

Tandem transfer hydrogenation-epoxidation of ketone substrates catalysed by alkene-tethered Ru(II)-NHC complexes

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Supporting Information

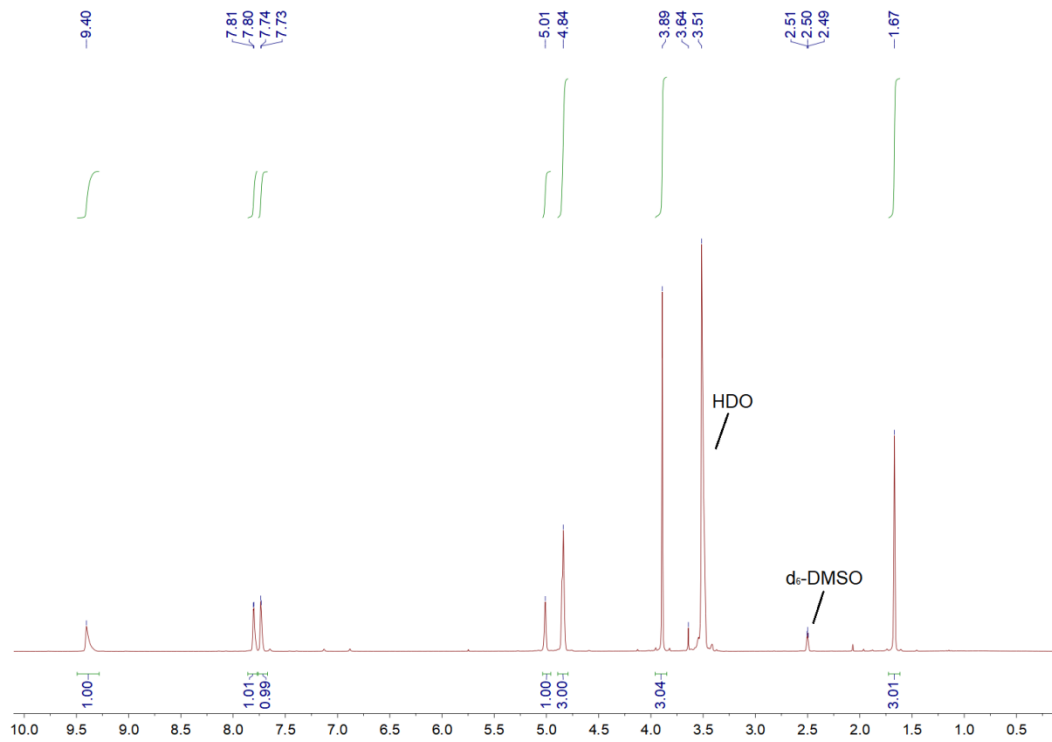
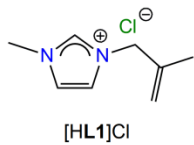
1. Synthesis, characterization, and NMR spectra of [H(L1-L6)]Cl
2. ¹H-, ¹³C-, ³¹P-NMR spectra of **1-9**
3. Variable temperature ¹H-NMR spectra (complex **4**)
4. Crystal data and structure refinement for [HL5]Cl, **P2**, **P3**, **1**, **2**, **4**, **5**, **8**, **9** (Tables **S1**, **S2**, **S3**)
5. Selected bond lengths and angles for [HL5]Cl, **P2**, **P3**, **1**, **2**, **4**, **5**, **8**, **9** (Table **S4**)
6. Time-resolved conversion profiles in the transfer hydrogenation-epoxidation catalysis
7. Optimization of transfer hydrogenation-epoxidation conditions (Table **S5**)
8. ¹H-NMR spectrum of catalysis reaction mixture

1. Synthesis, characterization, and NMR spectra of [H(L1-L6)]Cl

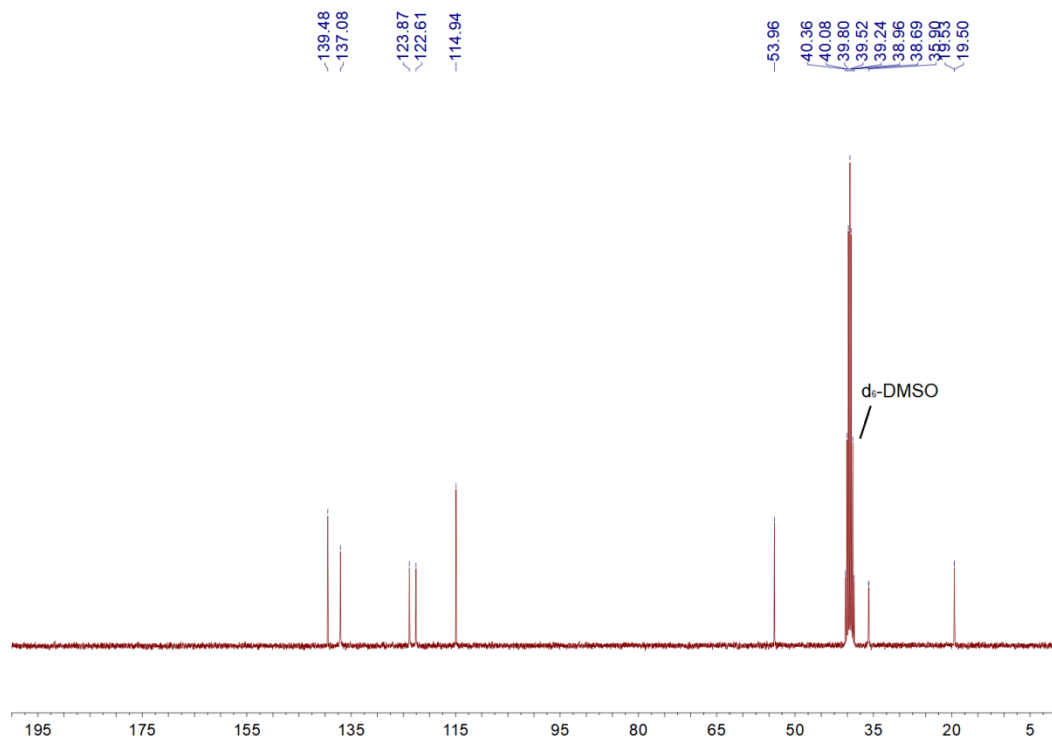
General synthesis of imidazolium salts: To an acetonitrile (20 mL) solution of the respective *N*-alkyl imidazole (42 mmol, [HL1]Cl, [HL2]Cl) was added 1-chloro-2-methylpropene (1 equivalent), and the resulting mixture heated under reflux overnight. After cooling, the reaction mixture was concentrated *in vacuo*, and washed with a 1:1 v/v Et₂O/Et₂OAc mixture (3 × 15 mL). The resulting oil/solid was concentrated *in vacuo* to give the respective ligands [H(L1-L6)]Cl.

[HL1]Cl: Yield: 94%. ¹H NMR ((CD₃)₂SO): δ_H = 1.67 (s, 3H, CH₃), 3.89 (s, 3H, NCH₃), 4.84 (s, 3H, =CH + NCH₂), 5.01 (s, 1H, =CH), 7.74 (d, ³J_{HH} = 2 Hz, 1H, NCH), 7.81 (d, ³J_{HH} = 2 Hz, 1H, NCH), 9.40 (s, 1H, NCHN). ¹³C{¹H} NMR ((CD₃)₂SO): δ_C = 19.5 (s, CH₃), 35.9 (s, NCH₃), 54.0 (s, NCH₂), 114.9 (s, =CH₂), 122.6 (s, NCH), 123.9 (s, NCH), 137.0 (s, CCH₂), 139.5 (s, NCN).

¹H-NMR

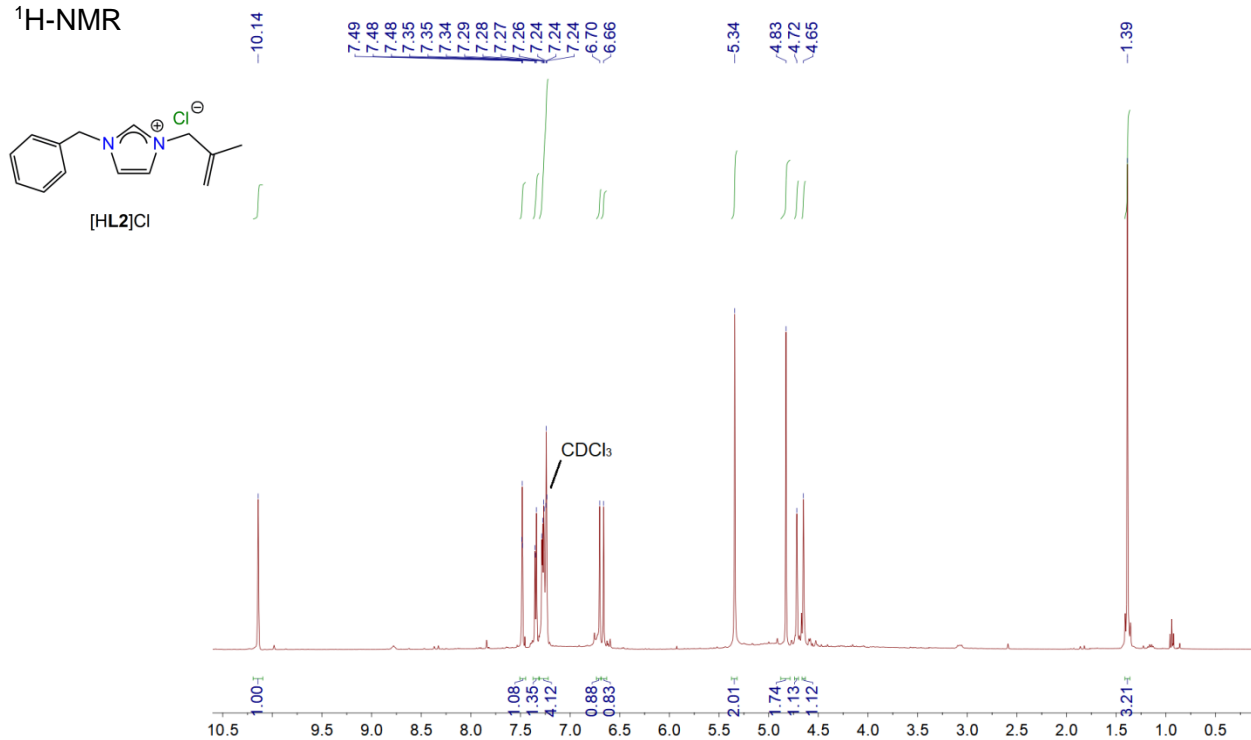


¹³C-NMR

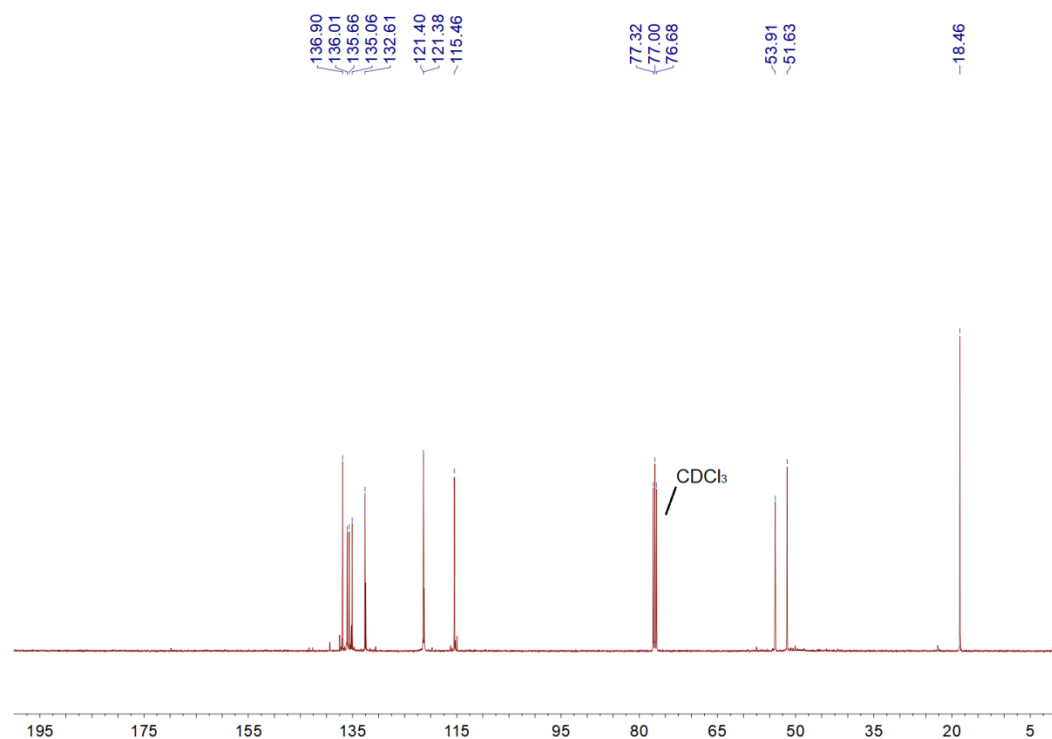


[HL2]Cl: Yield: 91%. ^1H NMR (CDCl_3): $\delta_{\text{H}} = 1.39$ (s, 3H, CH_3), 4.65 (s, 1H, $=\text{CH}_2$), 4.72 (s, 1H, $=\text{CH}_2$), 4.83 (s, 2H, CH_2), 5.34 (s, 2H, CH_2), 6.66 (s, 1H, NCH), 6.70 (s, 1H, NCH), 7.26 (m, 4H, C_6H_5), 7.49 (m, 1H, C_6H_5), 10.14 (s, 1H, NCHN). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): $\delta_{\text{C}} = 18.5$ (s, CH_3), 51.6 (s, CH_2), 53.9 (s, CH_2), 115.5 (s, $=\text{CH}_2$), 121.4 (s, NCH), 121.4 (s, NCH), 132.6 (s, C_6H_5), 135.1 (s, C_6H_5), 135.7 (s, *ipso* C_6H_5), 136.0 (s, CCH_2), 136.9 (s, NCHN).

