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## **Supporting Information**

#### Cholesterol linked benzothiazole: A versatile gelator for detection of picric acid

#### and metal ions such as Ag<sup>+</sup>, Hg<sup>2+</sup>, Fe<sup>3+</sup> and Al<sup>3+</sup> under different conditions

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Table	S1·	Results	of ge	lation	test	for	1
1 ant	<b>DI</b> .	Results	UI SU	iation	iest	101	

Solvent	1			
CHCl <sub>3</sub>	S			
DCM	S			
Nitrobenzene	G (mgc = 18 mg/mL,			
	$T_{gel} = 44 \text{ °C}$			
Benzene	G (mgc = 56 mg/mL, $T_{gel} = 40 \text{ °C}$ )			
1,4-Dioxane	$G (mgc = 25 mg/mL, T_{gel} = 52 \text{ °C})$			
Toluene	G (mgc = 45 mg/mL, $T_{gel} = 42 \text{ °C}$ )			
МеОН	Ι			
CH <sub>3</sub> CN	Ι			
$CHCl_3 - MeOH (1:1, v/v)$	Ι			
DMSO	Ι			
CHCl <sub>3</sub> -PET ether $(1:1,v/v)$	Ι			
Cyclohexane	PS			
n-Hexane	Ι			
Diethyl ether I				
Acetone	PS			
THF S				
$DMF \qquad \qquad G (mgc = 10 mg/2)$				
DMF : H <sub>2</sub> O (1:1, v/v)	P			
MeOH : H <sub>2</sub> O (1:1, v/v) I				
CH <sub>3</sub> CN : H <sub>2</sub> O (1:1, v/v) I				
S = solution; G = gel; I = insoluble; PS = partially soluble; PG = partial gelation. Gelation studies were carried out by taking 60 mg of compound 1 in 1 mL of each solvent. Gels were primarily characterized by inversion of vial method after ~5 min of sample preparation.				



Fig S1. Pictorial representation of gel of 1 in (a) nitrobenzene, (b) DMF, (c) toluene and (d) benzene.



**Fig S2.** Photograph representing the phase change of the nitrobenzene gel of 1 (20 mg/mL) upon successive addition of 1 equiv. amount of picric (PA), trifluoroacetic acid (TFA) and Et<sub>3</sub>N.



**Fig S3**. Change in (a) emission and (b) fluorescence ratio  $[I-I_0/I_0]$  of **1** ( $c = 2.50 \times 10^{-5} \text{ M}$ ) upon addition of 12 equiv. amounts of different guests ( $c = 1.0 \times 10^{-3} \text{ M}$ ) in CH<sub>3</sub>CN containing 1% CHCl<sub>3</sub>.



**Fig S4**. Partial <sup>1</sup>H NMR (400 MHz) of **1** ( $c = 3.07 \times 10^{-3} \text{ M}$ ) in absence and presence of 1 equiv. amount of PA.



Fig S5. (a) Benesai-Hildebrand plot and (b) detection limit of 1 for with picric acid ( $c = 1 \ge 10^{-3} \text{ M}$ ) in CH<sub>3</sub>CN containing 1% CHCl<sub>3</sub>.



**Fig S6**. Partial <sup>1</sup>H NMR (400 MHz) of **1** ( $c = 3.07 \times 10^{-3} \text{ M}$ ) in absence and presence of 1 equiv. amount of (a) Ag<sup>+</sup> and (b) Hg<sup>2+</sup> in CDCl<sub>3</sub>.



**Fig S7.** Change in UV-vis spectra of **1** with addition of 3 equiv. amounts of (a) all the metal ions, (b)  $Fe^{2+}$ , (c)  $Cu^{2+}$ , (d)  $Ag^+$  (e)  $Co^{2+}$ , (f)  $Ni^{2+}$ , (g)  $Zn^{2+}$ , (h)  $Cd^{2+}$  and (i)  $Cu^+$  ( $c = 1 \times 10^{-3} \text{ M}$ ) in CH<sub>3</sub>CN containing 1% CHCl<sub>3</sub>.



