

Supporting Information

Porous Cr₂O₃ bead with 3D continuous pore architecture: synthesis and its catalytic performances for decomposition of ammonium perchlorate

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Table S1

The products of Cr₂O₃ beads synthesized via various conditions.

Entry	Symbol	C _(K₂Cr₂O₇) /mol•L ⁻¹	Calcination temperature/°C
1	0.05 M-Cr ₂ O ₃ -MPB	0.05	750
2	0.1 M-Cr ₂ O ₃ -MPB	0.1	750
3	0.2 M-Cr ₂ O ₃ -MPB (750 °C-Cr ₂ O ₃ -MPB)	0.2	750
4	0.4 M-Cr ₂ O ₃ -MPB	0.4	750
5	450 °C-Cr ₂ O ₃ -MPB	0.2	450
6	550 °C-Cr ₂ O ₃ -MPB	0.2	550
7	Commercial Cr ₂ O ₃	—	—

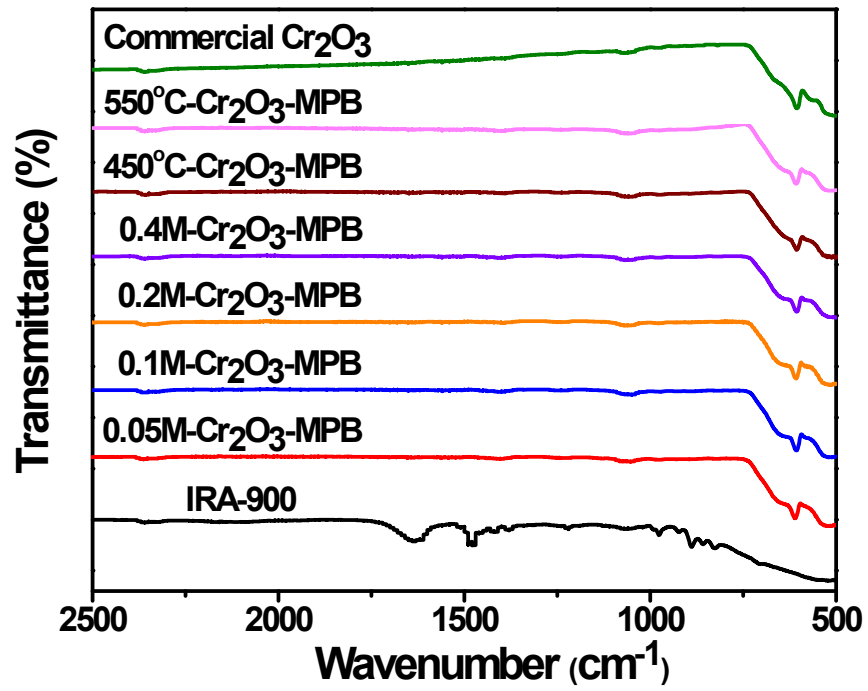


Fig. S1 FT-IR spectra of IRA-900 commercial Cr₂O₃ and synthesized Cr₂O₃-MPB materials.