^1H -NMR

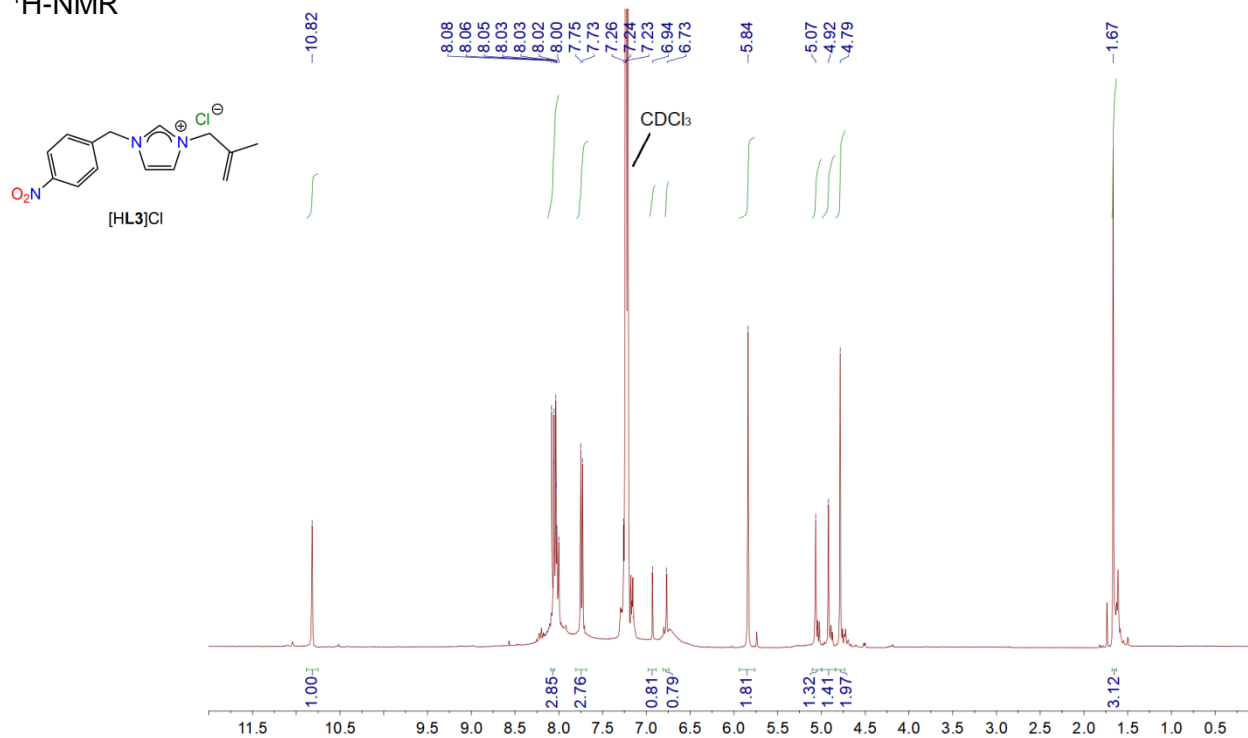


^{13}C -NMR

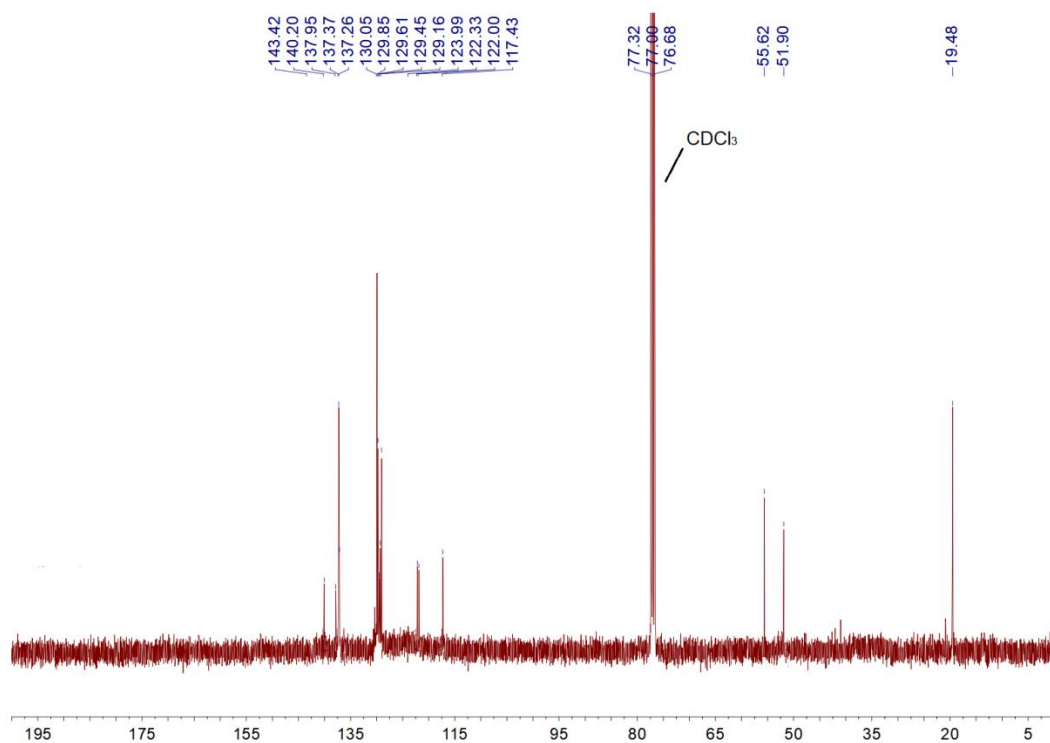


[HL3]Cl: Yield: 67%. ^1H NMR (CDCl_3): $\delta_{\text{H}} = 1.67$ (s, 3H, CH_3), 4.79 (s, 2H, CH_2), 4.92 (s, 1H, $=\text{CH}$), 5.07 (s, 1H, $=\text{CH}$), 5.84 (s, 2H, CH_2), 6.74 (s, 1H, NCH), 6.94 (s, 1H, NCH), 7.74 (d, $^3J_{\text{HH}} = 6$ Hz, 2H, C_6H_4), 8.05 (m, 2H, C_6H_4), 10.82 (s, 1H, NCHN). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): $\delta_{\text{C}} = 19.5$ (s, CH_3), 51.9 (s, CH_2), 53.6 (s, CH_2), 117.4 (s, $=\text{CH}_2$), 122.0 (s, NCH), 122.3 (s, NCH), 129.1 (s, C_6H_4), 129.5 (s, C_6H_4), 129.9 (s, *ipso* C_6H_4), 137.4 (s, CCH_2), 143.4 (s, NCN).

^1H -NMR

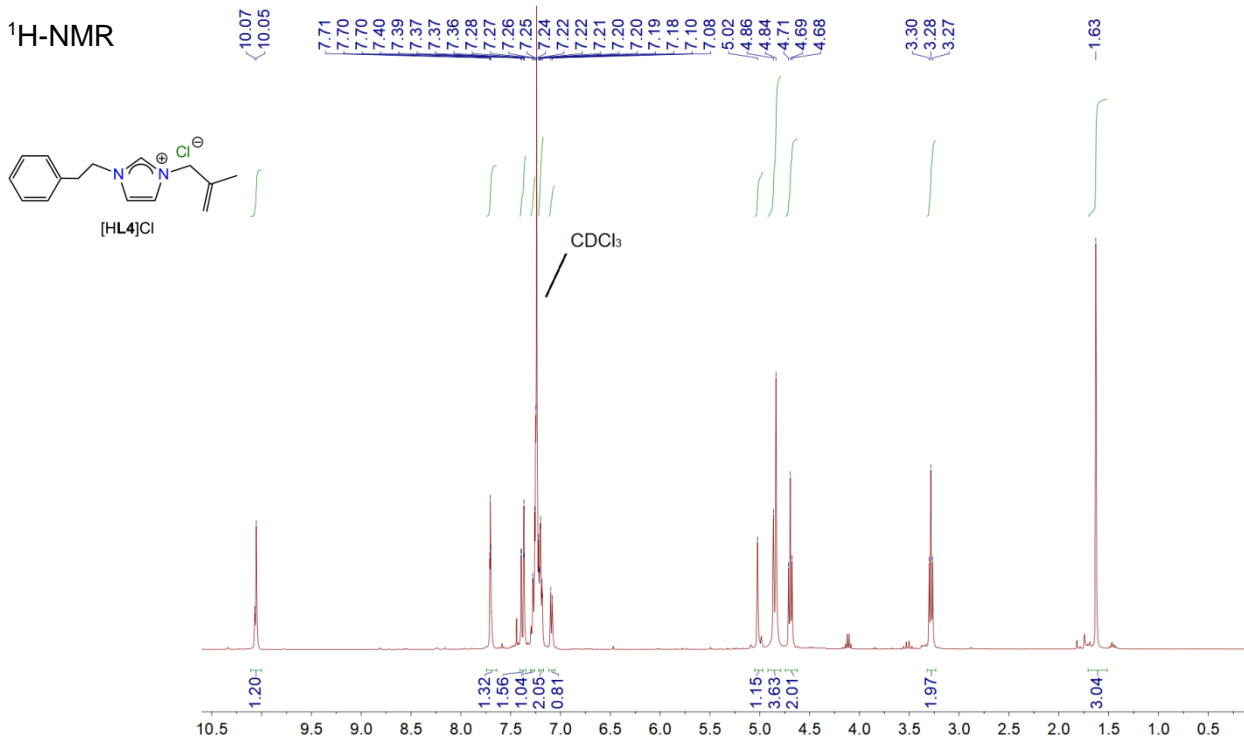


^{13}C -NMR

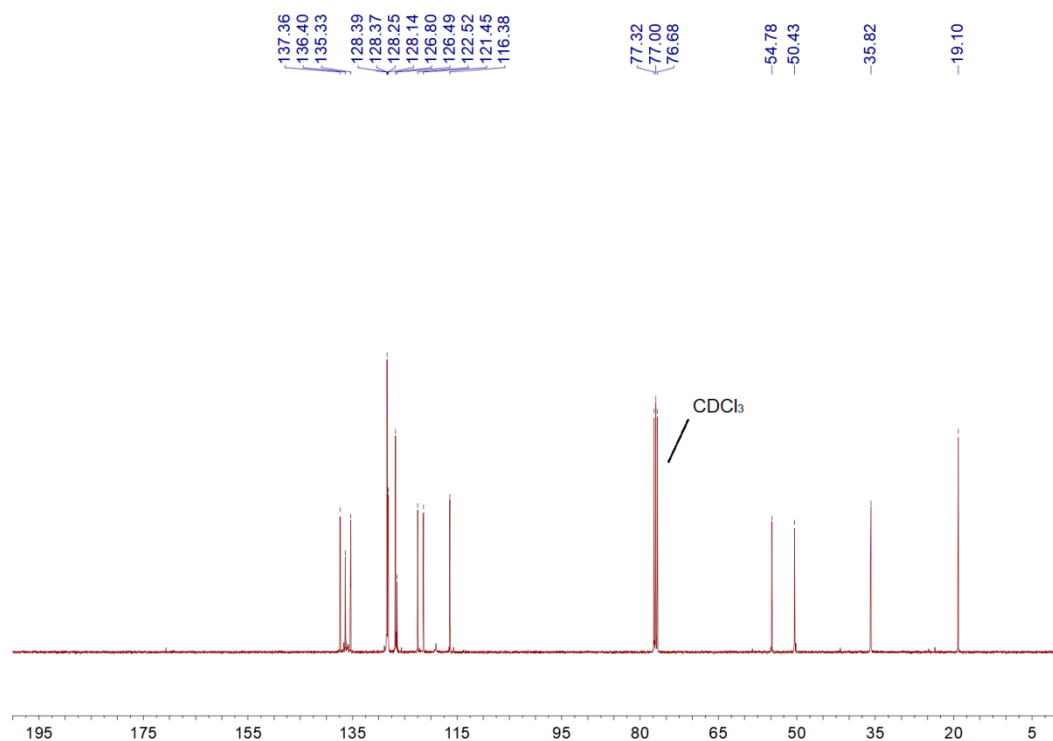


[HL4]Cl: Yield: 86%. $^1\text{H NMR}$ (CDCl_3): $\delta_{\text{H}} = 1.63$ (s, 3H, CH_3), 1.97 (t, $^3J_{\text{HH}} = 8$ Hz, 2H, CH_2), 4.69 (t, $^3J_{\text{HH}} = 8$ Hz, 2H, CH_2), 4.84 (m, 4H, $=\text{CH}_2 + \text{CH}_2$), 5.02 (s, 1H, $=\text{CH}_2$), 7.09 (d, $^3J_{\text{HH}} = 8$ Hz, 1H, NCH), 7.18-7.28 (m, 3H, NCH + C_6H_5), 7.37 (m, 1H, C_6H_5), 7.40 (m, 1H, C_6H_5), 7.70 (m, 1H, C_6H_5), 10.06 (s, 1H, NCHN). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): $\delta_{\text{C}} = 19.1$ (s, CH_3), 35.8 (s, CH_2), 50.4 (s, CH_2), 54.8 (s, CH_2), 116.4 (s, $=\text{CH}_2$), 121.5 (s, NCH), 122.5 (s, NCH), 126.5 (s, C_6H_5), 128.3 (s, C_6H_5), 128.4 (s, C_6H_5), 135.3 (s, *ipso* C_6H_5), 136.4 (s, CCH_2), 137.4 (s, NCHN).

$^1\text{H-NMR}$

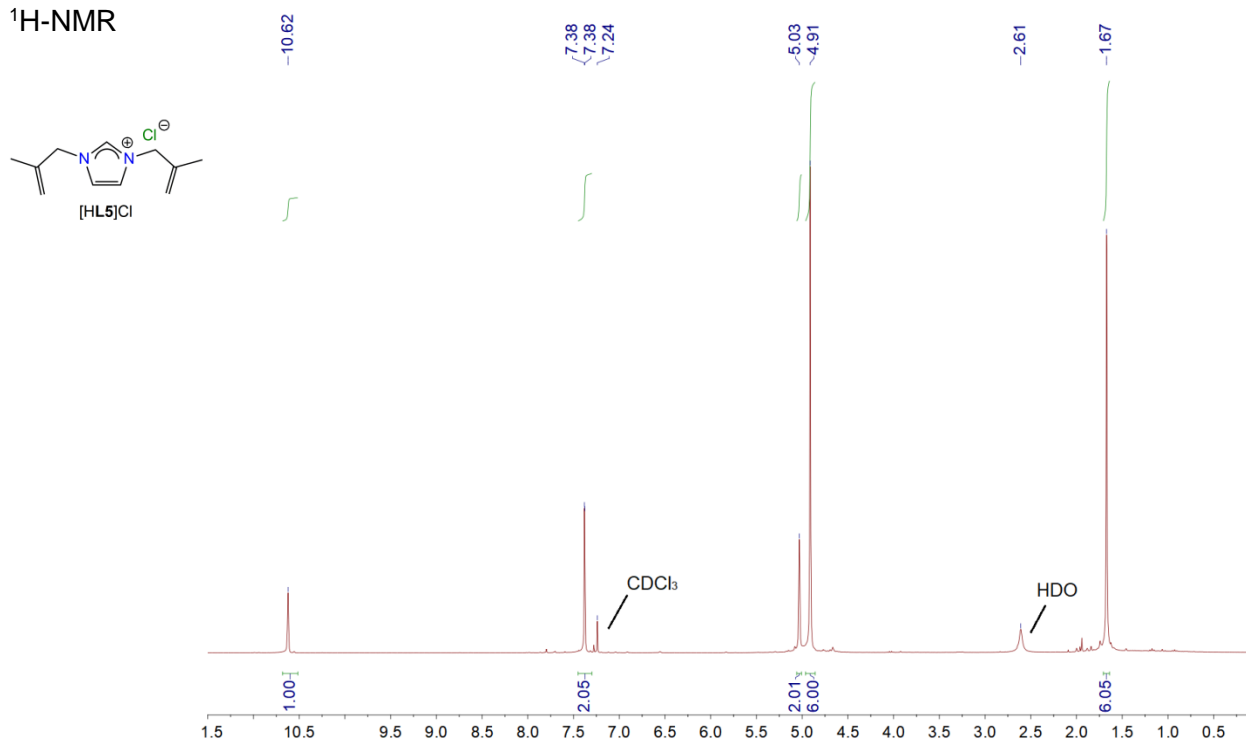


$^{13}\text{C-NMR}$

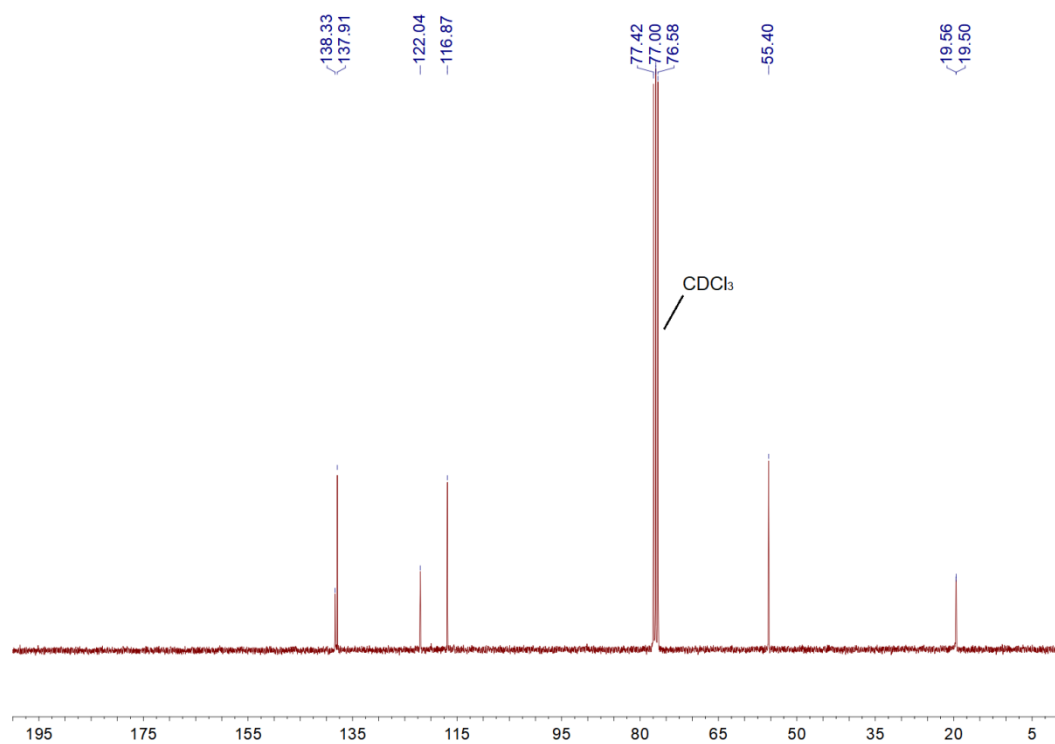


[HL5]Cl: Yield: 88%. ^1H NMR (CDCl_3): $\delta_{\text{H}} = 1.67$ (s, 6H, CH_3), 4.91 (s, 6H, $\text{CH}_2 + =\text{CH}_2$), 5.03 (s, 2H, $=\text{CH}_2$), 7.38 (s, 2H, NCH), 10.62 (s, 1H, NCH). $^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3): $\delta_{\text{C}} = 19.5$ (s, CH_3), 19.6 (s, CH_3), 55.4 (s, CH_2), 116.9 (s, $=\text{CH}_2$), 122.0 (s, NCH), 137.9 (s, $\text{C}(\text{CH}_3)_2$), 138.3 (s, NCH).