**Fig S8.** Change in emission of 1 with addition of 3 equiv. amounts of (a) all the metal ions, (b)  $Hg^{2+}$ , (c)  $Cu^{2+}$  and (d)  $Fe^{2+}$  ( $c = 1 \times 10^{-3} \text{ M}$ ) in CH<sub>3</sub>CN containing 1% CHCl<sub>3</sub>.



**Fig S9**. Change in fluorescence ratios of 1 ( $c = 2.5 \times 10^{-5} \text{ M}$ ) with (a) Fe<sup>3+</sup> and (b) Al<sup>3+</sup> ( $c = 1 \times 10^{-3} \text{ M}$ ) in the absence and presence of 3 equiv. amounts of different metal ions in CH<sub>3</sub>CN containing 1% CHCl<sub>3</sub>.



**Fig S10.** Benesai-Hildebrand plots of 1 with (a)  $Fe^{3+}$  and (b)  $Al^{3+}$ ; Detection limit plots of 1 for (c)  $Fe^{3+}$  and (d)  $Al^{3+}$  ( $c = 1 \times 10^{-3}$  M) in CH<sub>3</sub>CN containing 1% CHCl<sub>3</sub>.

Entry	Structure of compounds	Medium of Gelation	Sensing	Ref.	
1	$(1)^{4}$	Cyclohaxane and DCM/Haxane (1:4)	Detection of picric acid vapor through gel to sol transition	Langmuir 2011, <b>27</b> , 15275.	
2	N-M N-M N-M N-M N-M O O O O O O O O O O O O O O O O O O O	Toluene/DCM (8:2), Benzene/DCM (8:2) and o-Xylene/DCM (8:2)	Detection of picric acid through gel to sol transition	ACS Appl. Mater. Interfaces 2013, <b>5</b> , 672–679	
3	$H_{3}C(H_{2}C)_{15} O H H H H H H H H H H H H H H H H H H $	THF	Detection of picric acid through gel to sol transition	Appl. Mater. Interfaces 2013, <b>5</b> , 8394	
4	$H_2N$ $H_2N$ + CuCl <sub>2</sub> $H_2N$	Water	Detection of picric acid through gel to sol transition	ACS Appl. Mater. Interfaces 2014, <b>6</b> , 6308	
5	$ \begin{array}{c} \begin{array}{c} \begin{array}{c} \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} \\ \end{array} \\ \end{array} \\ \begin{array}{c} \end{array} \\ \end{array} $	Nitrobenzene, Toluene, benzene and 1,2-dichlorobenzene	Detection of picric acid through gel to sol transition	<i>ChemistrySelect</i> 2017, <b>2</b> , 4800.	
6	$H_{N}^{N} = N_{N}^{N} = N_{N$	DMSO	Detection of picric acid through gel to sol transition	New J. Chem. 2018, <b>42</b> , 5382-5394	
Our Work		Nitrobenzene	Detection of picric acid through gel to sol transition		

