

## **Water-assisted proton conductivity of two highly stable imidazole multi-carboxylate-based MOFs**

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### **Supporting Information**

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**Table S1.** Selected bond distances (Å) and angles (deg) for compound **1**

Mn(1)-N(1)	2.272(2)	Mn(1)-O(7)	2.171(2)
Mn(1)-N(3)	2.219(2)	Mn(1)-O(8)	2.187(2)
Mn(1)-O(1)	2.247(2)	O(5)#2-Mn(1)-O(1)	85.37(8)
Mn(1)-O(5)#2	2.161(2)	O(5)#2-Mn(1)-O7	172.90(8)
N(3)-Mn(1)-N(1)	96.14(9)	O(5)#2-Mn(1)-O(8)	86.54(8)
N(3)-Mn(1)-O(1)	170.04(9)	O(7)-Mn(1)-N(1)	93.47(9)
O(1)-Mn(1)-N(1)	74.38(8)	O(7)-Mn(1)-N(3)	94.71(9)
O(5)#2-Mn(1)-N(1)	89.19(8)	O(7)-Mn(1)-O(1)	89.00(8)
O(5)#2-Mn(1)-N(3)	91.55(9)	O(8)-Mn(1)-N(3)	92.16(9)
O(7)-Mn(1)-O(8)	89.86(9)	O(8)-Mn(1)-O(1)	97.09(8)
O(8)-Mn(1)-N(1)	170.77(8)		

Symmetry codes: #1: -X, 2-Y, 1-Z; #2: -X, 2-Y, -Z

**Table S2.** Hydrogen bonding parameters for **1**

D-H...A	d(H...A)	d(D...A)	∠(DHA)
N(2)H(2)...O(11) #1	2.00	2.850(4)	167.3
O(3)H(3)...O(2)	1.70(4)	2.516(3)	175(4)
O(7)H(7A)...O(10) #2	1.93	2.675(4)	142.5
O(7)H(7B)...O(3) #3	2.87	3.566(3)	154.0
O(7)H(7B)...O(4) #3	2.12	2.822(3)	153.0
O(8)H(8B)...O(1) #4	1.95	2.778(3)	159.6
O(10)H(10A)...O(9)	1.88	2.730(6)	161.9
O(10)H(10B)...O(2)#5	2.38	3.231(4)	155.7
O(11)H(11A)...O(6)#1	2.15	2.882(5)	141.4

Symmetry codes: #1: 1-X, 2-Y, 1-Z; #2: 1-X, 1-Y, 1-Z; #3: 1-X, 2-Y, -Z; #4: -X, 1-Y, -Z; #5 1+X, +Y, 1+Z

**Table S3.** Selected bond distances (Å) and angles (deg) for compound **2**

N(1)-Zn(1)	2.050(2)	O(1)-Zn(1)	2.149(2)
N(2)-Zn(2)	2.000(2)	O(5)-Zn(1)#2	1.956(2)
N(3)-Zn(1)	2.093(3)	O(7)-Zn(2)	2.488(3)
N(4)-Zn(1)	2.144(3)	O(10)-Zn(3)	2.303(2)
N(5)-Zn(2)	2.027(2)	Zn(1)-O(5)#2	1.956(2)
N(6)-Zn(3)	2.045(2)	Zn(3)-N(6)#1	2.045(2)
N(7)-Zn(2)	2.112(3)	Zn(3)-N(9)#1	2.275(7)
N(8)-Zn(2)	2.095(3)	Zn(3)-O(10)#1	2.303(2)
N(9)-Zn(3)	2.275(7)	N(2)-Zn(2)-N(5)	138.94(10)
N(1)-Zn(1)-N(3)	121.80(9)	N(2)-Zn(2)-N(7)	113.73(11)
N(1)-Zn(1)-N(4)	101.69(10)	N(2)-Zn(2)-N(8)	105.42(10)
N(1)-Zn(1)-O(1)	80.21(9)	N(2)-Zn(2)-O(7)	81.95(9)
N(3)-Zn(1)-N(4)	77.06(10)	N(5)-Zn(2)-N(7)	98.24(11)
N(3)-Zn(1)-O(1)	91.00(10)	N(5)-Zn(2)-N(8)	105.89(10)
N(4)-Zn(1)-O(1)	167.06(10)	N(5)-Zn(2)-O(7)	74.45(9)
O(5)#2-Zn(1)-N(1)	106.17(9)	N(7)-Zn(2)-O(7)	86.94(12)
O(5)#2-Zn(1)-N(3)	132.01(10)	N(8)-Zn2-N(7)	78.05(12)
O(5)#2-Zn(1)-N(4)	95.71(11)	N(8)-Zn(2)-O(7)	164.91(11)
O(5)#2-Zn(1)-O(1)	96.02(10)	N(6)1-Zn(3)-O(10)#1	76.85(9)
O(10)1-Zn(3)-O(10)	85.28(14)	N(6)-Zn(3)-N(6)#1	154.29(15)
N(9)-Zn(3)-N(9)#1	105.9(3)	N(6)#1-Zn(3)-N(9)#1	100.74(17)
N(9)#1-Zn(3)-O(10)	84.43(19)	N(6)-Zn(3)-N(9)#1	94.69(17)
N(9)#1-Zn(3)-O(10)#1	169.62(19)	N(6)-Zn(3)-N(9)	100.74(17)
N(9)-Zn(3)-O(10)#1	84.43(19)	N(6)#1-Zn(3)-N(9)	94.69(17)
N(9)-Zn(3)-O(10)	169.62(19)	N(6)-Zn(3)-O(10)	76.85(9)
N(6)-Zn(3)-O(10)#1	84.27(10)	N(6)#1-Zn(3)-O(10)	84.27(10)