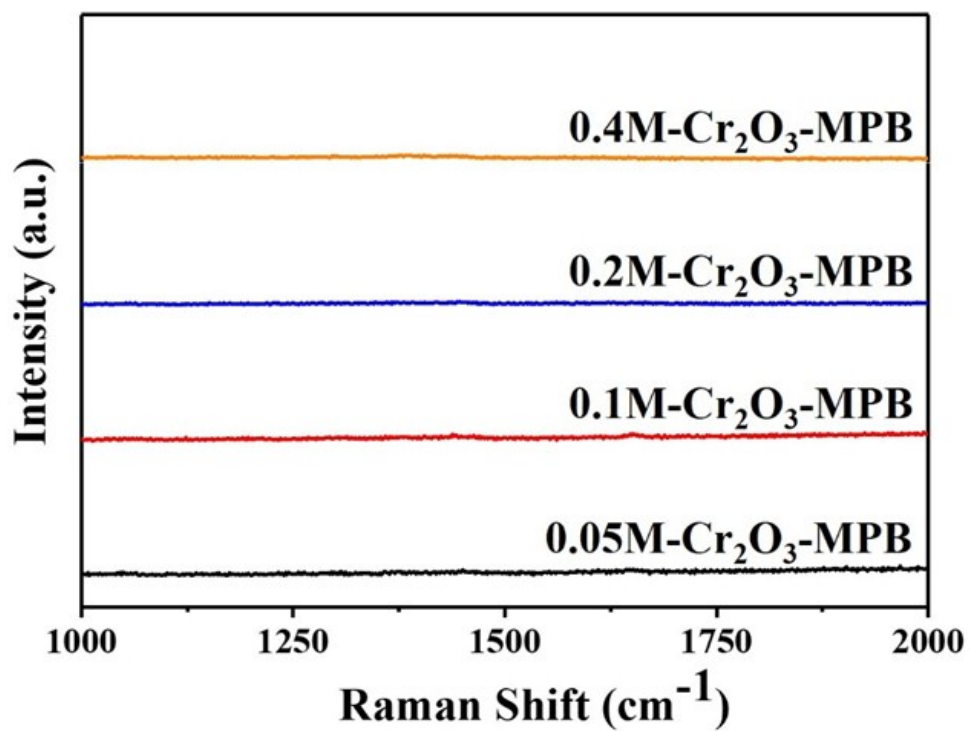


Fig. S2 Raman spectra of 0.05 M-Cr₂O₃-MPB, 0.1 M-Cr₂O₃-MPB, 0.2 M-Cr₂O₃-MPB and 0.4 M-Cr₂O₃-MPB. The spectra were measured at 532 nm laser wavelength.

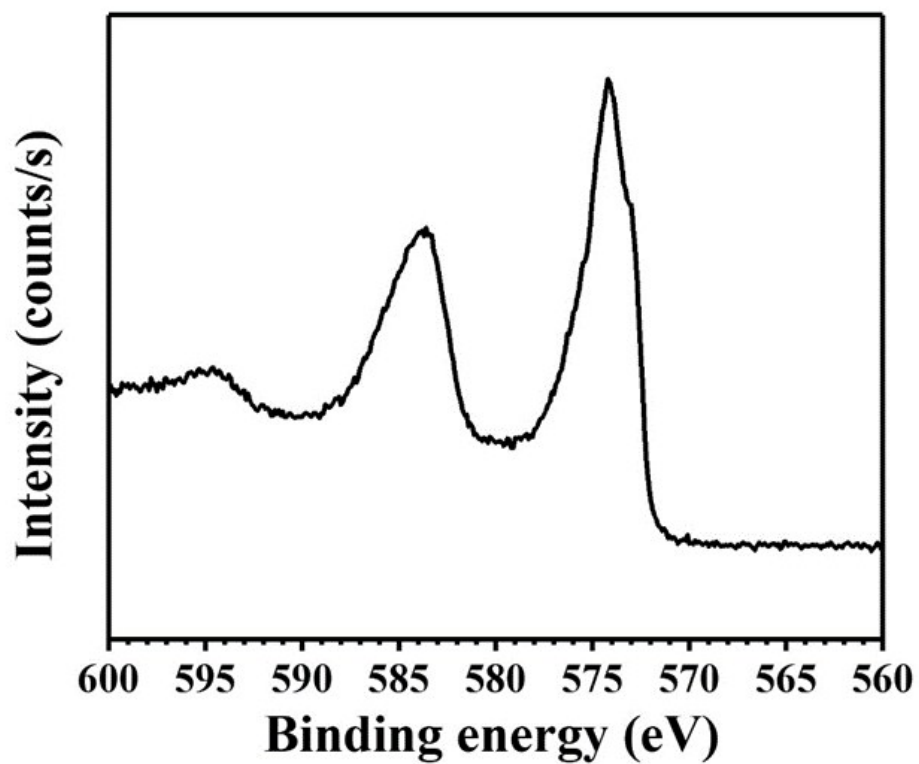


Fig. S3 XPS spectrum of 0.2 M-Cr₂O₃-MPB. The instrument (ESCALAB 250Xi, ThermoFisher) was used for the analysis.