## Table S2: Reported PA sensors in gel phase

Entry	Structure	Solvent	Phase transformation in presence of Hg <sup>2+</sup> ions	Interfering metal ions	Detection limit for Hg <sup>2+</sup> (M)	Ref.
1	$\left(\begin{array}{c} 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ $	1,2- dichloroethan e	Gel to Sol	-	-	<i>Org. Lett.</i> , 2011, <b>13</b> , 3372.
2	Act Content of the second seco	1,2- dichloroethan e	Gel to Sol	-	-	New J. Chem., 2013, <b>37</b> , 2419.
3	$ \begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & $	CHCl <sub>3</sub> :CH <sub>3</sub> O H (2:1, v/v)	Gel to Sol	Cu <sup>2+</sup> , Ag <sup>+</sup>	-	New. J. Chem., 2016, <b>40</b> , 3476.
4		0.2 N HCl	Sol to gel	-	-	<i>Chem. Commun.</i> , 2014, <b>50</b> , 734.
5	$\operatorname{cond}_N^H \operatorname{cond}_N^H$	DMSO : $H_2O$ = (1 : 6, v/v)	Gel to Sol	Cu <sup>2+</sup> , Ag <sup>+</sup>	2.02 x 10 <sup>-6</sup>	<i>Mater. Chem.</i> <i>Front.</i> , 2018, <b>2</b> , 385.
6		$DMF : H_2O =$ (1 : 1, v/v)	Gel to Sol	Cu <sup>2+</sup>	2.61 x 10 <sup>-6</sup>	New. J. Chem., 2018, <b>42</b> , 13718- 13725
7	$\left( \begin{array}{c} \left( \begin{array}{c} 1 \\ 1 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right) \right) = \left( \begin{array}{c} \left( \begin{array}{c} 1 \\ 1 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\$	DMF : H <sub>2</sub> O = (1 : 1, v/v)	Sol to gel Chemodosime tric approach	-	5.71 x 10 <sup>-6</sup>	Supramol. Chem., 2018, <b>30</b> , 722.
8	$R = -\frac{1}{2} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ N \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \end{array} \right)^{N} \left( \begin{array}{c} N \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\ 0 \\$	DMF : H <sub>2</sub> O = (1 : 1, v/v)	Sol to gel Chemodosime tric approach	-	5.51 x 10 <sup>-6</sup>	New J. Chem., 2019, DOI: 10.1039/c8nj0505 6b
		Nitrobenzene	Gel to Sol	-	-	Present work

**Table S3:** List of different Hg<sup>2+</sup> ion responsive supramolecular gelators

Entry	Structure	Solvent	Phase transformatio n in presence of Ag <sup>+</sup> ions	Interfering metal ions	Detection limit for Ag <sup>+</sup> (M)	Ref.
1	$\begin{array}{c} C_{18}H_{17} \\ & & \\ & $	EtOH	Gel to sol	-	-	<i>Tetrahedron</i> <i>Lett.</i> 2012, <b>53</b> , 1840.
2		МеОН	Sol to gel	-	-	<i>Chem.</i> <i>Commun.</i> 2013, <b>49</b> , 4181.
3	HO C17H35	MeOH:H <sub>2</sub> O (1:1, v/v)	Sol to gel	-	-	<i>Supramol.</i> <i>Chem.</i> 2014, <b>26</b> , 39.
4		THF/ H <sub>2</sub> O	Sol to gel	-	-	<i>Soft Matter</i> , 2011, <b>7</b> , 2412.
5		H <sub>2</sub> O	Sol to gel	-	-	<i>Soft Matter</i> , 2012, <b>8</b> , 6557.
6		DMF: H <sub>2</sub> O ( 2:3, v/v)	Sol to gel	-	-	<i>Cryst.</i> <i>Growth Des.</i> 2015, <b>15</b> , 4635.
7	$\begin{array}{c} & & & & \\ C_{12}H_{25}HN & & & C \\ & & & & C \\ & & & & C \\ & & & &$	CH <sub>2</sub> Cl <sub>2</sub> , CHCl <sub>3</sub> , THF	Sol to gel	-	-	<i>Langmuir</i> , 2012, <b>28</b> , 27.
8	$R = \underbrace{\begin{array}{c} HO \\ P \\ R \\ R$	Toluene: EtOH (99:1, v/v)	Sol to gel	-	-	Chem. Commun. 2015, <b>51</b> , 13929.
9	$R = \frac{1}{2} \frac{0}{100} \frac{0}{100} \frac{0}{100} \frac{1}{100} $	DMF, DMF/ H <sub>2</sub> O, DMSO/ H <sub>2</sub> O	Sol to gel	-	-	<i>Cryst.</i> <i>Growth Des.</i> 2015, <b>15</b> , 5360.
10	$R^{+} NH \qquad HN^{+} R$	EtOAc	Gel to sol	Li+	-	<i>Chem.</i> <i>Commun.</i> 2012, <b>48</b> , 2767.