^1H -NMR

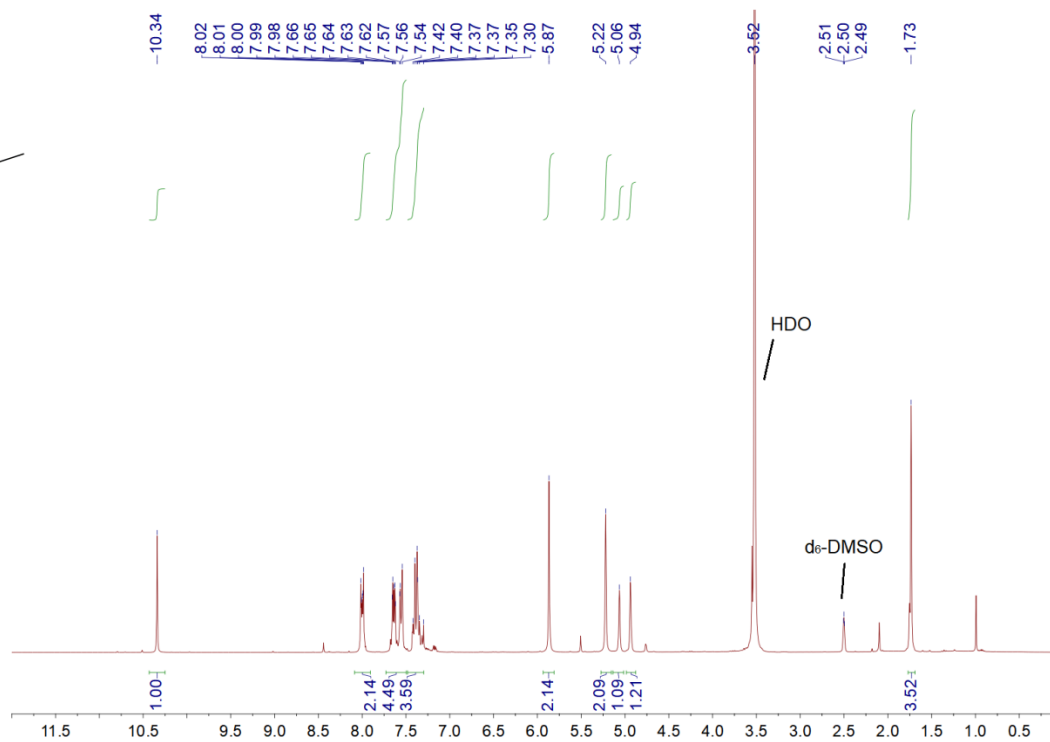
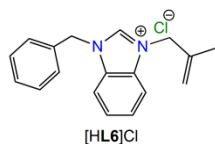


^{13}C -NMR

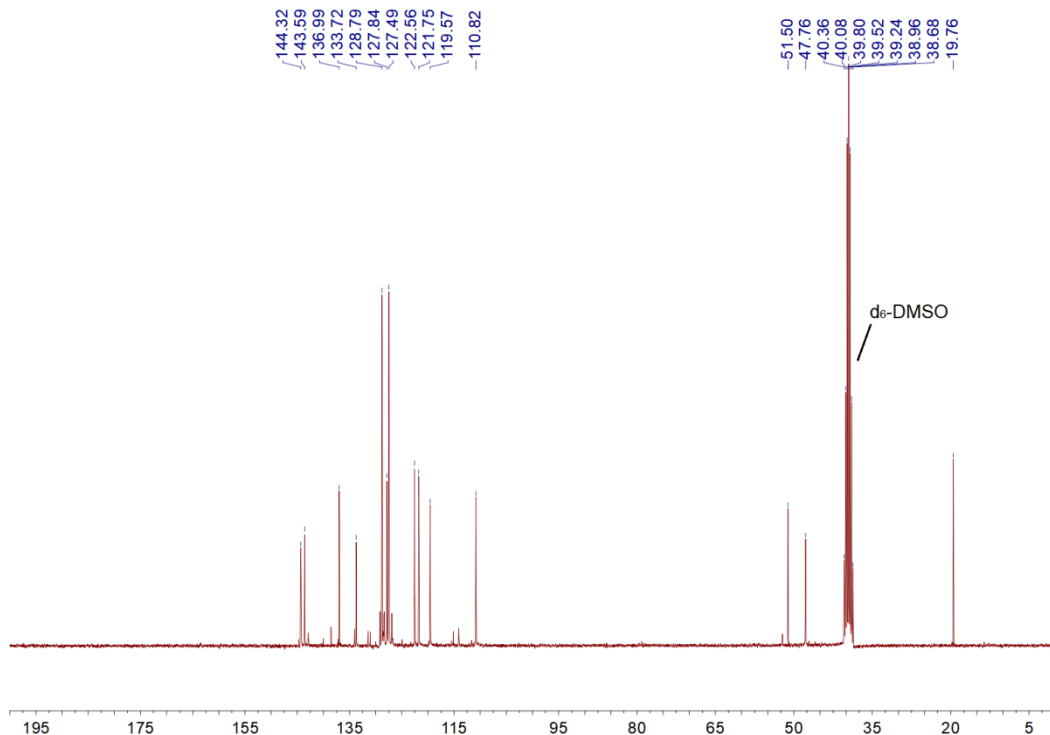


[HL6]Cl: Yield: 92%. $^1\text{H NMR}$ ($(\text{CD}_3)_2\text{SO}$): $\delta_{\text{H}} = 1.73$ (s, 3H, CH_3), 4.94 (s, 1H, $=\text{CH}_2$), 5.06 (s, 1H, $=\text{CH}_2$), 5.22 (s, 2H, NCH_2), 5.87 (s, 2H, NCH_2), 7.30-7.42 (m, 3H, C_6H_4), 7.54-7.66 (m, 4H, $\text{C}_6\text{H}_4 + \text{C}_6\text{H}_5$), 8.00 (m, 2H, C_6H_5), 10.34 (s, 1H, NCHN). $^{13}\text{C}\{^1\text{H}\}$ NMR ($(\text{CD}_3)_2\text{SO}$): $\delta_{\text{C}} = 19.8$ (s, CH_3), 47.8 (s, CH_2), 51.5 (s, CH_2), 110.8 (s, C_6H_4), 119.6 (s, $=\text{CH}_2$), 121.8 (s, NCH), 122.6 (s, NCH), 127.5 (s, C_6H_5), 127.8 (s, C_6H_5), 128.8 (s, C_6H_5), 133.7 (s, C_6H_4), 137.0 (s, *ipso* C_6H_5), 143.6 (s, CCH_2), 144.3 (s, NCN).

$^1\text{H-NMR}$



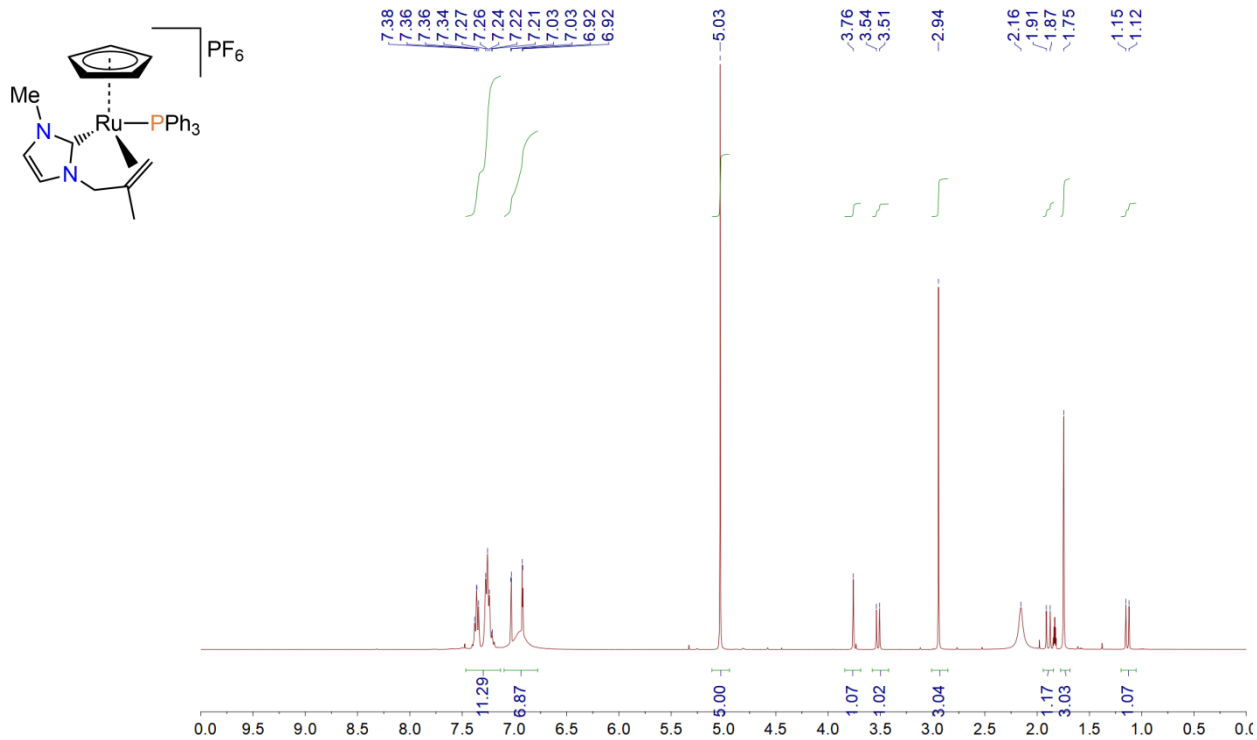
$^{13}\text{C-NMR}$



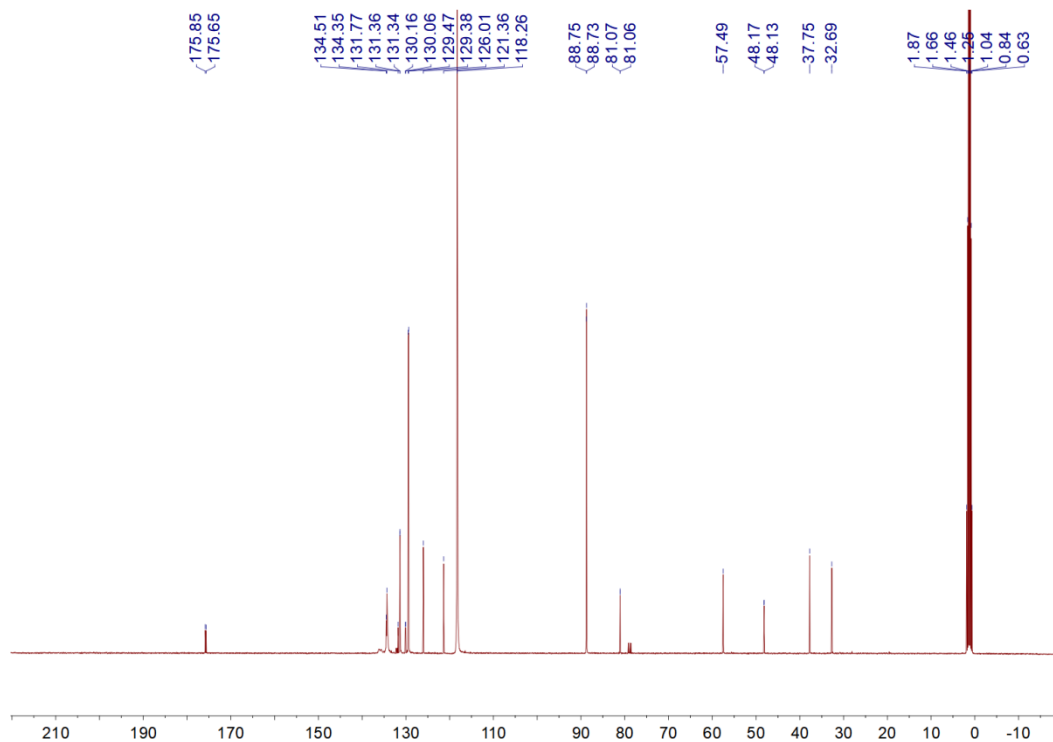
2. ^1H -, ^{13}C -, ^{31}P -NMR spectra of **1-9**