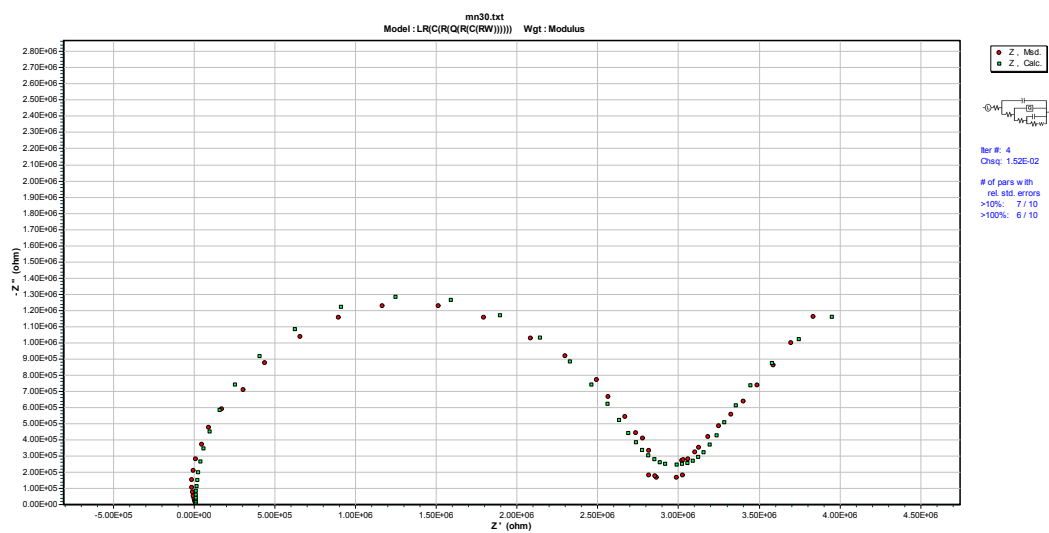
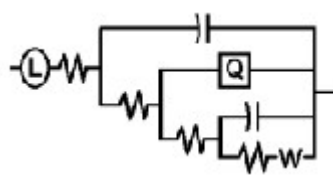
Symmetry transformations used to generate equivalent atoms: #1: +X,3/2-Y,1/2-Z; #2 1/2-X,1-Y,+Z.

**Table S4.** Hydrogen bonding parameters for **2**

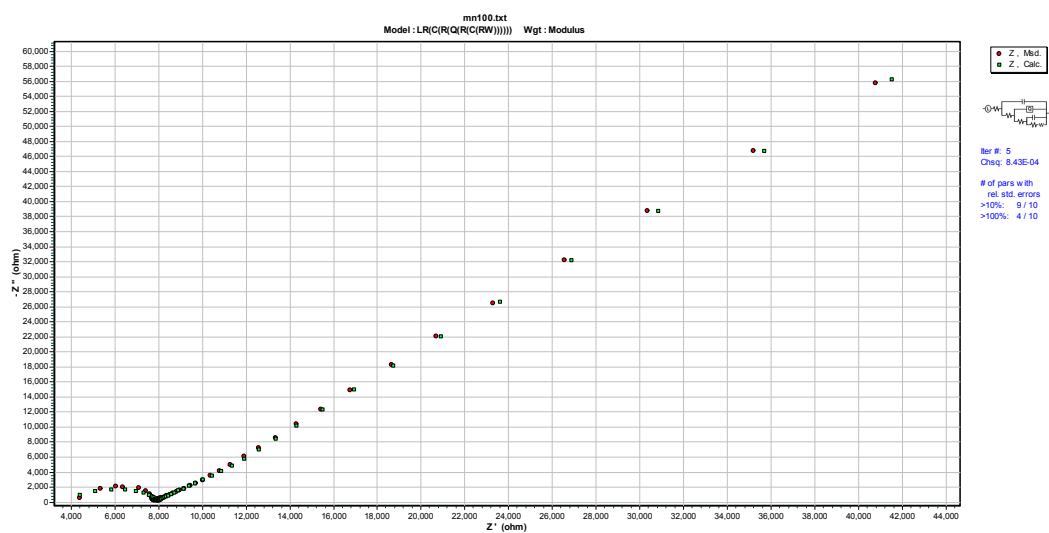
D-H...A	d(H...A)	d(D...A)	∠(DHA)
O(3)-H(3) ...O(2)	1.61(2)	2.499(3)	176(5)
O(12)-H(12) ...O(8)#1	2.38	3.297(6)	167.5
O(14)-H(14A) ...O(9)#2	1.92	2.751(5)	166.7
O(14)-H(14A) ...O(10)#2	2.81	3.487(5)	137.4
O(14)-H(14B) ...O(15)#1	2.06	2.823(7)	148.6
O(15)-H(15A) ...O(10)#3	2.16	2.958(5)	147.2
O(15)-H(15B) ...O(4)	2.39	3.014(5)	130.8
O(8)-H(8A) ...O(9)	1.62(10)	2.506(4)	160(9)