## **Table S4:** List of different Ag<sup>+</sup> ion responsive supramolecular gelators

11	HO = HO = HO	DMF : H <sub>2</sub> O (1:1, v/v)	Sol to gel	-	4.31 x 10 <sup>-5</sup>	<i>ChemistrySe</i> <i>lect</i> , 2017, <b>2</b> , 959.
12	$\begin{array}{c} X \\ Y \\ Y \\ Z \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$	DMSO: H <sub>2</sub> O	Sol to gel	-	-	Dalton Trans., 2017, <b>46</b> , 2793.
13	$N \longrightarrow NH \longrightarrow OC_{16}H_{33}$	Toluene/eth anol (10:1, v/v)	Sol to gel	-	-	<i>Langmuir,</i> 2007, <b>23</b> , 8217.
14	$R = \frac{1}{2} \left( \begin{array}{c} H \\ H \end{array} \right) \left( \begin{array}{c} H \\ H \\ H \end{array} \right) \left( \begin{array}{c} H \\ H \\ H \end{array} \right) \left( \begin{array}{c} H \\ H \\ H \end{array} \right) \left( \begin{array}{c} H \\ H \\ H \\ H \end{array} \right) \left( \begin{array}{c} H \\ H \\ H \\ H \\ H \\ H \end{array} \right) \left( \begin{array}{c} H \\ H $	Diphenyl ether	Sol to gel	-	-	<i>Chem. Lett.</i> , 2003, <b>32</b> , 12.
15		THF-H <sub>2</sub> O (3 : 2)	Sol to gel	-	-	New J. Chem., 2010, <b>34</b> , 2261.
16		H <sub>2</sub> O	Sol to gel	-	-	New J. Chem., 2014, <b>38</b> , 2470.
17	$R = \frac{1}{2} O \left( \begin{array}{c} & & & \\ & & & $	CHCl <sub>3</sub> :CH <sub>3</sub> OH (2:1, v/v)	Gel to Sol	Cu <sup>2+</sup> , Hg <sup>2+</sup>	-	New. J. Chem., 2016, <b>40</b> , 3476.
18		DMSO: H <sub>2</sub> O (1:1, v/v)	Sol to gel	Cu <sup>2+</sup>	-	New. J. Chem., 2018, <b>42</b> , 6488.
19		DMSO: H <sub>2</sub> O	Gel to sol	Cu <sup>2+</sup>	3.69 x 10 <sup>-6</sup>	
	$ \qquad \qquad$	DMSO: H <sub>2</sub> O	Gel to sol	Cu <sup>2+</sup> , Hg <sup>2+</sup>	3.34 x 10 <sup>-6</sup>	Mater. Chem. Front., 2018,
		DMSO: H <sub>2</sub> O	Sol to gel	-	1.93 x 10 <sup>-7</sup>	2, 385.
		DMSO: H <sub>2</sub> O	Sol to gel	-	1.28 x 10 <sup>-6</sup>	

20	R H	1,4-	Gel to sol	-	3.27 x 10 <sup>-5</sup>	
	$R = -\frac{5}{5} \left( \begin{array}{c} 0 \\ H \\ H \\ H \end{array} \right) \left( \begin{array}{c} 0 \\ H \\ H \\ H \\ H \\ H \end{array} \right) \left( \begin{array}{c} 0 \\ H \\$	dioxane- MeOH (1:1, v/v) 1,4- dioxane- H <sub>2</sub> O (1:1, v/v)	Sol to gel	-	9.27 x 10 <sup>-5</sup>	New J. Chem., 2019, DOI: 10.1039/c8nj 05056b
		Nitrobenze ne	Gel to Sol	-		Present work



## <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz) of 1



# Mass spectrum of 1.







## <sup>13</sup>C NMR (CDCl<sub>3</sub> containing 8% *d*<sub>6</sub>-DMSO, 100 MHz) of 3



## Mass spectrum of 3