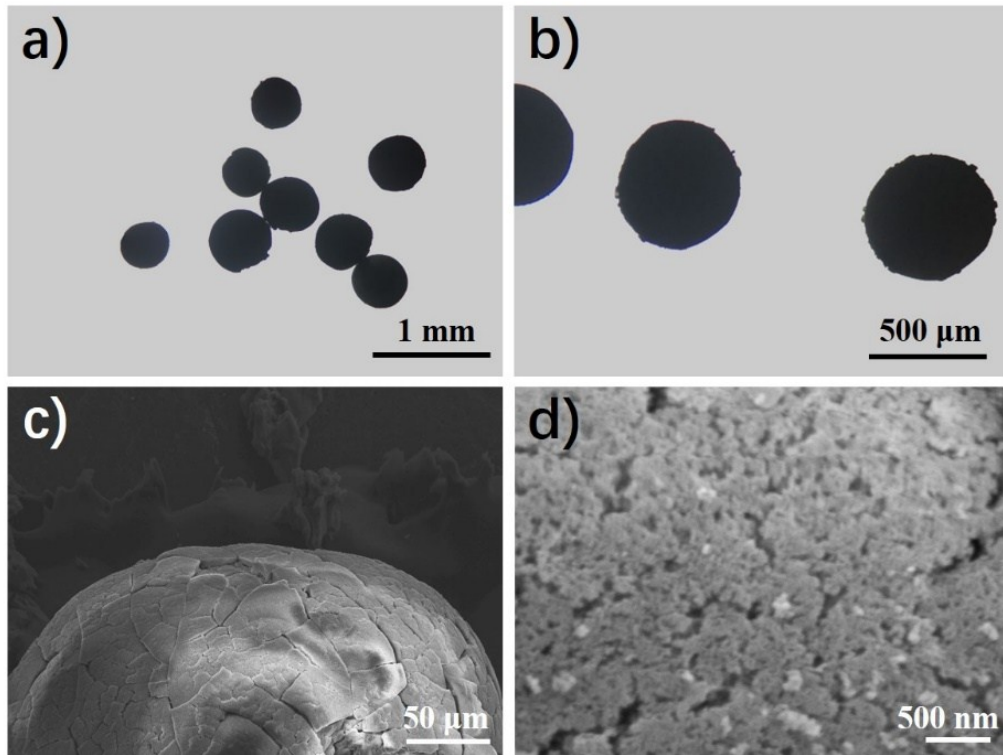


Fig. S4 a) and b) Microscopic images of 0.2 M-Cr₂O₃-MPB with different magnification; c) and d) SEM image of the surface of 0.2 M-Cr₂O₃-MPB.