Symmetry codes: #1: 1/2+X,+Y,1-Z; #2 : 1/2+X,3/2-Y,1/2+Z; #3 : +X,3/2-Y,1/2-Z

Equivalent circuit LR(C(R(Q(R(C(RW))))))):

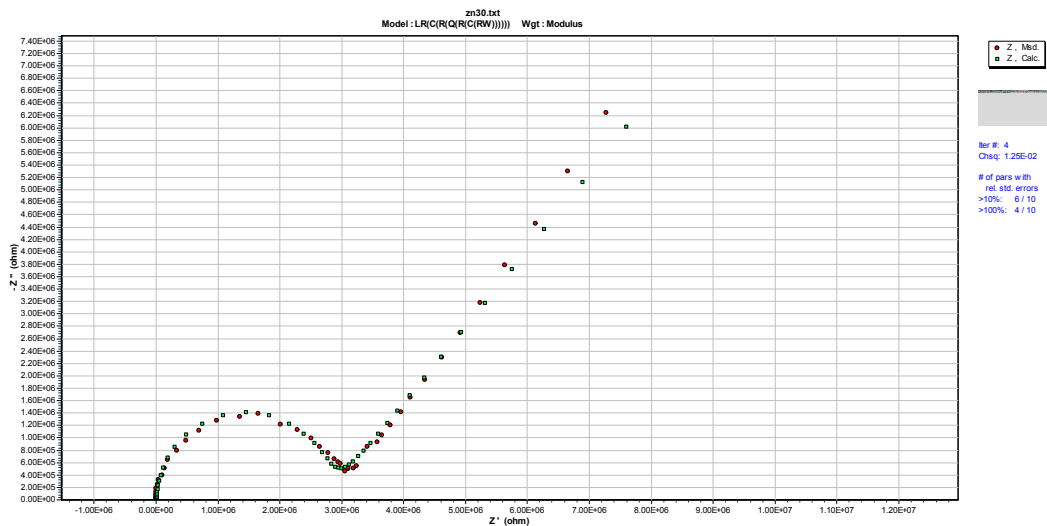


(a)

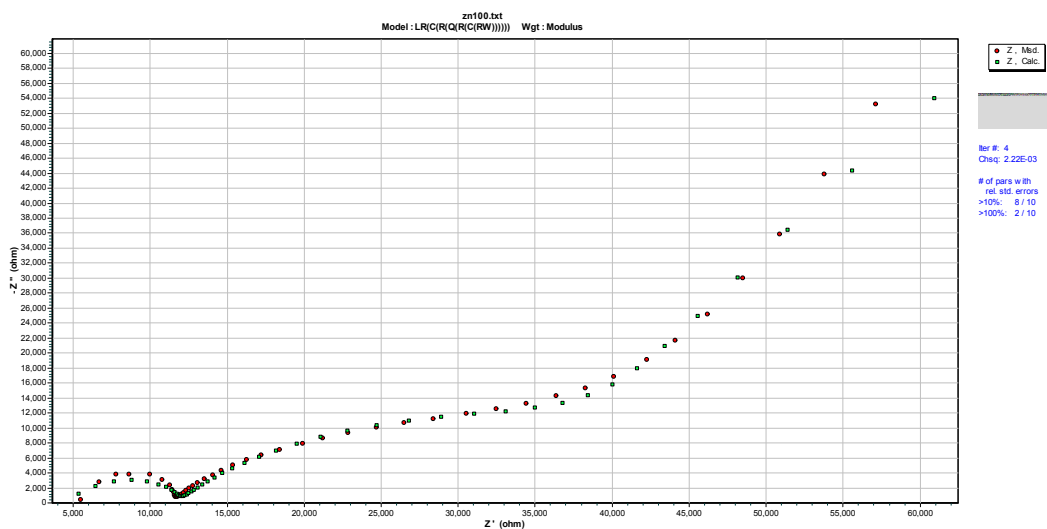


(b)

**Fig. S1** Nyquist plots for a polycrystalline sample of **1** at 30 °C (a) and 100 °C (b) and 98% RH. Red circle and green square are the measured impedance spectroscopy values and the fits of the impedance data to the equivalent circuit of LR(C(R(Q(R(C(RW)))))).

