

*Supplementary Information for*

**Hierarchical urchin-like Co<sub>9</sub>S<sub>8</sub>@Ni(OH)<sub>2</sub> heterostructures  
with superior electrochemical performance for hybrid  
supercapacitor**

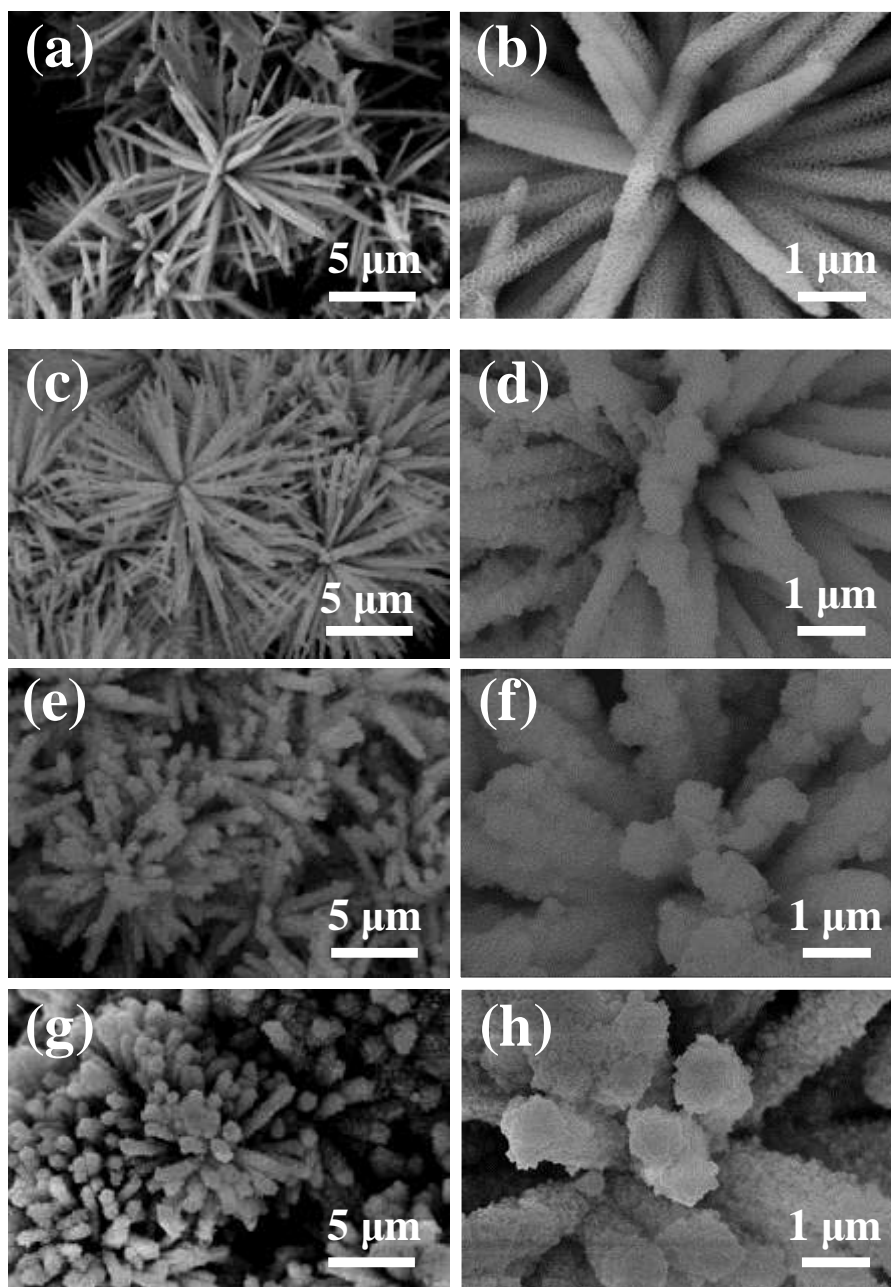
Qing Xu, Ping Xia, Yuyan Xu, Deli Jiang, and Min Chen\*