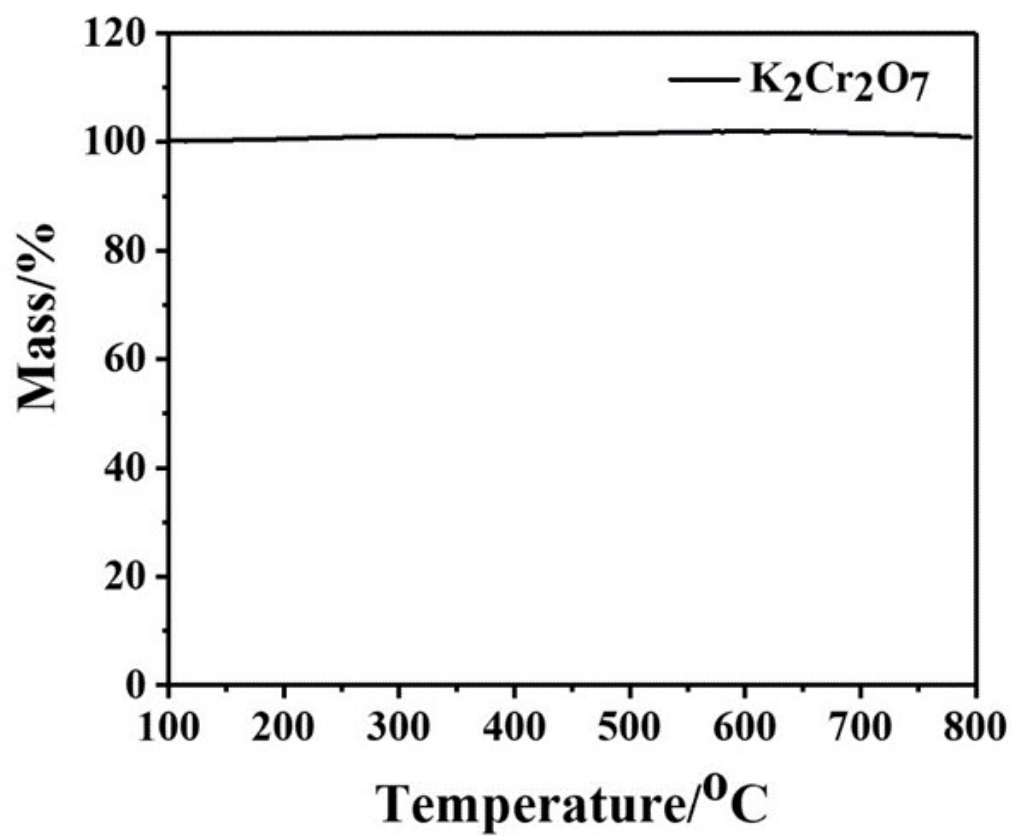


Fig. S5 TG curve of $K_2Cr_2O_7$ powders in air atmosphere.

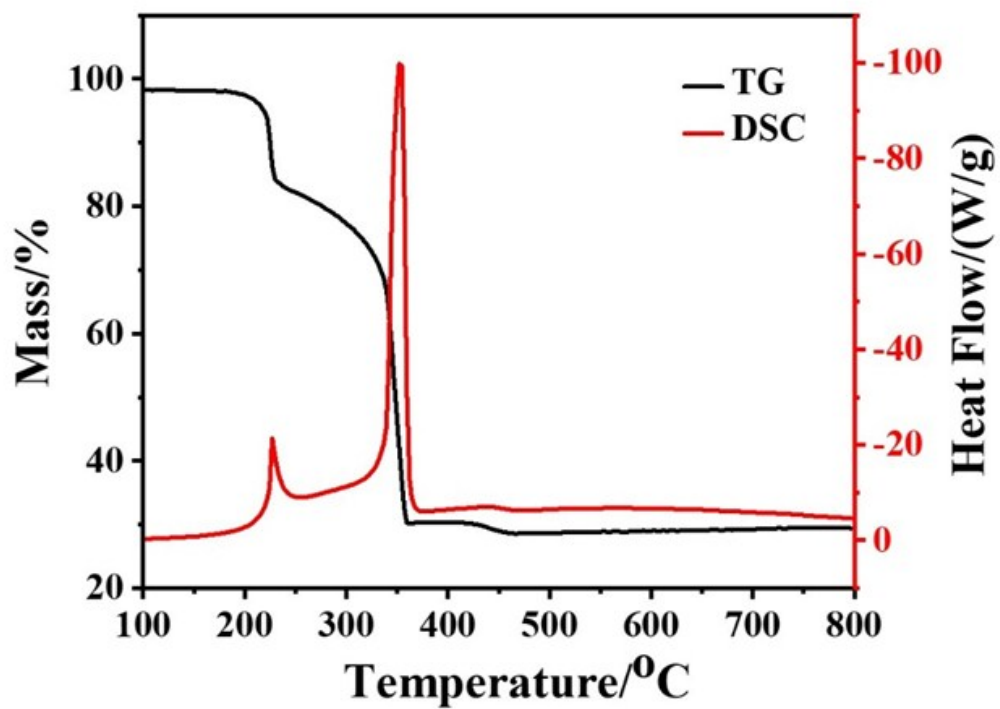


Fig. S6 TG-DSC curves of IRA-900 resin beads containing $\text{Cr}_2\text{O}_7^{2-}$ anions in air atmosphere.

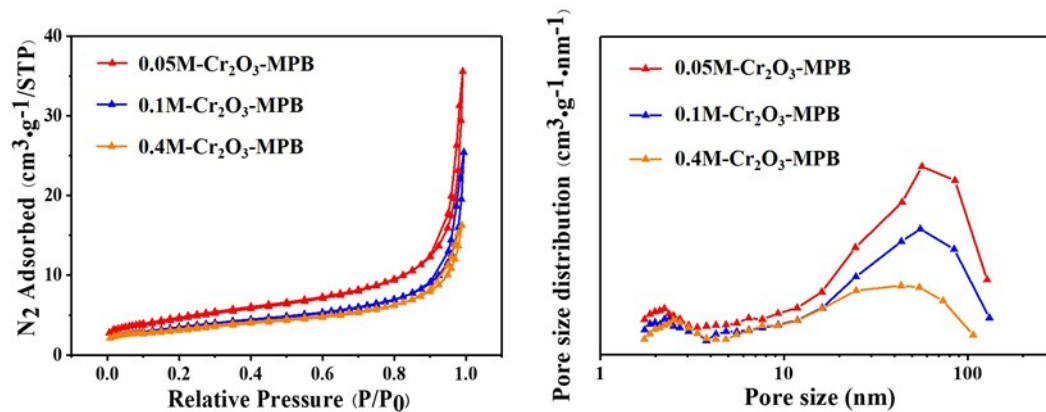


Fig. S7 Nitrogen adsorption-desorption isotherms and the homologous pore size distribution curves of series of Cr_2O_3 -MPB materials.