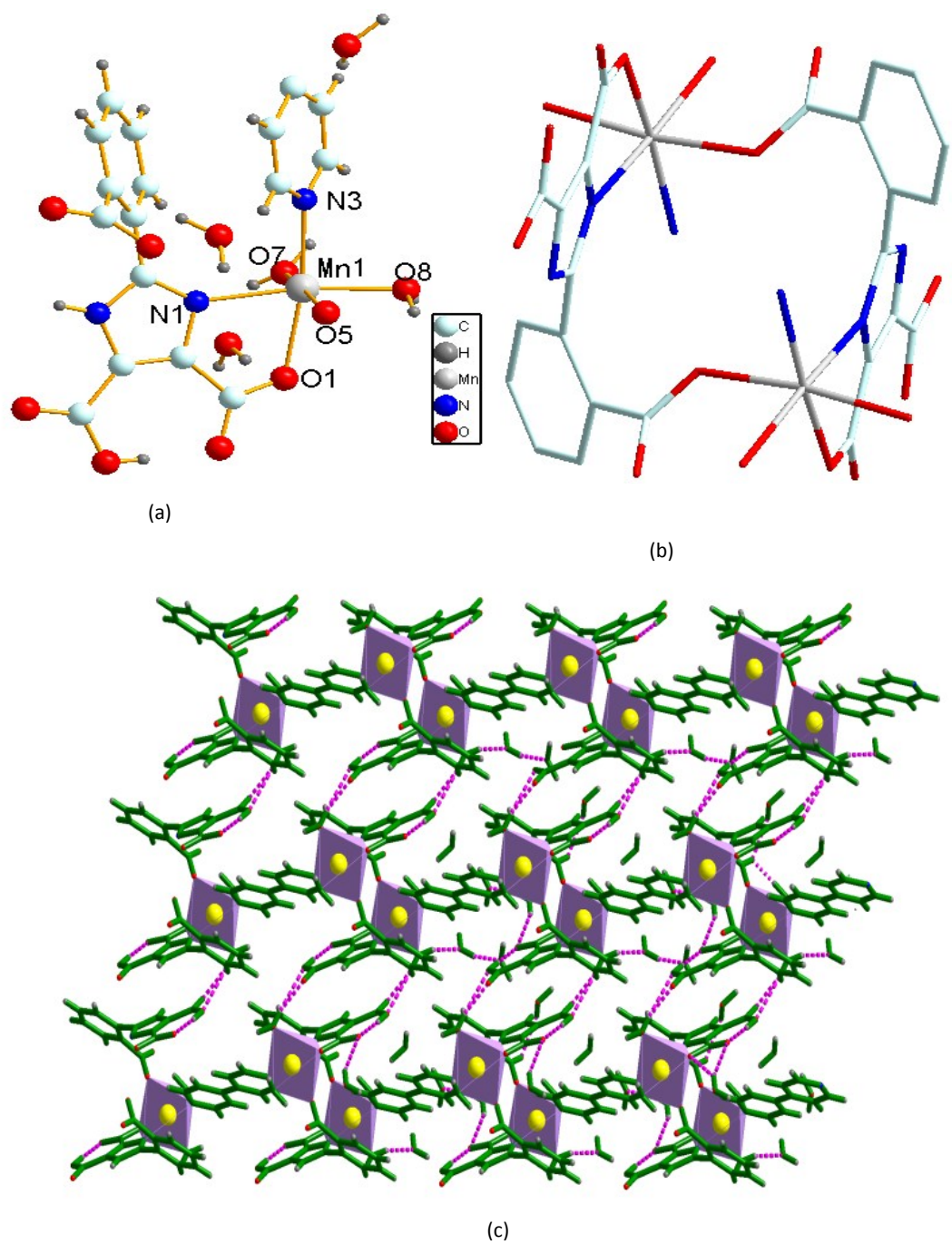


(a)



(b)

**Fig. S2** Nyquist plots for a polycrystalline sample of **1** at 30°C and 100°C and 98% RH. Red circle and green square are the measured impedance spectroscopy values and the fits of the impedance data to the equivalent circuit of LR(C(R(Q(RC(RW)))))).



**Fig. S3** (a) Structural unit of **1**. (b) Binary structure of **1** formed by bridging ligands.(c) Two-dimensional sheet of **1** connected by H-bonds.