Complex 1

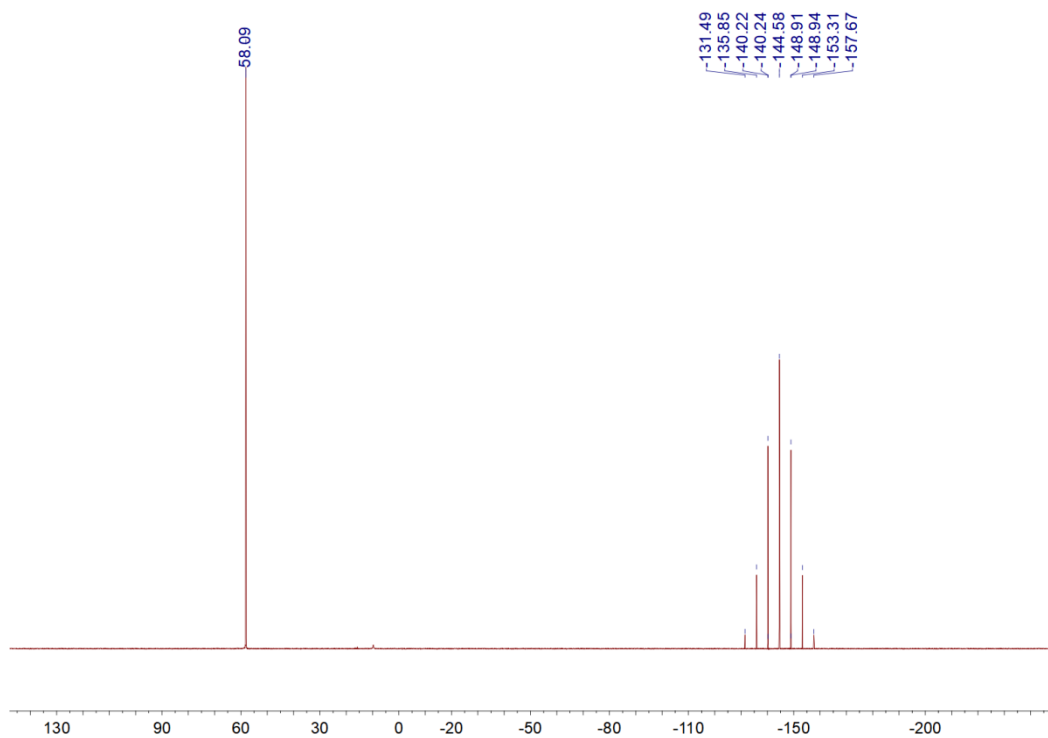
^1H -NMR



^{13}C -NMR

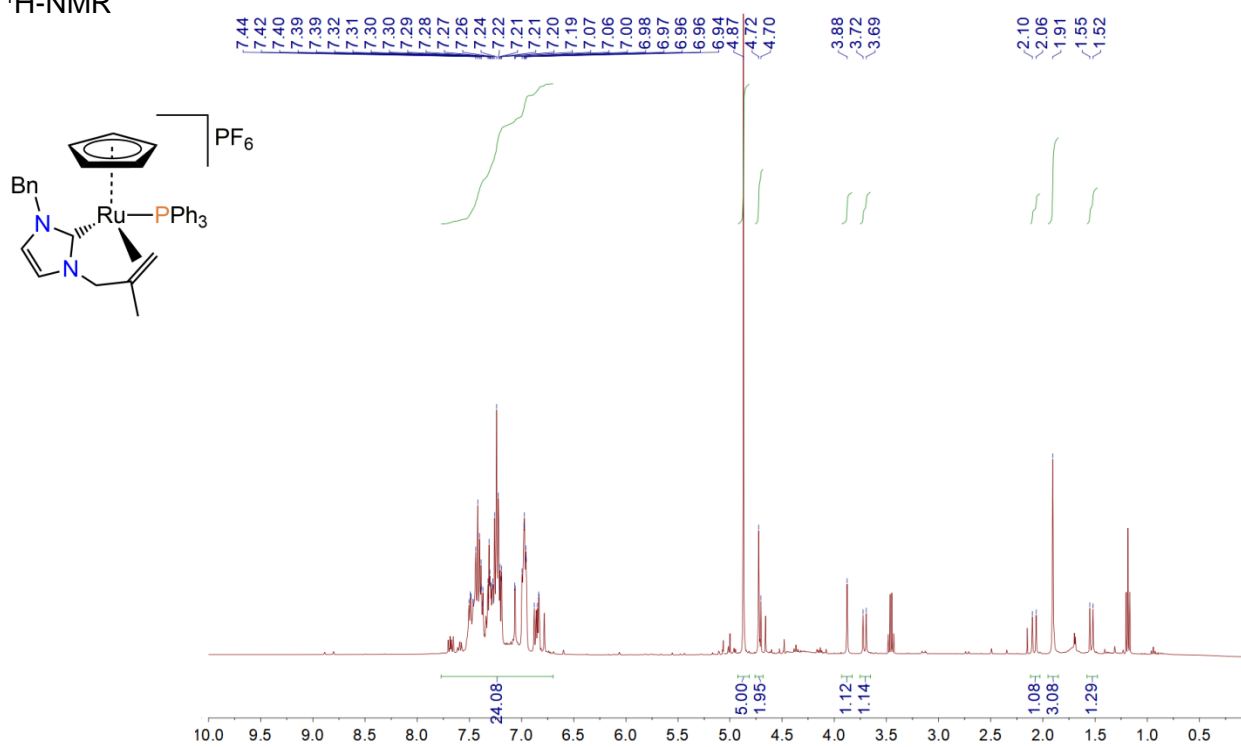


^{31}P -NMR

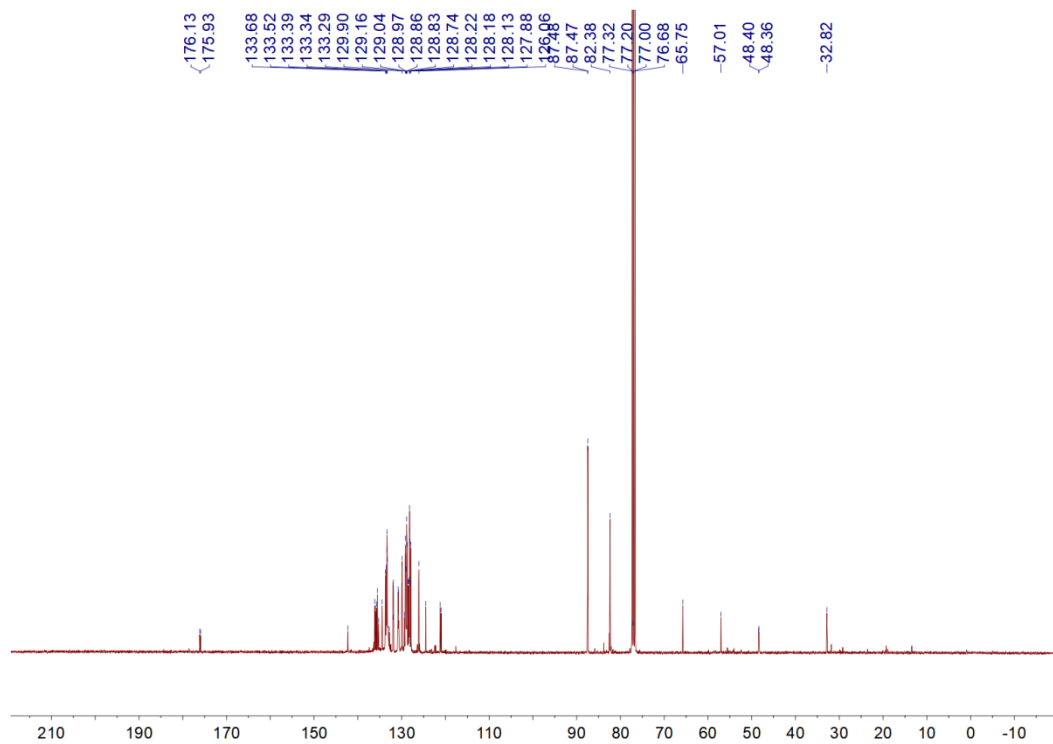


Complex 2

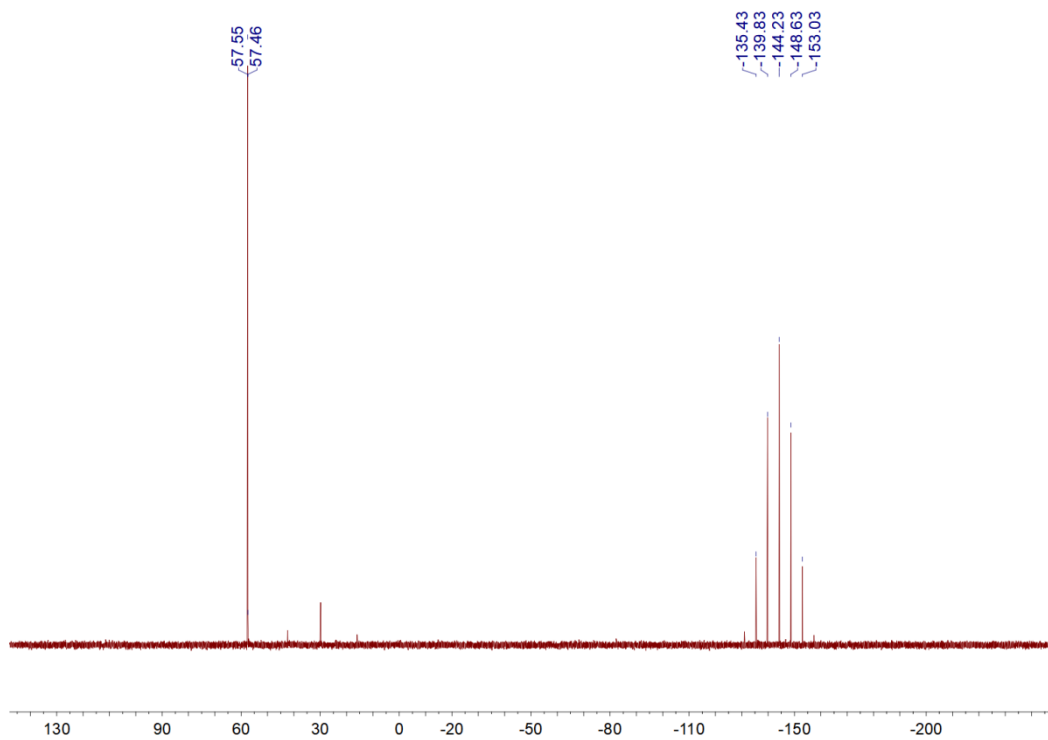
^1H -NMR



¹³C-NMR

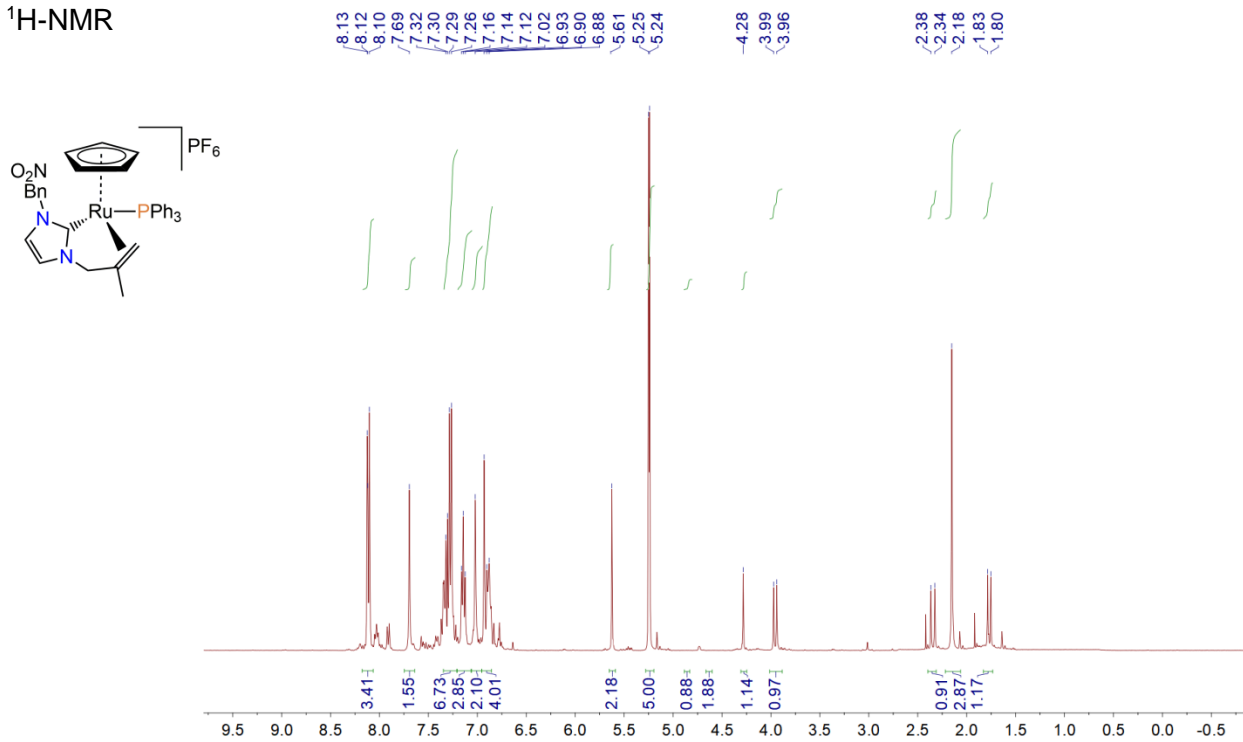


³¹P-NMR

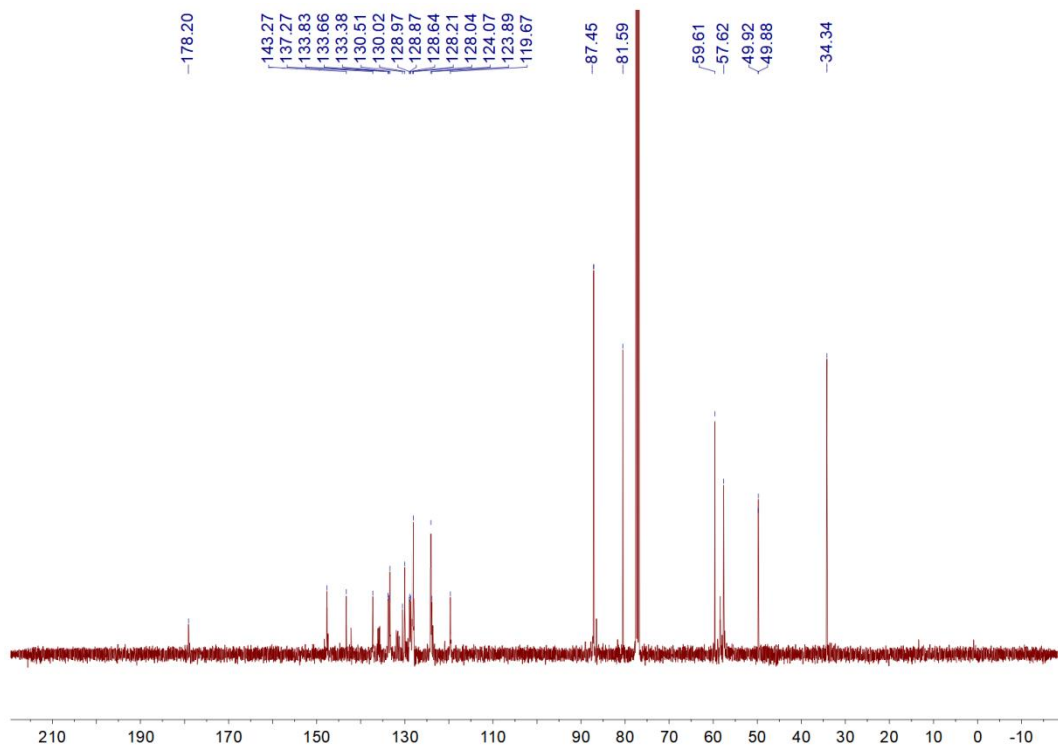


Complex 3

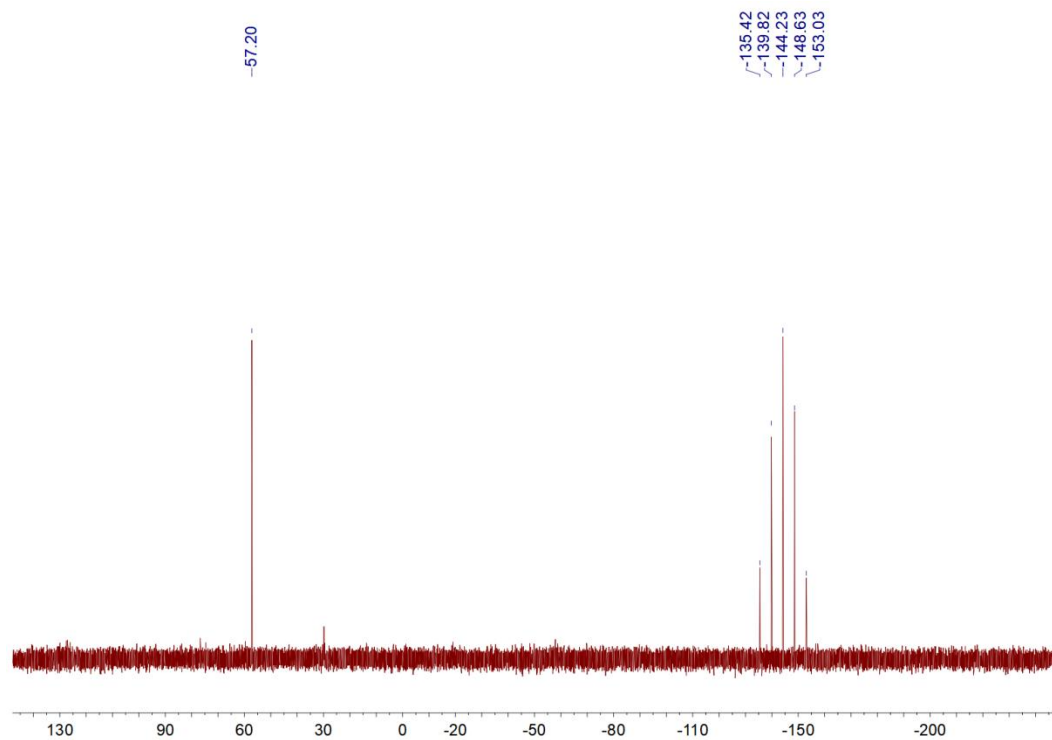