*School of Chemistry and Chemical Engineering, Jiangsu University, Zhenjiang*

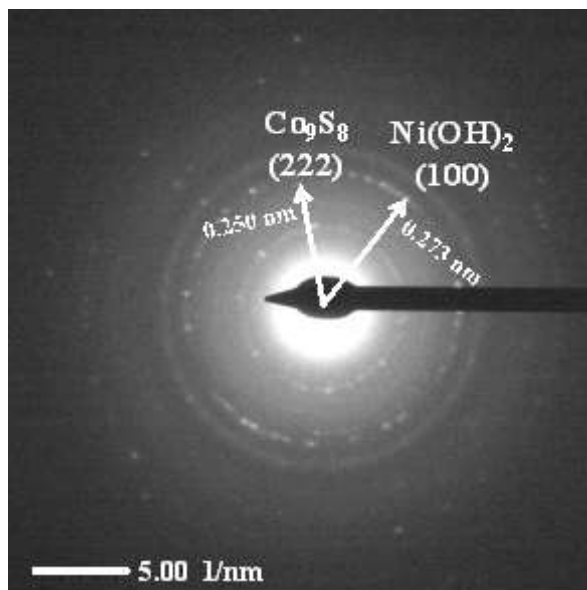
*212013, China*

Corresponding author: Min Chen

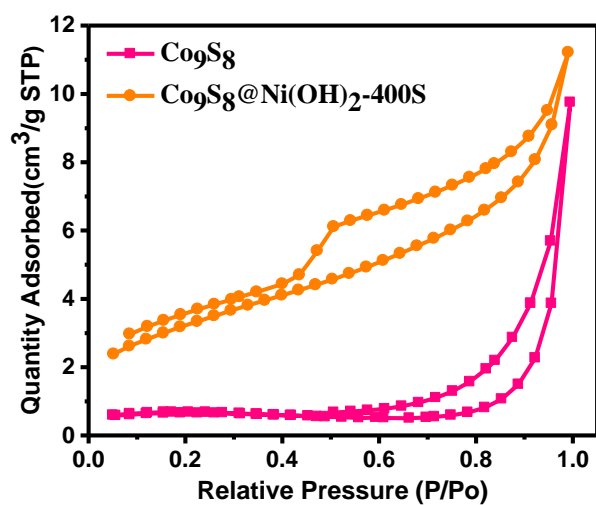
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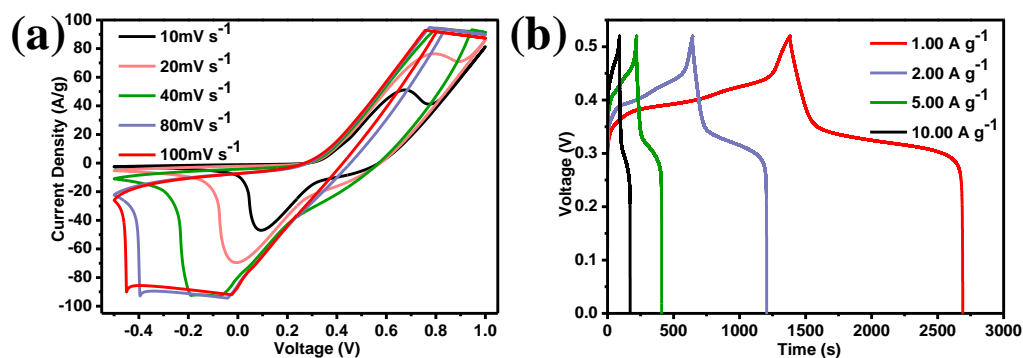
**Fig. S1** SEM images of (a,b)  $\text{Co}_9\text{S}_8@\text{Ni}(\text{OH})_2\text{-100S}$ , (c,d)  $\text{Co}_9\text{S}_8@\text{Ni}(\text{OH})_2\text{-200S}$ , (e,f)  $\text{Co}_9\text{S}_8@\text{Ni}(\text{OH})_2\text{-600S}$ , and (g,h)  $\text{Co}_9\text{S}_8@\text{Ni}(\text{OH})_2\text{-800S}$ .



**Fig. S2** Selected area electron diffraction pattern (SAED) of the  $\text{Co}_9\text{S}_8@\text{Ni}(\text{OH})_2\text{-400S}$  sample.



**Fig. S3** Nitrogen absorption-desorption isotherms of  $\text{Co}_9\text{S}_8$  and  $\text{Co}_9\text{S}_8@\text{Ni}(\text{OH})_2\text{-400S}$ .



**Fig. S4** (a) CV curves of the  $\text{Co}_9\text{S}_8\text{-Ni(OH)}_2\text{-400S}$  at various scan rates; (b) charge and discharge curves of the  $\text{Co}_9\text{S}_8\text{-Ni(OH)}_2\text{-400S}$  at different current densities.

**Table S1** Comparison studies for metal sulfide, hydroxide and their SC performances.

Electrocatalysts	Electrolyte (KOH)	Mass loading ( $\text{mg cm}^{-2}$ )	Power density ( $\text{W kg}^{-1}$ )	Energy density ( $\text{W h kg}^{-1}$ )	Reference
$\text{Co}_9\text{S}_8$	3M	1.9	800	15	<i>New J. Chem.</i> 2017, 41, 1142-1148
$\text{NiCo}_2\text{S}_4/\text{Co}_9\text{S}_8$	3M	3.5	800	42	<i>Appl. Surf. Sci.</i> 2018, 434, 861-870
$\text{Co}_9\text{S}_8/\alpha\text{-MnS}@N\text{-C}@MoS_2$	2M	/	729	64	<i>Small</i> , 2018, 1800291.
$\text{Co}_9\text{S}_8\text{-NSA/NF}$	1M	2.8	828	20	<i>Nanoscale</i> , 2018, 10, 2735
$\text{CoMoO}_4/\text{Co}_9\text{S}_8$	3M	4.5	800	38	<i>Electrochim. Acta</i> , 2017, 252, 470-481
$\text{Co}_9\text{S}_8@\text{Ni(OH)}_2/\text{CF}$	6M	2	253	31	<i>J. Mater. Chem. A</i> , 2017, 5, 22782
$\text{Ni}_3\text{S}_2@\text{CoS}$	2M	13	800	16	<i>J. Name.</i> 2013, 00, 1-3
$\text{NiCo}_2\text{S}_4@\text{Ni(OH)}_2$	2M	2.8	6420	32	<i>Electrochim. Acta</i> , 2016, 193, 116-127
$\text{Co}_3\text{O}_4@\text{Ni(OH)}_2$	3M	4.5	347	40	<i>Chem. Eng. J.</i> 2017, 315, 35-45
$\text{NiCo}_2\text{S}_4@\text{Ni(OH)}_2@\text{PPy}$	2M	3.2	800	16	<i>J. Mater. Chem. A</i> , 2018, 6, 2482
$\text{Ni(OH)}_2/3D\text{-Ni}$	1M	/	500	40	<i>Nano Energy</i> 2017, 39, 639-646
$\text{Co}_9\text{S}_8\text{-Ni(OH)}_2/\text{NF}$	2M	4.5	800	48	<b>This work</b>