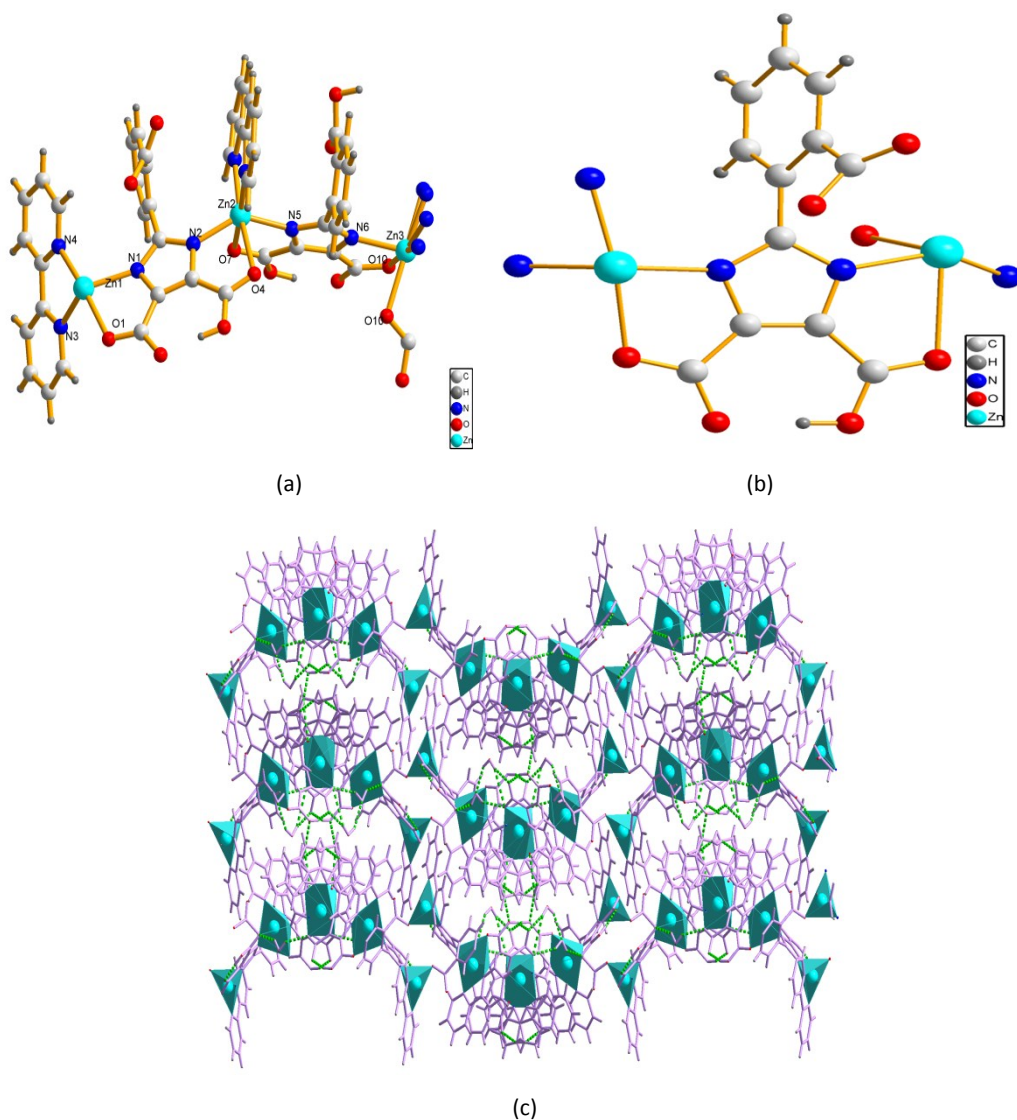


Fig. S4 (a) Structural unit of **2**. (b) Coordination mode of the imidazole multi-carboxylate ligand of **2**. (c) Two-dimensional sheet of **2** connected by H-bonds.

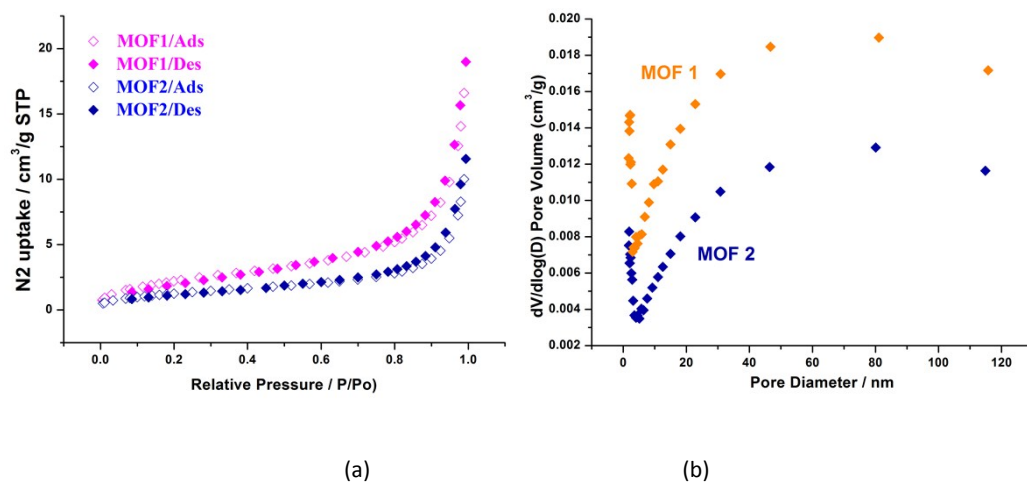


Fig. S5 Nitrogen gas adsorption and desorption isotherms at 77 K (a) and pore size distribution profiles (b) for **1**

and 2.

