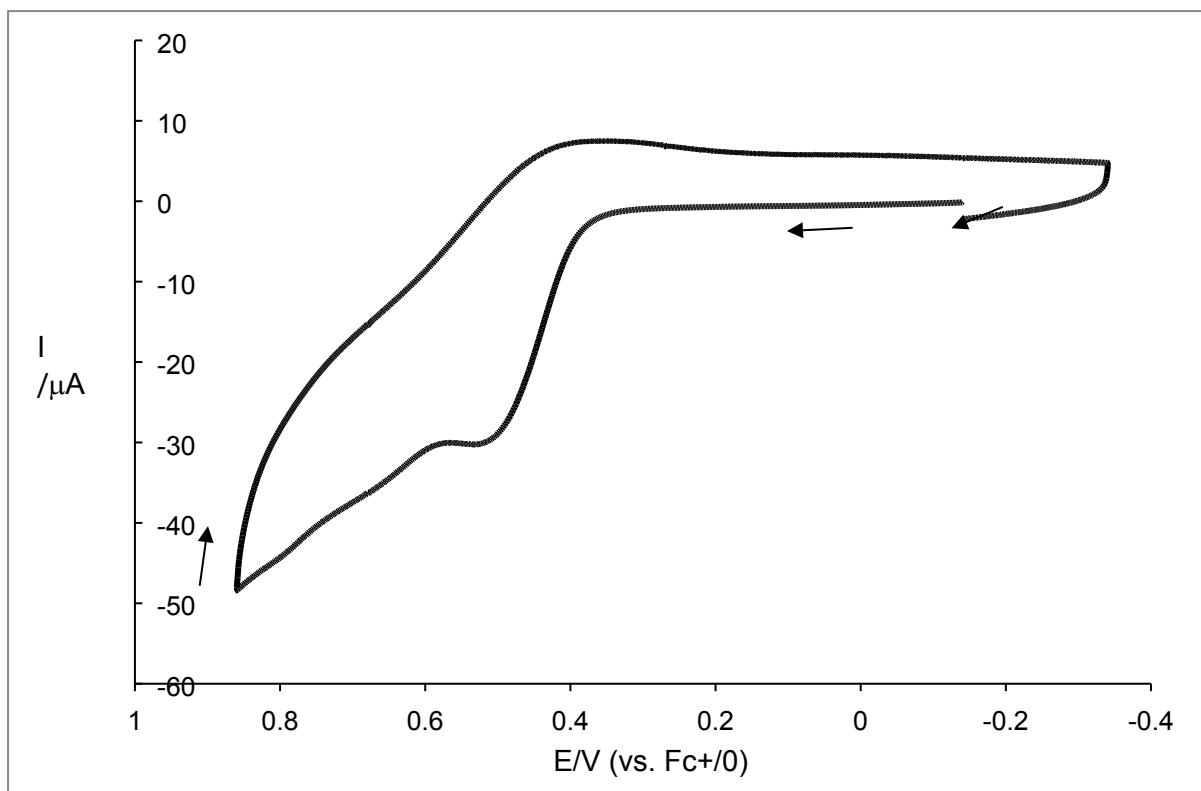


Figure S11. CV of  $[\text{nBu}_4\text{N}][\text{Os}(\text{N})(\text{cat})_2]$ ; measured at a glassy carbon electrode in  $\text{CH}_2\text{Cl}_2$ , supporting electrolyte: 0.2 M of  $[\text{nBu}_4\text{N}][\text{PF}_6]$ , scan rate =  $100 \text{ mVs}^{-1}$ .