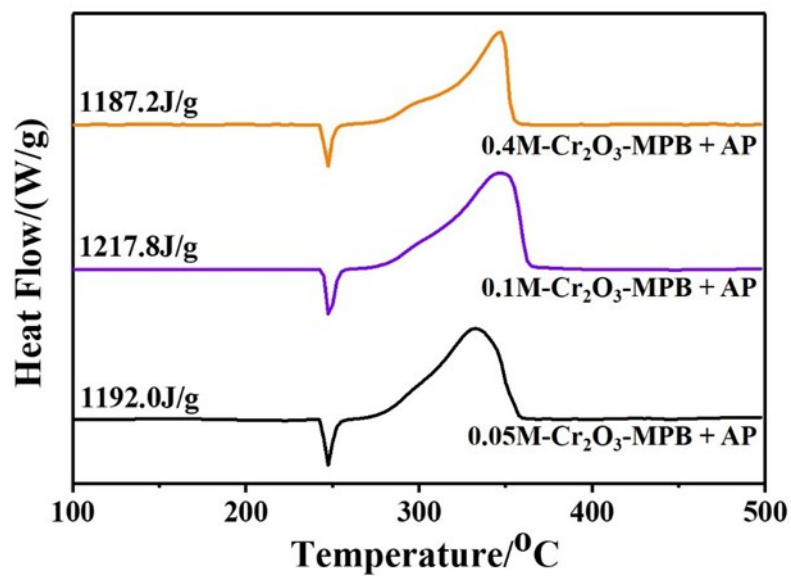


Fig. S8 DSC curves of mixtures of AP and series of Cr₂O₃-MPB materials.

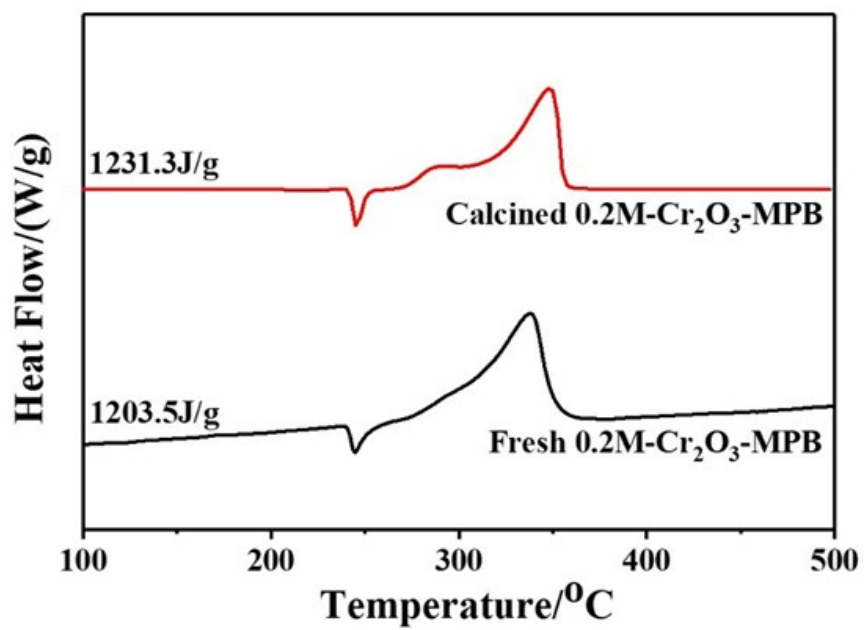


Fig. S9 DSC curves of mixtures of AP and fresh/calcined 0.2 M-Cr₂O₃-MPB.