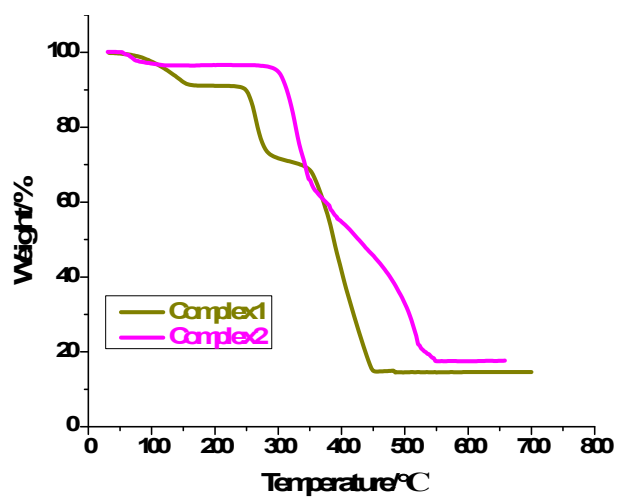
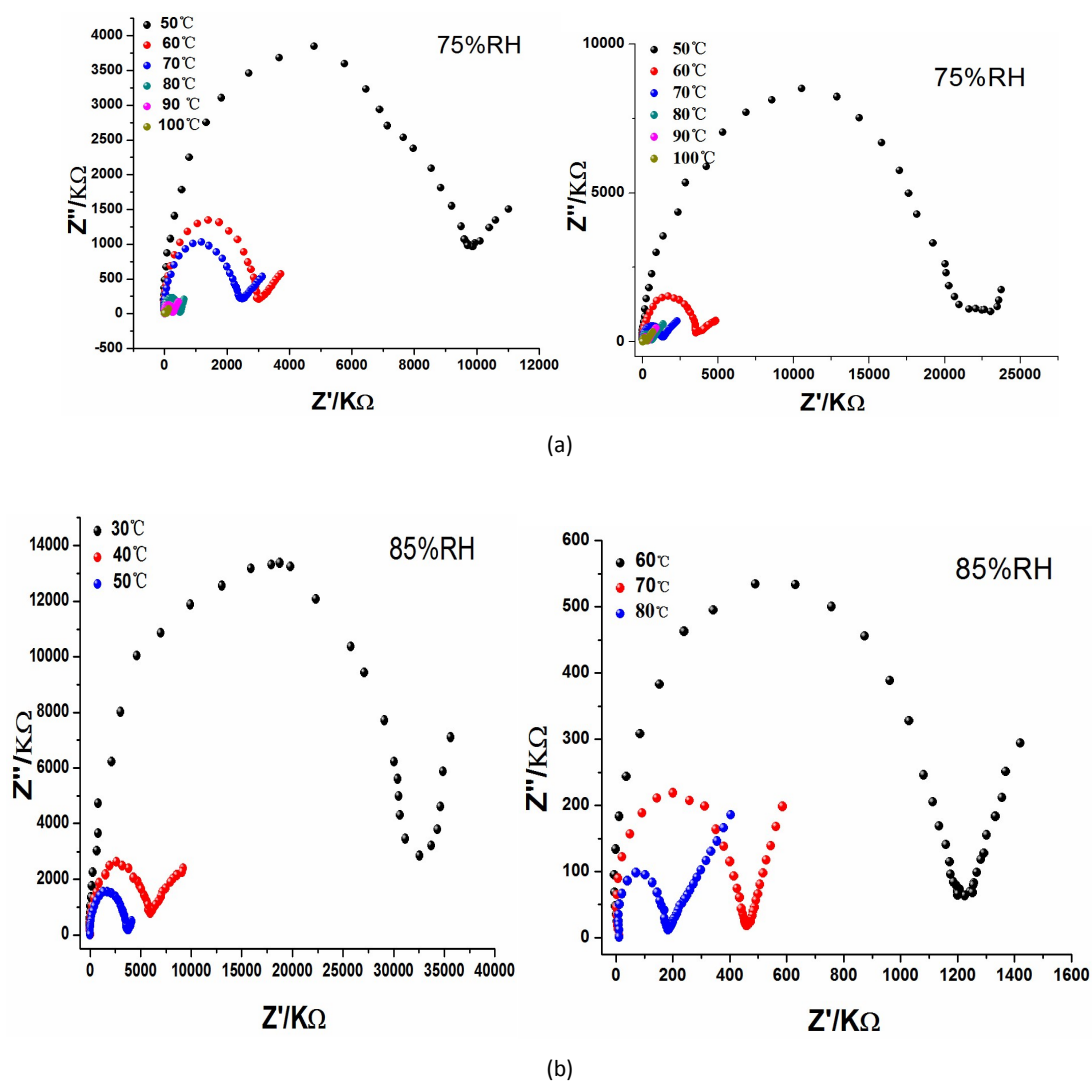
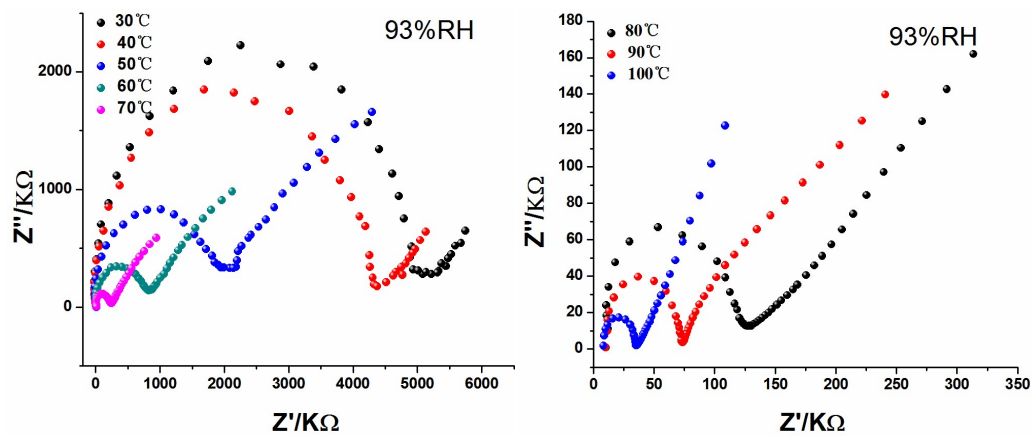


Fig. S6 The TG curves of MOFs 1 and 2.