¹H-NMR



¹³C-NMR

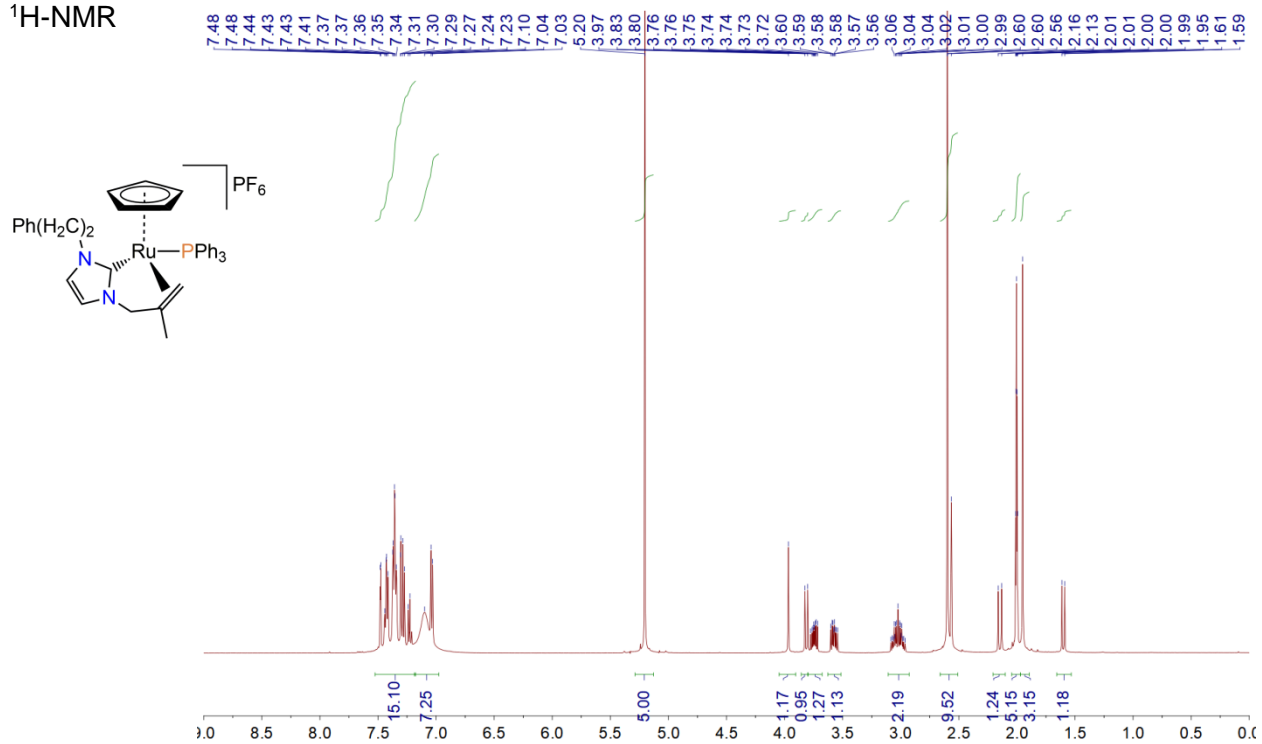


³¹P-NMR

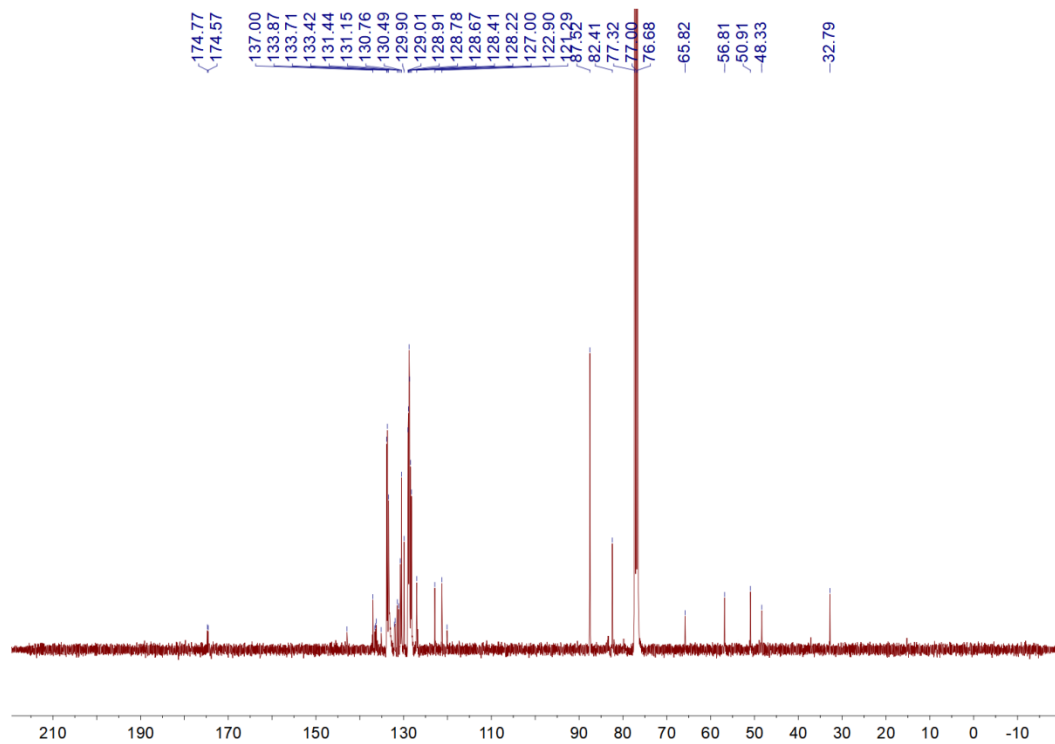


Complex 4

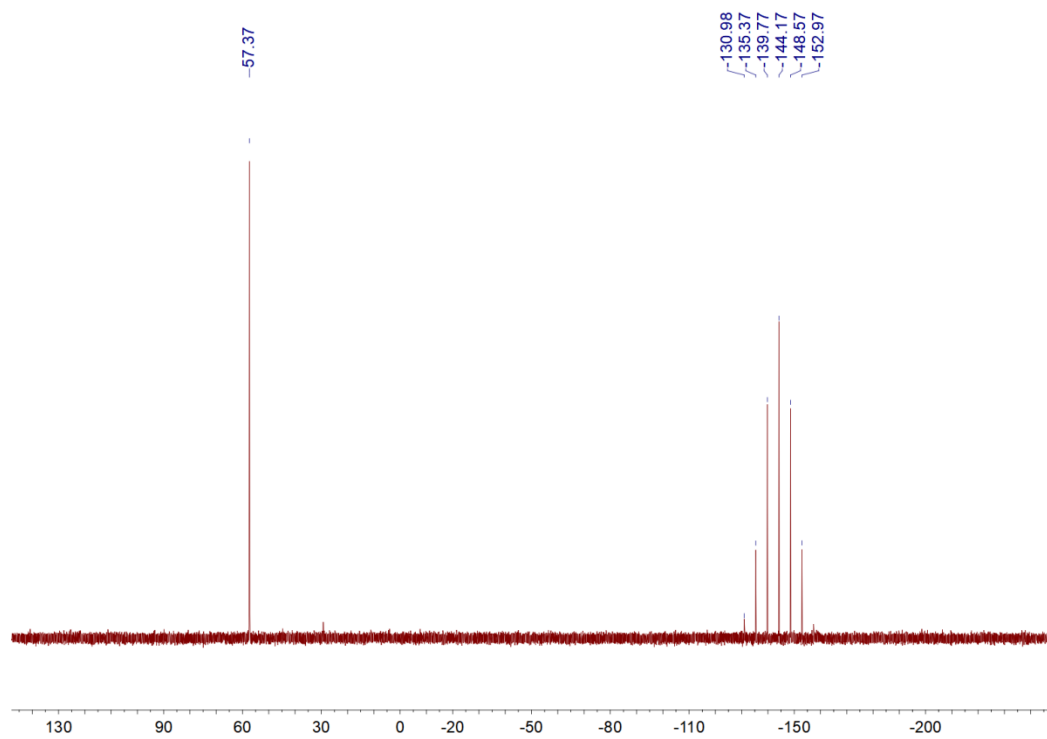
¹H-NMR



^{13}C -NMR

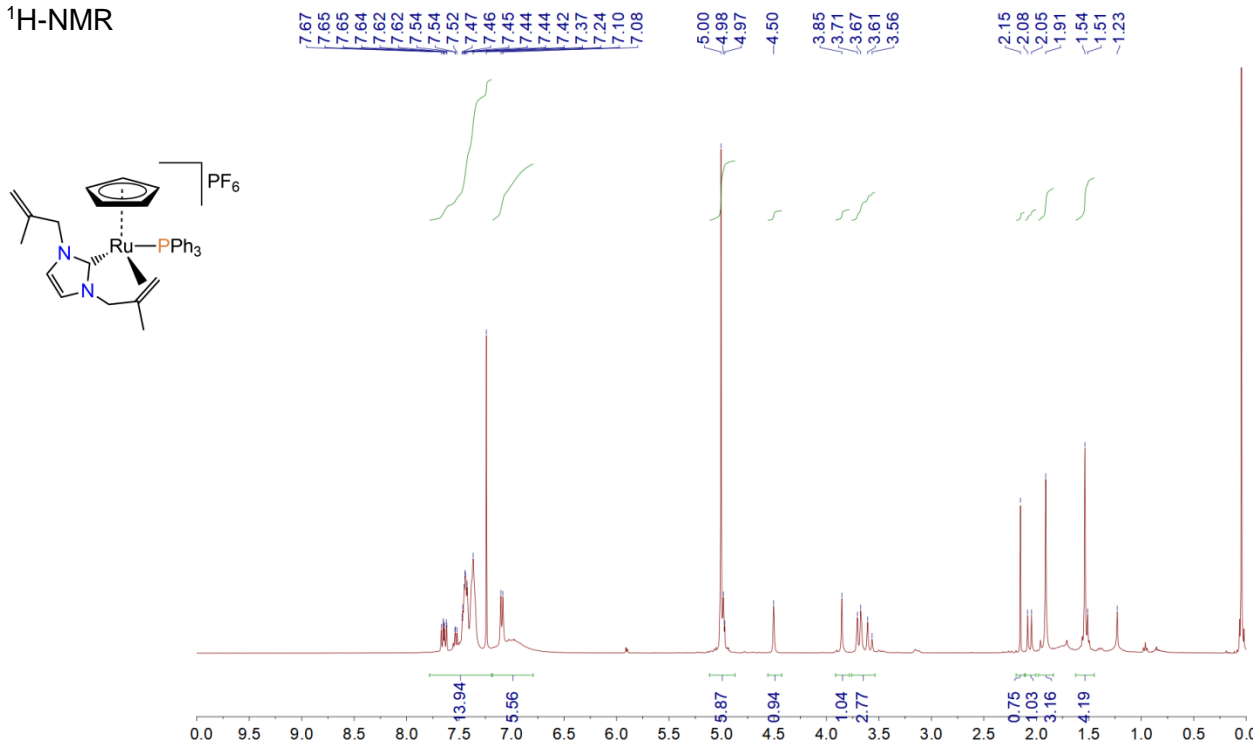


^{31}P -NMR

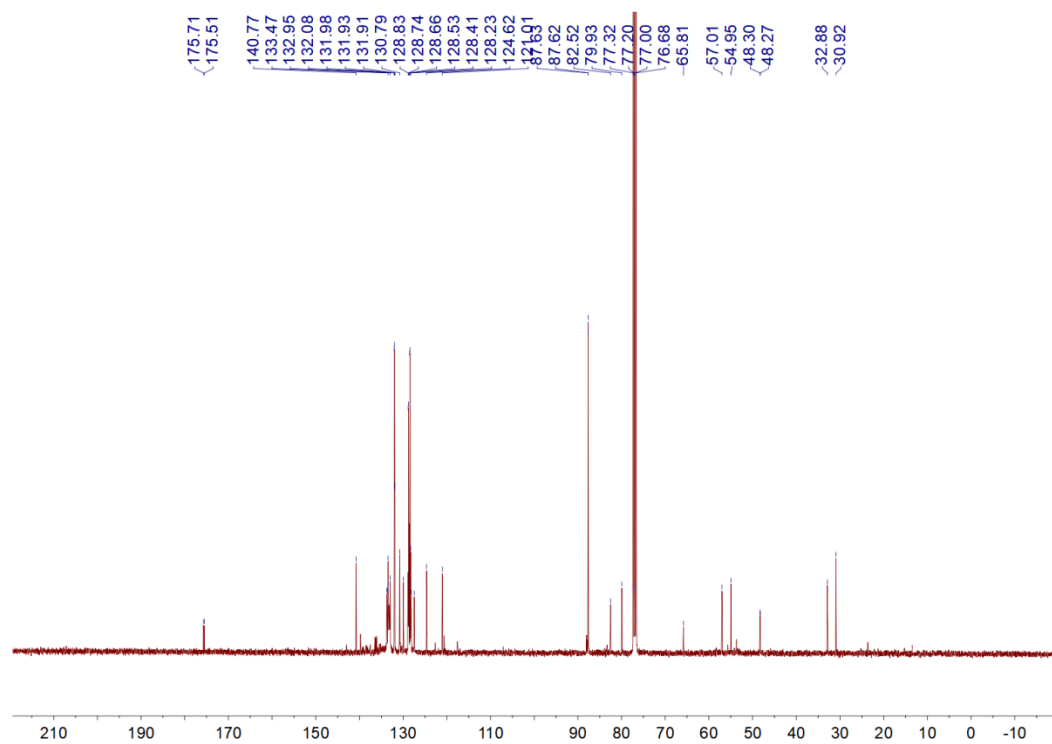


Complex 5

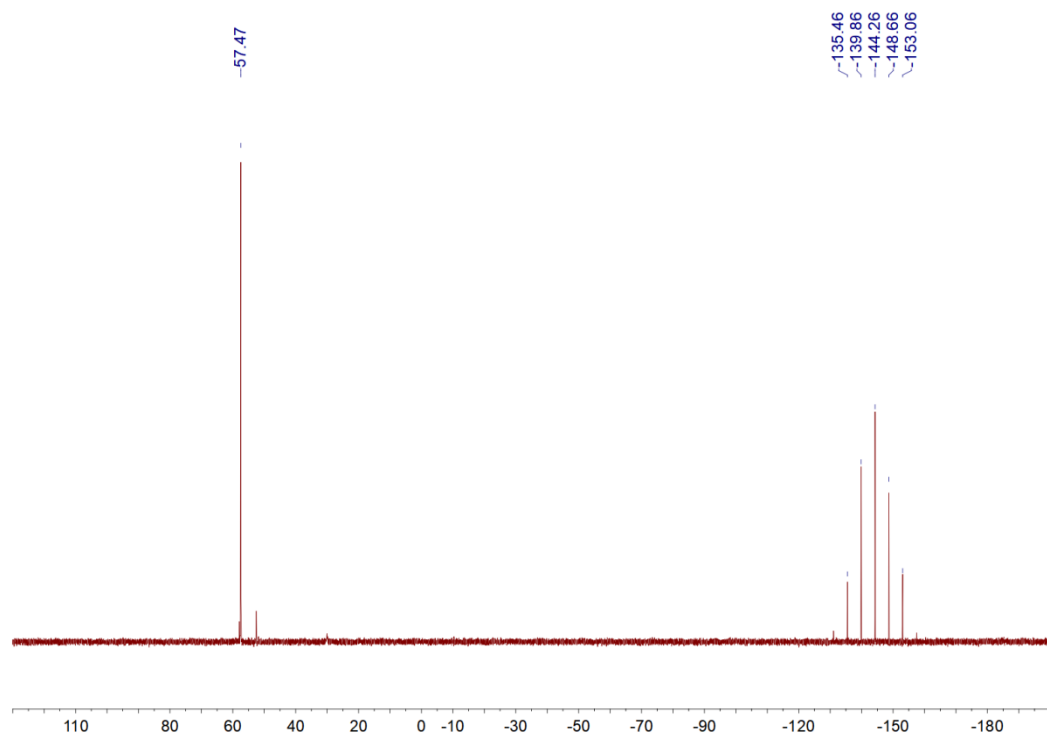
¹H-NMR



¹³C-NMR