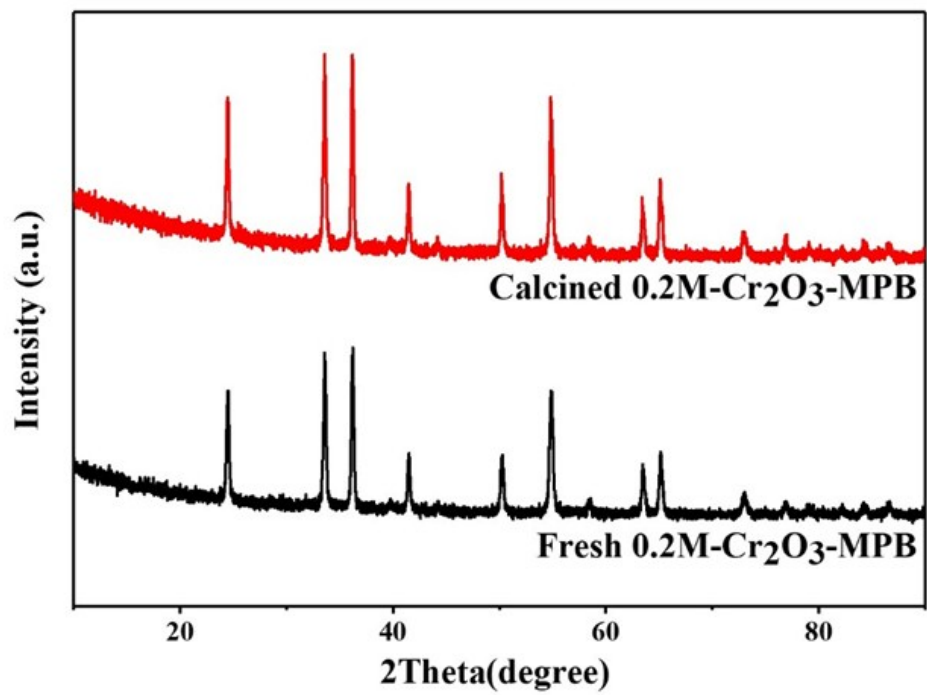


Fig. S10 XRD patterns of fresh and calcined 0.2 M-Cr₂O₃-MPB.

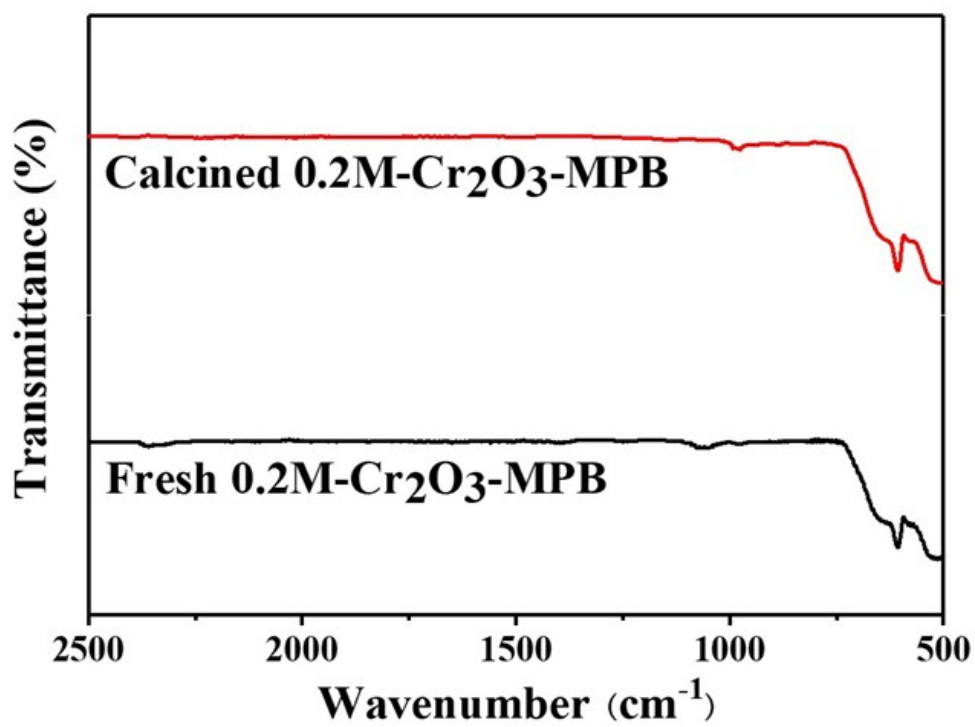


Fig. S11 FT-IR spectra of fresh and calcined 0.2 M-Cr₂O₃-MPB.