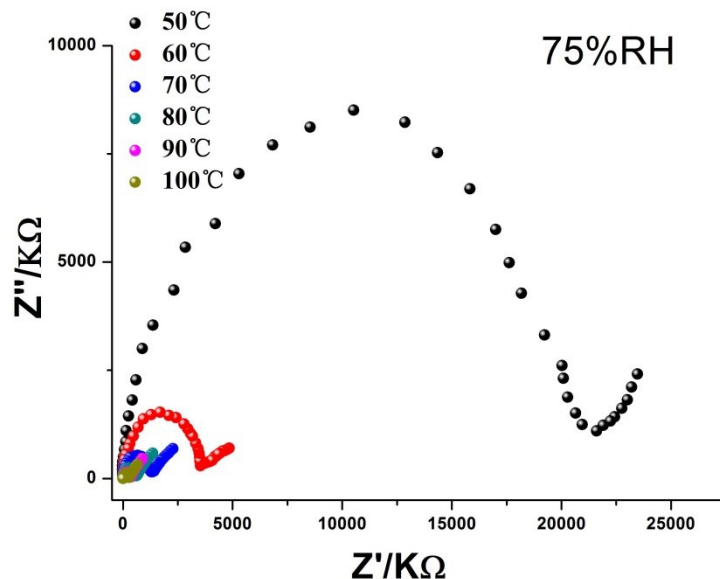




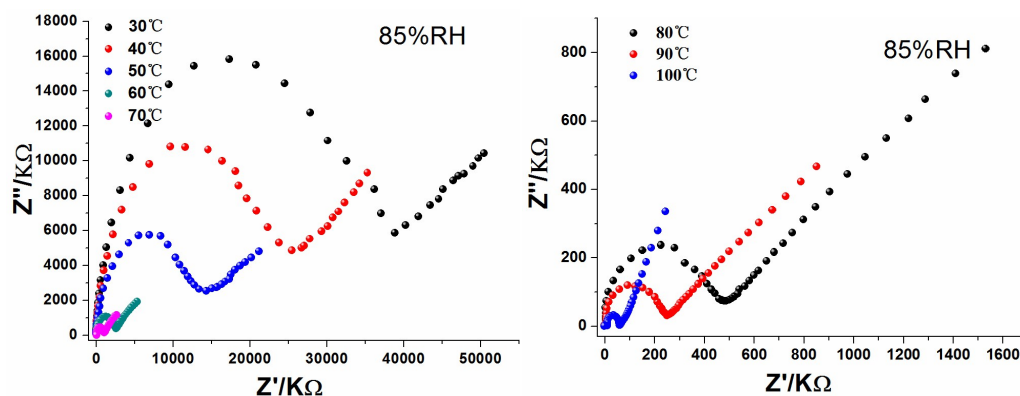


(c)

Fig. S7 Impedance spectra of 1 at 75% (a), 85% (b) and 93% RHs (c) and 30-100°C.



(a)



(b)

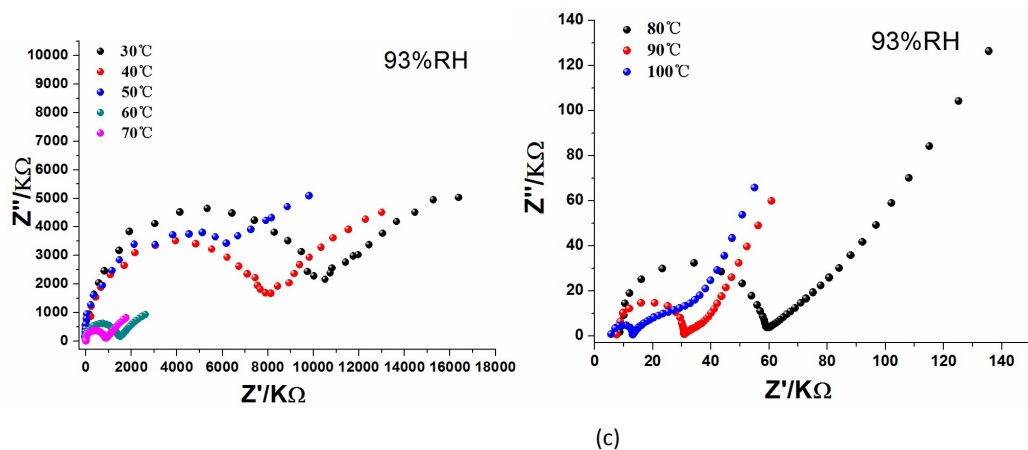


Fig. S8 Impedance spectra of **2** at 75% (a), 85% (b) and 93% RH (c) and 30-100°C.

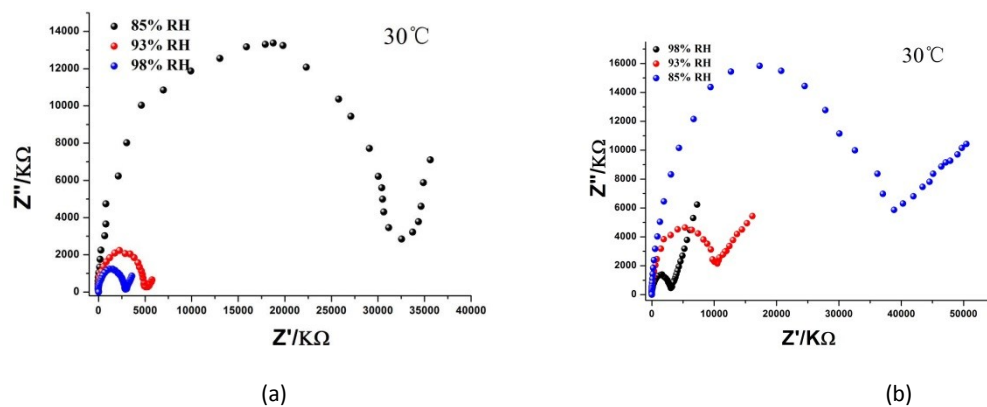


Fig. S9 Impedance spectra of **1** (a) and **2** (b) at different RHs and 30 °C.