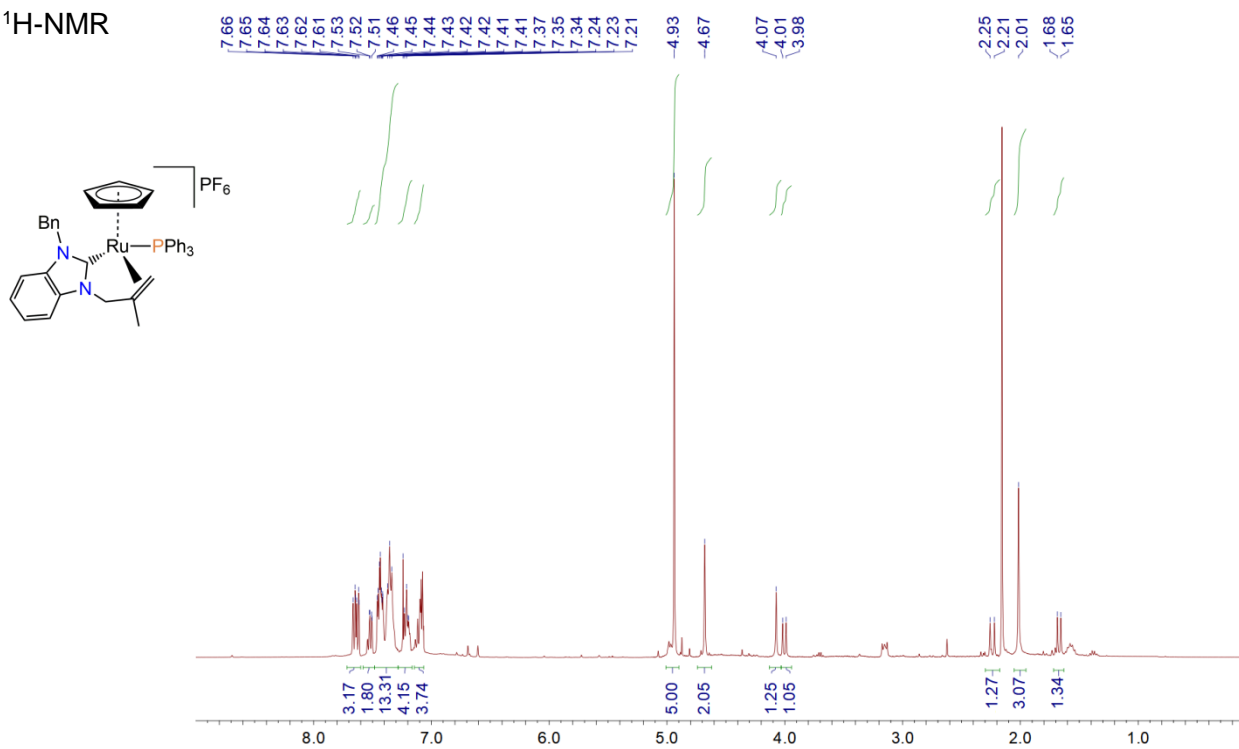


³¹P-NMR

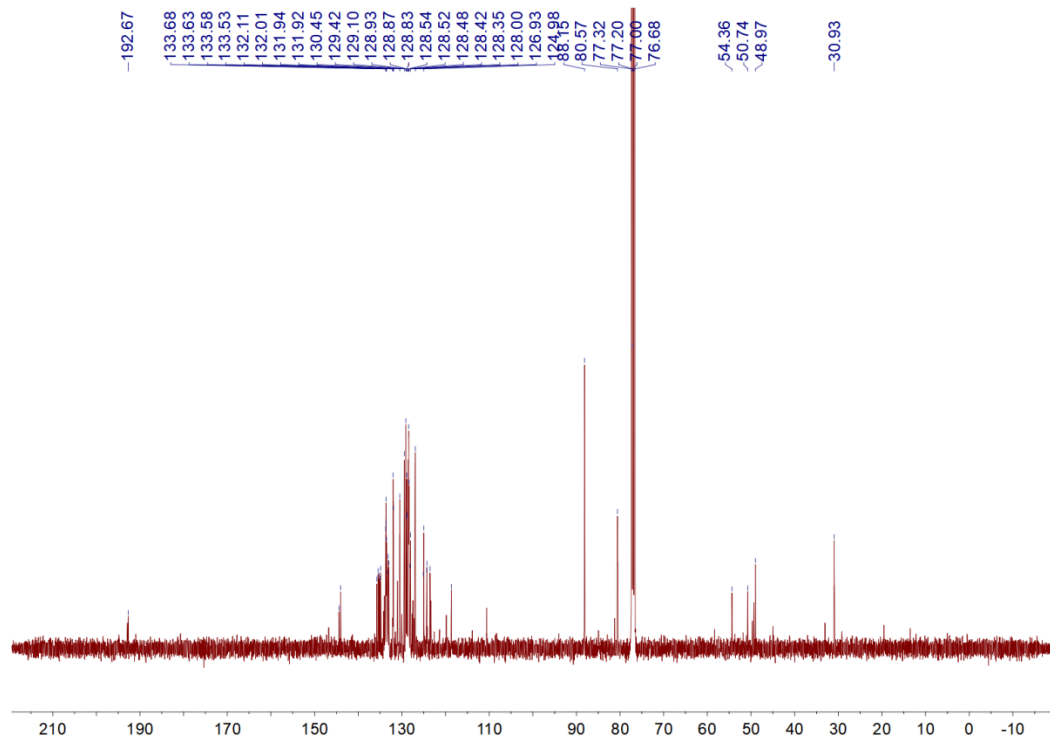


Complex 6

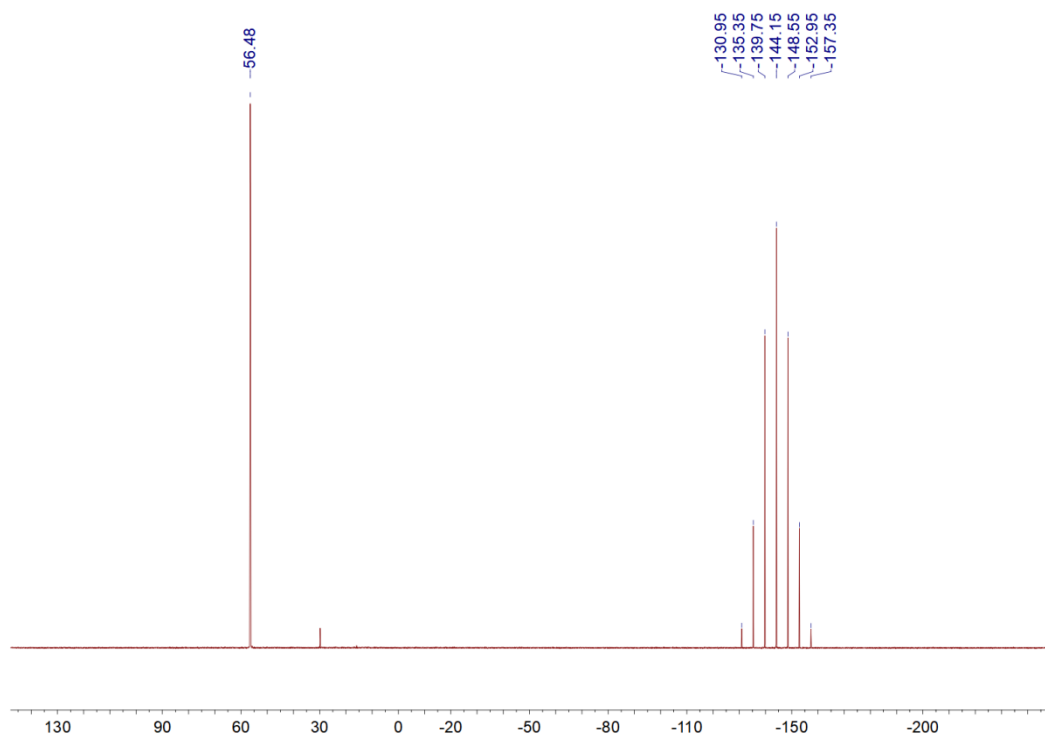
¹H-NMR



¹³C-NMR

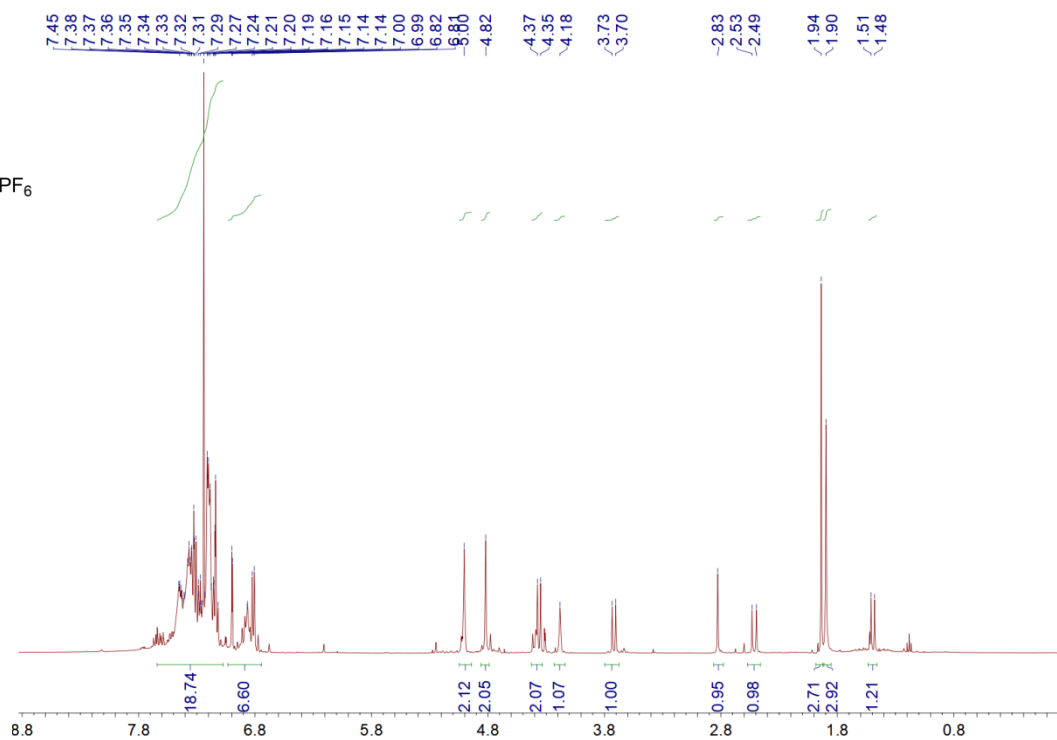
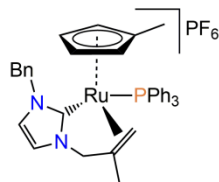


³¹P-NMR

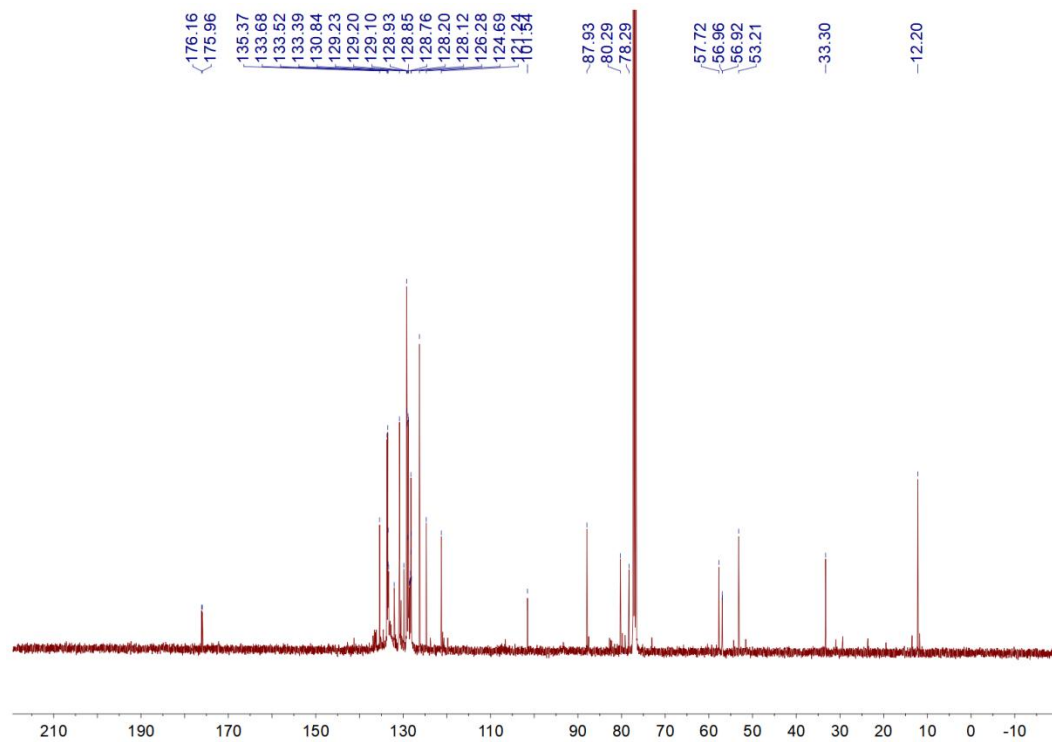


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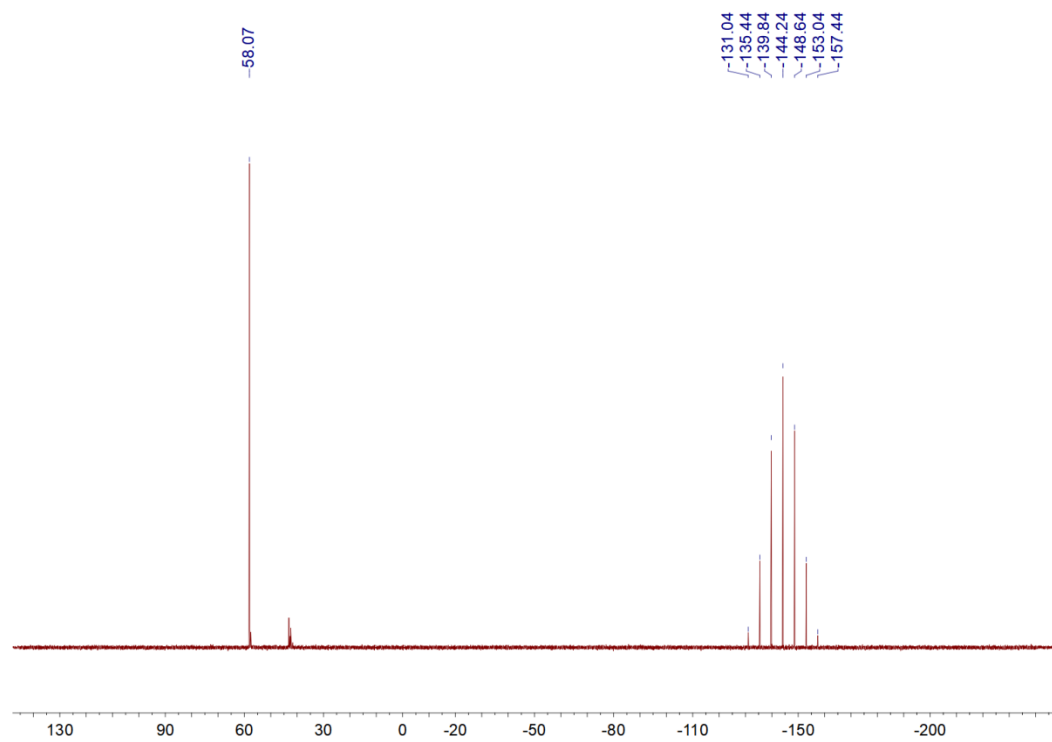
¹H-NMR