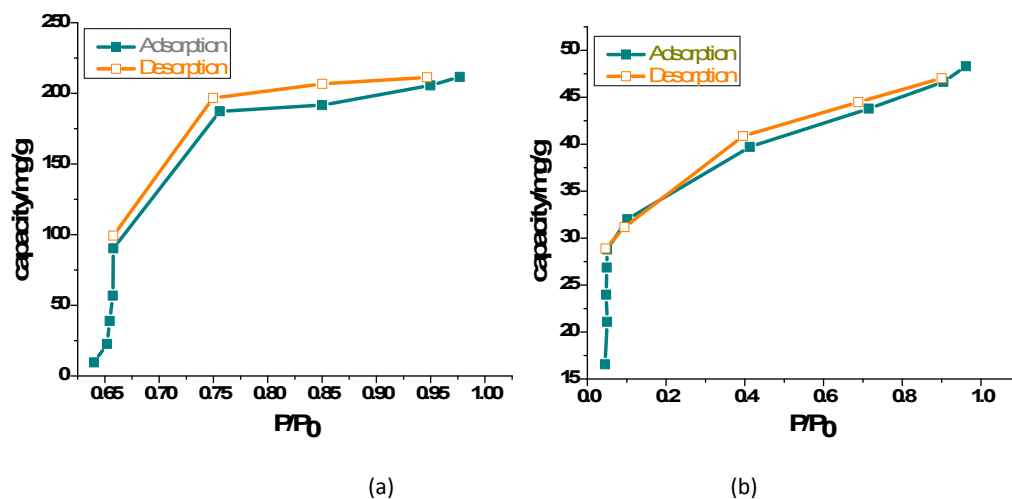
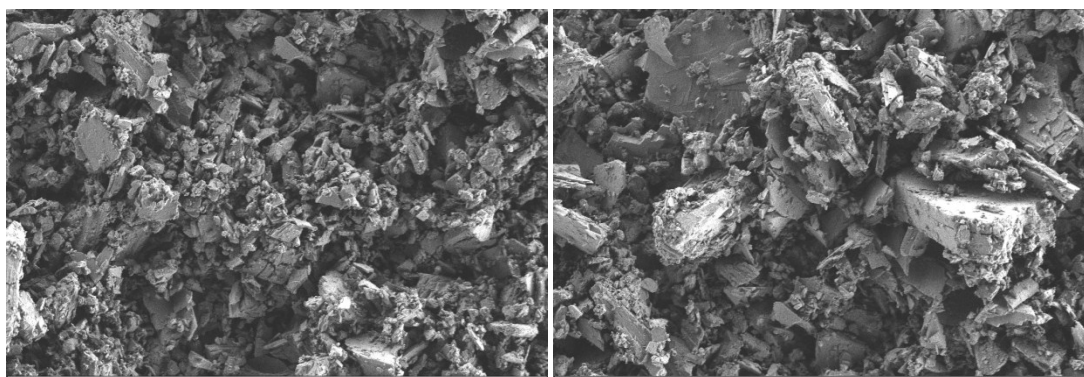


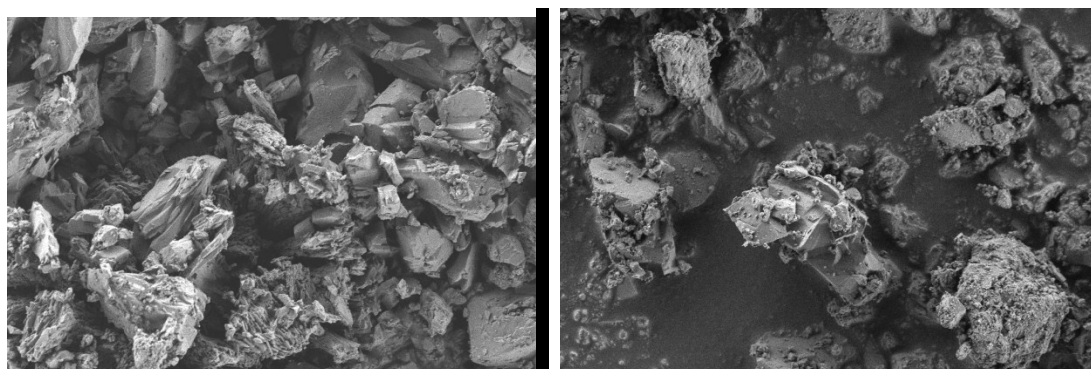
Fig. S10 The adsorption curves of MOFs **1** and **2**.



(a)

(b)

**Fig. S11** SEM images for ground powders of **1** before (a) and after (b) impedance test.



(a)

(b)

**Fig. S12** SEM images for ground powders of **2** before (a) and after (b) impedance test.