¹³C-NMR

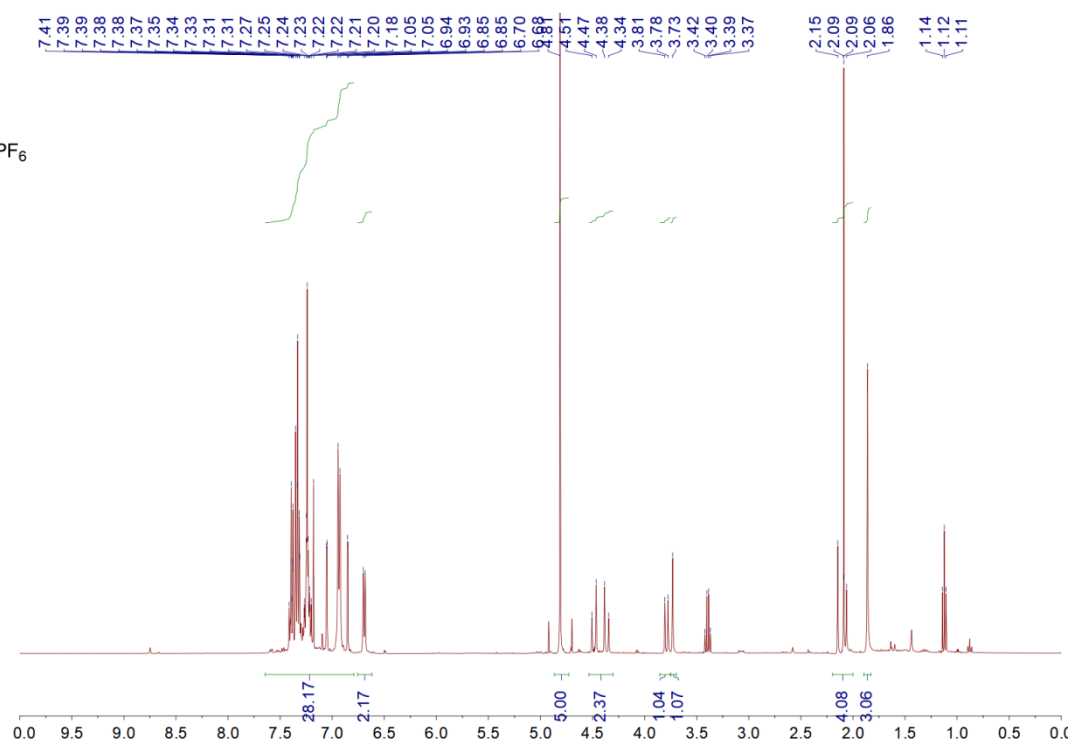
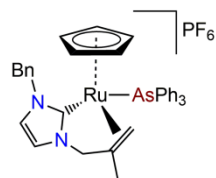


^{31}P -NMR

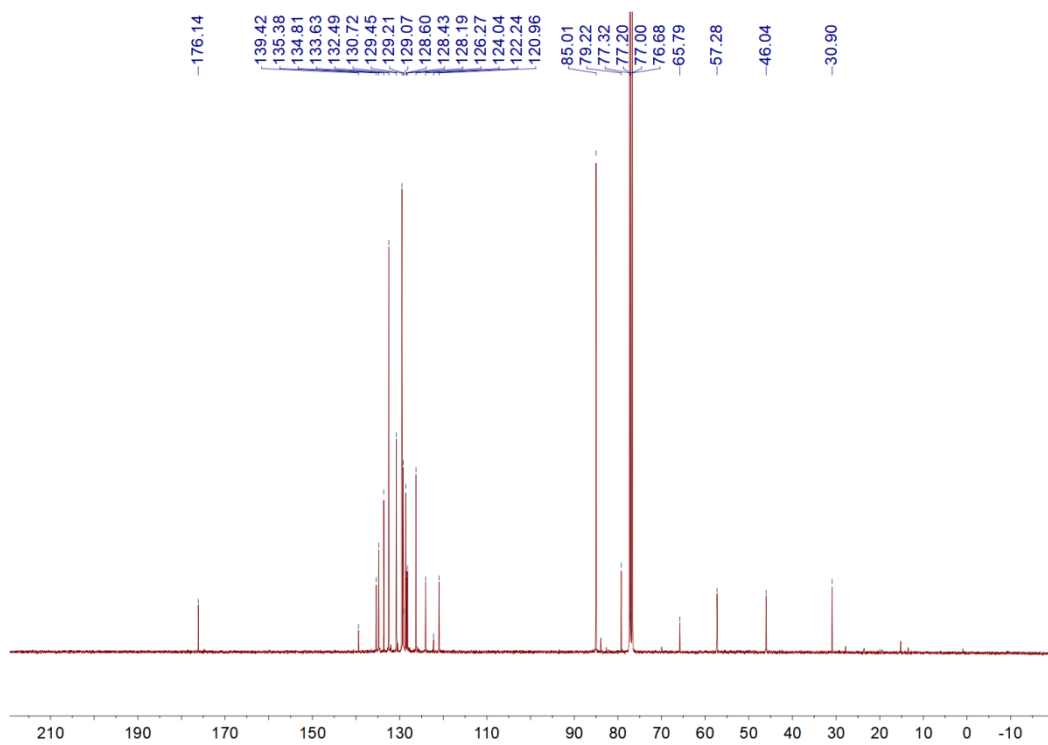


Complex 8

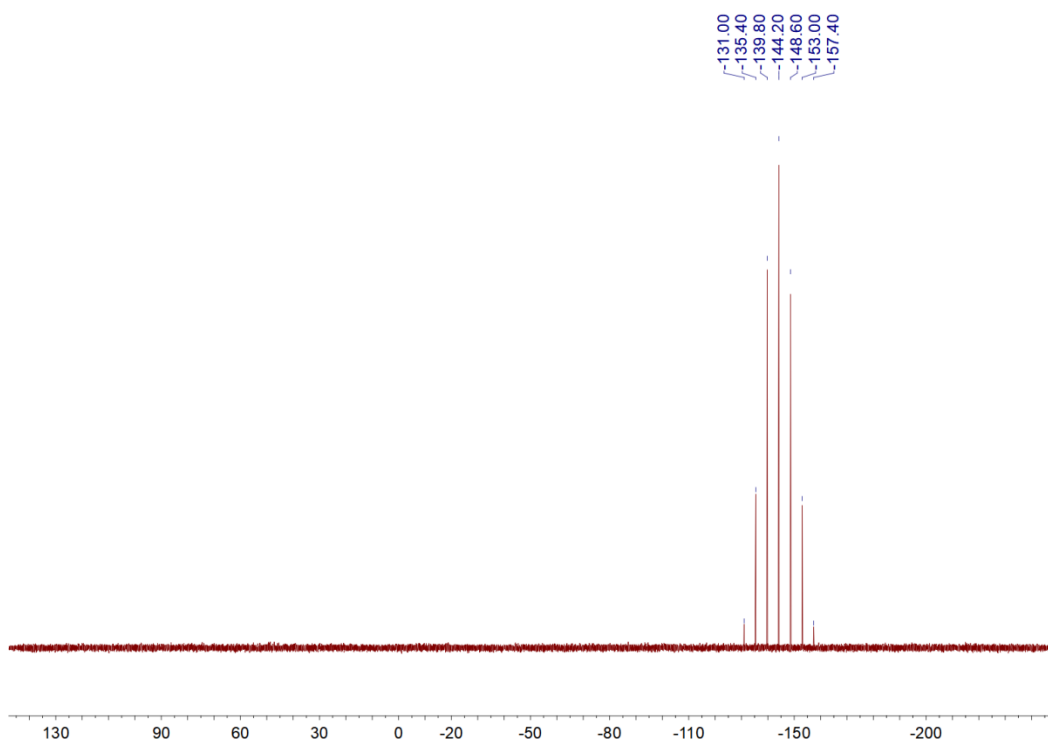
^1H -NMR



¹³C-NMR

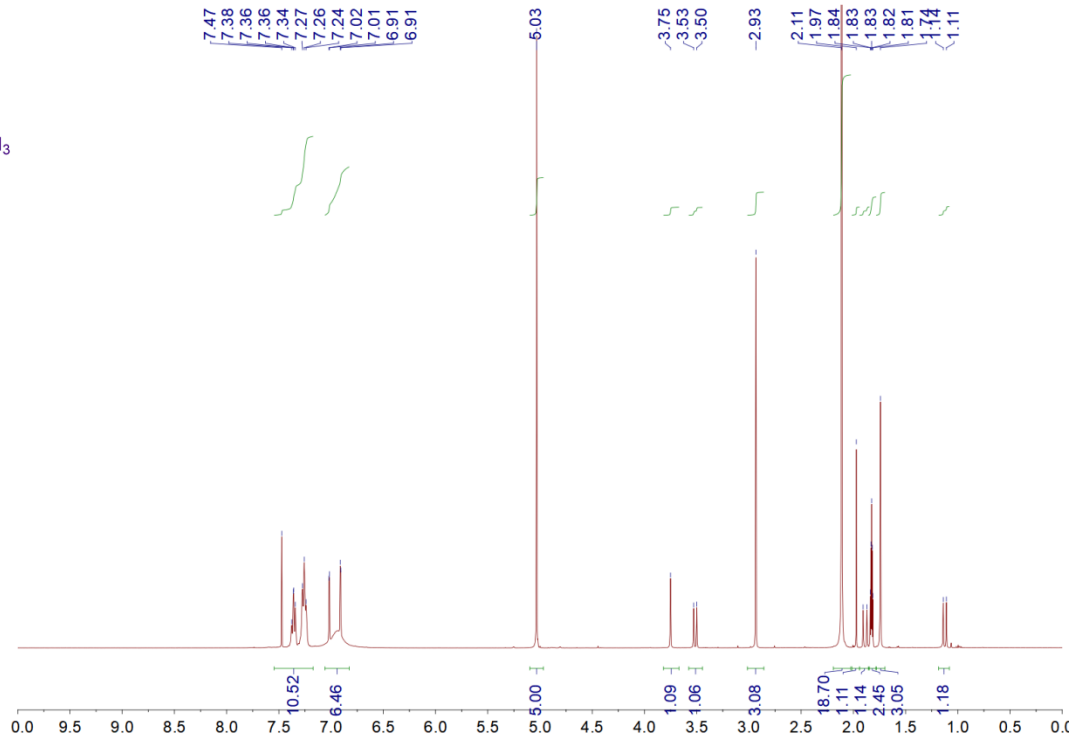
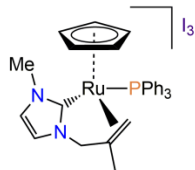


³¹P-NMR

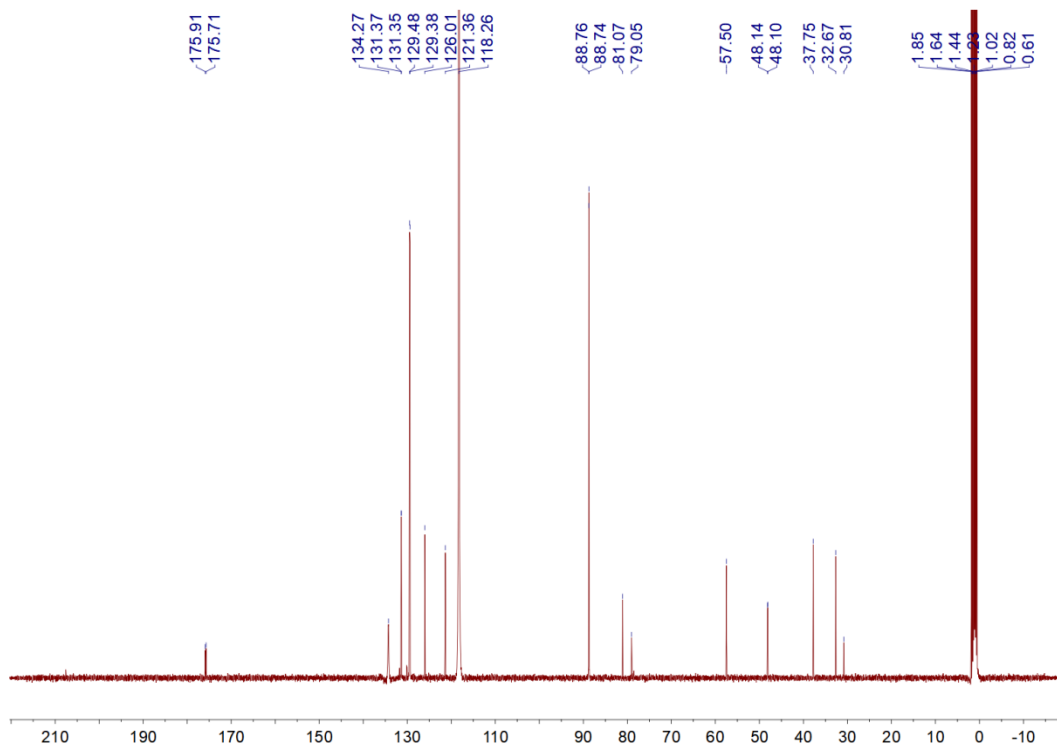


Complex 9

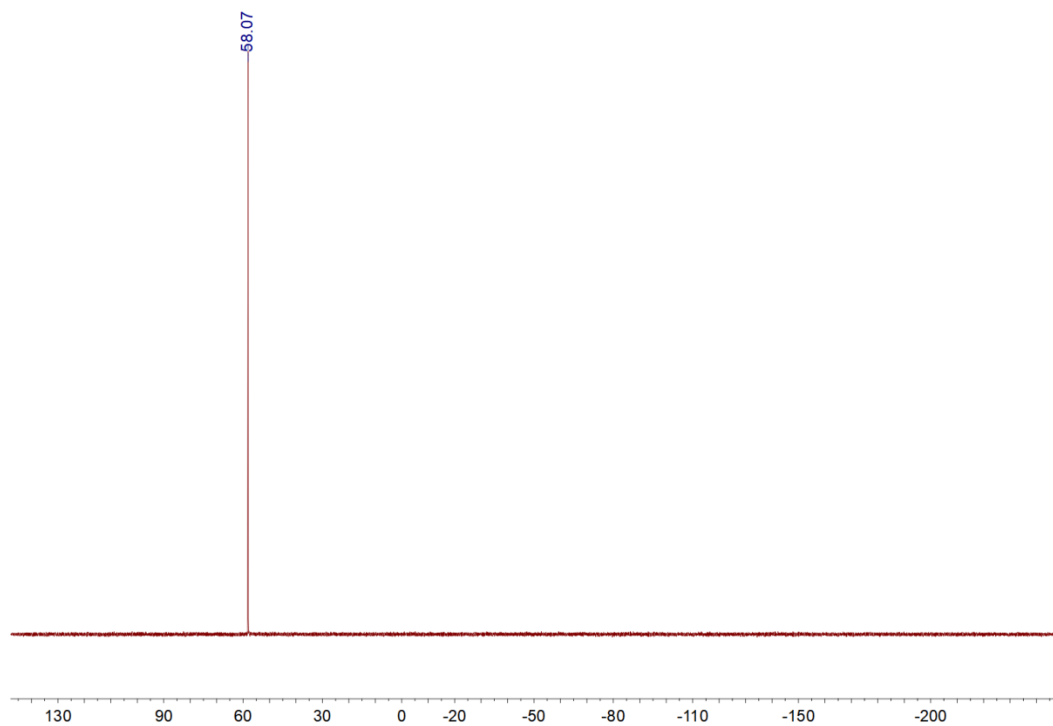
¹H-NMR



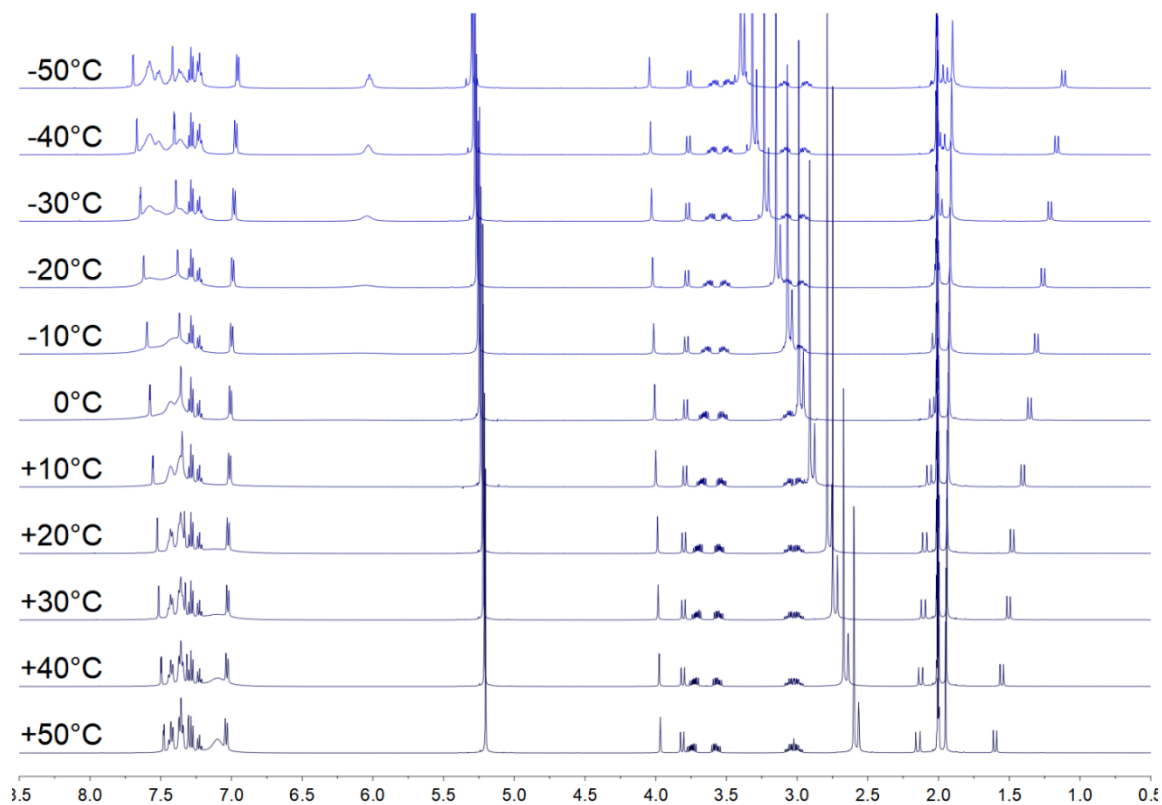
¹³C-NMR



^{31}P -NMR



3. Variable temperature ^1H -NMR (400 MHz) spectra (complex 4). [Solvent = $(\text{CD}_3)_2\text{CO}$.]



4. Crystal data and structure refinement for [HL5]Cl, P2,P3,1,2,4,5,8 (Tables S1,S2, S3)

Table S1

Complex	[HL5]Cl	P2	P3	1
Emp. formula	C _{14.7} H ₂₄ N _{2.7} O _{0.7} Cl _{1.3}	C ₄₃ H ₃₉ Cl ₃ P ₂ Ru	C ₄₂ H ₃₇ Cl ₃ As ₂ Ru	C ₃₁ H ₃₂ N ₂ F ₆ P ₂ Ru
Form. wt. (g.mol ⁻¹)	295.64	740.17	898.97	709.59
Crystal system	orthorhombic	triclinic	triclinic	monoclinic
Space group	<i>Pbcn</i>	<i>P-1</i>	<i>P-1</i>	<i>P2₁/c</i>
Crystal descr.	colourless blade	yellow block	orange block	yellow plate
a (Å)	16.1278(8)	9.8196(2)	9.931(7)	16.2788(2)
b (Å)	12.2682(5)	14.155(3)	14.018(1)	9.1026(9)
c (Å)	12.5817(6)	14.978(3)	14.536(1)	19.875(2)
α (°)	90	73.264(8)	102.99(2)	90
β (°)	90	71.924(7)	104.958(2)	96.086(3)
γ (°)	90	78.353(7)	98.90(2)	90
Volume (Å ³)	2489.4(2)	1880.8(6)	1856.0(2)	2928.5(5)
Z	6	2	2	4
Abs. coeff. (m.mm ⁻¹)	0.280	0.600	2.438	0.708
F(000)	953.5	760.0	900.0	1440.0
Independent refl.	2565	7775	9240	6046
Completeness (%)	99.9	99.2	99.6	99.6
Data/Restr/Para	2565/0/138	7775/0/443	9240/0/433	6046/0/381
Goodness of fit on F ²	1.034	0.963	1.037	1.045
Final R ₁ indexes	0.0376	0.0318	0.0264	0.0307
wR ₂ indices (all data)	0.1331	0.1153	0.0524	0.0666
Largest diff. peak and hole (e.Å ⁻³)	0.31/-0.32	1.25/-0.97	0.47/-0.63	1.83/-1.02

Table S2

Complex	2	4	5	8
Emp. formula	C ₃₈ H ₃₈ N ₂ F ₆ P ₂ Cl ₂ Ru	C ₃₈ H ₃₈ N ₂ F ₆ P ₂ Ru	C ₃₄ H ₃₆ N ₂ F ₆ P ₂ Ru	C ₁₅₀ H ₁₄₈ N ₈ F ₂₄ P ₄ As ₄ Cl ₄ Ru
Form. wt. g.mol ⁻¹)	870.61	799.71	749.66	3488.51
Crystal system	monoclinic	triclinic	monoclinic	monoclinic
Space group	<i>P2₁/n</i>	<i>P-1</i>	<i>C2/c</i>	<i>P2₁</i>
Crystal descr.	yellow block	yellow needle	yellow block	yellow block
a (Å)	13.8600(6)	11.0881(7)	18.0319(9)	9.910(2)
b (Å)	13.9831(7)	13.1574(8)	13.2824(6)	24.910(5)
c (Å)	18.9389(9)	14.6115(8)	29.0663(1)	15.236(3)
α (°)	90	113.867(2)	90	90
β (°)	93.545(2)	95.538(2)	99.842(2)	108.51(3)
γ (°)	90	111.568(2)	90	90
Volume (Å ³)	3663.4(3)	1735.89(2)	6859.1(6)	3566.6(1)
Z	4	2	8	1
Abs. coeff. (m.mm ⁻¹)	0.723	0.607	0.609	1.542
F(000)	1768.0	816.0	3056.0	1753.0
Independent refl.	7590	7167	7046	14652
Completeness (%)	99.7	99.9	99.8	99.7
Data/Restr/Para	7590/0/461	7167/0/443	7046/0/408	14652/1/905
Goodness of fit on F ²	1.535	1.096	1.132	0.843
Final R ₁ indexes	0.0532	0.0367	0.0351	0.0289
wR ₂ indices (all data)	0.1895	0.1288	0.0842	0.0637
Largest diff. peak and hole (e.Å ⁻³)	1.98/-1.82	1.08/-0.87	0.73/-0.80	0.63/-0.54

Table S3

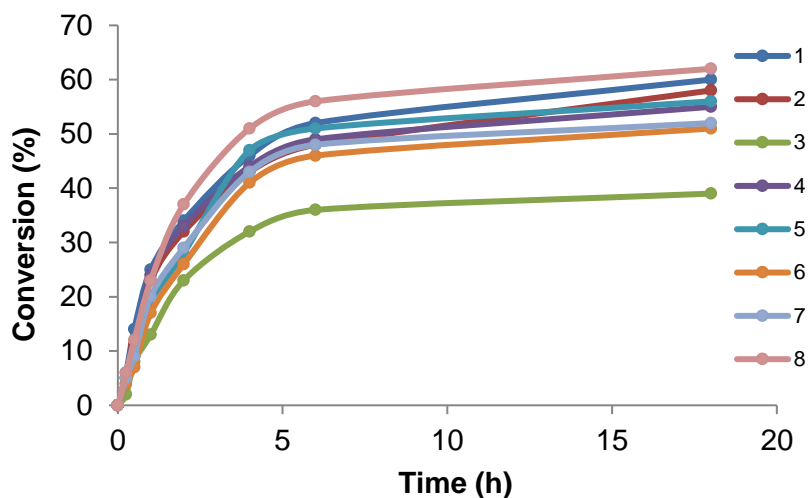
Complex	9
Emp. formula	C _{15.5} H ₁₆ NP _{0.5} 1.5Ru _{0.5}
Form. weight (g.mol ⁻¹)	472.66
Crystal system	triclinic
Space group	<i>P</i> -1
Crystal descr.	yellow block
a (Å)	9.4358(2)
b (Å)	9.7889(2)
c (Å)	18.948(4)
α (°)	91.365(6)
β (°)	94.352(6)
γ (°)	114.451(5)
Volume (Å ³)	1585.6(5)
Z	4
Abs. coeff. (m.mm ⁻¹)	3.488
F(000)	900.0
Independent refl.	6706
Completeness (%)	98.4
Data/Restr/Para	6706/0/348
Goodness of fit on F ²	1.047
Final R ₁ indexes	0.0218
wR ₂ indices (all data)	0.0467
Largest diff. peak and hole (e.Å ⁻³)	1.31/-1.27

5. **Table S4:** Selected bond lengths and angles for [HL5]Cl, P2,P3,1,2,4,5,8,9

Description	[HL5]Cl	P2	P3	1	2	4	5	8	9
Ru1-C2	-	-	-	2.033(2)	2.042(4)	2.040(3)	2.038(3)	2.035(4)	2.033(3)
Ru1-Cg ^a	-	1.844(3)	1.813(4)	1.899(3)	1.894(3)	1.892(4)	1.889(6)	1.874(6)	1.907(7)
Ru-E1 ^b	-	2.3222(7)	2.4229(2)	2.3262(6)	2.2993(1)	2.3135(7)	2.3168(6)	2.4190(7)	2.3155(8)
Ru1-C2-N1	-	-	-	120.23(2)	119.90(3)	119.95(2)	119.60(2)	119.7(3)	120.81(2)
C2-Ru1-E1 ^b	-	-	-	87.32(7)	88.98(1)	86.84(7)	87.19(7)	85.92(1)	87.51(7)
E1-Ru1-Ca ^{b,c}	-	-	-	93.99(7)	95.33(1)	95.34(7)	96.08(8)	94.50(9)	93.14(7)
C2-Ru1-Ca ^c	-	-	-	89.65(1)	89.58(1)	89.71(1)	89.66(1)	89.38(2)	89.38(1)
C2-N1-C4	125.22(1)			130.7(2)	118.0(3)	129.6(2)	118.4(2)	118.8(3)	117.6(2)
C2-N2-C8	126.14(2)	-	-	126.30(2)	127.00(3)	125.8(2)	126.4(2)	125.5(3)	126.1(2)
C2-N1-C4-C5	104.65(2)	-	-	28.3(3)	26.6(5)	24.7(3)	27.5(3)	-22.6(7)	-26.8(3)
C2-N2-C8-C9	-68.92(2)	-	-	-	-91.4(5)	-104.4(3)	-99.9(3)	100.9(7)	-

^a Cg = centroid of arene/cyclopentadienyl moiety. ^b E = P (1-7), As (8). ^c Average position between two carbon atoms belonging to the alkene moiety.

6. Time-resolved conversion profiles in the transfer hydrogenation-epoxidation catalysis



General conditions: 4'-bromophenylphenacylbromide (BPAB, 0.6 mmol), iPrOH (4 mL), KO^tBu (1.2 eq.), [Ru] (2 mol%), 110 °C. Determined by ¹H-NMR, based on the average of at least two runs.

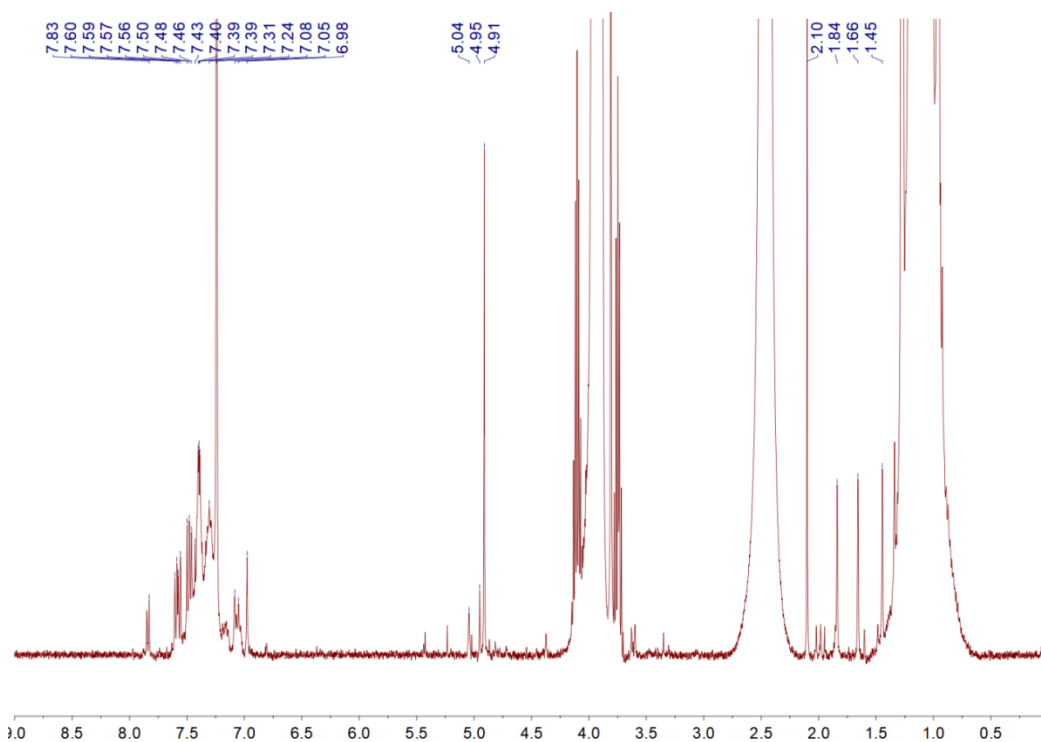
7. Optimization of transfer hydrogenation-epoxidation conditions (Table S5)

Table S5: Optimization of transfer hydrogenation-epoxidation conditions.

Entry	Complex	Temp (°C)	Base	Conversion ^a (%)			Selectivity ^b (%)
				2h	6h	18h	
1	-	110	-	0	1	2	0:0:100
2	-	110	KO ^t Bu	0	2	3	0:0:100
3	1	110	KOH	19	31	38	49:51:0
4	1^c	110	KOH	20	36	41	90:3:7
5	1 (3 mol%)	110	KO ^t Bu	37	55	63	29:71:0
6	1 (1 mol%)	110	KO ^t Bu	21	37	43	47:53:0
7	1	110	KOH	26	44	49	43:57:0
8	1	25	KOH	7	8	9	89:11:0

General conditions: 4'-bromophenylphenacylbromide (BPAB, 0.6 mmol), iPrOH (4 mL), base (1.2 eq.), [Ru] (2 mol%), 110 °C. ^a Determined by ¹H-NMR, based on the average of at least two runs. ^b Selectivity (A:E:O) = alcohol:epoxide:other, after 18 hours. ^c Two equivalents of base used.

8. $^1\text{H-NMR}$ spectrum of catalysis reaction mixture



General conditions: 4'-bromophenylphenacylbromide (BPAB, 0.6 mmol), $i\text{PrOH}$ (4 mL), KO^tBu (1.2 eq.), [1] (2 mol%), anisole (0.6 mmol), 110 $^\circ\text{C}$. Aliquot taken after 6 hours reaction time analysed using CDCl_3 .